Hydrophilic Extractives in Heartwood of European Larch (Larix decidua Mill.)

Janja Zule¹, Katarina Čufar², Vesna Tišler²

ABSTRACT • The heartwood of two European larch trees was examined for the content of hydrophilic extractives. Ethanol was used as a solvent for extractions of adequately pretreated samples, while gas chromatography with flame ionisation detection (GC-FID), gas chromatography coupled to mass spectrometry (GC-MS) and high performance size exclusion chromatography (HPLC-SEC) were applied for analyses. Different phenolic compounds, such as flavonoids and lignans were identified and quantitatively evaluated. The most abundant flavonoids were taxifolin and dihydrokaempferol, while the main lignan was secoisolariciresinol. Contents of flavonoids and lignans, measured at different stem heights, varied between 10.6 and 15.7 mg/g. At lower and medium heights flavonoids prevailed, while at upper stem positions the concentrations of lignans significantly increased. In addition, some trimeric and tetrameric phenolic structures were also detected. There was very little variation in the composition of hydrophilics between the two examined trees. The identified flavonoids and lignans are classified as very strong antioxidants.

Key words: European larch, heartwood, hydrophilic extractives, flavonoids, lignans, chromatographic techniques

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Ključne riječi: europski ariš, srž, hidrofilni ekstraktivi, flavonoidi, lignani, kromatografske tehnike
1 INTRODUCTION
1. UVOD

Hydrophilic extractives are nonstructural wood components, which are readily soluble in polar solvents, such as acetone, methanol, ethanol and water. In contrast to lipophilic compounds in resin canals and parenchyma cells of sapwood (Zule et al., 2015), they mainly originate from the transition zone between heartwood and sapwood. Their concentrations are the highest at the boundary, after which they gradually decrease in radial direction towards the pith. Hydrophilics are mostly composed of phenolic compounds. Their synthesis is genetically regulated so each tree species contains specific compounds by which it differs from other species. Phenolic type compounds are comprised of aromatic components from simple phenols to complex phenolic polymers.

The most important are compounds with two phenolic groups, e.g. stilbenes, flavonoids and lignans (Hillis, 1962; Shen et al., 1986; Kolbit et al., 1996; Debell et al., 1997; Celimene et al., 1999; Eaton and Hale, 1999; Kleist and Schmitt, 1999; Stenius, 2000, Willför et al., 2004). Flavonoids have a characteristic diphenylpropane (C_6C_3C_6) carbon skeleton. Both aromatic rings in the molecule usually contain hydroxyl groups. The most abundant flavonoids in wood are dihydrokaempferol (C_{15}H_{12}O_{6}, M_w=288), taxifolin (C_{15}H_{12}O_{7}, M_w=304), narigin (C_{15}H_{12}O_{6}, M_w=272) and catechin (C_{15}H_{12}O_{7}, M_w=290) (Hillis, 1962; Sjöstrom, 1981; Stenius, 2000). In addition to simple flavonoids, there are also flavonoid polymers with 3 to 8 flavonoid units, which are known as condensed tannins.

Lignans are an important group of plant phenols and are widespread in the plant world. Typical for them is oxidative coupling of two phenylpropene (C_5C_3C_3) units via linkage between β-positions on propane side chains. In addition to dimeric lignans, minor quantities of trimers (sesquilignans) and tetramers (dilignans) can also be found in wood. They are generally called oligolignans. Several tens of lignans have been identified so far in wood tissues. However, the predominating are dimeric components, e.g. hydroxyimatairesinol (C_{20}H_{22}O_{7}, M_w=374), pinoresinol (C_{20}H_{22}O_{6}, M_w=358), dimeric components, e.g. hydroxyimatairesinol so far in wood tissues. However, the predominating are flavonoids. Their concentrations, measured in radial direction at representative stem height (1.5 m), were highest at the heartwood/sapwood boundary.

Some researchers studied the correlation between the content of hydrophilics and natural durability of larch heartwood (Doi et al., 1998; Ohmura et al., 1999; Windiesen et al., 2002; Windiesen and Wegener, 2003; Gierlinger et al., 2002, 2003, 2004). Strong connection between extractive structure and rotting resistance has been confirmed.

Recently, many studies have also been dedicated to evaluation of larch wood as potential source of technologically important highly bioactive phenolic compounds, such as taxifolin, dihydrokaempferol and secoisolariciresinol (Babkin et al., 1999; Babkin et al., 2001; Aleksandrova et al., 2002; Ivanova et al., 2012; Ostroukhova et al., 2012; Wen-jie et al., 2005, Willför et al., 2005).

Most authors focused on chemical characterization of larch heartwood at the representative stem height (1.5 m), while there is almost no data on the content of extractives at different positions along the stem and in branches. The aim of the present work was to make a detailed chemical characterization of hydrophilics in the heartwood of European larch (Larix decidua Mill.) and to determine their variability in longitudinal direction within a tree and between two trees selected for analyses. This article is the continuation of the previously published work (Zule et al., 2015), where lipophilic extractive compounds of the same samples were characterized.

2 MATERIALS AND METHODS
2. MATERIJALI I METODE

2.1 Samples
2.1. Uzorci

Two larch trees (Larix decidua Mill.) were felled at the end of June, 2009 in the Alpine region of Slovenia at the altitude of 1000 m. They originated from a mixed forest where beech (Fagus sylvatica L.) and spruce (Picea abies Karst.) predominated. Both larch trees were 180 years old and about 30 m high. They were healthy without visual damage.

About 5 cm thick discs were cut from the trunks at the base (50 cm above ground) as well as at the heights of 8, 18 and 28 m. The discs were debarked and representative sections of heartwood (circular sectors from the pith to the heartwood/sapwood boundary) were cut into smaller pieces, which were subsequently frozen at -24°C prior to analysis.

2.2 Drying and grinding
2.2. Sušenje i mljevenje

Frozen samples were first conditioned at room temperature, after which they were cut into short splin-
2.3 Extraction

Extractions were conducted by means of an accelerated solvent extraction system (ASE) using the instrument Dionex ASE 200. About 5 g of each freeze-dried powdered sample was weighed into a metal extraction cell and sequentially extracted first with hexane (V-50 ml) to remove lipophilic fraction (Zule et al., 2015) and afterwards with 95 % ethanol (V-50 ml) to isolate hydrophilic extractive portion. The temperature of the first extraction step was 90 °C, pressure 13.8 MPa and extraction time 10 minutes (2 static cycles with static time of 5 minutes). Ethanol extraction was carried out at 100 °C under the same experimental conditions. Both extractions were performed under the stream of nitrogen.

2.4 Derivatization of extractives

All ethanol extracts were derivatized prior to chromatographic analyses (GC-FID, GC-MS) by which components with hydroxyl groups, such as flavonoids and lignans, were converted to the corresponding trimethylsilyl (TMS) derivatives, which were less polar and so more convenient for subsequent chromatographic analyses. 2 ml of internal standard solution were added to each extract, containing about 0.5 mg of extractive compounds. Internal standards were hexenicosanoic acid (S1), betulinol (S2), cholesterol esterase and 1.3-dipalmitoyl-2-oleyl glycerol (S4), all having concentration of 0.02 mg/ml. The mixture of all ethanol extracts were derivatized nonderivatized ethanol extracts of heartwood samples were dried under the stream of N2, after which they were redissolved in THF, so that the concentration of the last was 1 mg/ml. The analyses were carried out on the HP 6890-5973 GC-MSD instrument. The separation was carried out on the HP-1 (25 m x 0.20 mm x 0.11 μm) capillary column under the following experimental conditions: temperature program of column heating: 120 °C, 6 °C/min, 300 °C (10 min); carrier gas He (0.8 ml/min); split injector (1:20) – 160 °C, 8 °C/min, 260 °C (15 min); FID detector: 320 °C; injection volume 1 μl. Betulinol (S2) served as standard in determination of phenolic compounds. The latter were calculated by comparison of the corresponding peak areas using correction factor 1.2. All results are expressed as milligram per gram of dry sample weight, where the limit of quantification was about 0.01 mg/g (Willfor, 2007).

2.6 Gravimetric determination of the content of hydrophilic extractives

10 ml aliquots of ethanol extracts were dried under the stream of N2, and in vacuum desiccator at 40 °C until a constant weight was reached. The gravimetric amount of extractives was calculated as milligram per gram of dry sample weight, where the limit of quantification was about 0.01 mg/g (Willfor, 2007).

2.5 Chromatographic analyses of extractives

2.5.1 Identification of extractive compounds by GC-MS

Characteristic components of the representative ethanol extracts were identified by means of gas chromatography coupled to mass spectroscopy (GC-MS). The analyses were performed on the HP 6890-5973 GC-MSD instrument. The separation was carried out on the HP-1 (30 m x 0.25 mm x 0.25 μm) capillary column under the following experimental conditions: temperature program of column heating 80 °C, 8 °C/min, 290 °C; carrier gas He (0.9 ml/min); split injector (1:15) – 260 °C; MS-EI detector (source temp. 280 °C, 70 eV, quadrupole temp. 180 °C). Mass range (m/z) was between 10 and 1050. For positive identification of individual compounds, the mass spectra of their chromatographic peaks were compared with spectra of pure compounds from the Wiley Registry NIST 2008 Mass Spectral Library. On both GC-FID and GC-MS chromatograms there was practically the same sequence of chromatographic peaks of individual compounds as similar long capillary columns were used, by which reliability of identification was ensured and quantitative work facilitated (Willfor, 2007).

2.5.2 GC-FID analysis on long capillary column

The determination of individual flavonoids, lignans and other hydrophilic compounds was accomplished by means of gas chromatography on the Perkin Elmer AutoSystem XL instrument. HP-1 (25 m x 0.20 mm x 0.11 μm) capillary column was used for separation of compounds under the following experimental conditions: temperature program of column heating: 120 °C, 6 °C/min, 300 °C (10 min); carrier gas He (0.8 ml/min); split injector (1:20) – 160 °C, 8 °C/min, 260 °C (15 min); FID detector: 320 °C; injection volume 1 μl. Betulinol (S2) served as standard in determination of phenolic compounds. The latter were calculated by comparison of the corresponding peak areas using correction factor 1.2. All results are expressed as milligram per gram of dry sample weight, where the limit of quantification was about 0.01 mg/g (Willfor, 2007).

2.5.3 Characterization of phenolic compounds by HPLC-SEC

The presence of dimeric, trimeric and tetrameric phenols in representative ethanol extracts was thus confirmed as higher phenols can not be detected by gas chromatography due to their low volatility. The selected nonderivatized ethanol extracts of heartwood samples were dried under the stream of N2, after which they were redissolved in THF, so that the concentration was 1 mg/ml. The analyses were carried out on the chromatographic system, composed of the following units: autosampler Spark Holland Marathon-XT, pump Shimadzu LC – 10ATVP, chromatographic columns 2x Jordi Gel DVB 500A (300 mm x 7.8 mm) and detector Sedere SEDEX 85 ELSD, which is a low temperature evaporative light scattering detector. Tetrahydrofuran (THF) was used as eluent with the flow of 0.8 ml/min. The injection volume was 50 μl.

2.6 Gravimetric determination of the content of hydrophilic extractives

The gravimetric amount of extractives was calculated as milligram per gram of dry sample weight, where the limit of quantification was about 0.01 mg/g (Willfor, 2007).
gram of dry sample weight. All quantitative determinations of hydrophilic extractives in the samples of heartwood were performed in two parallels. The presented results are average values of individual determinations.

3. RESULTS AND DISCUSSION
3. REZULTATI I RASPRAVA

3.1 Gravimetric determination of ethanol extracts
3.1. Gravimetrijsko određivanje etanolnih ekstrakata

Hexane extracted heartwood samples, from which lipophiles (free fatty and resin acids, diterpenoids, triglycerides, steryl esters and sterols) had been removed, were consecutively extracted in the same way by a polar solvent ethanol (95 %) to obtain hydrophilic extractives. Gravimetrically determined content of ethanol extract in the heartwood of both trees is presented in Figure 1. Evidently, the concentrations were slightly higher in the base of the trees (0.5 m), however they remained more or less constant further up the stems. The average values were 28.0 ± 2.8 mg/g for Tree 1 and 30.5 ± 2.7 mg/g for Tree 2, calculated on dry mass of heartwood. Obviously, there was no significant difference between the two trees.

Ethanol proved to be a suitable polar solvent. It is less toxic than methanol and less flammable than acetone. It can be easily recycled. The ASE extraction method has many advantages over commonly used Soxhlet method. It is automated and computer controlled. The whole process is very quick due to the application of elevated temperature and pressure, while solvent consumption is significantly reduced. It is indispensable method for sequential analyses of large numbers of samples. The results are comparable to Soxhlet.

3.2 Identification of individual phenolic compounds (GC-MS)
3.2. Identifikacija pojedinačnih fenolnih spojeva (GC-MS)

Identification was performed by GC-MS analyses of two typical ethanol extracts of the heartwood of tree 1 at 0.5 and 28 m of height. The following flavonoids and lignans were identified: naringenin, taxifolin (2 isomers), dihydrokaempferol, secoisolariciresinol, isoliovil, lariciresinol, todolactol A and nortrachelogenin. In addition, some monomeric and dimeric sugar units were also detected in the extracts.

3.3 Determination of the composition of ethanol extracts related to stem height
3.3. Određivanje sustava etanolnih ekstrakata u odnosu prema visini stabla

The composition of ethanol extracts was established from GC-FID chromatograms, recorded on a 25 m long capillary column. Qualitative composition of phenols was almost the same for all heartwood samples of both trees up to the stem height of 18 m, while it considerably changed at the height of 28 m near the top of the trees. The flavonoids dihydrokaempferol and taxifolin predominated in all samples at lower and middle stem heights, where lignans were present only in trace amounts. On the other hand, the concentrations of lignans significantly increased at the top of both trees at 28 m. Secoisolariciresinol was far the most abundant lignin in the heartwood of both trees.
Total flavonoid and lignan concentrations, which were calculated from the corresponding GC-FID chromatograms, varied between 10.6 and 12.6 mg/g in tree 1, while the corresponding values in tree 2 were slightly higher – between 12.4 and 15.7 mg/g. Concentrations of dihydrokaempferol were in the range between 1.8 and 7.1 mg/g and those of taxifolin between 3.3 and 8.4 mg/g.

Secoisolariciresinol was the most abundant lignan and it was even the predominant phenolic compound at the top of tree 1 (Figure 4). The concentration of secoisolariciresinol in the heartwood of tree 1 at 28 m was 4.2 mg/g and in the heartwood of tree 2 at the same height it was 3.4 mg/g. It is very interesting that its concentration never exceeded 0.2 mg/g in the lower sections of heartwood.

The content and composition of phenols in the heartwood of trees 1 and 2 in relation to stem height is shown in Figures 4 and 5.

From the comparison of gravimetric and chromatographic results, it was evident that the contents of ethanol extracts were at least twice as high as the chromatographically determined concentrations of flavonoids and lignans in those extracts. This could be partly ascribed to the presence of different sugar units, such as mono- and disaccharides in ethanol extracts, as they are readily soluble in polar solvents. However, some substances, which were simultaneously extracted by ethanol, could not be detected by GC-FID due to their...
higher molecular mass and thus lower volatility. The presence of higher phenolic structures was unambiguously confirmed by HPLC-SEC (Figure 6).

Figure 6 shows the molecular mass distribution of a THF redissolved typical ethanol extract. The most intensive peak (L2) with the retention time of 21 min represents dimeric phenols, e.g., flavonoids and lignans, while weaker peaks with retention times between 18 and 20 min (L3 and L4) symbolize higher phenols having three and four phenolic groups in their molecular structure. According to the peak area ratio, it could be estimated that typical ethanol extracts of larch heartwood contained averagely about 20% of higher phenols. Sugar units and other non phenolic components were not detected by HPLC-SEC under specified experimental conditions.

The most abundant phenolic compounds of European larch, e.g., dihydrokaempferol, taxifolin and secoisolariciresinol exhibit, according to available literature data, very strong antioxidative properties, which can be ascribed to their specific molecular structure. The structural formulas of the most typical larch flavonoids and lignans are presented in Figure 7.

The biological activity of taxifolin could be attributed to the relatively high content of phenolic hydroxyl groups in the right ring of its molecular structure, while on the other hand secoisolariciresinol is very efficient antioxidant on account of its butanediol structure (Figure 7). Presence of higher trimeric and tetrameric phenols also positively affects antioxidative properties of extractives and thus biological resistance of wood tissues against rotting (Pietarinen et al. 2006; Willför et al. 2003; Scalbert, 1991; De Bruyne et al. 1999).

The most significant finding in the case of larch heartwood was that chemical structure of hydrophilic extractive fraction changed towards the top of both examined trees, which was not the case with lipophilic fraction of the same tissues (Zule et al., 2015). While the flavonoids taxifolin and dihydrokaempferol predominated in the majority of heartwood, lignans appeared at the top in abundant concentrations with secoisolariciresinol as the main phenolic compound. Such distribution points to the fact that lignans are synthesized during early growth period, while later on, during wood aging, the synthesis proceeds more and more in the direction of flavonoid formation (Willför, 2002).

Larch forest residues, such as tree tops, damaged wood, cuttings, sawdust, knots and branches from wood processing could serve as relevant source for large scale isolation of valuable bioactive compounds. The latter may be applied as “green chemicals” or natural preservatives in pharmaceutic, food, chemical and other industries. The remaining extracted wood may be further chemically converted to different platform chemicals and biofuel, which is the main idea of wood biorefineries.

4 CONCLUSIONS

Nine different phenolic compounds, such as flavonoids and lignans were determined in ethanol ex-

![Figure 6](image-url)  
**Figure 6** HPLC-SEC chromatogram of ethanol extract of heartwood of tree 1 (8 m)  
*Slika 6.* HPLC-SEC kromatogram etanolnog ekstrakta srži stabla 1. (8 m)

![Figure 7](image-url)  
**Figure 7** Structural formulas of taxifolin (1), dihydrokaempferol (2), secoisolariciresinol (3) and lariciresinol (4)  
*Slika 7.* Strukturne formule taksifolina (1), dihidrokemferola (2), sekoizolaricirezinola (3) i laricirezinola (4)
tracts of the heartwood of two European larch trees. All substances identified, e.g. naringenin, taxifolin (2 isomers), dihydroykaempferol, secoisolariciresinol, la ricresinol, isolovil, norlarchogenin and todolactol A, are typical for larch species (European, Siberian, Japanese, Western, Tamarack and others) and are commonly not found in other conifers in any significant amounts. Their average concentrations in the heartwood of the two examined trees were 12 ± 2 mg/g and 15 ± 2 mg/g, and did not change essentially in vertical direction along the stems. Flavonoids predominated at lower and middle positions while lignans were more abundant at the top of both trees. The most important flavonoid taxifolin and lignan secoisolariciresinol are classified as very powerful antioxidants which, in combination with other phenolic substances, most likely provides efficient chemical protection of larch heartwood against rotting and harsh environmental conditions. Our study may contribute to better understanding of the chemistry of wood tissues.

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