

Synthesis and potential cytotoxic activity of some new benzoxazoles, imidazoles, benzimidazoles and tetrazoles

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The present work deals with the synthesis of some novel heterocyclic compounds such as benzoxazoles **2**, **7**, **13** and **19**, imidazoles **3**, **8**, **14** and **20**, benzimidazoles **4**, **9**, **15** and **21**, and tetrazoles **10**, **16**, and **22**. The synthesized compounds were characterized by IR, ¹H NMR, mass spectrometry and elemental analysis. The compounds were evaluated for cytotoxicity against human cancer cell lines such as MCF-7 (breast cancer) and HT-29 (colon cancer) by the MTT assay method. Among the tested compounds, 4,4'-sulfonylbis(*N*-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)aniline (**9**), *N*-bis(2-(benzo[*d*]oxazol-2-yl)-ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**13**), *N*-bis(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**15**) and *N*-tris(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (**21**) showed potent cytotoxicity.

Keywords: benzimidazoles, imidazoles, benzoxazoles, tetrazoles, synthesis, cytotoxicity, MTT assay

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Tumor can be defined as an abnormal mass of tissue, the growth of which exceeds and is uncoordinated with that of normal tissue and continues in the same manner after cessation of the stimuli that have initiated it (1). Control of disseminated tumor growth by systemically active chemotherapeutic agents remains a major challenge for cancer chemotherapy despite decades of focused efforts. Although there has been some notable success with certain forms of cancer, drug therapy has only a limited impact upon the three major killers: carcinoma of the lung, breast, and colorectal system (2).

Heterocyclic compounds are rich source of diverse physical, chemical and biological properties (3). Good therapeutic properties of the imidazole related drugs have encouraged medicinal chemists to synthesize a large number of novel chemotherapeutic agents. Medicinal properties of imidazoles include anticancer, inhibiting β -lactamase, 20-HETE (20-hydroxy-5,8,11,14-eicosatetraenoic acid) synthase, carboxypeptidase, heme oxygenase, antiaging, anticoagulating, anti-inflammatory, antibacterial, antifungal, antiviral,

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antitubercular, antidiabetic and antimalarial activities (4). Benzimidazoles are heterocyclic compounds having various biological activities (5), among which albendazole, mebendazole and thiabendazole are widely used anthelmintic drugs. Furthermore, benzimidazole compounds are also reported to possess biological activities such as inhibition of the angiotensin receptor TIE-2 and tyrosine kinase receptor VEGFR-2 (vascular endothelial growth factor receptor-2), antitumor activity, thrombopoietin receptor agonistic activity, gamma-aminobutyric acid (GABA) agonistic activity, antimicrobial activity, topoisomerase inhibition, neuropeptide Y1 receptor antagonistic activity and inhibition of angiotensin II. Tetrazoles can act as pharmacophores for the carboxylate group and increase its utility (6, 7). Angiotensin II blockers often contain tetrazoles such as losartan and candesartan. A well-known tetrazole is dimethyl thiazolyl diphenyl tetrazolium salt (MTT), which is used in the MTT assay to quantify the respiratory activity of live cells in cell cultures, although it kills the cells in the process. Recent observations suggest that substituted benzoxazoles and related heterocycles possess potential activity with lower toxicity in the chemotherapeutic approach in man (8). Careful literature survey revealed that targets containing benzoxazole moiety, either isolated from plants or accessed by total synthesis, have remarkable biological activities, such as antimicrobial, antihistaminic, antiparasitic, herbicidal, antiallergic and antihelmintic activities. They were screened for cytotoxic activity by the MTT assay.

EXPERIMENTAL

Melting points were determined with a digital melting point apparatus (LAB India MR-VIS, India) and are uncorrected. Elemental analyses were performed on a Perkin-Elmer 2400 analyzer (Perkin-Elmer, USA) and were found within ± 0.4 % of the theoretical values. FTIR spectra were recorded on a Shimadzu FT-IR model spectrophotometer (Japan). ^1H NMR spectra were recorded in DMSO on a Bruker AV III 500 MHz (USA) using TMS as internal standard. Mass spectra were recorded on a JEOL GCmate instrument (Japan). The purity of the compounds was checked by TLC on pre-coated silica gel G (HF₂₅₄) aluminium plates (Merck, USA) using chloroform/methanol (4:1) and visualized in a UV chamber.

Compounds 3-(4-(2-methyl-4-oxoquinazolin-3(4*H*)-yl)phenylamino)propanenitrile (1), 3,3'-(4,4'-sulfonylbis(4,1-phenylene)bis(azanediyl))dipropanenitrile (6), 3,3'-(6-phenyl-1,3,5-triazine-2,4-diyl)bis(azanediyl)dipropanenitrile (12) and 3,3,3',1,3,5-triazine-2,4,6-triyl)tris(azanediyl)tripropanenitrile (18) were prepared from primary amine by treatment with acrylonitrile and Triton B following reported literature (9).

Syntheses of benzoxazoles, imidazoles and benzimidazoles. – 3-(4-(2-(benzo[*d*]oxazol-2-yl)-ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (2), 4,4'-sulfonylbis(*N*-(2-(benzo[*d*]oxazol-2-yl)ethyl)aniline (7), *N*-bis(2-(benzo[*d*]oxazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (13), *N*-tris(2-(benzo[*d*]oxazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (19), 3-(4-(2-((1*H*-imidazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (3), 4,4'-sulfonylbis(*N*-(1*H*-imidazol-2-yl)ethyl)aniline (8), *N*-bis(2-1*H*-imidazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (14), *N*-tris(2-1*H*-imidazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (20), 3-(4-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-

Table I. Physical and analytical data of synthesized compounds

Compd.	Mol. formula	M_r	Yield (%)	M. p. (°C)	Elemental analysis (calcd./found, %)		
					C	H	N
2	$C_{24}H_{20}N_4O_2$	396.16	74	310–313	72.71	5.08	14.13
					72.77	5.18	14.17
3	$C_{20}H_{19}N_5O$	345.16	77	257–259	69.55	5.54	20.28
					69.61	5.36	20.32
4	$C_{24}H_{21}N_5O$	395.17	75	299–301	72.89	5.35	17.71
					72.91	5.41	17.66
6	$C_{18}H_{18}N_4O_2S$	354.12	68	99–102	61.00	5.12	15.81
					61.13	5.17	15.90
7	$C_{30}H_{26}N_4O_4S$	538.17	72	240–242	66.90	4.87	10.40
					66.63	4.89	10.22
8	$C_{22}H_{24}N_6O_2S$	436.17	71	197–200	60.53	5.54	19.25
					60.61	5.63	19.27
9	$C_{30}H_{28}N_6O_2S$	536.20	76	230–232	67.14	5.26	15.66
					66.95	5.18	15.61
10	$C_{18}H_{20}N_{10}O_2S$	440.15	69	99–103	49.08	4.58	31.80
					49.17	4.61	31.97
12	$C_{15}H_{15}N_7$	293.33	64	99–102	61.42	4.15	33.43
					61.56	4.21	33.48
13	$C_{27}H_{23}N_7O_2$	477.19	67	167–169	67.91	4.85	20.53
					67.96	4.83	20.47
14	$C_{19}H_{21}N_9$	375.19	69	197–199	60.78	5.64	33.58
					60.74	5.67	33.61
15	$C_{27}H_{25}N_9$	475.22	71	219–220	68.19	5.30	26.51
					68.12	5.34	26.44
16	$C_{15}H_{17}N_{13}$	379.17	68	180–200	47.49	4.52	48.00
					47.44	4.49	48.04
18	$C_{12}H_{15}N_9$	285.15	63	95–97	50.52	5.30	44.18
					50.49	5.27	44.25
19	$C_{30}H_{27}N_9O_3$	561.22	72	138–139	64.16	4.85	22.45
					64.12	4.82	22.41
20	$C_{18}H_{24}N_{12}$	408.22	66	210–212	52.93	5.92	41.65
					52.89	5.97	41.69
21	$C_{30}H_{30}N_{12}$	558	75	184–185	64.50	5.41	30.09
					64.46	5.32	30.18
22	$C_{12}H_{18}N_{18}$	414.20	73	248–250	34.78	4.38	60.84
					34.73	4.42	60.87

Table II. Spectral data of synthesized compounds

Compd.	IR (KBr) (ν , cm^{-1})	^1H NMR (DMSO- d_6) (δ , ppm)	MS, m/z (%)
2	3306 (C-H), 2921 (CH_2), 1507 (C=N), 3400 (-NH)	2.08 (s, 3H, CH_3), 7.53 (s, 12H, Ar-H), 3.34 (s, 2H, CH_2), 3.02 (s, 2H, CH_2)	395.54 (M-1, 62), 348.98 (21), 207.20 (52), 56.11 (100)
3	3221 (C-H), 2926 (CH_2), 1609 (C=N), 3346 (-NH)	6.7–7.5 (m, 8H, Ar-H), 3.37 (s, 2H, CH_2), 2.48 (s, 3H, CH_3), 8.2 (s, NH)	346.78 (M+1, 12), 141.11 (100), 215.91 (16), 129.21 (64), 115.15 (52), 105.19 (29)
4	3055 (C-H), 2926 (CH_2), 1627 (C=N), 3315 (-NH)	6–6.4 (m, 8H, Ar-H), 3.6 (s, 2H, CH_2 -C), 2.9 (s, 2H, CH_2), 2.48 (s, 3H, -N=C- CH_3), 8.1 (s, NH benzimidazole)	394.54 (M-1, 70), 294.91 (72), 279.98 (75), 269.23 (78), 206.26 (100)
7	3100 (C-H), 2929 (CH_2), 1591 (C=N), 3350 (-NH)	7.31–7.5 (m, 8H, benzoxazole), 7.5 (d, 4H, $J = 11.5$, Ar-H), 6.57–6.58 (d, 4H, $J = 8.5$, Ar-H), 3.58 (s, 2H, CH_2), 2.89 (s, 2H, - CH_2) 7.95 (s, 1H, NH)	537.92 (M-1, 17), 425.06 (6), 126.84 (100), 254.62 (37), 182.86 (52)
8	3085 (C-H), 2928 (CH_2), 1634 (C=N), 3421 (-NH)	6.56–6.58 (d, 4H, $J = 8.5$, Ar-H), 7.42–7.44 (d, 4H, $J = 8.5$, Ar-H), 3.38 (s, 2H, - CH_2 -N), 3.8 (s, C-NH), 5.98 (s, 4H, CH_2 imidazole ring), 8.2 (s, NH, imidazole)	435.72 (M-1, 6), 423.75 (20), 247.61 (50), 101.76 (100), 139.77 (37)
9	3223 (C-H), 2929 (CH_2), 1592 (C=N), 3360 (-NH)	6.58–6.69 (m, 4H, Ar-H), 7.1–7.5 (m, 4H, Ar-H), 3.30 (s, 2H, CH_2), 2.80 (s, 2H, CH_2), 8.0 (s, 1H, NH benzimidazole)	535.77 (M-1, 7), 252.77 (15), 148.89 (100), 122.88 (47)
10	3098 (C-H), 3005 (CH_2), 1602 (C=N), 3426 (-NH), 1424 (-N=N), 1225 (-N=N), 1102 and 1145 (tetrazole ring)	7.5–7.8 (m, 4H, Ar-H), 6.68–6.69 (d, 4H, $J = 18$, Ar-H), 3.38 (s, 2H, - CH_2), 2.8 (s, 2H, CH_2) 8.3 (s, 1H, NH) 5.7 (s, NH tetrazole)	440.63 (M^+ , 27), 260.60 (100), 247.61 (77), 300.52 (32)
13	3050 (C-H), 2921 (CH_2), 1529 (C=N), 3400 (-NH)	7.5 (s, 12H, Ar-H), 3.4 (s, 2H, CH_2), 2.8 (s, 2H, - CH_2)	478.76 (M+1, 5), 442.31 (25), 262.25 (74), 182.62 (100), 167.77 (63)
14	3005 (C-H), 2643 (CH_2), 1697 (C=N), 3417 (-NH)	3.4 (s, 2H, - CH_2), 2.8 (s, 2H, CH_2), 5.72 (s, 4H, - CH_2 imidazole ring), 8.1 (s, NH, imidazole), 7.5 (s, 8H, Ar-H)	374.56 (M-1, 9), 172.23 (100), 164.67 (42)
15	3057 (CH), 2924 (CH_2), 1625 (C=N), 3380 (-NH)	3.42 (s, 2H, CH_2), 2.89 (s, 2H, CH_2); 4.5 (s, 1H, -NH), 7.45 (s, 12H, Ar-H), 12.33 (s, 1H, -NH benzimidazole)	475.05 (M^+ , 5), 442.32 (16), 262.26 (37), 182.72 (67), 54.50 (100)
16	3102 (C-H), 2932 (CH_2), 1590 (C=N), 3397 (-NH), 1430 (-N=N), 1231 (-N=N), 1099 and 1169 (tetrazole ring)	2.5 (s, 2H, CH_2), 3.5 (s, 2H, CH_2), 4 (s, 1H, NH), 7.3–8.5 (s, 5H, Ar-H), 5.5 (s, 1H, NH tetrazole ring)	378.78 (M-1, 7), 262.26 (19), 247.30 (41), 182.72 (36), 167.79 (100), 121.98 (19)
19	3090 (C-H), 2926 (CH_2), 1639 (C=N), 3402 (-NH)	3.58 (s, 2H, CH_2), 2.89 (s, 2H, CH_2) 7.1–7.6 (m, 8H, Ar-H)	561.10 (M^+ , 10), 550.32 (8), 518.11 (15), 246.33 (100), 229.36 (25), 213.38 (17)

20	3099 (C-H), 2883 (CH ₂), 1631 (C=N), 3261 (-NH)	2.5 (s, 2H, CH ₂), 3.4 (s, 2H, CH ₂), 6 (s, 1H, NH), 7.9 (s, 1H, NH imidazole)	408.10 (M ⁺ , 7), 393.34 (5), 58.71 (100), 201.11 (72), 217.95 (52)
21	3164 (CH), 2930 (CH ₂), 1634 (C=N), 3411 (-NH)	2.5 (s, 2H, CH ₂), 3.4 (s, 2H, CH ₂), 7.5 (s, 12H, Ar-H), 12.53 (s, 1H, -NH benzimidazole)	558.10 (M ⁺ , 5), 63.13 (100), 175.29 (17), 145.73 (45)
22	3205 (C-H), 2939 (CH ₂), 1655 (C=N), 3408 (-NH), 1404 (-N=N), 1225 (-N=N=N), 1101 and 1174 (tetrazole ring)	2.5 (s, 2H, CH ₂), 3.4 (s, 2H, CH ₂), 7.5–8.0 (s, 5H, Ar-H), 5.5 (s, 1H, NH tetrazole ring)	414.13 (M ⁺ , 11), 70.83 (100), 251.43 (32), 373.20 (17), 110.75 (38), 96.81 (58)

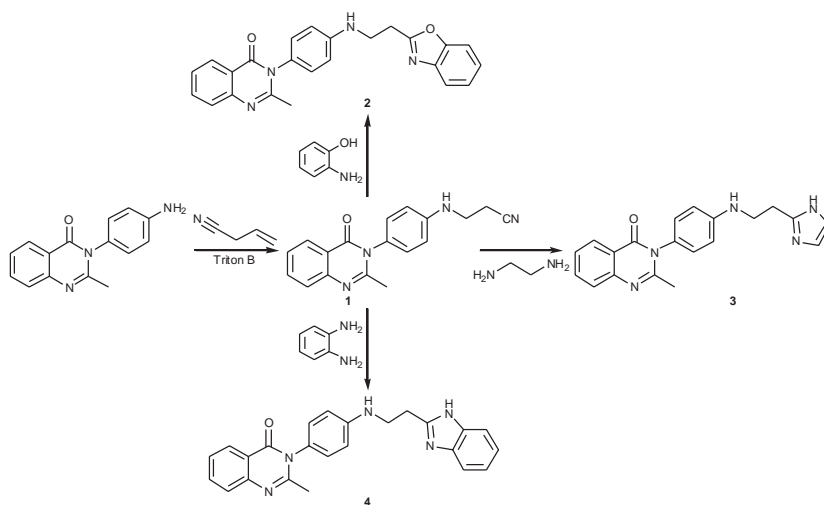
-4(3H)-one (4), 4,4'-sulfonylbis(*N*-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)aniline (9), *N*-bis(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (15), *N*-tris(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (21).

To a mixture of (3-(4-(2-methyl-4-oxoquinazolin-3(4*H*)-yl)phenylamino)propanenitrile (1) (0.01 mol, 2.13 g), 3,3'-(4,4'-sulfonylbis(4,1-phenylene)bis(azanediyl))dipropanenitrile (6) (0.01 mol, 3.54 g) or 3,3'-(6-phenyl-1,3,5-triazine-2,4-diyl)bis(azanediyl)dipropanenitrile (12) (0.01 mol, 2.93 g) and 3,3,3',1,3,5-triazine-2,4,6-triyl)tris(azanediyl)tripropanenitrile (18) (0.01 mol, 2.85 g) and *o*-aminophenol (0.01 mol, 1.09 g) or ethylenediamine (0.01 mol, 0.60 g) or *o*-phenylenediamine (0.01 mol, 1.08 g), 5 mL of conc. HCl was added and refluxed at 160 °C on an oil bath for ten hours. The hydrochloride was precipitated after cooling for four hours. The precipitate was filtered and washed with an ethanol/ether mixture (1:5), suspended in acetone and made alkaline with a strong ammonia solution. Excess of water was added to liberate the base as a solid. The solid was filtered, washed with water and recrystallized from methanol to get benzoxazole or imidazole or benzimidazole by reacting the nitriles with ethylenediamine and *o*-phenylenediamine, respectively.

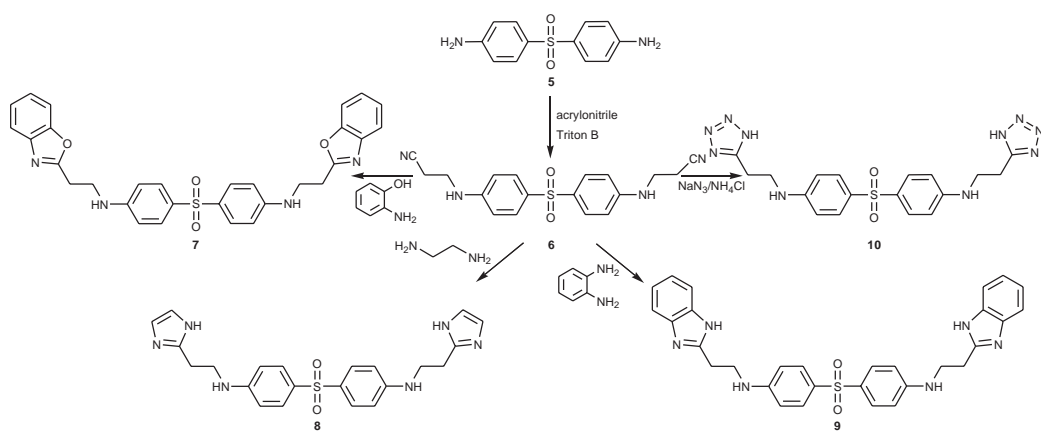
Syntheses of tetrazole (4,4'-sulfonylbis(N-(2-(1H-tetrazole-5-yl)ethyl)aniline) (10), N-bis(2-(1H-tetrazol-5-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (16) and N-tris(2-(1H-tetrazol-5-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (22).

To a mixture of compound (3-(4-(2-methyl-4-oxoquinazolin-3(4*H*)-yl)phenylamino)propanenitrile (1) (0.01 mol, 2.13 g), 3,3'-(4,4'-sulfonylbis(4,1-phenylene)bis(azanediyl))dipropanenitrile (6) (0.01 mol, 3.54 g), 3,3'-(6-phenyl-1,3,5-triazine-2,4-diyl)bis(azanediyl)dipropanenitrile (12) (0.01 mol, 2.93 g) or 3,3,3',1,3,5-triazine-2,4,6-triyl)tris(azanediyl)tripropanenitrile (18) (0.01 mol, 2.85 g), sodium azide (0.01 mol, 0.65 g), dimethylformamide (10 mL) and NH₄Cl (0.1 mol, 5.3 g) was placed in a 100-mL round-bottomed flask. The content was heated in oil bath for seven hours at 125 °C. The solvent was removed at reduced pressure. The reaction mixture was dissolved in 100 mL distilled water and carefully acidified with dil. HCl (1:1) (2 mL) to make solution of pH 2. The solution was cooled to 5 °C in an ice bath. The product was isolated by filtration, washed with several portions of water and dried. The crude product was recrystallized from DMF.

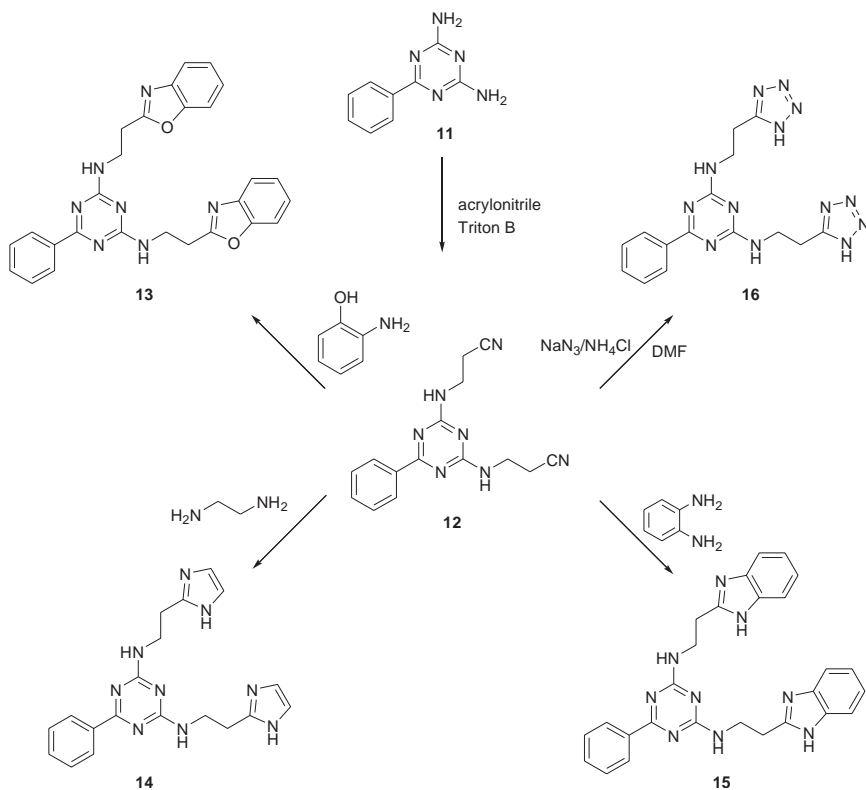
Physicochemical and spectral data for the synthesized compounds are given in Tables I and II. Synthetic routes are presented in Schemes 1–4.



Scheme 1.



Scheme 2.

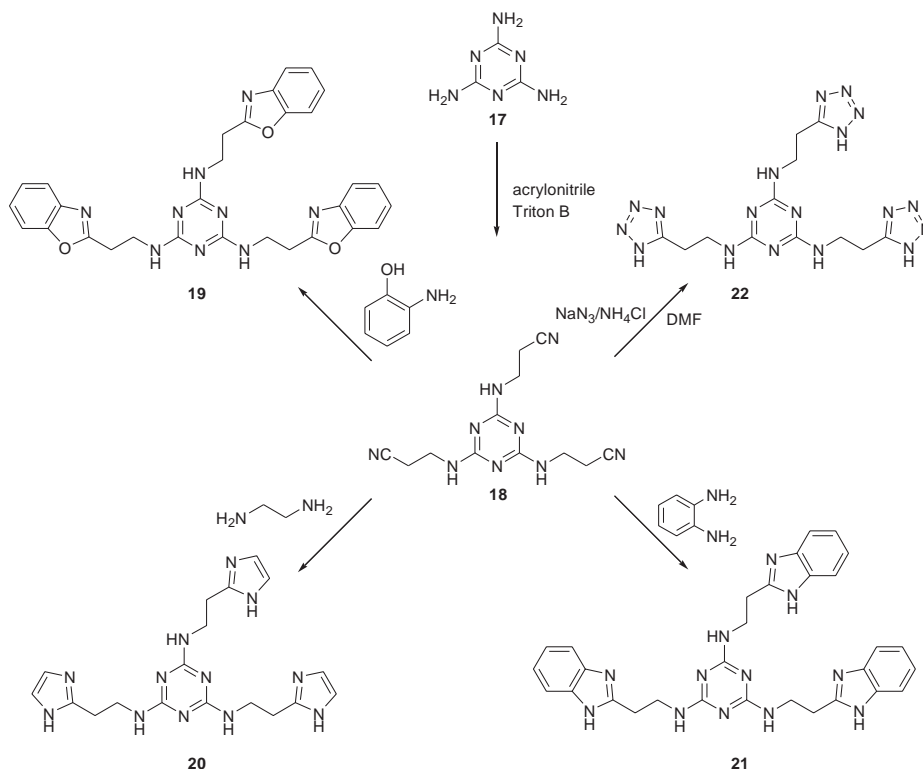


Scheme 3.

MTT assay

The tumor cell growth inhibition activities of the newly synthesized compounds were assessed *in vitro* on the human cancer cell line MCF-7 (breast cancer) and HT-29 (colon cancer) obtained from the National Centre for Cell Sciences (NCCS), Pune, India. 5-Fluorouracil (Merck, Germany) was used as a standard.

Cultured cells were trypsinized and resuspended in growth medium. The cells were seeded at 1×10^5 cells per well in a 96-well plate and incubated for 24 h in a CO₂ incubator. The cells were treated with different concentrations (1, 5, 10, 12.5, 25, 50, 75, 100, 125 and 250 $\mu\text{g mL}^{-1}$) of test compounds and then incubated for 24 h in the CO₂ incubator. To each well, 10 μL of MTT reagent was added and they were again incubated for 4 h at 37 °C. Each experiment was done in triplicate. The content of each well was transferred to a different Eppendorf tube and centrifuged at 2000 rpm for 10 minutes. The pellets obtained were dissolved by adding 200 μL of DMSO and were made up to 2 mL. Absorbance was measured at 550 nm and the % viability was calculated from the mean absorbance of the test compound in respect to the mean absorbance of the control.



Scheme 4.

The IC_{50} values were reported as mean \pm standard deviation of three independent experiments. One-way analysis of variance (ANOVA) and Student *t*-test were used to compare data using the Graph Pad Prism 5.0 software at a 95 % confidence limit.

RESULTS AND DISCUSSION

Chemistry

3-(4-(2-Methyl-4-oxoquinazolin-3(4*H*)-yl)phenylamino)propanenitrile (1) on treatment with *o*-aminophenol, ethylenediamine and *o*-phenylene diamine yielded the compounds 3-(4-(2-(benzo[*d*]oxazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (2), 3-(4-(2-((1*H*-imidazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (3) and 3-(4-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (4), respectively (Scheme 1). A similar reaction was carried out to synthesize compounds 7-9 (Scheme 2), 14-16 (Scheme 3) and 20-22 (Scheme 4) from compound 3,3'-(4,4'-sulfonylbis(4,1-phenylene)bis(azanediyl))dipropenenitrile (6), 3,3'-(6-phenyl-1,3,5-triazine-2,4-

-diyl)bis(azanediy) dipropanenitrile (**12**) and 3,3,3',1,3,5-triazine-2,4,6-triyl)tris(azanediy) tripropanenitrile (**18**) respectively. The structures of the compounds were confirmed by IR, ^1H NMR, mass spectra and elemental analysis.

Benzoxazoles (3-(4-(2-(benzo[*d*]oxazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (**2**), 4,4'-sulfonylbis(*N*-(2-(benzo[*d*]oxazol-2-yl)ethyl)aniline) (**7**), *N*-bis(2-(benzo[*d*]oxazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**13**) and *N*-tris(2-(benzo[*d*]oxazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (**19**) were prepared by treating compounds **1**, **6**, **12** and **18** with *o*-aminophenol in the presence of conc. HCl at 160 °C. In general, infrared spectral data of all benzoxazoles (**2**, **7**, **13**, **19**) revealed bands at 3350–3402 (–NH), 2921–2929 (CH₂), 1507–1639 (C=N) and 3050–3306 cm⁻¹ (CH). The ^1H NMR spectrum of the compounds showed a singlet at δ 3.4–3.58 ppm which can be assigned to methylene protons. A multiplet at δ 7.1–7.6 ppm was attributed to aromatic protons. A singlet was observed for the –NH proton at δ 7.95 ppm. Compound **2** showed a singlet at δ 2.08 ppm for CH₃ protons. Molecular ion peaks were observed at *m/z* 395.54, 537.92, 478.76 and 561.10 for compounds **2**, **7**, **13**, **19**, respectively.

Imidazoles (3-(4-(2-(1*H*-imidazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (**3**), 4,4'-sulfonylbis(*N*-(1*H*-imidazol-2-yl)ethylaniline) (**8**), *N*-bis(2-(1*H*-imidazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**14**), and *N*-tris(2-(1*H*-imidazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (**20**) were prepared by treating compounds **1**, **6**, **12** and **18** with ethylenediamine in the presence of conc. HCl at 160 °C. The molecular ion peaks observed at *m/z* 346.78, 435.72, 374.56 and 408.10 for compounds **3**, **8**, **14** and **20** confirm the molecular mass of the compounds. The IR spectrum of imidazoles (**3**, **8**, **14** and **20**) showed characteristic bands at 3050–3223 (CH), 2921–2926 (CH₂), 1507–1639 (C=N) and 3315–3402 cm⁻¹ (–NH). The ^1H NMR spectra of all imidazole compounds showed singlets at δ 2.5 and 3.4 ppm which can be assigned to methylene protons. Compound **3** showed a multiplet at δ 6.7–7.5 ppm, which may be assigned to aromatic protons. The three singlets appearing at δ 8.2, 3.37 and 2.48 ppm are attributed to NH (imidazole), methylene and methyl protons, respectively.

Benzimidazoles (3-(4-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethylamino)phenyl)-2-methylquinazolin-4(3*H*)-one (**4**), 4,4'-sulfonylbis(*N*-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)aniline) (**9**), *N*-bis(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**15**), *N*-tris(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (**21**) were prepared by treating compounds **1**, **6**, **12** and **18** with *o*-phenylenediamine in the presence of conc. HCl at 160 °C. Molecular ion peaks were obtained at *m/z* 394.54, 535.77, 475.05 and 558.10 for compounds **4**, **9**, **15** and **21**, respectively. The IR spectra of benzimidazoles showed characteristic bands at 3055–3223 (CH), 2924–2930 (CH₂), 1592–1634 (C=N) and 3315–3411 cm⁻¹ (–NH). The ^1H NMR spectrum of all the four benzimidazole compounds showed singlets at δ 3.3 and δ 2.9 ppm, which can be assigned to methylene protons. The strong peaks observed as a singlet at δ 12.33 to 12.53 ppm are due to NH proton of the benzimidazole ring of compounds **15** and **21**, respectively. Compound **4** showed a multiplet at δ 6–6.4 ppm, which can be assigned to aromatic protons. The methyl protons attached to the quinazoline ring showed a singlet at δ 2.48 ppm. The singlet appearing at δ 8.1 ppm can be attributed to NH (benzimidazole) protons.

Compounds 4,4'-sulfonylbis(*N*-(2-(1*H*-tetrazol-5-yl)ethyl)aniline) (**10**), *N*-bis(2-(1*H*-tetrazol-5-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**16**), and *N*-tris(2-(1*H*-tetrazol-5-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (**22**) were prepared by treating 3,3'-(4,4'-sulfonyl-

bis(4,1-phenylene)bis(azanediyl)dipropanenitrile (**6**), 3,3'-(6-phenyl-1,3,5-triazine-2,4-diyl)bis(azanediyl)dipropanenitrile (**12**) and 3,3,3',1,3,5-triazine-2,4,6-triyl)tris(azanediyl)tripropanenitrile (**18**) with NaN_3 and NH_4Cl (Schemes 2-4). The infrared spectral data of all the three tetrazoles (**10**, **16** and **22**) revealed bands at μ 3397–3426 ($-\text{NH}$), 2932–3005 (CH_2), 1590–1655 ($\text{C}=\text{N}$), 3098–3205 (CH), 1404–1430 ($\text{N}=\text{N}$), 1225–1231 ($\text{N}-\text{N}=\text{N}$) and 1099–1174 (tetrazole ring) cm^{-1} . The ^1H NMR spectrum of the compounds showed a singlet at δ 2.5–2.8 and 3.38–3.5 ppm, which can be assigned to methylene protons. The singlet observed at δ 5.5–5.7 ppm was due to the $-\text{NH}$ protons of the tetrazole ring. Molecular ion peaks obtained at m/z 440.63, 378.78 and 414.13 for compounds **10**, **16** and **22** confirm the molecular weight of the compounds.

Cytotoxicity screening

Induction of cell death or inhibition of cell proliferation is an important property for chemotherapeutic agents. The cytotoxic effect of compounds **2–4**, **7–10**, **13–16** and **19–22** was evaluated by measuring the level of cell proliferation after incubation of the cells with the test samples using the MTT colorimetric assay (10). The results are expressed as percentage of cell proliferation compared to the cell control (cells treated with vehicle,

Table III. Cytotoxic activity of newly synthesized compounds

Compd.	IC_{50} ($\mu\text{g mL}^{-1}$)	
	MCF-7	HT-29
2	22.5	21.7
3	11.0	18.2
4	16.9	19.5
7	17.9	19.3
8	15.6	18.7
9	5.7	5.2
10	37.0	52.7
13	6.12	5
14	14.8	40.3
15	7.2	5.9
16	102.2	113.8
19	12.5	15.6
20	19.2	22.5
21	6.0	5.5
22	40.1	60.2
5 FU	6.2	5.5

IC_{50} – concentration required to inhibit cell viability by 50 %:
Control: DMSO.

DMSO 0.1 %). The results of IC_{50} values of MCF-7 (breast cancer) and HT-29 (colon cancer) of test compounds are presented in Table III.

It is evident from the study that all the heterocyclic compounds synthesized in the present work had a cytotoxic effect on the tested cell lines. It is worth noting that compounds **9**, **13**, **15** and **21** exhibit IC_{50} values comparable to the standard drug 5-fluorouracil. From the results of the study, it is inferred that among the new heterocyclic compounds synthesized in the present research work, compounds with benzimidazole and benzoxazole moieties exhibit good cytotoxic activity. It is interesting to note that among the four benzimidazole compounds synthesized (**4**, **9**, **15** and **21**), three compounds, *viz.*, 4,4'-sulfonylbis(*N*-(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)aniline (**9**), *N*-bis(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**15**) and *N*-tris(2-(1*H*-benzo[*d*]imidazol-2-yl)ethyl)-1,3,5-triazine-2,4,6-triamine (**21**) exhibit potent cytotoxicity with IC_{50} values of 5.7 (MCF-7), 5.2 (HT-29), 7.22 (MCF-7), 5.9 (HT-29) 6.0 $\mu\text{g mL}^{-1}$ (MCF-7), and 5.5 $\mu\text{g mL}^{-1}$ (HT-29), respectively. Among the four benzoxazole derivatives (**2**, **7**, **13** and **19**), *N*-bis(2-(benzo[*d*]oxazol-2-yl)ethyl)-6-phenyl-1,3,5-triazine-2,4-diamine (**13**), shows potent cytotoxic activity against the two cancer cell lines taken for the study. The least potency was shown by compounds **10**, **16** and **22** containing tetrazole moiety.

From the data of cytotoxic activity, compounds **9**, **15** and **21** were found to possess promising cytotoxic activity. This activity of the compounds may be due to the presence of dapsons, triazine and benzimidazole moieties. The remarkable antiproliferative activity of compound **13** may be due to the presence of triazine and benzoxazole moieties.

CONCLUSIONS

In the present study, syntheses of some new benzimidazoles, benzoxazoles, imidazoles and tetrazoles is described. All the compounds were evaluated for cytotoxic activity. Benzimidazoles **9**, **13**, **15** and **21** and benzoxazole **13** were found to be the most active compounds with a promising cytotoxic activity. It can be deduced from the results that benzimidazole and benzoxazole moiety can create potent cytotoxic compounds, whereas imidazole and tetrazole compounds show lower cytotoxic effects. Further studies and structural modifications are needed to increase the cytotoxic activity.

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