Antioxidant and Angiotensin-Converting Enzyme Inhibitory Activity of *Eucalyptus camaldulensis* and *Litsea glaucescens* Infusions Fermented with Kombucha Consortium

**Claudia I. Gamboa-Gómez**, Rubén F. González-Laredo, José Alberto Gallegos-Infante, Mª del Mar Larrosa Pérez, Martha R. Moreno-Jiménez, Ana G. Flores-Rueda and Nuria E. Rocha-Guzmán*

1Durango Institute of Technology, Felipe Pescador 1830 Ote., MX-34080 Durango, Durango, Mexico
2European University of Madrid, Calle Tajo, s/n, Villaviciosa de Odón, ES-28670 Madrid, Spain

Received: January 26, 2016
Accepted: May 11, 2016

**Summary**

Physicochemical properties, consumer acceptance, antioxidant and angiotensin-converting enzyme (ACE) inhibitory activities of infusions and fermented beverages of *Eucalyptus camaldulensis* and *Litsea glaucescens* were compared. Among physicochemical parameters, only the pH of fermented beverages decreased compared with the unfermented infusions. No relevant changes were reported in consumer preference between infusions and fermented beverages. Phenolic profile measured by UPLC MS/MS analysis demonstrated significant concentration changes of these compounds in plant infusions and fermented beverages. Fermentation induced a decrease in the concentration required to stabilize 50 % of DPPH radical (*i.e.* lower IC₅₀). Additionally, it enhanced the antioxidant activity measured by the nitric oxide scavenging assay (14 % of *E. camaldulensis* and 49 % of *L. glaucescens*); whereas relevant improvements in the fermented beverage were not observed in the lipid oxidation assay compared with unfermented infusions. The same behaviour was observed in the inhibitory activity of ACE; however, both infusions and fermented beverages had lower IC₅₀ than positive control (captopril). The present study demonstrated that fermentation has an influence on the concentration of phenolics and their potential bioactivity. *E. camaldulensis* and *L. glaucescens* can be considered as natural sources of biocompounds with antihypertensive potential used either as infusions or fermented beverages.

**Key words:** herbal infusions, fermented beverages, kombucha, antioxidant activity, polyphenols

**Introduction**

In recent years, the consumption of herbal infusions around the world has increased due to their beneficial health effects. These beverages are prepared by placing a small amount of the selected plant material in freshly boiled water, allowing the preparation to steep for a short period of time (1). Although herbal infusions do not have any particular nutritional value, they represent an important source of bioactive compounds such as polyphenols. It has been shown that these compounds can act by diverse mechanisms providing significant protection against chronic diseases (2). For example, the consumption of some herbal polyphenols with antioxidant activity may regulate hypertension through inhibition of the angiotensin-converting enzyme (ACE), a key component in the renin-angiotensin aldosterone system which regulates blood pressure (3).

*Corresponding author: Phone: +52 (618) 818 5402; 818 6936 Ext. 112; E-mail: nrocha@itdurango.edu.mx
Recently, several research studies have focused on enhancing the beneficial health potential of herbal infusions. For example, Aloulou et al. (4) reported that fermentation process with kombucha enhanced the phenolic content, antioxidant activity and α-amylase inhibitory activity in five commonly consumed teas. Similar results were observed by Velicánskii et al. (5) in fermented lemon balm (Melissa officinalis L.) infusion; they reported an enhancement of phenolic compounds such as rosmarinic, caffeic and ferulic acids (1.3-, 1.9- and 4.6-fold higher, respectively), as well as major antioxidant activity against DPPH radicals than in unfermented infusions.

The beneficial effects of these fermented beverages are attributed to the presence of restructuring polyphenols, gluconic acid, glucuronic acid, lactic acid, vitamins, amino acids, antibiotics and a variety of micronutrients produced during fermentation (6).

The fermentation process involves the activity of yeasts that ferment glucose and fructose to ethanol, which is then oxidized to acetic acid by acetic acid bacteria (AAB). The main source of carbon in this process is sucrose. The sugar is hydrolyzed by the enzyme invertase from the yeast present in the kombucha consortium, producing ethanol via the metabolic pathway of glycolysis, with a preference for fructose as the substrate. Subsequently, AAB convert glucose and ethanol into gluconic and acetic acids (7).

In occidental countries, the consumption of tea prepared from Camellia sinensis and analogues, such as fermented beverages, is not that common; on the contrary, the consumption of infusions made with plant materials typical of each region predominates. In this same regard, in a previous study we analyzed phenolic composition for its bactericidal, antiseptic, expectorant and anti-inflammatory or antibiotic, among other purposes, effects. This is unknown and has also been reported (10).

The fermentation process was performed using kombucha culture from a Mexican trading house (Healthy, Natural Life, Tlaquepaque, Jalisco, Mexico). A previous study reported that kombucha consortium contained yeast strains such as: Saccharomyces cerevisiae, Saccharomyces bisporus, Saccharomyces ludwigii, Torulopsis sp., Zygosaccharomyces sp., Dekkera, Kazachstania and Pichia, among others. Bacteria such as Acetobacter, Gluconacetobacter, Lactobacillus, Lactococcus, Leuconostoc, Bifidobacterium, Thermus, Allobaculum, Ramino cocciaceae incertae sedis and Propionibacterium, among others, have also been reported (10).

Preparation of herbal infusions and fermented beverages

Both materials were dried at room temperature in shade, ground (1.13 mm) and stored in the dark for later determinations. Infusions (1%, by mass per volume) were prepared by adding 200 mL of freshly boiled water to 2 g of dried ground sample. Infusions were left to stand for 10 min and filtered.

Fermentation was performed using kombucha culture from a Mexican trading house (Healthy, Natural Life, Tlaquepaque, Jalisco, Mexico). A previous study reported that kombucha consortium contained yeast strains such as: Saccharomyces cerevisiae, Saccharomyces bisporus, Saccharomyces ludwigii, Torulopsis sp., Zygosaccharomyces sp., Dekkera, Kazachstania and Pichia, among others. Bacteria such as Acetobacter, Gluconacetobacter, Lactobacillus, Lactococcus, Leuconostoc, Bifidobacterium, Thermus, Allobaculum, Ramino cocciaceae incertae sedis and Propionibacterium, among others, have also been reported (10).

The fermentation of infusions was performed under the conditions reported by Vázquez-Cabr al et al. (7). The infusions were cooled at room temperature, and then sweetened by dissolving 10 g/L of sucrose. In small bioreactors, 1 L of sweetened infusions was inoculated and acidified with 2.5 g of inoculum (biofilm with a symbiotic culture of bacteria and yeasts) and 100 mL of black tea vinegar, respectively. The bioreactors were covered with cheesecloth, and the fermentation at (25±1) °C was monitored for seven days. On the last fermentation day, the biomass was removed from the beverage by filtration.

For preparation of black tea vinegar, inoculum was grown at 25 °C in sweetened (sucrose, 10 %) black tea for 20 days. Freshly cultured kombucha was used for further subcultures or for fresh fermentation batches.

Physicochemical evaluation

The pH was measured using a digital pH meter (pH meter session™; Hach, Loveland, CO, USA) calibrated at pH=4, 7 and 10 at 25 °C. Colour characteristics of infusions were evaluated by measuring their colour coordinates with a colourimeter (chroma meter CR-400/410; Konica Minolta, Osaka, Japan). Colour intensity measurements were expressed as L* (lightness), a* (red/green colour), and b* (yellow/blue colour) values.

Consumer preference tests

The consumer preference evaluation of both, infusions and fermented beverages was performed by 15 untrained panellists who declared drinking a cup of any herbal infusion at least once a week. The panellists ranked the samples according to the following scale: 1=very unpleasant to 9=very pleasant. The samples were coded with random three digit numbers. Black tea was used as a positive control.
Sugar content in fermented beverages

Sample analysis was carried out with an Acquity ultra-performance liquid chromatography (UPLC) system (Waters Corp., Milford, MA, USA) coupled with an evaporative light scattering detector (ELSD; Waters Corp., Wexford, Ireland). The LC system consisted of a sample manager (5 °C) and a binary solvent manager. The column used to determine fructose, glucose and sucrose was an Acquity UPLC BEH Amide, 100 mm×2.1 mm, 1.7 μm (Waters Corp., Wexford), operated at 40 °C. The elution profile included two solvents (both J.T.Baker, Mexico City, Mexico): acetonitrile/water (80:20) with 0.1 % NH₄OH (solvent A), and acetonitrile/water (30:70) with 0.1 % NH₄OH (solvent B) and an initial gradient of 5–60 % B in 5 min, isocratic by 1 min, reset and equilibrated for 5.5 min. The flow rate was 0.25 mL/min. The UPLC control and data processing were performed using the MassLinx software.

Ultra-performance liquid chromatography coupled with tandem mass spectrometry analysis

Sample analysis was carried out with an Acquity UPLC system (Waters Corp., Milford) coupled with a tandem Xevo TQ-S triple quadrupole mass spectrometer (Waters Corp., Wexford). The LC system consisted of a sample manager (20 °C) and a binary solvent manager. The column used to determine flavon-3-ols was an Acquity UPLC BEH C18, 50 mm×2.1 mm, 1.7 μm (Waters Corp., Wexford) operated at 40 °C. The elution profile included two solvents (both J.T.Baker): acidified water with 1 % formic acid (solvent A) and methanol LC-MS grade (solvent B); initial 98 % A, 0–2 min, 68 % A, 2–3.8 min, 55 % A, 3.8–4.5 min, 45 % A, 4.5–6.0 min, 5 % B (linear gradient) for column washing and subsequently 6.0–9.5 min, 98 % A for column stabilization. Multiple reaction monitoring (MRM) data were collected from 0 to 9.5 min. Negative ionization mode was used for MS assays. Electron spray ionization (ESI) conditions were as follows: capillary voltage 2.85 kV, desolvation temperature 500 °C, source temperature 150 °C, desolvation and cone gas 794 and 151 L/h, respectively, and collision gas 0.14 mL/min. For identification and quantification, a multiple reaction monitoring (MRM) was employed to check the stability of the ionization efficiency of mass spectrometer and a mixture of phenolic compound standards (20 ng/μL; Sigma Corp., St. Louis, MO, USA) for monitoring retention time and m/z values. The UPLC and tandem Xevo TQ-S triple quadrupole mass spectrometer control and data were processed using MassLinx v. 4.1 (Waters Corp., Milford) software.

Radical scavenging and chain-breaking assay

Antioxidant capacity of herbal infusions and fermented beverages was evaluated using the stable radical 2,2’-diphenyl-1-picrylhydrazyl (DPPH; Sigma Corp.) method described by Brand-Williams et al. (11). The percentage of scavenging or quenching radicals, Q, was calculated using the following equation:

\[ Q = \frac{A_0 - A}{A_0} \times 100 \]

where \( A_0 \) is the absorbance of the blank, and \( A \) is the absorbance of the mixture of radical and sample at different concentrations. For the calculation of the half maximal inhibitory concentration (IC₅₀ in μg/μL), the logarithm of the concentration against the inhibition percentage was plotted.

For chain-breaking kinetics, 3 mL of 3·10⁻⁵ mol/L of DPPH in methanol solution was used. The reaction began with the addition of 10 μL of herbal infusion or fermented beverage at previously determined IC₅₀ (μg of total content of analyzed flavonoids per μL). The DPPH bleaching was monitored spectrophotometrically at 515 nm with a spectrophotometer (model 50-BIO; Varian Cary®, Melbourne, Australia) at 25 °C for at least 60 min (12). The following equation was used to obtain the reaction rate, \( k \):

\[ \frac{1}{A} = \frac{1}{A_0} - \frac{k t}{A_0} \]

where \( A \) is the absorbance at increasing time (t), and \( A_0 \) is the initial absorbance. Results on dry mass basis were expressed as both IC₅₀ and chain-breaking activity (k/mg). Catechin was used as positive control.

Lipid peroxidation assay

Lipid peroxidation was determined according to Jozwik et al. (13) with several modifications. Inhibition of the formation of thiobarbituric acid reactive substances (TBARS), a lipid oxidation product, was quantified with a spectrophotometer (model 50-BIO; Varian Cary®), using healthy human plasma donated by the blood bank of the Public General Hospital (City of Durango, Durango, Mexico). Inhibition of plasma oxidation was evaluated using Fenton’s reagent with the addition of phosphate buffer (150 μL of 500 mM H₂O₂ and 150 μL of 100 mM FeCl₃ for 3 h at 37 °C; Caledon Laboratories Ltd, Georgetown, ON, Canada). Catalyzed plasma (600 μL) oxidation was evaluated for each herbal infusion and fermented beverage samples (100 μL) at several concentrations using phosphate buffer (pH=7.4) to reach the final volume (1 mL). Prior to oxidation phase, a thiobarbituric acid (TBA) reagent (MP Biomedicals, LLC, Illkirch, France) was prepared consisting of 40.5 mL of 20 % acetic acid buffered at pH=3.5 with 1 M NaOH, 13.2 mL of 8.2 % SDS (Sigma Corp.), 40.5 mL of 0.8 % TBA, and made up to 100 mL with double distilled water. Samples (1 mL) were mixed with 4.0 mL of TBA reagent, incubated at 96 °C for 80 min, and cooled down the samples on ice. Then, they were mixed with n-butanol (5 mL) and centrifuged at 3000×g for 15 min. The ability of the samples to inhibit plasma oxidation was determined by the absorbance decrease at 532 nm. For the calculation of the IC₅₀ (μg/μL), the logarithm of the concentration against the inhibition percentage was plotted. The calculation of the IC₅₀ (μg/μL) was included as positive control.

Nitric oxide scavenging assay

Infusions and fermented beverages were screened for antioxidant activity potential using the nitric oxide scavenging assay described by Balakrishnan et al. (14). The absorbance of the chromophores formed during the diazotization of nitrite with sulphanilamide, subsequently coupled with naphthylethylenediamine dichloride (Griess reagent, Sigma Corp.) was read at 546 nm (spectrophotome-
ter model 50-BIO; Varian Cary®). Catechin was used as a positive control and results were expressed as IC₅₀.

**Angiotensin-converting enzyme inhibitory activity**

The angiotensin-converting enzyme (ACE) inhibitory activity of infusions and fermented beverages was screened using the methodology described by Actis-Goretta et al. (15). The activity of ACE from rabbit lung (0.05 U/mg of protein; Sigma Corp.) was evaluated following the hydrolysis of hippuryl-l-histidyl-l-leucine to hippuric acid. The formed hippuric acid was separated and quantified with Waters HPLC system consisting of Waters 717 plus autosampler and Waters 600 pump (Waters Corp., Milford). Chromatographic separations were performed with a Supelcosil LC-18-DB column, 150 mm×4.6 mm, 5 μm i.d. (Waters Corp., Wexford), and a mobile phase composed of 0.1 % (by volume) trifluoroacetic acid in a volume ratio of H₂O and acetonitrile of 75:25 (J.T.Baker). The flow rate used to achieve a retention time of 2.9 min was 1 mL/min. The hippuric acid detection was carried out at 228 nm (Waters 2487 UV spectrophotometric detector; Waters Corp., Milford). Commercial hippuric acid was used as the standard (Sigma Corp.). For the inhibition of ACE activity by infusions and fermented beverages, a mixture of 10 μL of ACE (0.05 U/mg of protein) in 50 mM HCl-Tris, 300 mM NaCl (pH=8.3), and different concentrations of samples to reach the volume of 90 μL was pre-incubated for 5 min at 37 °C. A volume of 10 μL of 3 mM hippuryl-l-histidyl-l-leucine (1 mM final concentration; Sigma Corp.) was added to the above mixture and incubated for 30 min at 37 °C. The reaction was stopped by placing the sample in a water bath at 100 °C for 5 min. For the calculation of the IC₅₀ (μg/mL) the logarithm of the concentration against the inhibition activity percentage was plotted. Captopril (Sigma Corp.) was used as a positive control.

**Statistical analysis**

Data were expressed as mean value±standard error (SE). Statistical significance was determined by one-way variance analysis (ANOVA) (p<0.05) followed by the Tukey’s test, where p<0.05 was considered significant. Statistical analysis was made using JMP v. 5.0.1 software (SAS Institute, Cary, NC, USA).

**Table 1. Physicochemical evaluation and consumer acceptance of herbal infusions and fermented beverages of Litsea glaucescens and Eucalyptus camaldulensis**

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Litsea glaucescens</th>
<th></th>
<th>Eucalyptus camaldulensis</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>Infusion</td>
<td>Fermented beverage</td>
<td>Infusion</td>
<td>Fermented beverage</td>
</tr>
<tr>
<td>pH</td>
<td>(5.10±0.05)</td>
<td>(5.00±0.05)</td>
<td>(3.00±0.05)</td>
<td>(3.00±0.05)</td>
</tr>
<tr>
<td>L*</td>
<td>(22.00±0.05)</td>
<td>(22.10±0.01)</td>
<td>(22.00±0.05)</td>
<td>(22.10±0.02)</td>
</tr>
<tr>
<td>a*</td>
<td>(-0.13±0.01)</td>
<td>(-0.14±0.02)</td>
<td>(-0.14±0.05)</td>
<td>(-0.14±0.01)</td>
</tr>
<tr>
<td>b*</td>
<td>(0.30±0.01)</td>
<td>(0.30±0.01)</td>
<td>(0.33±0.02)</td>
<td>(0.33±0.02)</td>
</tr>
<tr>
<td>(% consumed sucrose)</td>
<td>N.D.</td>
<td>(70.40±0.06)</td>
<td>N.D.</td>
<td>(75.60±0.3)</td>
</tr>
<tr>
<td>(mg/L) fructose</td>
<td>(37.50±0.02)</td>
<td>N.D.</td>
<td>(37.20±0.02)</td>
<td>N.D.</td>
</tr>
<tr>
<td>(mg/L) glucose</td>
<td>(43.9±0.2)</td>
<td>N.D.</td>
<td>(40.1±0.3)</td>
<td>N.D.</td>
</tr>
<tr>
<td>Consumer preference</td>
<td>(4.0±0.4)</td>
<td>(4.0±0.4)</td>
<td>(4.0±0.6)</td>
<td>(3.0±0.2)</td>
</tr>
</tbody>
</table>

Values are means of duplicate determinations±standard error. Different letters in each row indicate statistical difference (p≤0.05) by Tukey’s test. N.D.=not determined

**Results and Discussion**

Results of physicochemical characterization are shown in Table 1. It was observed that the pH of fermented beverages decreased by 2 units compared with the unfermented infusions. The overall decrease in pH is attributable to the increased concentration of organic acids produced during fermentation by bacteria and yeasts from the tea fungus. In concordance, Velickanski et al. (16) reported decreased values of pH (between 2 and 3 units) in analogues of kombucha at the same fermentation times.

On the other hand, several authors reported changes in colour as a consequence of fermentation. For example, a progressive lightening of colour in fermented teas has been reported (17,18). However, in this work we did not observe significant differences between infusions and fermented beverages (Table 1). This could be attributed to the nature of plant materials and their chemical composition. Contrary to the species studied in this work, epigallocatechin-3-gallate (EGCG) is the main polyphenol found in green tea (about 59 % of total catechins) (19). In fermentation of kombucha, EGCG is the major precursor for other compounds such as theaflavins (20,21). These other compounds, including bisflavanols and thearubigins are mainly responsible for the colour change in kombucha beverages (22).

During fermentation, the sucrose is hydrolyzed to glucose and fructose by yeast. Glucose is mainly converted into gluconic acid via the pentose phosphate pathway by acetic acid bacteria, whereas fructose is metabolized to acetic acid, and in minor proportion to gluconic acid (23). The results of sucrose consumption can be observed in Table 1. In both fermented beverages, the kombucha consortium utilized more than 70 % of sucrose for fermentation process. This is consistent with the results of residual glucose and fructose (Table 1), suggesting an active process of fermentation in these herbal infusions (Table 1).

Plant infusions such as E. camaldulensis are traditionally consumed due to their pleasant smell and taste; however, the changes that occur during fermentation process can alter the consumer preference despite their nutraceutical potential. For this reason, we compared the acceptability of infusions and fermented beverages. For this evalu...
evaluation black tea infusion and fermented beverage were included as positive controls, resulting in 5 and 4 score points on a hedonic scale for consumer preference, respectively. Results for *E. camaldulensis* and *L. glaucescens* infusions and fermented beverages are shown in Table 1. Both herbal infusions have similar consumer preferences, but 1 point lower than the black tea infusion. Among fermented beverages, *E. camaldulensis* had lower acceptability, whereas *L. glaucescens* had similar results to the control. It has been reported that biotransformation of several compounds produced during fermentation increases the preference by consumers compared with certain herbal infusions (7). Nevertheless, our results have not shown any differences in the acceptability between herbal infusions and fermented beverages; even when a minor decrease (1 point on hedonic scale) for fermented *E. camaldulensis* beverage was observed. This could be attributed to the formation of other compounds that can influence flavour; however, more chemical studies are needed.

**Phenolic profile and concentrations of herbal infusions and fermented beverages**

Results of phenolic profile of infusions and fermented beverages are shown in Fig. 1. Compounds such as gallic acid, epigallocatechin, gallic acid gallate, catechin, rutin, kaempferol and quercetin were identified in both beverages. The most abundant detected compound was gallic acid gallate (GCG), being higher in *L. glaucescens* infusions and in fermented *E. camaldulensis* beverage.

Chromatogram results (Fig. 1) and concentrations (Table 2) of each identified compound demonstrated that the fermentation influenced the polyphenols present in the herbal infusions of each plant differently. For instance, in fermented *L. glaucescens* beverage, the concentration of GCG decreased significantly (p<0.05; 1.8-fold), whereas in fermented *E. camaldulensis* beverage, the concentration increased (1.8-fold), both compared with their unfermented infusions. A similar trend was observed by Jayabalain et al. (18,24), who reported an increased concentration of total phenolics of kombucha beverages after the fermentation. This could be attributed to the degradation of polyphenol complexes, as a consequence of the increased acidity during fermentation and to the enzymes produced by the kombucha consortium (25).

On the other hand, the decreased concentration of polyphenolic compounds observed in fermented *L. glaucescens* beverage could be explained by the partial oxidation of polyphenols to form polymerized compounds with higher molecular mass (16), and thus lower concentration of these compounds.

**Antioxidant capacity of beverages**

Results of DPPH radical scavenging activity showed that both herbal infusions had antioxidant activity, especially that of *E. camaldulensis*, with its IC$_{50}$ closer to the positive control (Table 3). Despite the chain-breaking DPPH activity results, the fermentation process induced a decrease in the concentration required to stabilize 50 % of
In concordance with our results, Jayabalan et al. (24) reported an enhancement in DPPH scavenging activity in fermented beverages with respect to unfermented tea. Previous studies with kombucha beverages attributed the antioxidant activity to the polyphenols and catechins, with a statistically significant correlation between the phenolic content and the DPPH scavenging capacity (R=0.7180; 24). However, in our work a lower correlation by Spearman’s test (p<0.05) was observed (R=0.3162). This could be explained by the existence of a partial oxidation of polyphenols, forming macromolecular compounds, and therefore decreasing their relative concentration. However, these compounds could still provide a noteworthy radical scavenging activity as we demonstrated in this work.

In order to complement the analysis of antioxidant activity of plant infusions and fermented beverages, we evaluated the inhibition of lipid peroxidation in pre-oxidized healthy human serum. Results showed that both herbal infusions have antioxidant capacity with IC50 values close to the positive control (catechin). Improvement of antioxidant capacity of fermented beverages was not observed, showing similar IC50 values to the unfermented infusions. Jayabalan et al. (18) reported that the antioxidant activity is extremely dependent on the used conditions and the analyzed substrates or products; therefore, not all methods give the same response.

Authors such as Arnao (26) reviewed limitations of antioxidant activity assays, suggesting that the activity must be measured using more than one method, includ-
ing primary and secondary oxidation products, and using tests that measure specific substrates or products. Therefore, we used a third method: the nitric oxide (NO) scavenging assay. Nitric oxide reacts with superoxide (O₂⁻) to form the peroxynitrite anion, which is a potentially strong oxidant as the decomposition of this molecule produces hydroxyl radical and nitrogen dioxide, contributing significantly to oxidative cell damage (21). The results of the NO assay showed that infusions had weaker effect on NO scavenging activity than catechin, used as a positive control (Table 3). However, the antioxidant activity increased with the fermentation (49% of L. glaucescens and 14% of E. camaldulensis), unlike the DPPH test results, where we did not find any statistically significant correlation between the phenolic content and antioxidant activity. A high correlation (R=0.8000) between polyphenol content and NO scavenging response by Spearman’s test (p<0.05) was observed. The same was observed by Jayabalan et al. (24), reporting that during fermentation, polyphenol structural modifications occurred, such as glycosylation, resulting in better scavenging performance on nitrogen radicals.

**Antihypertensive activity in vitro**

Finally, the antihypertensive potential of fermented beverages was evaluated in vitro through the inhibition of the angiotensin-converting enzyme (ACE) activity. ACE is a carboxypeptidase, and it participates in regulating blood pressure by converting an inactive form of the decapeptide angiotensin I to a potent vasoconstrictor octapeptide, angiotensin II. Therapeutic ACE inhibitors represent a carboxypeptidase, and it participates in regulating blood pressure. On the other hand, fermented beverages did not improve fasting and postprandial dyslipidemia and reduce oxidative stress: a randomized controlled trial. Am J Clin Nutr. 2014;99:463–71.

http://dx.doi.org/10.3945/ajcn.113.073445


http://dx.doi.org/10.1186/1472-6882-12-63


http://dx.doi.org/10.17113/ftb.52.04.14.3611


http://dx.doi.org/10.1080/23311932.2015.1059033


**Conclusions**

The present study has demonstrated that fermentation of Litsea glaucescens and Eucalyptus camaldulensis infusions with kombucha consortium modifies their concentration of phenolic compounds, and their antioxidant and antihypertensive activities. Fermented beverages exhibited increased free radical scavenging activities, but this response depends on the used medium, substrate and mainly on the plant material used for infusion preparation. Additionally, it is interesting to note that E. camaldulensis and L. glaucescens can be considered as natural sources of biocompounds with antihypertensive potential (especially the latter) either as infusions or fermented beverages. Therefore, they can be considered for further investigations, especially for *in vitro* assays.

**Acknowledgements**

This project was supported by Redes Temáticas de Colaboración Académica: Nanotecnología y Omics para el estudio de Nutracéuticos (SEP-PRODEP). We acknowledge the Area of Ecology and Systemics from CIIDIR-IPN, Unidad Durango for their support. Also we appreciate the technical support of Ignacio Velázquez-Jiménez. The authors gratefully acknowledge Bogar Vallejo and Tradu-c services for language revision.

**References**


http://dx.doi.org/10.1021/jf010759r


http://dx.doi.org/10.3945/ajcn.113.073445


http://dx.doi.org/10.1186/1472-6882-12-63


http://dx.doi.org/10.17113/ftb.52.04.14.3611


http://dx.doi.org/10.1080/23311932.2015.1059033


http://dx.doi.org/10.1080/23311932.2015.1059033