Dedicated to Prof. dr. LJUDEVIT ILIJANIČ on the occasion of his 70th birthday.

# Chemical composition of the plants of loess steppes in Hungary

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The first two stages of the successional series on loess (Agropyro-Kochietum, Salvio-Festucetum rupicolae) differ from each other in their floristic composition, the total element content of their soils and plant species as well as their water demand. Chrysopogon gryllus, a characteristic species of the loess steppe, belongs to the calciotroph physiotype according to the Ca/K ratio in its leaves. The leaves of this species contain relatively high concentrations of Ca and Mg and a low concentration of N. The high C/N ratio in the leaves and roots of Chrysopogon gryllus should indicate that the mineralization dynamics is slowed down in the stands dominated by this species. Barbula cordata, a moss species of the Salvio-Festucetum rupicolae community, accumulates very high amounts of heavy metals from its surroundings. In the Agropyro-Kochietum community, Bromus species accumulated the highest amounts of heavy metals in their leaves.

**Key-words:** semidesert, loess, steppe, grassland, plant community, chemical pomposition, *Chrysopogon gryllus*, Hungary

# Introduction

Characteristic plant communities of Hungary are the semidesert community on loess walls (*Agropyro-Kochietum*) and the loess steppe grassland (*Salvio-Festucetum rupicolae*) developing on chernozem or calcic chernozem soils. The floristic composition of these communities was described by ZÓLYOMI (1958) and ZÓLYOMI and FEKETE (1994).

On steep loess walls, the post-glacial, xerotherm semidesert community (*Agropyro-Kochietum*) has developed with a relict character. Its field layer has a 60–70% coverage. In addition to annuals it is composed of perennial herbs,

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shrubs and undershrubs in a significant number. Its floristic composition is relatively stable, having *Agropyron pectiniforme* and *Bassia prostrata* as characteristic species. The dynamics of the community is rather slow and on steep loess walls it seems to have been stable for thousands of years (MOLNÁR 1999).

If this community develops on plain loess areas, it can be considered the first stage of a successional series. In this case the second stage of the successional series is loess steppe grassland (Salvio-Festucetum rupicolae) with an 80–90% field layer coverage and the frequent presence of the species Elymus hispidus and Festuca rupicola. Recently, in some places Chrysopogon gryllus has been spreading. Nowadays, characteristic loess steppe grasslands can be found already on very limited areas, several of which have become degraded.

## Materials and methods

The investigation of the chemical element composition of the plant components of loess steppes in Hungary was carried out under the organization of the International Association for Ecology (INTECOL) and the International Union of Biological Sciences (IUBS) in the context of the international research program «Element Concentration Cadasters in Ecosystems» (ECCE).

The frequent and characteristic species of the two plant communities as well as the soil profiles belonging to them, were sampled at the border of the village Érd, to the south of Budapest. Sampling was carried out according to the prescriptions of the international program (Lieth and Markert 1985, 1988; Markert 1986). In addition, sampling of soils and plants in grassland areas dominated by *Festuca valesiaca* and *Chrysopogon gryllus* was carried out in 8 patches, each having the same soil type and spread over several square meters in contact with each other.

The following species of the Agropyro-Kochietum community were subjected to multielement analysis: Ephedra distachya, Artemisia austriaca, Erodium cicutarium, Bassia prostrata, Linaria genistifolia, Sedum telephium subsp. maximum, Xeranthemum annuum, Iris pumila, Agropyron pectiniforme, Elymus hispidus, Bromus squarrosus, B. tectorum, Festuca valesiaca and Stipa capillata. Characteristic species of the communities are underlined.

The species of the Salvio-Festucetum rupicolae community investigated were as follows: Ephedra distachya, Achillea pannonica, Artemisia campestris, Aster linosyris, Brassica elongata. Cephalaria transsylvanica, Chrysopogon gryllus, Chamaecytisus austriacus, Nepeta nuda. Peucedanum alsaticum, Phlomis tuberosa. Plantago media, Potentilla incana, Salvia nemorosa, Thymus odoratissimus, Allium sphaerocephalon, Agropyron pectiniforme, Elymus hispidus, Festuca rupicola, Festuca valesiaca and Barbula cordata.

Following separation of the different organs (roots, stems and leaves) of the plants, they were subjected to a thorough tap-water wash. Samples were dried at 70°C. Ashing was carried out with nitric acid and hydrogen peroxide in teflon-bombs. With an ICP-AES, at the Department of Chemistry of the Horticultural University in Budapest, the amounts of the following elements in the soil and plant samples were established: Al, As, B. Ba, Ca, Cd, Co, Cr, Cu, Fe, Ga, K,

Li, Mg, Mn, Mo, Na, Ni, P, Pb, Se, Si, Sr, Ti, V and Zn. Minimal detection limits of the separate elements were in  $\mu g$  g $^{-1}$  as follows: Al: 0.0015; As: 0.0003; B: 0.0003; Ba: 0.0001; Ca: 0.0010; Cd: 0.0003; Co: 0.0006; Cr: 0.0005; Cu: 0.0003; Fe: 0.0003; Ga: 0.0024; K: 0.0050; Li: 0.0020; Mg: 0.0015; Mn: 0.0001; Mo: 0.0002; Na: 0.0500; Ni: 0.0003; P: 0.0750; Pb: 0.0025; Se: 0.0030; Si: 0.0025; Sr: 0.0001; Ti: 0.0002; V: 0.0002; Zn: 0.0001. The overall amount of these elements in the plant samples is mentioned in the present work as «total element content«. The total nitrogen content was measured with the standard Kjehldahl method. The amount of this element was not added to the «total element content«. The present evaluation is based chiefly on the results of the leaf investigations. All data are presented in  $\mu g$  g $^{-1}$  dry weight.

# Results and discussion

The Agropyro-Kochietum and Salvio-Festucetum rupicolae communities are situated on skeleton and chernozem soils, respectively, both formed on loess. The total element contents of the uppermost soil layers of both communities are high. The uppermost 10 cm soil layer is particularly rich in Ca (Tab. 1, 2).

**Tab. 1.** Chemical composition (in μg g<sup>-1</sup>) of the soil of the *Agropyro-Kochietum* community

| Soil layers in cm | 0-10  | 10–20   | 20-30   | 30-40   | 40-50   | 5060                |
|-------------------|---|---|---|---|---|---------------------|
| Al                | 4525  | 4316  | 5106  | 4553  | 6183  | 5048                |
| As                | <d *< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><di*< td=""></di*<></td></dl*<></td></dl*<></td></dl*<></td></dl*<></td></d *<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><di*< td=""></di*<></td></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><di*< td=""></di*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><di*< td=""></di*<></td></dl*<></td></dl*<> | <dl*< td=""><td><di*< td=""></di*<></td></dl*<> | <di*< td=""></di*<> |
| В                 | 2.0   | <d!*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<></td></d!*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""></dl*<></td></dl*<> | <dl*< td=""></dl*<> |
| Ba                | 64.4  | 53.9  | 52.7  | 51.6  | 56.8  | 61.7                |
| Ca                | 68264   | 8614  | 8061  | 9147  | 9521  | 10252               |
| Cd                | 1.3   | 1.1   | 1 2   | 1.1   | 1.4   | 1.2                 |
| Со                | 5.1   | 4.9   | 5.4   | 4.9   | 5.9   | 5.4                 |
| r)                | 8.6   | 7.8   | 2.3   | 8.2   | 11.0  | 9.3                 |
| Си                | 17.8  | 12.8  | 16.2  | 11.1  | 12.4  | 11.0                |
| Fe                | 5208  | 5014  | 5892  | 5271  | 7320  | 5921                |
| Ga                | 5.9   | 5.1   | 5.6   | 5.2   | 6.4   | 5.5                 |
| К                 | 1668  | 1451  | 1450  | 1138  | 1203  | 969                 |
| li li             | 9.7   | 10.9  | 12.5  | 11.9  | 16.4  | 14.2                |
| Mg                | 7175  | 7151  | 7630  | 7661  | 8628  | 8368                |
| Mn                | 452   | 355   | 372   | 323   | 348   | 355                 |
| Mo                | 0.1   | 0.02  | <d *< td=""><td><d *< td=""><td><dl*< td=""><td><di*< td=""></di*<></td></dl*<></td></d *<></td></d *<> | <d *< td=""><td><dl*< td=""><td><di*< td=""></di*<></td></dl*<></td></d *<> | <dl*< td=""><td><di*< td=""></di*<></td></dl*<> | <di*< td=""></di*<> |
| Na                | 61.2  | 51.9  | 53.4  | 56.6  | 62.8  | 66.2                |
| Ni                | 21.2  | 17.6  | 19.6  | 18.4  | 22.0  | 19.5                |
| P                 | 1165  | 648   | 561   | 503   | 554   | 497                 |
| РЬ                | 25.6  | 9.1   | 11.1  | 7.4   | 7.3   | 7.2                 |
| Se                | <d!*< td=""><td><d *< td=""><td><dl*< td=""><td><d *< td=""><td><dl*< td=""><td><d *< td=""></d *<></td></dl*<></td></d *<></td></dl*<></td></d *<></td></d!*<> | <d *< td=""><td><dl*< td=""><td><d *< td=""><td><dl*< td=""><td><d *< td=""></d *<></td></dl*<></td></d *<></td></dl*<></td></d *<> | <dl*< td=""><td><d *< td=""><td><dl*< td=""><td><d *< td=""></d *<></td></dl*<></td></d *<></td></dl*<> | <d *< td=""><td><dl*< td=""><td><d *< td=""></d *<></td></dl*<></td></d *<> | <dl*< td=""><td><d *< td=""></d *<></td></dl*<> | <d *< td=""></d *<> |
| Si                | 1053  | 822   | 925   | 776   | 918   | 825                 |
| Sr                | 95.6  | 112   | 115   | 12  | 128   | 142                 |
| Ti                | 22.1  | 13.2  | 15.7  | 11.9  | 13.7  | 11.7                |
| V                 | 18.9  | 7.5   | 7.9   | 6.9   | 8.2   | 7.2                 |
| Zn                | 71.9  | 33.7  | 40.4  | 30.6  | 38.0  | 32.6                |

< dl\*: element concentration under the detection limit of the instrument

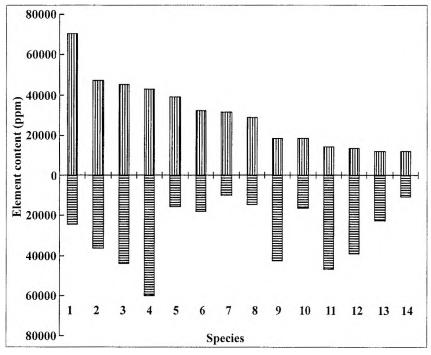
**Tab. 2.** Chemical composition (in  $\mu$ g g<sup>-1</sup>) of the soil of the *Salvio-Festucetum* community

| Soil layers in cm | 0-10  | 10-20   | 20-30   | 30-40   | 4050  | 5060                |
|-------------------|---|---|---|---|---|---------------------|
| Al                | 3762  | 3492  | 4761  | 3517  | 7433  | 3423                |
| As                | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""></dl*<></td></dl*<> | <dl*< td=""></dl*<> |
| В                 | 0.21  | 0.08  | <dl*< td=""><td><dl*< td=""><td><d *< td=""><td>&lt; d *</td></d *<></td></dl*<></td></dl*<>            | <dl*< td=""><td><d *< td=""><td>&lt; d *</td></d *<></td></dl*<>            | <d *< td=""><td>&lt; d *</td></d *<>            | < d *               |
| Ba                | 43.4  | 46.7  | 44 3  | 41.9  | 45.2  | 48.4                |
| Ca                | 65343   | 5832  | 5894  | 9847  | 10371   | 10593               |
| Cd                | 0.95  | 0.96  | 1.2   | 0.88  | 0.93  | 0.86                |
| Co                | 5.1   | 4.9   | 5.7   | 5.2   | 5.4   | 5.3                 |
| Ct                | 7.0   | 6.8   | 9.1   | 6.9   | 7.1   | 6.6                 |
| Cu<br>Cu          | 8.9   | 9.2   | 8.2   | 5.9   | 5.3   | 5.1                 |
| Fe                | 4547  | 4314  | 5855  | 4832  | 4817  | 4424                |
| Ga                | 5.1   | 4.9   | 5.7   | 4.6   | 4.9   | 4.5                 |
| K                 | 606   | 705   | 654   | 497   | 493   | 471                 |
| i<br>Li           | 7.8   | 7.6   | 9.8   | 8.3   | 9.3   | 8.7                 |
| Mg                | 6505  | 6835  | 7072  | 6889  | 7930  | 7771                |
| Mn<br>Mn          | 292   | 284   | 307   | 271   | 277   | 276                 |
| Mo                | <dl*< td=""><td>0.2</td><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<></td></dl*<>                 | 0.2   | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""></dl*<></td></dl*<> | <dl*< td=""></dl*<> |
| Na                | 30.4  | 35.9  | 36.3  | 36.3  | 39.4  | 42.8                |
| Ni                | 16.5  | 17.6  | 18.3  | 16.2  | 17.2  | 16.4                |
| P                 | 574   | 587   | 591   | 516   | 563   | 634                 |
| Pb                | 11.0  | 16.9  | 7.8   | 5.7   | 5.0   | 4.9                 |
| Se                | <di*< td=""><td><di*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<></td></di*<></td></di*<> | <di*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<></td></di*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""><td><dl*< td=""></dl*<></td></dl*<></td></dl*<> | <dl*< td=""><td><dl*< td=""></dl*<></td></dl*<> | <dl*< td=""></dl*<> |
| Si                | 962   | 967   | 1182  | 878   | 862   | 809                 |
| Sr                | 702   | 79.8  | 77.5  | 87.4  | 99.5  | 105                 |
| Ti                | 16.5  | 17.3  | 18.7  | 13.7  | 12.7  | 11.9                |
| V                 | 10.9  | 7.6   | 7.5   | 5.8   | 6.0   | 5.6                 |
| v<br>Zn           | 9.5   | 29.7  | 28.7  | 22.6  | 22.9  | 22.1                |
| - LII             |   | L1.1  | 20.1  | 22.0  | 44.7  | LL.I                |

< di\*: element cancentration under the detection limit of the instrument

The soil profile of the *Agropyro-Kochietum* community contains more Al, Cu, Fe, K, Li, Mg, Mn, Na and Sr, than the same layer of the *Salvio-Festucetum rupicolae* community. The heavy metals are of geogenic origin, although the elevated Pb and Zn contents of the upper layers suggest that some anthropogenic impacts also contributed to the element composition of these soils. The upper soil layers contain only low amounts (0.2–0.4%) of nitrogen.

The plant species of the two communities can be well separated on the basis of the total element contents of their leaves. The species of the semidesert community on loess walls, which is considered to be the first stadium of the successional series, contain less total amounts of elements (Fig. 1) than those of the loess steppe grassland (Fig. 2). The total element content is lowest in the annual grass species, which are followed, in order of increasing element contents, by the perennial grasses as well as dicotyledons (Figs. 1, 2). The leaves of the gymnosperm dwarf shrub *Ephedra distachya* contain moderate amounts of elements. The chemical constitutions of *Agropyron pectiniforme* and *Bassia prostrata* are significantly different; the leaves of the latter contain much more Ca, K, Mg, Fe and P, than those of the first. The higher Fe content of the skeleton soil is also indicated by the investigated exemplars of *Bassia prostrata* as well as of *Agropyron pectiniforme* occurring in the *Agropyro-Kochietum* community (Tab. 3).



**Fig. 1.** Total element contents in the leaves (above the abscissa) and roots (under the abscissa) of the species of an *Agropyro-Kochietum* plant community in Hungary

5. Ephedra distachya

#### Dicotyledonopsida

- 1. Sedum telephium subsp. maximum
- 2. Bassia prostrata
- 4, Erodium cicutarium
- 6. Artemisia austriaca
- 7. Xeranthemum annuum
- 8. Linaria genistifolia

# Monocotyledonopsida

- 3. Iris pumila
- 9. Elymus hispidus
- 10. Stipa capillata
- 11. Festuca valesiaca
- 12. Agropyron pectiniforme
- 13. Bromus squarrosus
- 14. Bromus tectorum

Bromus species accumulated in their leaves the highest amounts of heavy metals amongst the investigated species in the Agropyro-Kochietum community (Fig. 3), whereas in the Salvio-Festucetum rupicolae community, besides Barbula cordata, a moss species, Thymus odoratissimus, Potentilla incana and Aster linosyris were the best accumulators of these elements (Fig. 4).

The total element content of the leaves of *Sedum telephium* subsp. *maximum* (70345  $\mu g g^{-1}$ ) is the highest amongst all the investigated species as a consequence of the large amount of Ca (44969  $\mu g g^{-1}$ ) accumulated by this plant. The K/Ca ratios within the leaves of the plants indicate that the wooded steppe plants (e.g. *Nepeta nuda* with a K/Ca ratio of 0.41 and *Phlomis tuberosa* with a K/Ca ratio of 0.74) belong to the calciotroph physiotype (KINZEL 1982).

**Tab. 3.** Chemical composition (in µg g<sup>-1</sup>, in the case of N in %) of the leaves of Agropyron pectiniforme and Bassia prostrata

|       | Agropyron pectiniforme<br>(Agropyro-Kochietum)               | Agropyron pectiniforme<br>(Salvio-Festucetum rupicolae) | Bassia<br>prostrata |
|-------|--|---|---------------------|
| Al    | 172  | 137   | 241                 |
| As    | <d *< td=""><td>&lt; d *</td><td>&lt; d *</td></d *<>        | < d *   | < d *               |
| В     | 40.5   | 6.7   | 85.8                |
| Ba    | 6.2  | 5.8   | 15 2                |
| Ca    | 4816   | 4584  | 18851               |
| Cd    | 0.14   | < d!*   | 0.44                |
| Со    | 0.17   | 0.23  | 0.66                |
| Cr    | 1.1  | 1.1   | 2.3                 |
| Cu    | 3.6  | 6.3   | 6.9                 |
| Fe    | 165  | 107   | 1141                |
| Ga    | 1.9  | 0.16  | 3.1                 |
| K     | 6443   | 6707  | 18288               |
| lı lı | 8.5  | 4.0   | 14.3                |
| Mg    | 718  | 640   | 4543                |
| Mn    | 65.9   | 31.8  | 230                 |
| Mo    | <dl*< td=""><td>0.21</td><td><d\*< td=""></d\*<></td></dl*<> | 0.21  | <d\*< td=""></d\*<> |
| Na    | 53.5   | 40.8  | 109                 |
| Ni    | 2.6  | 1.7   | 5.8                 |
| Р     | 679  | 549   | 1004                |
| Pb    | 3.0  | 1.7   | 6.7                 |
| Se    | 0.44   | <d *< td=""><td><d *< td=""></d *<></td></d *<>         | <d *< td=""></d *<> |
| Si    | 162  | 176   | 139                 |
| Sr    | 22.5   | 9.9   | 46.0                |
| Ti    | 3.9  | <d *< td=""><td>22.8</td></d *<>                        | 22.8                |
| V     | 1.5  | 0.86  | 4.9                 |
| Zn    | 30.2   | 55.8  | 51.2                |
| N%    | 1.32   | 1.32  | 1 65                |

<sup>&</sup>lt;d|\* element concentration under the detection limit of the instrument

As a consequence of the relatively low N concentration in the soils the N content of the species is between 0.73% (*Iris pumila*) and 2.18% (*Erodium cicuta-rium*) in the *Agropyro-Kochietum*, and between 0.79% (*Plantago media*) and 2.57% (*Chamaecytisus austriacus*) in the *Salvio-Festucetum rupicolae* communities, respectively.

The total element contents in the roots of the investigated species of the Agropyro-Kochietum community vary within a relatively wide range (9870 µg g<sup>-1</sup> in the case of Xeranthemum annuum, and  $60172 \, \mu g \, g^{-1}$  in the case of Erodium cicutarium). This phenomenon is presumably in connection with the openness of the community, its low number of species, as well as the absence of concurrence amongst the roots.

In the loess steppe grassland, where the concurrence amongst the roots is relatively strong, the total element contents of the roots vary within a narrower range (with the extremes of 12895  $\mu g$  g<sup>-1</sup> in *Chamaecytisus austriacus* and 41641  $\mu g$  g<sup>-1</sup> in *Agropyron pectiniforme*). In both communities there is a con-

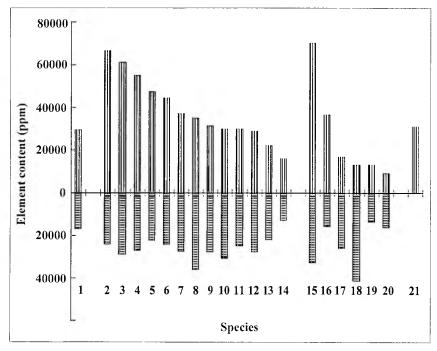


Fig. 2. Total element contents in the leaves (above the abscissa) and roots (under the abscissa) of the species of a *Salvio-Festucetum rupicolae* plant community in Hungary

1. Ephedra distachya

#### Dicotyledonopsida

- 2. Nepeta nuda
- 3. Plantago media
- 4. Peucedanum alsaticum
- 5. Phlomis tuberosa
- 6. Aster linosyris
- 7. Artemisia campestris
- 8. Cephalaria transsylvanica
- 9. Thymus odoratissimus
- 10. Potentilla incana
- 11. Achillea pannonica

- 12. Salvia nemorosa
- 13. Brassica elongata
- 14. Chamaecytisus austriacus

## Monocotyledonopsida

- 15. Festuca rupicola
- 16. Allium sphaerocephalon
- 17. Elymus hispidus
- 18. Agropyron pectiniforme
- 19. Festuca valesiaca

# Briophyta

20. Barbula cordata

spicuous accumulation of relatively high amounts of heavy metals by the roots of some grass species (Figs. 3, 4) as compared with their total element contents (Figs. 1, 2).

The characteristic species of the loess steppe grassland contain in their leaves higher amounts of elements (Tab. 4) than those of the semidesert community on loess walls, first of all as a consequence of their higher Ca and Mg contents. The chemical composition of the two woody steppe species (*Nepeta nuda*, *Phlomis tuberosa*) seems to be similar to that of those exemplars investigated previously

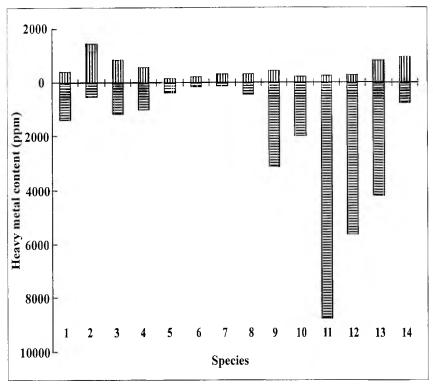


Fig. 3. Heavy metal contents in the leaves (above the abscissa) and roots (under the abscissa) of the species of an Agropyro-Kochietum plant community in Hungary

5. Ephedra distachya

#### Dicotyledonopsida

- 1. Sedum telephium subsp. maximum
- 2. Bassia prostrata
- 4. Erodium cicutarium
- 6. Artemisia austriaca
- $7.\ Xeranthemum\ annuum$
- 8. Linaria genistifolia

## Monocotyledonopsida

- 3. Iris pumila
- 9. Elymus hispidus
- 10. Stipa capillata
- 11. Festuca valesiaca
- 12. Agropyron pectiniforme
- 13. Bromus squarrosus
- 14. Bromus tectorum

in the Aceri tatarico-Quercetum community (climax stage of the successional series) on a chernozem brown forest soil (Kovács et al. 1993).

The higher total element content in species of the Salvio-Festucetum rupicolae community than in those of the previous successional stage (Figs. 1, 2) can be explained by the findings of ZÓLYOMI and FEKETE (1994), who established that the species of the Agropyro-Kochietum and Salvio-Festucetum rupicolae communities are different in respect of their W- (water requirement) values. A favourable water supply may contribute to the accumulation of more elements by the plants. Consequently, it seems that plant species constituting the different

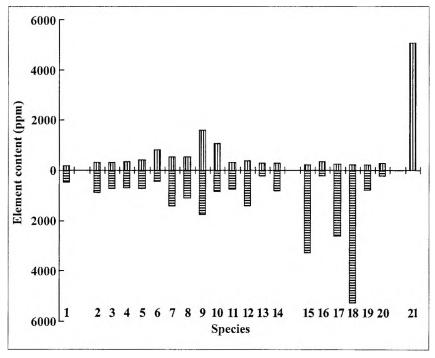


Fig. 4. Heavy metal contents in the leaves (above the abscissa) and roots (under the abscissa) of the species of a *Salvio-Festucetum rupicolae* plant community in Hungary

1. Ephedra distachya

#### Dicotyledonopsida

- 2. Nepeta nuda
- 3. Plantago media
- 4. Peucedanum alsaticum
- 5. Phlomis tuberosa
- 6. Aster linosyris
- 7. Artemisia campestris
- 8. Cephalaria transsylvanica
- 9. Thymus odoratissimus
- 10. Potentilla incana
- 11. Achillea pannonica

- 12. Salvia nemorosa
- 13. Brassica elongata
- 14. Chamaecytisus austriacus

#### Monocotyledonopsida

- 15. Festuca rupicola
- 16. Allium sphaerocephalon
- 17. Elymus hispidus
- 18. Agropyron pectiniforme
- 19. Festuca valesiaca

#### Briophyta

20. Barbula cordata

stages of the successional series are, besides their ecological demands, different in their chemical composition.

Soils in Hungary are very poor in Se. However, this element can be detected in small concentrations (0.008–1.405  $\mu$ g g<sup>-1</sup>) in the stems and leaves of the species of some plant communities occurring on sand or loess (Kovács et al. 1998).

Barbula cordata is a moss species growing on the soil surface of loess steppe grasslands. The total element content of the moss is higher than that of the inves-

Tab. 4. Chemical composition (in μg g<sup>-1</sup>) of the leaves of some characteristic species of the *Salvio-Festucetum rupicolae* community

|    | Brassica elongata | Nepeta nuda   | Phlomis tuberosa  | Salvia nemorosa                                 | Barbula cordata     |
|----|-------------------|---|---|---|---------------------|
| Al | 157               | 191   | 256   | 294   | 3380                |
| As | < d *             | $< dI^*$  | 1.0   | <dl*< td=""><td><dl*< td=""></dl*<></td></dl*<> | <dl*< td=""></dl*<> |
| В  | 29.0              | 33.1  | 39 <b>9</b>   | 65 0  | 25 8                |
| Ва | 11.8              | 87.7  | 90 0  | 31 5  | 40.8                |
| Ca | 26349             | 41945   | 23045   | 18021   | 15359               |
| Cd | 0.11              | 10.0  | 0 24  | 0 07  | 1.1                 |
| Со | 0.08              | 0.02  | 0.45  | < d *   | 2 9                 |
| (r | 0.47              | <di*< td=""><td>1.5</td><td>0.33</td><td>7 7</td></di*<>  | 1.5   | 0.33  | 7 7                 |
| Си | 7.4               | 5.4   | 6.1   | 5.1   | 142                 |
| Fe | 151               | 221   | 308   | 284   | 4642                |
| Ga | 0.24              | 1.9   | 5.4   | 0.5   | 7.2                 |
| K  | 38964             | 17010   | 17163   | 20904   | 2507                |
| li | 4.6               | 2.6   | 3.2   | 16  | 8 3                 |
| Mg | 2698              | 5737  | 4079  | 5148  | 2879                |
| Mn | 36.9              | 35.9  | 54 1  | 39.0  | 186                 |
| Mo | 0.19              | <d *< td=""><td><dl*< td=""><td><dl*< td=""><td><d *< td=""></d *<></td></dl*<></td></dl*<></td></d *<> | <dl*< td=""><td><dl*< td=""><td><d *< td=""></d *<></td></dl*<></td></dl*<> | <dl*< td=""><td><d *< td=""></d *<></td></dl*<> | <d *< td=""></d *<> |
| Na | 41.9              | 47.7  | 54.9  | 68.8  | 60.2                |
| Ni | 1.3               | 0.73  | 2.1   | 1.2   | 21.5                |
| Р  | 1399              | 1062  | 2460  | 1539  | 1269                |
| Pb | 1.5               | 3.4   | 3.6   | 2.3   | 25.3                |
| Se | 0.7               | <d *< td=""><td><d *< td=""><td><d *< td=""><td><d *< td=""></d *<></td></d *<></td></d *<></td></d *<> | <d *< td=""><td><d *< td=""><td><d *< td=""></d *<></td></d *<></td></d *<> | <d *< td=""><td><d *< td=""></d *<></td></d *<> | <d *< td=""></d *<> |
| Si | 154               | 219   | 98.6  | 104   | 167                 |
| Sr | 78.1              | 52.4  | 62.6  | 44.4  | 30.2                |
| Ťì | 2.5               | 4.4   | 0.34  | 1.6   | 71.6                |
| ٧  | 0 72              | 1.3   | 1.6   | 1.6   | 49.5                |
| Zn | 21.2              | 50.5  | 32.8  | 36.8  | 113                 |
| N% | 1.78              | 0.97  | 1.32  | 1.25  | 1 19                |

< dl\*: element concentration under the detection limit of the instrument

tigated grasses (Tab. 4). 19% of the total element content is made up by heavy metals (Fig. 4). In higher concentrations first of all Fe, Mn and Zn were detected. The chemical composition of the moss, which is capable of accumulating ions through its entire body surface, is very similar to that of the uppermost soil layer.

Chrysopogon gryllus, which is an indigenous steppe species in Hungary with a southern Eurasian area, often occurs not only on loess soils, but also in dry grasslands on sandy soils. Recently this species – similarly to other grass species of the  $C_4$  photosynthetic type – has been spreading in the loess steppe grasslands at the cost of Festuca valesiaca. As a consequence of its specific root system the species possesses a capacity for a concurrence as well as a significant drought tolerance.

The chemical compositions of the leaves of *Festuca valesiaca* and *Chrysopogon gryllus* are different (Tab. 5). The leaves of *Chrysopogon gryllus* – like other perennial  $C_4$  type grass species (Kovács et al. 1998) – contain higher amounts of Ca and Mg and a lower concentration of N than the perennial  $C_3$  type grasses. Characteristic of the leaves of *Festuca valesiaca* is a relatively high con-

**Tab. 5.** Chemical element content of the leaves, roots and soils of *Festuca valesiaca* ( $C_3$ ) of the *Salvio-Festucetum rupicolae* community and *Chrysopogon gryllus* ( $C_4$ ) of the *Agropyro-Kochietum* community on a loess steppe in Erd, Hungary (n=8)

|                           | Chrysopogon gryllus | Festuca valesiaca |
|---------------------------|---------------------|-------------------|
| Leaf                      |                     |                   |
| (a (µg g ¹)               | 3230                | 2674              |
| Mg (μ.g.g <sup>-1</sup> ) | 1346                | 860               |
| K (μ.g g <sup>-1</sup> )  | 3144                | 5880              |
| C (%)                     | 42 41               | 41 57             |
| N (%)                     | 0.81                | 1 10              |
| Ç/N                       | 52.6                | 38 3              |
| Root                      |                     |                   |
| (a (µg g <sup>-1</sup> )  | 3638                | 18488             |
| Mg (μg g <sup>-1</sup> )  | 1729                | 3083              |
| K (μg g ¹)                | 2029                | 1718              |
| C (%)                     | 42.3                | 33 3              |
| N (%)                     | 0.91                | 1 66              |
| C/N                       | 46.5                | 20 0              |
| Soil                      |                     |                   |
| Ca (µ g g ¹)              | 68234               | 57409             |
| Mg (μg g <sup>-1</sup> )  | 8307                | 8940              |
| K (μ.g. g <sup>-1</sup> ) | 1253                | 1353              |
| C (%)                     | 5.06                | 5.99              |
| N (%)                     | 0.35                | 0.37              |
| C/N                       | 17.9                | 17.0              |

centration of K and P. In the leaves of the two species the K/Ca ratio is also different (2.19 in the case of *Festuca valesiaca*, and 0.96 in the case of *Chrysopogon gryllus*). This latter ratio is known to be characteristic of the calciotroph physiotype (KINZEL 1982).

Whereas the carbon content of the two species is similar, the lower nitrogen content of the leaves of *Chrysopogon gryllus* results in an increased C/N ratio. The relatively high C/N ratio in both the leaves (52.3) and roots (46.5) of *Chrysopogon gryllus* is presumably the consequence of the slow mineralization dynamics, which may result in a decreased rate of the biogeochemical cycle of the community and also an accumulation of organic matter within it.

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