

Correlation Between Phytochemical and Mineral Contents and Antioxidant Activity of Black Glutinous Rice Bran, and Its Potential Chemopreventive Property

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Summary

In this work total anthocyanin content (TAC), total flavonoid content (TFC), total phenolic content (TPC) and minerals found in five black glutinous rice cultivars (MS, SK, PY, PC and KK) from Thailand were analyzed. The antioxidant activity of anthocyanin-rich black glutinous rice bran extracts against nitric oxide radical (NO[•]), superoxide radical (O₂^{•-}) and lipid peroxy radical (LOO[•]) was also determined. Potential chemopreventive property of rice bran extract was screened based on cellular bioassays for phase II detoxification enzyme induction. Quinone reductase (QR) induction in murine hepatoma cells was used as a marker for this effect. Rice bran extract of cultivar KK had the highest TAC, of SK the highest TFC and of PC the highest TPC. The best antioxidants against NO[•], O₂^{•-} and LOO[•] were cultivars KK, MS, and SK, respectively. Overall, TAC, TFC and TPC had a combinatorial effect on the antioxidant activities of all extracts; none of them dominated. Minerals may not play a role in the antioxidant activity of the extracts because most correlations between them and the antioxidant activity were unpredictable. However, rice bran contained high mass fractions of some essential minerals on dry mass basis, including Zn (103–133 µg/g), Se (11–18 µg/g) and Cu (3.8–7.1 µg/g). Chemopreventive study indicated that PC cultivar was the most potent chemopreventor with the lowest concentration of an inducer needed to double the QR activity (CD value) of 0.7 µg/mL. These findings showed that black glutinous rice bran is rich in phytochemicals and some essential minerals, and has a potential chemopreventive property.

Key words: anthocyanins, black glutinous rice bran, mineral content, antioxidant activity, quinone reductase induction

Introduction

Health concerns have led to an increased trend in the adoption of phytochemical-rich diets. Phytochemicals such as phenolics and flavonoids (including anthocyanins) have been reported to have potential of preventing reactive oxygen species (ROS)-induced diseases like diabetes, cancer and cardiovascular diseases (1–3). Phenolics, flavo-

noids and anthocyanins have been found in various plants and rice (*Oryza sativa* L.) is one of the predominant sources of phytochemicals consumed daily, especially in Asian countries.

Glutinous rice with different bran colours, such as brown, red and black is commonly consumed in Thailand (4,5) as a regular steamed rice or in sweet dishes. Among

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these rice types, black glutinous rice has received more interest as it has been found to contain higher contents of phenolics, flavonoids and especially anthocyanins than others (6). These phytochemicals showed potential for inducing phase II detoxification enzymes, a group of important cell defense enzymes that convert reactive and harmful molecules into inactive water-soluble ones which are more easily eliminated (7). Hence, consumption of black rice or black rice products increases cell defensive actions and prevents ROS-induced diseases.

Rice is also a rich source of minerals, including Ca, Cu, Fe, K, Mg, Mn, Se and Zn (8,9). It was found in an *in vitro* study that Mn contributes to the antioxidant activity of rice (10). Concentrations of phenolics, flavonoids, anthocyanins and minerals are different depending on variety, growing conditions and soil components (11,12). Various cultivars of black glutinous rice are grown in the northern region of Thailand and they may have different antioxidant components that need to be investigated. For these reasons, antioxidant activities of rice bran may not only depend on the contents of phenolics, flavonoids and anthocyanins, but also on the mineral content. However, few studies reporting the correlation between antioxidant activity and mineral content have been published (10,13).

In this study, the contents of phytochemicals, including phenolics, flavonoids and anthocyanins, and minerals in black glutinous rice bran samples from Thailand were identified. *In vitro* antioxidant activities against NO^{\cdot} , $\text{O}_2^{\cdot-}$ and LOO^{\cdot} of black glutinous rice bran extracts were examined. The relationship between antioxidant activity and the content of phytochemicals and some trace minerals was evaluated. Potential chemopreventive properties of the extracts were tested in order to evaluate the health-protective benefits of black glutinous rice bran.

Materials and Methods

Chemicals and reagents

Mineral standards were purchased from Merck (Darmstadt, Germany). Commercial and natural antioxidant standards were of reagent grade and were purchased from Sigma-Aldrich (St. Louis, Mo, USA), together with Folin-Ciocalteu reagent, phenazine methosulphate (PMS), β -nicotinamide adenine dinucleotide (NADH), sodium nitroprusside dihydrate (SNP), Griess-Ilosvay's nitrite reagent, nitroblue tetrazolium chloride (NBT) and all other reagents.

Samples

Five black glutinous rice cultivars grown in the northern Thailand: Kaw Kum Doi Mooser (MS) from Tak Province, Kaw Kum Doi Saked (SK) from Chiang Mai Province, Kaw Kum Phayao (PY) from Phayao Province, Kaw Neaw Dum Phichit (PC) from Phichit Province and Kaw Kum Khao Kho (KK) from Phetchabun Province were used. Rough rice samples harvested from October to December 2011 were purchased directly from farmers. All samples were stored at $(4\pm 1)^\circ\text{C}$ until milling.

Extraction of active compounds from black glutinous rice bran

Black glutinous rice bran powder was defatted with hexane and dried at 40°C in a hot air oven for 3 h. Moisture content of the defatted rice bran powder was analyzed using the loss on drying test. A mass of 0.1 g of the bran powder was extracted with 1 M HCl methanol/water (15:85, by volume) using magnetic stirrer (C-MAG HS7; IKA® Werke GmbH & Co. KG., Staufen, Germany). Anthocyanin-rich black glutinous rice bran extract solution was concentrated using a rotary evaporator (Rotavapor® R-100; BÜCHI Labortechnik AG, Flawil, Switzerland) at $(40\pm 2)^\circ\text{C}$ and the volume was adjusted to 25 mL with methanol before storage at $(-20\pm 1)^\circ\text{C}$ for further study.

Total anthocyanin content analysis

Total anthocyanin content (TAC) was quantified using the pH differential method reported by Giusti and Wrolstad (14). Briefly, 100 μL of rice bran extract were filled up to 1 mL either with 0.025 M potassium chloride buffer (pH=1.0) or 0.4 M sodium acetate buffer solution (pH=4.5). Absorbance of each mixture was measured at 510 and 700 nm using a UV-Vis spectrophotometer (DR/4000 U; Hach, Loveland, CO, USA) and total absorbance was calculated using the following equation:

$$A=(A_{510\text{ nm}} - A_{700\text{ nm}})_{\text{pH}=1.0} - (A_{510\text{ nm}} - A_{700\text{ nm}})_{\text{pH}=4.5} \quad /1/$$

where $A_{510\text{ nm}}$ and $A_{700\text{ nm}}$ are the absorbances measured at 510 and 700 nm, respectively. TAC (in mg/L) was calculated using the following equation:

$$\text{TAC}=(A \cdot M \cdot \text{DF} \cdot 1000)/(\varepsilon \cdot 1) \quad /2/$$

where A is the absorbance from Eq. 1, M is the molecular mass of cyanidin-3-O-glucoside ($M=449.2$), DF is the dilution factor (100 μL of sample is diluted to 1 mL, $\text{DF}=10$), and ε is the molar absorption coefficient of cyanidin-3-O-glucoside ($\varepsilon=26\,900\text{ L}/(\text{mol}\cdot\text{cm})$).

Total flavonoid content analysis

Total flavonoid content (TFC) was determined following the method of Pękal and Pyrzynska (15) with some modifications. Briefly, 0.5 mL of rice bran extract was mixed with 2 mL of distilled water, 0.15 mL of 5 % NaNO_2 , 0.15 mL of 10 % $\text{AlCl}_3 \cdot 6\text{H}_2\text{O}$ and 1 mL of 1 M NaOH. The absorbance was determined by a UV-Vis spectrophotometer (DR/4000 U; Hach) at 415 nm after 15 min of incubation. Flavonoid content was expressed in gram of catechin equivalent (CE) per kg of rice bran on dry mass (dm) basis.

Total phenolic content analysis

Total phenolic content (TPC) of rice bran extract was measured using Folin-Ciocalteu method with some modifications (16). Briefly, 0.1 mL of rice bran extract was mixed with 0.5 mL of Folin-Ciocalteu reagent. After 5 min, 1.5 mL of 7.5 % sodium carbonate were added. The mixture was diluted with 5 mL of distilled water and left to react for 2 h before measuring the absorbance at 765 nm by a UV-Vis spectrophotometer (DR/4000 U; Hach). Gallic acid was used as a standard and the TPC was reported in gram of gallic acid equivalent (GAE) per kg of rice bran on dm basis.

Mineral content analysis

To quantify the mineral content, 500 mg of rice bran powder were digested with 5 mL of 60 % HClO₄ and 62 % HNO₃ (1:3 by volume). The solution was diluted with deionized water before analysis. Mineral content was analyzed using atomic absorption spectrophotometer (model GBC Avanta; GBC Scientific Equipment, Hampshire, IL, USA) with the detection limit of 4.4 µg/kg.

Antioxidant analyses

Nitric oxide scavenging activity

The experiment was conducted by mixing 0.5 mL of different concentrations of rice bran extract (4–90 mg/mL) with 125 µL of 5 mM SNP solution in phosphate-buffered saline (PBS), 50 mM, pH=7.4 (17). After 2 h, 150 µL of Griess-Ilosvay's nitrite reagent were added and the absorbance was measured at 550 nm by a UV-Vis spectrophotometer (DR/4000 U; Hach). Quercetin was used as a positive control. Percentage of nitric oxide scavenging activity was calculated by the following equation:

$$\text{NO}^{\cdot} \text{ scavenging activity} = (A_0 - (A_1/A_0)) \cdot 100 \quad /3/$$

where A_0 is the absorbance of the control and A_1 is the absorbance of the analyte in the presence of rice bran extract or quercetin. Antioxidant activity was reported as the effective concentration needed to scavenge NO[·] free radicals by 50 % compared to the control (EC₅₀).

Superoxide scavenging activity

Scavenging activity against O₂^{·-} was performed using PMS/NADH system (18). A volume of 3 mL of Tris-HCl buffer (100 mM, pH=7.4), 750 µL of NBT (300 µM), 750 µL of NADH (936 µM), 750 µL of PMS (120 µM) and 300 µL of different concentrations of rice bran extract (15–250 mg/mL) were mixed together. The absorbance of the solution was measured after 15 min of incubation at 560 nm by a UV-Vis spectrophotometer (DR/4000 U; Hach). L-Ascorbic acid was used as a positive control. Percentage of scavenging was calculated using the following equation:

$$\text{O}_2^{\cdot-} \text{ scavenging activity} = (A_0 - (A_1/A_0)) \cdot 100 \quad /4/$$

where A_0 is the absorbance of the control and A_1 is the absorbance of the analyte in the presence of rice bran extract or L-ascorbic acid. Antioxidant activity was reported as EC₅₀.

Inhibition of lipid peroxidation

Lipid peroxidation (LOO[·]) was determined using ammonium thiocyanate method with some modifications (19). Linoleic acid emulsion (20 mM) was prepared in PBS using Tween 20 as an emulsifier. The emulsion was mixed with various concentrations of rice bran extract (4–45 mg/mL). After incubation for 20 h in the dark, 100 µL of the mixture were mixed with 1 mL of absolute methanol, 100 µL of ammonium thiocyanate (30 % in distilled water), and 100 µL of ferrous chloride solution (0.02 M in 3.5 % HCl). The developed red colour was measured at 500 nm by a UV-Vis spectrophotometer (DR/4000 U; Hach). Butylated hydroxytoluene (BHT) was used as a positive control. LOO[·] inhibition activity (in %) was calculated using the following equation:

$$\text{Inhibition of LOO}^{\cdot} = (A_0 - (A_1/A_0)) \cdot 100 \quad /5/$$

where A_0 is the absorbance of the control and A_1 is the absorbance in the presence of rice bran extract or BHT. The activity was reported as EC₅₀.

Quinone reductase induction assay

The quinone reductase (QR) induction was used as a biomarker for the induction of phase II detoxifying enzymes (20). Murine hepatoma cell line (Hepa 1c1c7 cells; ATCC, Rockville, MD, USA) were cultured with α-MEM (Gibco[®] by Life Technology[™], Thermo Fisher Scientific, Carlsbad, CA, USA) in identical duplicate 96-well plates (Falcon[®]; Corning, Tewksbury, MA, USA) for 24 h. Then, the culture medium was replaced with fresh medium containing the rice bran extract and the cells were incubated for an additional 48 h. The QR induction activity was analyzed on a 96-well plate by dissolving the treated cells with 50 µL of digitonin solution per well and shaking for 30 min. A volume of 200 µL of reactant cocktail (21 mL of Tris-HCl enzyme buffer, pH=7.4, 1.5 mL of MTT (3-(4,5-dimethylthiazo-2-yl)-2,5-diphenyltetrazolium bromide) and 22.5 µL of menadione solution) per well was added (20). The development of dark purple colour was measured as the absorbance at 490 nm using a microplate reader (SpectraMax Plus 384; Molecular Devices, Sunnyvale, CA, USA). Protein assay was measured on another identical 96-well plate (Falcon[®]; Corning). The culture medium was discarded and the treated cells were stained with 100 µL of crystal violet stain per well. The stained protein was dissolved with 150 µL of sodium dodecyl sulfate per well and the absorbance was measured at 610 nm by a UV-Vis spectrophotometer (DR/4000 U; Hach). The relative QR activity can be determined from the ratio of ΔA_{490 nm} (QR assay) and A_{610 nm} (protein assay) with the ratio for control cells set to 1.0.

The results of QR induction were expressed as CD value, IC₅₀ and chemopreventive index (CI, calculated as IC₅₀/CD).

Statistical analysis

Results were expressed as mean value ± standard deviation of three replications. Statistical differences were analyzed using one-way ANOVA followed by Duncan's test at 95 % confidence level (p ≤ 0.05). Correlation analysis was obtained using bivariate correlations and expressed as Pearson's correlation coefficient (*r*). Statistical analysis was performed using SPSS v. 16.0 (SPSS Inc., Chicago, IL, USA).

Results and Discussion

Total anthocyanin, total flavonoid and total phenolic contents

The moisture content of the defatted rice bran powder ranged from 8.5 to 9.4 %. Fig. 1 shows the results of total anthocyanin content (TAC), total flavonoid content (TFC) and total phenolic content (TPC) measurements. The value of TAC in rice bran extract samples ranged between 10 (in MS cultivar) and 23 (in KK cultivar) g/kg.

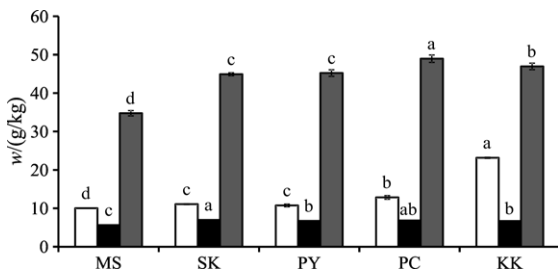


Fig. 1. Total anthocyanin (□), phenolic (■) and flavonoid (▒) contents on dry mass basis of black glutinous rice bran samples (MS, SK, PY, PC and KK; for sample abbreviations see Materials and Methods). Different letters in the same test indicate statistically significant difference ($p \leq 0.05$) according to Duncan's multiple range test ($N=3$)

The content of TFC expressed as CE ranged between 0.6 (in MS cultivar) and 6.9 (in KK cultivar) g/kg. TPC expressed as GAE was found in the range of 3.4 (in MS cultivar) to 4.7 (in PC cultivar) g/kg. Overall, the extract of MS cultivar had the lowest TAC, TFC and TPC.

Among the five rice bran samples, KK cultivar had the highest TAC, which was about two times higher than that in the other rice bran extracts. The bioactive components in the rice bran extract were different due to several factors, including cultivation techniques, ripening environments and growing conditions (11,12).

Antioxidant activities of black glutinous rice bran crude extracts

Fig. 2 shows the results of antioxidant activity tests and the EC_{50} values of rice bran extracts. A lower EC_{50} value indicates higher antioxidant capacity. The best NO^{\cdot} scavenger was quercetin, indicated by the lowest EC_{50} value (1.9 mg/L), followed by the extracts of KK (62 mg/L), MS (74 mg/L), PY (91 mg/L), SK (95 mg/L) and PC cultivars (98 mg/L). Among the rice bran extracts, KK cultivar had the best antioxidant activity ($p \leq 0.05$), while MS

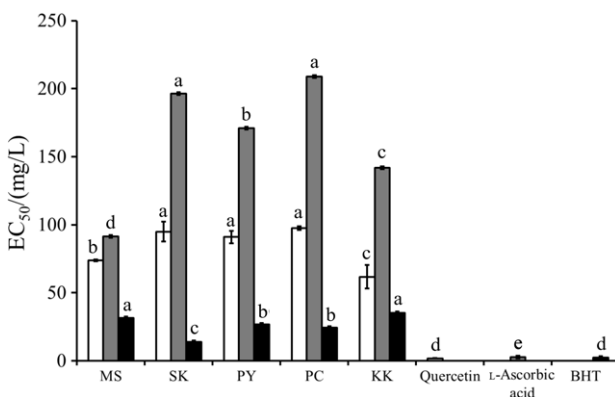


Fig. 2. Antioxidant activities of black glutinous rice bran crude extracts (MS, SK, PY, PC and KK; for sample abbreviations see Materials and Methods). Quercetin, L-ascorbic acid and BHT were positive controls for nitric oxide (□), superoxide (■), and lipid peroxidation (▒) assays, respectively. EC_{50} =effective concentration needed to scavenge the free radicals by 50 % compared to the control. Different letters in the same test indicate statistically significant difference ($p \leq 0.05$) according to Duncan's multiple ranges test ($N=3$)

cultivar had better activity than SK, PY and PC cultivars, although it had lower TAC, TFC and TPC.

L-Ascorbic acid had the best $O_2^{\cdot-}$ scavenging activity ($EC_{50}=2.8$ mg/L). The extract of MS cultivar had significantly better activity ($p \leq 0.05$) with regard to EC_{50} value (31 mg/L) than the extracts of KK (142 mg/L), PY (171 mg/L), SK (196 mg/L) and PC cultivars (208 mg/L).

The best inhibitor against LOO^{\cdot} was BHT ($EC_{50}=1.9$ mg/L). The extract of SK cultivar had significantly lower EC_{50} value (14 mg/L) with $p \leq 0.05$, indicating stronger LOO^{\cdot} inhibition than by PC (24 mg/L), PY (26 mg/L), MS (32 mg/L) and KK extracts (35 mg/L).

All rice bran extracts had significantly lower ($p \leq 0.05$) antioxidant activities than the corresponding positive controls. Our preliminary study results (data not shown) indicated that acidified methanol/water rice bran extracts contained various compounds, both active (phenolic acids, anthocyanins (5) and proanthocyanidins (21)) and inactive (sugars and soluble oligosaccharides (22)). The active components in rice bran extracts were more dilute than in the highly pure positive controls. Different rice bran extracts show different dominant antioxidant activities. In general, TAC does not seem to play a major antioxidant role against NO^{\cdot} , LOO^{\cdot} and $O_2^{\cdot-}$, as indicated by the extracts of KK, which had the highest TAC but did not show dominant antioxidant activity in all tested assays. It has been reported previously that the chemical structure of anthocyanins is changed or possibly degraded at pH higher than 4.0 (23). The structurally changed anthocyanins may lose their ability to donate proton (24). Protocatechuic acid is the major product of cyanidin-3-O-glucoside (major anthocyanin in rice bran extract) degradation at low pH values (25). However, *in vitro* antioxidant activity of protocatechuic acid was the weakest compared to other phenolic acids (26). This indicates that antioxidant activities of rice bran extracts are affected by the ratios and types of other active components (TFC, TPC and minerals) and the products of anthocyanin degradation.

Mineral profile of black glutinous rice bran

Three major minerals (Ca, K and Mg) and six trace elements (Cu, Fe, Mn, Na, Se and Zn) were quantified in black rice bran powder (Table 1). Mineral contents ($\mu\text{g/g}$) in black rice bran were found in the following order: $K > Mg > Zn > Na > Fe > Se$ or $Mn > Cu > Ca$. The high content of K (16642 $\mu\text{g/g}$ in KK to 21884 $\mu\text{g/g}$ in PY cultivars) and Mg (1894 $\mu\text{g/g}$ in MS to 2411 $\mu\text{g/g}$ in KK cultivars) was found in all the rice bran samples. Some essential minerals were found in black rice bran: Zn (103–133 $\mu\text{g/g}$), Se (11–18 $\mu\text{g/g}$) and Cu (3.8–7.1 $\mu\text{g/g}$). This result is in agreement with the report of Parengam *et al.* (9), who found that the two preventing minerals in various samples were K or Mg. However, their contents varied even in the same rice cultivar due to many factors and growing conditions, including soil type, fertilizers, herbicides and fungicides (27,28).

Although rice bran is a by-product of rice production, this study shows that it is a promising source of concentrated essential minerals. Deficiency of some minerals is an important problem that has led to a study of mineral

Table 1. Mineral contents on dry mass basis in different black glutinous rice bran extracts (MS, SK, PY, PC and KK; for sample abbreviations see Materials and Methods)

Mineral	<i>w</i> /($\mu\text{g/g}$)				
	MS	SK	PY	PC	KK
Ca	(0.8 \pm 0.1) ^b	(1.6 \pm 0.1) ^a	(1.9 \pm 0.2) ^a	(1.9 \pm 0.3) ^a	(1.9 \pm 0.1) ^a
Cd	(0.7 \pm 0.1) ^a	N.D.	(0.6 \pm 0.1) ^b	(0.9 \pm 0.2) ^a	N.D.
Cr	N.D.	N.D.	N.D.	N.D.	N.D.
Cu	(5.9 \pm 0.6) ^b	(3.8 \pm 0.2) ^c	(3.9 \pm 0.3) ^c	(5.2 \pm 0.6) ^b	(7.1 \pm 0.3) ^a
Fe	(16.0 \pm 1.4) ^c	(17.0 \pm 0.6) ^c	(17.0 \pm 2.2) ^c	(24.0 \pm 1.1) ^b	(30.0 \pm 0.9) ^a
K	(17300 \pm 738) ^b	(20796 \pm 2020) ^a	(21885 \pm 451) ^a	(21823 \pm 854) ^a	(16643 \pm 242) ^b
Mg	(1894 \pm 63) ^b	(2123 \pm 6) ^b	(1979 \pm 41) ^b	(2409 \pm 188) ^a	(2412 \pm 191) ^a
Mn	(12.0 \pm 0.3) ^c	(15 \pm 1) ^c	(16.0 \pm 0.4) ^b	(21.0 \pm 0.6) ^a	(13.0 \pm 0.1) ^d
Na	(97 \pm 13) ^a	(85 \pm 14) ^{ab}	(84 \pm 6) ^{ab}	(73 \pm 12) ^b	(84 \pm 12) ^{ab}
Se	(12.0 \pm 1.3) ^{bc}	(14.0 \pm 1.2) ^b	(14.0 \pm 0.4) ^b	(18.0 \pm 1.4) ^a	(11.0 \pm 1.5) ^c
Zn	(133 \pm 3) ^a	(113 \pm 8) ^b	(120 \pm 3) ^b	(103 \pm 1) ^c	(117 \pm 2) ^b

The data are presented as mean value \pm standard deviation of triplicate analyses. Different letters in the same row indicate statistically significant values ($p\leq 0.05$). N.D.=not detected, below detection limit of the instrument at 0.04 ppm

fortification, such as with Fe, which has been found at low concentrations (6–10 $\mu\text{g/g}$) in some brown rice cultivars in Thailand (29). However, higher concentrations of Fe were found in the five black glutinous rice bran samples used in this study in the range of 16–30 $\mu\text{g/g}$.

Some trace elements such as Zn, Se, Cu and Mn were reported to be included in human cell defense mechanisms against reactive oxygen species (ROS) (30,31). These minerals act as cofactors of several antioxidant enzymes. For example, superoxide dismutase works by incorporating Zn, Cu and Mn in order to convert $\text{O}_2^{\cdot-}$ to a less harmful H_2O_2 (31). Consequently, glutathione peroxidase eliminates H_2O_2 by conjugating with selenium (32). Since high contents of Zn, Se and Cu were found in all of

the rice samples, black glutinous rice bran can be an important source of essential minerals that could promote health benefits to consumers.

Correlation between antioxidant activities and TAC, TFC, TPC and mineral content

Pearson's correlation coefficient (r) (Table 2) was obtained from bivariate correlation analysis and used to describe the correlation between the antioxidant activities against NO^{\cdot} , $\text{O}_2^{\cdot-}$ and LOO^{\cdot} and the content of antioxidant components (TAC, TFC, TPC, Cu, Fe, Mn and Zn). It needs to be noted that antiradical values (calculated as $1/\text{EC}_{50}$) were employed for the correlation analysis instead

Table 2. Pearson's correlation coefficients (r) between TAC, TFC, TPC and mineral content and antioxidant activity ($1/\text{EC}_{50}$) of rice bran extracts (MS, SK, PY, PC and KK; for sample abbreviations see Materials and Methods)

Extract	$1/\text{EC}_{50}$	r ($N=3$)						
		TAC	TFC	TPC	Cu	Fe	Mn	Zn
MS	NO^{\cdot}	0.994	0.842	0.129	-0.532	-0.932	-1.000*	-0.127
	$\text{O}_2^{\cdot-}$	-0.380	-0.874	0.811	-0.493	0.762	0.475	0.926
	LOO^{\cdot}	0.777	1.000*	-0.432	0.016	-0.979	-0.839	-0.632
SK	NO^{\cdot}	0.631	0.160	-0.330	-0.998*	-0.998*	-0.014	-0.237
	$\text{O}_2^{\cdot-}$	-0.970	-0.721	-0.302	0.839	0.764	-0.589	-0.393
	LOO^{\cdot}	-0.949	-0.666	-0.230	0.877	0.811	-0.527	-0.323
PY	NO^{\cdot}	0.153	0.724	-0.981	-0.176	-0.933	0.024	-0.688
	$\text{O}_2^{\cdot-}$	-0.944	0.268	0.292	-0.786	0.756	-0.893	-0.317
	LOO^{\cdot}	-0.736	0.976	-0.705	-0.916	-0.220	-0.818	-0.986
PC	NO^{\cdot}	0.846	-0.679	0.307	-0.865	-0.998*	0.907	-0.720
	$\text{O}_2^{\cdot-}$	-0.227	-0.916	-0.799	-0.757	-0.392	0.694	0.420
	LOO^{\cdot}	0.955	0.116	0.922	-0.182	-0.597	0.271	-0.996
KK	NO^{\cdot}	-0.701	0.422	0.984	0.732	0.468	-0.181	0.765
	$\text{O}_2^{\cdot-}$	-0.888	0.121	0.881	0.492	0.719	0.170	0.927
	LOO^{\cdot}	0.913	0.702	-0.177	0.376	-0.992	-0.878	-0.870

*correlation is significant at the 0.05 level (2-tailed). TAC=total anthocyanin content, TFC=total flavonoid content, TPC=total phenolic content

of EC₅₀ value since direct analysis using EC₅₀ gave negative *r* value, which confused the interpretation of the results. Three correlation levels were defined as strong ($r=(+/-)0.600-1.000$), moderate ($r=(+/-)0.400-0.599$), and weak ($r=(+/-)0.000-0.399$) (33).

Both positive and negative correlations of NO[•] scavenging activity with TAC: strong positive (MS, $r=0.994$; SK, $r=0.631$ and PC, $r=0.846$) and strong negative (KK, $r=-0.701$), TFC: strong positive (MS, $r=0.842$ and PY, $r=0.724$) and strong negative (PC, $r=-0.679$) and TPC: strong positive (KK, $r=0.984$) and strong negative (PY, $r=-0.981$) were found in rice bran extracts from all rice cultivars. These results indicate unpredictable correlation between the antioxidant contents and NO[•] scavenging activity. The content of Cu, Fe, Mn and Zn may not affect the NO[•] scavenging activity as both positive and negative correlations were found. However, low content of Fe tends to promote the NO[•] scavenging activity as most correlations were strong negative in MS ($r=-0.932$), SK ($r=-0.998$), PY ($r=-0.933$) and PC ($r=-0.998$) cultivars, with the exception of a moderate positive correlation in KK cultivar ($r=0.468$).

TAC, TFC and TPC may not contribute to the O₂^{•-} scavenging since most *r* values were negative or weak positive. Similarly, Cu, Mn and Zn may also not be involved in the scavenging of O₂^{•-} as both positive and negative correlations with the O₂^{•-} scavenging activity were found. However, the presence of Fe tends to promote the scavenging of O₂^{•-} as most *r* values were strong positive (*r* values from 0.719 to 0.764), with only one moderately negative *r* value in PC cultivar ($r=-0.392$). Similar results were found in the report of Kaneda *et al.* (10), where various levels of correlation between the content of trace elements and O₂^{•-} scavenging activity were found depending on rice samples.

The inhibition of LOO[•] activity did not correlate with TAC, TFC and TPC since both strong positive and strong negative correlations were found as follows: TAC: strong positive (MS, $r=0.777$; PC, $r=0.955$ and KK, $r=0.913$) and strong negative (SK, $r=-0.949$ and PY, $r=-0.736$), TFC: strong positive (MS, $r=1.000$; PY, $r=0.976$ and KK, $r=0.702$) and strong negative (SK, $r=-0.666$) and TPC: strong positive (PC, $r=0.922$) and strong negative (PY, $r=-0.705$). Similar results were also found between LOO[•] inhibition activity and the content of Cu and Fe, which indicated that these two elements did not affect the inhibition of LOO[•]. However, low contents of Mn and Zn tended to promote the LOO[•] inhibition activity as negative correlations, except a weak positive correlation between Mn and LOO[•] inhibition activity in PC cultivar ($r=0.271$), were found in all rice bran extracts.

In general, this correlation study indicated that none of TAC, TFC or TPC solely contributed to the antioxidant activities against NO[•], O₂^{•-} and LOO[•], since various levels of both positive and negative correlations were presented. In the case of TAC, antioxidant activity of anthocyanin may be influenced by the pH of the solution in the antioxidant assay, since the chemical structure of anthocyanin is degraded at higher pH values (34). Low and moderate correlations of TFC and TPC with *in vitro* antioxidant activities were previously reported for a number of plant extracts (6,13). For these reasons, antioxidant activity of rice bran extracts was considered to be caused by the

combined effects of TAC, TFC and TPC. These findings suggest the necessity of choosing a suitable *in vitro* antioxidant assay for the compounds of interest. The pH of the test solution may influence the antioxidant activity of some compounds, especially anthocyanins, which are susceptible to pH.

Mineral content of rice bran does not seem to affect NO[•], O₂^{•-} and LOO[•] antioxidant activities as both positive and negative *r* values were determined. However, low levels of Mn and Zn had an effect on the inhibition of LOO[•] as the correlations were mostly negative. These results contradict previous studies where these minerals were found to contribute to cell defense against ROS (30,31). For example, Zn showed protective effects against oxidative damage and lipid peroxidation in cellular environment (35,36). The results in this study also indicate that high content of Fe tended to promote the scavenging of O₂^{•-}. However, excess of Fe was reported to catalyze the generation of ROS in a cellular study (35). These conflicting results could be explained by the protective mechanism of these minerals that was achieved by forming a complex structure with some specific cellular molecules, such as superoxide dismutase, catalase and glutathione peroxidase (30–32,36). Thus, the chemical reactions in the antioxidant assays used in this experiment may limit the activities of trace elements and prevent their interaction with ROS. This is in agreement with the previous report which suggested that antioxidants investigated *in vitro* were not affected by the content of trace elements (10,13).

Induction of quinone reductase

The quinone reductase (QR) activity of rice bran extracts is shown in Table 3. The results were calculated based on the TAC value. The chemopreventive effect of all the crude rice bran extracts expressed as CD values ranged from 0.7 µg/mL in PC cultivar (strongest QR inducer) to 5.6 µg/mL in KK cultivar (weakest QR inducer). The IC₅₀ values, which show the risk of cytotoxicity, ranged from 5.1 µg/mL in SK cultivar (more toxic) to 32.0 µg/mL in KK cultivar (less toxic), whereas MS cultivar was not cytotoxic at the highest concentration used (1500 µg/mL). Overall, the PC extract had the highest QR induction ability indicated by the highest chemopreventive index (CI) value (risk to benefit ratio) of 8.3 (calculated as IC₅₀ (5.9 µg/mL) divided by CD value (0.7 µg/mL)).

Table 3. Quinone reductase (QR) activities of black glutinous rice bran extracts (MS, SK, PY, PC and KK; for sample abbreviations see Materials and Methods)

Rice bran	CD/(µg/mL)	IC ₅₀ /(µg/mL)	CI
MS	(3.9±0.1) ^b	N/A	N/A
SK	(0.9±0.1) ^c	(5.1±0.2) ^b	5.7
PY	(0.8±0.1) ^c	(5.5±0.0) ^b	7.2
PC	(0.7±0.1) ^c	(5.9±0.1) ^b	8.3
KK	(5.6±0.2) ^a	(32.0±1.6) ^a	5.5

CD=the concentration of anthocyanins required to double the QR activity, IC₅₀=the concentration that causes 50 % cell death, CI=chemopreventive index (calculated as IC₅₀/CD), N/A=data not applicable due to the very low toxicity of the extract. Different letters in the same column indicate statistically different values ($p\leq 0.05$)

TAC may not play a major role in QR induction since KK cultivar, which had the highest TAC, was not the best QR inducer. Regarding the CD values, PC, PY, SK and MS cultivars were significantly ($p \leq 0.05$) stronger QR inducers than KK cultivar. The possible reason is that the assay was conducted in neutral pH environment for two days, during which the anthocyanin fraction in rice bran extract was degraded to cyanidin and protocatechuic acid (25). Consequently, the QR induction activity of anthocyanins decreased. This indicated that anthocyanins were not directly involved in this cell assay. Thus, only the compounds in the rice bran extracts that were more stable at neutral pH (*i.e.* phenolic acids, with the exception of protocatechuic acid) had the effect on QR induction activity (26). This finding is in agreement with our previously published results (37), which reported that phenolic acid-rich fraction isolated from the crude rice bran extract had stronger QR induction activity than the anthocyanin-rich fraction from the same source.

Conclusions

Crude extracts of anthocyanin-rich black glutinous rice bran have different phytochemical contents. However, the correlation study suggested that the antioxidant activity of these extracts seems to be the combination of activities of these phytochemicals. Minerals did not contribute to the antioxidant activity of the extracts since positive and negative correlations of different strength were found. Although some samples contained significantly high content of anthocyanins, they were not dominant antioxidants because they were affected by the pH values. A screening of chemoprevention effect indicated that anthocyanin-rich black glutinous rice bran extracts had potential anticarcinogenic properties since the products of anthocyanin degradation were considered to be involved in the antioxidant activity and chemopreventive property of the extracts. Further studies on the effects of pH on antioxidant activity of anthocyanins, isolation and identification of products of anthocyanin degradation and their antioxidant activity should be performed.

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