INTRODUCTION

The reduction of area in the through thickness direction is an essential mechanical property of thick steel heavy plates. By a routine control, a very small through thickness reduction of area was found for tensile specimen of a 90 mm plate. Careful investigations of the fracture and section of specimens cut from the as solidified continuously cast 250 mm thick slab showed that the cause was the presence of coarse particles of niobium carbonitride as constituent of the quasi eutectic Fe-Nb(C, N) that form because of the centerline segregation of niobium.

Key words: structural steel, heavy plates, through thickness reduction of area, eutectic niobium carbonitride

The mass fraction of sulphur in the tested steel, which was only 0.003 mass %, excludes the possibility of low reduction of area because of sulphide inclusions. A small content of niobium is added to the investigated structural steel to achieve the required mechanical properties [4]. The addition of Nb could also affect the through thickness ductility of heavy plates because of the formation of coarse niobium carbonitride particles as constituents of the degenerated eutectic Fe-Nb(C, N), that may form at high content of niobium or because of a defective steel solidification [5, 6].

Table 1. Chemical composition of the S 355 J2+N steel grade

<table>
<thead>
<tr>
<th>Element</th>
<th>C</th>
<th>Si</th>
<th>Mn</th>
<th>P</th>
<th>S</th>
<th>Cr</th>
<th>Cu</th>
<th>Ni</th>
<th>Al</th>
<th>Nb</th>
<th>Ti</th>
<th>N</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mass/ %</td>
<td>0.15</td>
<td>0.49</td>
<td>1.1</td>
<td>0.018</td>
<td>0.003</td>
<td>0.14</td>
<td>0.29</td>
<td>0.12</td>
<td>0.033</td>
<td>0.022</td>
<td>0.005</td>
<td>0.0071</td>
</tr>
</tbody>
</table>

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To identify the cause for the low reduction of area a detailed investigations of specimens cut at different distances from the surface of the as solidified slab were carried out [7].

EXPERIMENTAL WORK

The structural steel S 355 J2+N was melted in electric arc furnace (EAF), vacuum degassing (VD) treated, continuously cast, and cut into slabs of dimensions 250 × 1085 × 4770 mm. The slabs were cooled to ambient temperature and after surface grinding reheated in pusher-type furnace to the temperature 1250 °C and hot rolled to 90-mm thick plates. The chemical composition of the steel is listed in Table 1. First, samples perpendicular to the slab casting direction were examined after grinding and deep-etching for 40 minutes in 25 % H₂SO₄ at 70 °C that revealed the as cast macrostructure.

From this specimen, samples A, B and C in thickness direction were cut out for metallographical examination, as shown in Figure 2. From the 90 mm heavy plate specimens were cut out in thickness direction and submitted to tensile testing and examination with electron scanning microscopy (SEM) as well as energy dispersive X-ray spectroscopy (EDXS).

RESULTS AND DISCUSSION

In Figure 3 the secondary electron image of fracture surface of one specimen with coarse niobium carbide nitride inclusions and small MnS inclusions is shown. The spots of the EDXS analysis of three inclusions are marked with arrows in Figure 3, and the results are given in Table 2.

In the mapping micrographies in Figure 4 bright areas represent the element in the particles and shows the morphology of the particles and the main element in large inclusions [8]. Most of the particles observed on the fracture surface showed a high content of niobium. On the base of fractographies it was concluded that niobium containing particles (Nb, Ti)(C, N) were the main cause for the poor through thickness reduction of area of the steel plate.

<table>
<thead>
<tr>
<th>Element</th>
<th>C</th>
<th>Fe</th>
<th>Mn</th>
<th>S</th>
<th>Ti</th>
<th>Nb</th>
<th>Al</th>
<th>Pb</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mass/ %</td>
<td>(Nb, Ti) (C, N)</td>
<td>3,18</td>
<td>33,09</td>
<td>-</td>
<td>-</td>
<td>3,76</td>
<td>59,97</td>
<td>-</td>
</tr>
<tr>
<td>MnS</td>
<td>-</td>
<td>3,56</td>
<td>61,89</td>
<td>33,09</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Pb</td>
<td>2,03</td>
<td>8,72</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>0,96</td>
<td>88,29</td>
</tr>
</tbody>
</table>

Table 2. Results of the spot EDXS analyses (see Figure 3)

<table>
<thead>
<tr>
<th>Element</th>
<th>C</th>
<th>Fe</th>
<th>Mn</th>
<th>S</th>
<th>Ti</th>
<th>Nb</th>
<th>Al</th>
<th>Pb</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mass/ %</td>
<td>(Nb, Ti) (C, N)</td>
<td>4,27</td>
<td>26,95</td>
<td>-</td>
<td>-</td>
<td>2,47</td>
<td>66,31</td>
<td>-</td>
</tr>
<tr>
<td>MnS</td>
<td>-</td>
<td>3,56</td>
<td>61,89</td>
<td>33,09</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>-</td>
</tr>
<tr>
<td>Pb</td>
<td>1,73</td>
<td>5,14</td>
<td>1,1</td>
<td>-</td>
<td>-</td>
<td>-</td>
<td>1,62</td>
<td>90,41</td>
</tr>
</tbody>
</table>

Table 3. Results of the spot EDXS analyses (see Figure 5)
From the location of the fracture of tensile specimen shown in Figure 1, we assumed that the source of the coarse precipitates was very strong centreline segregation during solidification of the steel slab. This conclusion is confirmed by the fact that from three samples, as shown in Figure 2 the niobium rich precipitates were found only in the specimen cut from the slab centre in the sample A. The precipitates are very similar to the Fe-Nb(C, N) eutectic called as »Chinese script« [9]. Beside the niobium rich particles, a minor number of very small manganese sulphide inclusions and lead droplets were found, also. All those phases were found only in the centreline of the cast slab. Results of the spot EDXS analyses from the cast slab are presented in Table 3 and the spots of analyses are marked in Figure 5.

The analyses show, that niobium carbonitride particles contain also approximately 2 to 4 mass % of Ti despite of only 0,005 mass % of titanium in steel originating from the steel scrap used. A similar composition of niobium carbonitride was reported in Nb-Ti microalloyed steels [10]. The solubility of niobium carbonitride of approximate composition Nb(C_{0.9}, N_{0.1}) in structural steel is given by the equation [11,12]:

\[
\log \left( \frac{[\text{Nb}]}{[\text{C}]^2 \cdot [\text{N}]^{14}} \right) = \frac{6770 - 226}{7} T (1)
\]

with [Nb], [C], [N] as weight contents of the elements in the steel and T as temperature in K. Considering the actual contents of niobium, carbon and nitrogen, the solution temperature of 1140 °C was deduced indicating that the slab soaking temperature was sufficient for a complete, solution in austenite of the niobium carbonitride of approximate composition Nb(C_{0.9}, N_{0.1}).

The fact that coarse niobium rich precipitates were found also in the hot rolled plate after heating the slabs to 1250 °C indicates that their composition differs from that of soluble niobium carbinitride. The solubility for niobium carbide in austenite is greater than the solubility for niobium nitride [13]. It is assumed that the stability of particles in the investigated steel is due to their high content of nitrogen [14]. The shape and size of coarse carbonitride particles suggest that they are constituents of a degenerated quasi eutectic Fe-Nb(C, N). The location of the eutectic in the centre of the slab and the composition of the steel suggest that its formation is an improper solidification process related to a high casting temperature, a high slab solidification rate or a deficiency in the secondary slab cooling.

CONCLUSIONS

Considering the content of carbon, nitrogen and niobium in the steel, all carbonitride phase with the approximate composition Nb(C_{0.9}, N_{0.1}) is in solid solution in austenite at 1140 °C. Since the slab soaking temperature was 1250 °C, it is evident that the carbonitride found in the examined steel has not the quoted composition and that it has a higher content of nitrogen and correspondingly higher solution temperature in austenite.

The shape and size of the niobium rich particles suggest that they are the constituents of a degenerated eutectic Fe-Nb(C, N) that formed during solidification process of continuous cast slabs due to micro segregation of niobium.

The dissolution of this part of carbonitride eutectic does not follow the solubility product as is the case for niobium carbinitride formed by precipitation and is not dissolved during reheating even at low initial Nb contents. This is the fact that niobium carbonitride eutectic results in reduced ductility particularly in the through thickness direction.

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REFERENCES


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