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## Transannular Photocyclization of (*E*)-3 $\beta$ -Acetoxy-5,10-seco-1(10)-cholesten-5-one\*

Ljubinka Lorenc, Vladimir Pavlović, and Mihailo Lj. Mihailović

Department of Chemistry, Faculty of Science, University of Belgrade, Studentski trg 16. P. O. Box 550, YU-11001 Belgrade, Yugoslavia, and Institute of Chemistry, Technology and Metallurgy, Belgrade

Bernard Tinant, Jean-Paul Declercq, and Maurice Van Meerssche

Laboratoire de Chimie physique et de Cristallographie, Université de Louvain, Bâtiment Lavoisier, Place Louis Pasteur, 1, B-1348 Louvain-la-Neuve, Belgium

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UV irradiation of the (*E*)-5,10-seco-steroidal ketone **1** in acetone solution results, in addition to the previously described *E/Z* isomerization and intramolecular Paterno-Büchi reaction (leading to the 1 $\alpha$ ,5 $\alpha$ -oxetane **4** in about 42% yield and 1 $\beta$ ,5 $\beta$ -oxetane **5** in 2—3% yield), in a transannular cyclization (accompanied by acetic acid elimination) producing the anthrasteroidal enone **3** (in about 7% yield). The structure of **3** was deduced from spectral data and confirmed by X-ray analysis.

### INTRODUCTION

We reported previously that UV irradiation of (*E*)-3 $\beta$ -acetoxy-5,10-seco-1(10)-cholesten-5-one (**1**) in dioxane<sup>2</sup> or acetone solution<sup>3</sup> with a high pressure mercury lamp leads to an intramolecular Paterno-Büchi reaction (cycloaddition of the carbonyl chromophore to the olefinic double bond) to give two oxetane derivatives, one, **4**, with the 1 $\alpha$ ,5 $\alpha$ -configuration, as the major product (in 32—42% yield), and the other, **5**, with the 1 $\beta$ ,5 $\beta$ -configuration, as a minor component (in 2—3% yield); in addition, *E/Z* isomerization of the olefinic  $\Delta^{1(10)}$ -double bond was observed. In the present paper we investigated, in more detail, the photolytic behaviour of the (*E*)-5,10-seco-ketone **1** when subjected to UV irradiation in acetone solution.

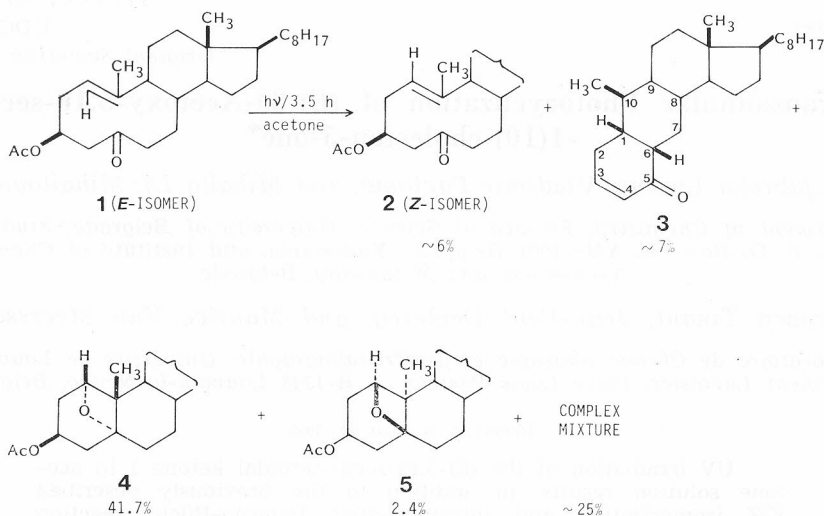
### RESULTS AND DISCUSSION

A  $\sim 0.9 \times 10^{-2}$  M acetone solution of the (*E*)-stereoisomeric 5,10-seco-ketone **1** was irradiated with a high pressure mercury lamp TQ 150 Z2 for 3.5 hours at room temperature. Product analysis revealed that under these conditions, besides the above mentioned transformations (*i. e.* *E/Z* isomerization and intramolecular Paterno-Büchi reaction), a transannular cycli-

\* Part XXVIII in the series »Synthesis structure and reactions of seco-steroids containing a medium-sized ring«. For Part XXVII see reference 1.

zation (accompanied by acetic acid elimination) producing the anthrasteroidal enone **3** also took place as a minor process. The results are given in Scheme 1.\*

Scheme 1



The photolysis products were separated by column chromatography. Actually, due to similar adsorption properties, (Z)-3 $\beta$ -acetoxy-5,10-seco-ketone **2** and the anthrasteroidal enone **3** were eluted from the column as a mixture. Their successful separation (on a SiO<sub>2</sub> column) was performed only after the mixture was treated with a ~1% methanolic KOH solution, which transformed product **2** into its more polar derivative, *i.e.* (Z)-3 $\beta$ -hydroxy-5,10-seco-1(10)-cholesten-5-one<sup>4</sup>, leaving **3** unchanged (see Experimental).

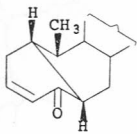
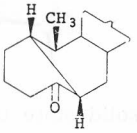
The structure of compound **3** was deduced from elemental microanalysis (C<sub>27</sub>H<sub>44</sub>O) and spectral data. Thus, MS ( $m/z$  384), IR and <sup>1</sup>H-NMR spectra of **3** revealed that the 3 $\beta$ -OAc group present in **1** was lost during irradiation (in the form of acetic acid) to produce a conjugated enone function (UV maximum at 227 nm, IR absorption at 1670 and 1620 cm<sup>-1</sup>, and signals at 5.97 and 6.95 ppm in the <sup>1</sup>H-NMR spectrum). Also, its <sup>1</sup>H-NMR and <sup>1</sup>H-decoupled <sup>13</sup>C-NMR spectra indicated that the original  $\Delta^{1(10)}$ -double bond was missing (no signals of the H—C(1) proton and the CH<sub>3</sub>(19) group attached to an isolated olefinic bond). Moreover, the appearance of the CH<sub>3</sub>(19) group as a doublet at 0.97 ppm in the <sup>1</sup>H-NMR spectrum of **3** and the number of primary, secondary, tertiary and H-free C-atoms detectable in the <sup>1</sup>H-decoupled <sup>13</sup>C-NMR spectrum of this compound (5 quartets, 9 triplets, 11 doublets and 2 singlets) strongly suggested that the olefinic  $\Delta^{1(10)}$ -double bond took part in an intramolecular cyclization leading to the anthrasteroidal structure **3**.

Catalytic hydrogenation of the olefinic double bond in **3** afforded the corresponding 3-saturated analogue **6**, the spectral characteristic of which were in complete agreement with the proposed structure (see Experimental).

\* Yields refer to crude products (see Experimental).

In an attempt to determine the stereochemistry at the newly formed chiral centres of **3** (particularly at the C(1) and C(6) bridgehead positions), CD measurement was performed on compound **3** and its saturated derivative **6**. The CD maxima are summarized in Table I. However, they proved to be unsatisfactory for a precise configurational assignment of **3**. Namely, from the molecular models it follows that the enone moiety in configuration-

TABLE I  
CD Data for the anthrasteroidal ketones **3** and **6** (in acetonitrile)<sup>a</sup>

Compound	$\lambda_{\max}$ ( $\Delta\epsilon$ )
 <b>3</b>	351(+2.17), 339(+2.27), 254(-0.48) 224(-3.78), 216(-3.61), 194(+6.53)
 <b>6</b>	301(+2.69), 243(-0.34), 232(-0.46) 222(-0.53), 199(-1.92), 198(-1.86)

<sup>a</sup>  $\lambda$  given in nm

ally different anthrasteroid systems of type **3** is not, or at least not very strongly, twisted, making predictions of the Cotton-effect sign for the unsaturated molecules difficult; on the other hand, in the case of the saturated analogue **6** a positive CD for the  $n \rightarrow \pi^*$  band (found experimentally, Table I) is consistent with the  $6\beta$ -configuration in both the  $1\alpha,6\beta$ - and  $1\beta,6\beta$ -stereoisomers. Thus, the  $1\beta,6\beta,10\beta(\text{CH}_3)$ -configuration of the cyclization product **3** was determined by X-ray analysis.

#### X-Ray Analysis and Structure Determination of the Anthrasteroidal Enone **3**

Crystal data of enone **3** are as follows:  $\text{C}_{27}\text{H}_{44}\text{O}$ ,  $M_r = 384.65$ , orthorhombic, space group  $P2_12_12_1$  with  $a = 33.173(18)$ ,  $b = 6.604(3)$ ,  $c = 11.172(4)$  Å;  $V = 2447(2)$  Å<sup>3</sup>,  $D_x = 1.04$  g cm<sup>-3</sup> for  $Z = 4$ . The intensities of 1523  $hkl$  independent reflections were collected on a Huber four circle diffractometer using  $\text{CuK}\alpha$  graphite monochromatized radiation ( $\lambda = 1.54178$  Å) up to  $2\theta = 130^\circ$ . 739 reflections having  $I \geq 2.5\sigma(I)$  were considered as observed and used in the structure refinement. The structure was solved by MULTAN 80<sup>5</sup> and refined using  $F$  values first with isotropic and then with anisotropic temperature factors for the non-H atoms with SHELX 76<sup>6</sup>. The positions of the H atoms were calculated with C—H distances of 1.08 Å and H—C—H angles of  $109.4^\circ$ . The final  $R$  value is 0.051 for 739 observed reflections. Lists of the atomic coordinates and of the geometrical parameters have been deposited with the Cambridge Crystallographic Data Centre.

Figure 1 is a stereoscopic view of molecule 3, showing the numbering of the atoms (PLUTO<sup>7</sup>).

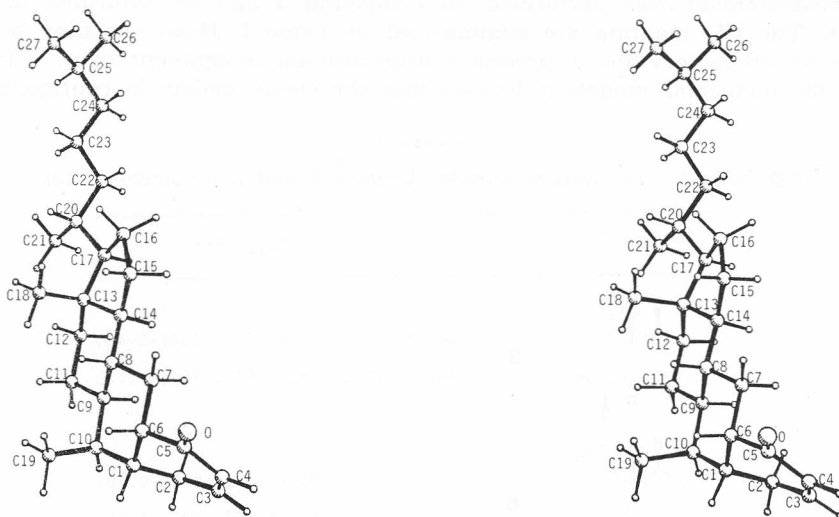
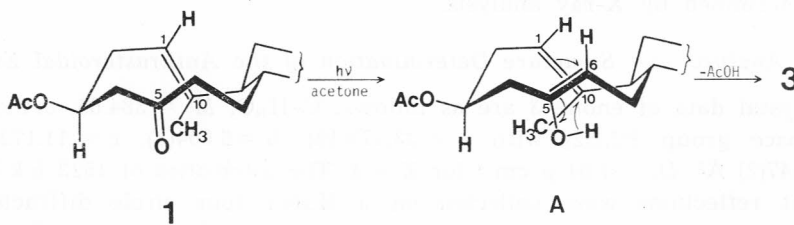


Figure 1. Stereoscopic view of enone 3 in its solid state conformation.

The C(1)—C(2)—C(3) = C(4)—C(5)—C(6) ring has a half-chair conformation with torsion angles 46, -22, 0, -2, 27 and -49°. The two other 6-membered rings are in parallel chair conformations with an approximate local mirror plane containing C(8) and C(9). The pentagonal ring exhibits an envelope conformation with C(13) at the flap.

A plausible mechanistic course for the photochemically induced cyclization of the (*E*)-5,10-*seco*-ketone 1 to enone 3 is shown in Scheme 2.

Scheme 2



The first step involves photoenolization of ketone 1\* (towards the C(6) atom) to produce species A, which then undergoes intramolecular rearrangement *via* a cyclic six-membered transition state (along with acetic acid elimination), resulting in the stereospecific formation of the 1 $\beta$ ,6 $\beta$ ,10 $\beta$ (CH<sub>3</sub>)-anthrasteroidal enone 3 as the final product. Formally, this rearrangement can be considered as a retro-Norrish Typ II process, not observed as yet. In this case the reaction is possible, probably due to entropy factors.

\* About photoenolization of saturated ketones see, for example, ref. 8.

## EXPERIMENTAL

## General

Prep. column chromatography: Silica gel 0.063—0.200 mm. TLC: control of reactions and separation of products on silica gel G (Stahl) with benzene/AcOEt 9 : 1, detection with 50% aq. H<sub>2</sub>SO<sub>4</sub> soln. Melting points are uncorrected. CD spectra: ISA-Jobin-Yvon dichrograph model Mark III at room temperature in acetonitrile at conc. of approximately 0.2 mg/ml. UV spectra: Varian UV Super Scan 3 spectrophotometer:  $\lambda_{\max}$  nm ( $\epsilon$ ). IR spectra: Perkin-Elmer-337 spectrophotometer;  $\nu_{\max}$  in cm<sup>-1</sup>. NMR spectra: Bruker AM-360 (<sup>1</sup>H at 360 MHz, <sup>13</sup>C at 90.55 MHz), CDCl<sub>3</sub> soln. at room temperature, TMS as internal standard; chemical shifts in ppm as  $\delta$  values. Mass spectra (MS): Varian CH7 instrument; in *m/z*.

UV Irradiation of (E)-3 $\beta$ -Acetoxy-5,10-seco-1(10)-cholesten-5-one (1)<sup>4</sup>

A solution of the (E)-5,10-seco-ketone (1) (1.00 g) in acetone (250 ml) was irradiated with a high pressure mercury lamp TQ 150 Z2 (Hanau) at room temp for 3.5 hours. It was then evaporated *in vacuo* and the oily residue (1.2 g) chromatographed on silica gel (50 g). Benzene-diethyl ether (99 : 1) eluted a mixture of (Z)-3 $\beta$ -acetoxy-5,10-seco-1(10)-cholesten-5-one (2) and anthrasteroidal enone 3 (132 mg). Benzene-diethyl ether (98 : 2) eluates gave first 180 mg (18%) of unchanged (E)-seco-ketone 1, *m.p.* 136 °C (from acetone-MeOH) (lit.<sup>4</sup> *m.p.* 136 °C), followed by 24 mg (2.4%) 19 $\alpha$ -methyl-1 $\beta$ ,5-epoxy-5 $\beta$ -cholestan-3 $\beta$ -ol acetate (5). Oxetane 5 was rechromatographed on SiO<sub>2</sub> column to afford a chromatographically (TLC) pure sample (16 mg, 1.6%), oil (lit.<sup>2</sup> oil);  $[\alpha]_D^{20} = +32.5$  ( $c = 1.93\%$ , CHCl<sub>3</sub>); IR (CH<sub>2</sub>Cl<sub>2</sub>): 1730, 1235, 1028; <sup>1</sup>H-NMR: 0.68 (s, CH<sub>3</sub>-18), 0.84 (d, CH<sub>3</sub>-26 and CH<sub>3</sub>-27), 0.87 (d, CH<sub>3</sub>-21), 0.88 (s, CH<sub>3</sub>-19), 2.03 (s, AcO), 3.02 (m, H-C(1)), 5.10 (m, H-C(3)); <sup>13</sup>C-NMR: 171.2 (s, CH<sub>3</sub>COO), 74.3 (d, C-3), 70.0 (s, C-5), 58.5 (d, C-1), 56.6 (d, C-17), 55.0 (d, C-14), 49.3 (d, C-9), 43.5 (s, C-13), 41.0 (s, C-10), 40.3 (t, C-12), 39.5 (t, C-24), 37.8 (d, C-8), 37.5 (t, C-4), 36.1 (t, C-22), 35.8 (d, C-20), 31.8 (t, C-6), 30.2 (t, C-7), 29.7 (t, C-2), 28.0 (d, C-25), 27.9 (t, C-16), 24.6 (t, C-11), 24.1 (t, C-15), 23.9 (t, C-23), 22.8 (q, C-27), 22.6 (q, C-26), 21.4 (q, CH<sub>3</sub>COO), 18.6 (q, C-21), 16.5 (q, C-19), 12.4 (q, C-18). MS: *m/z* = 444 (M<sup>+</sup>).

*Anal.* Calcd. for C<sub>29</sub>H<sub>48</sub>O<sub>3</sub> (*M<sub>r</sub>* = 444.70): C 78.32, H 10.88%; found: C 78.17, H 10.64%.

Benzene-diethyl ether (97 : 3 and 96 : 4) eluted 1 $\alpha$ ,5-epoxy-5 $\alpha$ -cholestan-3 $\beta$ -ol acetate (4)<sup>2-3</sup> (417 mg, 41.7%), *m.p.* 101—102 °C (from acetone) (lit.<sup>2-3</sup> *m.p.* 101—102 °C);  $[\alpha]_D^{20} = +20.0$  ( $c = 1.00\%$ , CHCl<sub>3</sub>); IR (CH<sub>2</sub>Cl<sub>2</sub>): 1732, 1238, 1025; <sup>1</sup>H-NMR: 0.66 (s, CH<sub>3</sub>-18), 0.84 (s, CH<sub>3</sub>-19), 0.85 (d, CH<sub>3</sub>-26 and CH<sub>3</sub>-27), 0.91 (d, CH<sub>3</sub>-21), 2.01 (s, AcO), 3.94 (d, *J* = 6 Hz, H-C(1)), 5.20 (m, H-C(3)); <sup>13</sup>C-NMR: 170.5 (s, CH<sub>3</sub>COO), 88.5 (s, C-5), 83.2 (d, C-1), 66.7 (d, C-3), 56.1 (d, C-14 and C-17), 47.0 (d, C-9), 45.5 (s, C-18), 42.5 (t, C-4), 39.9 (t, C-12), 39.5 (t, C-24), 38.7 (d, C-8), 36.2 (t, C-22), 35.8 (d, C-20), 34.2 (s, C-10), 31.6 (t, C-2), 31.0 (t, C-6), 28.1 (t, C-16), 28.0 (d, C-25), 27.3 (t, C-7), 24.4 (t, C-15), 23.9 (t, C-23), 23.1 (t, C-11), 22.8 (q, C-27), 22.6 (q, C-26), 21.2 (q, CH<sub>3</sub>COO), 18.8 (q, C-21), 11.8 (q, C-18), 11.7 (q, C-19).

*Anal.* Calcd. for C<sub>29</sub>H<sub>48</sub>O<sub>3</sub> (*M<sub>r</sub>* = 444.70): C 78.32, H 10.88%; found: C 78.18, H 10.87%.

Further elution with benzene-diethyl ether (95 : 5, 90 : 10 and 80 : 20) afforded a complex mixture (260 mg, about 25%), from which no definite product could be isolated.

Separation of the Anthrasteroidal Enone 3 from (Z)-3 $\beta$ -Acetoxy-5,10-seco-1(10)-cholesten-5-one (2)

A mixture of 2 and 3 (isolated with benzene-diethyl ether (99 : 1) in the above chromatography procedure) was dissolved in 5 ml MeOH to which 5% methanolic KOH (1 ml) was added, and the solution left overnight at room temperature. The residue (126 mg) obtained after the usual work-up was chromatographed on silica gel (5 g). Elution with benzene gave the anthrasteroidal enone 3 (62 mg, 7.2%),

which was twice recrystallized from acetone-MeOH (43 mg, 5.0<sup>0</sup>%), *m. p.* 136–137 °C;  $[\alpha]_D^{20} = +102.6$  ( $c = 0.30\%$ ,  $\text{CHCl}_3$ ); UV (EtOH): 227 (7700); IR ( $\text{CH}_2\text{Cl}_2$ ): 1670, 1620; <sup>1</sup>H-NMR: 0.66 (s,  $\text{CH}_3$ -18), 0.86 (d,  $\text{CH}_3$ -26 and  $\text{CH}_3$ -27), 0.90 (d,  $\text{CH}_3$ -21), 0.97 (d,  $\text{CH}_3$ -19), 5.97 (d x d,  $J = 10, 2.4$  Hz, H-C(4)), 6.95 (d x d x d,  $J = 10, 6.2, 2.8$  Hz, H-C(3)); <sup>13</sup>C-NMR: 203.4 (s, C-5), 150.2 (d, C-3), 128.5 (d, C-4), 56.3 (d, C-17), 55.9 (d, C-14), 44.3 (d, C-6), 43.3 (s, C-13), 41.2 (d, C-1), 41.0 (d, C-9), 40.1 (t, C-12), 39.5 (t, C-24), 36.3 (d, C-8), 36.2 (t, C-22), 35.7 (d, C-20), 33.5 (d, C-10), 30.2 (t, C-7), 28.2 (t, C-16), 27.9 (d, C-25), 26.4 (t, C-11), 23.9 (t, C-15 and C-23), 23.8 (t, C-2), 22.8 (q, C-27), 22.6 (q, C-26), 18.7 (q, C-21), 14.5 (q, C-19), 12.0 (q, C-18). MS: *m/z* 384 (*M*<sup>+</sup>).

*Anal.* Calcd. for  $\text{C}_{27}\text{H}_{44}\text{O}$  ( $M_r = 384.65$ ): C 84.31, H 11.53<sup>0</sup>%; found: C 84.14, H 11.38<sup>0</sup>%.

Benzene-diethyl ether (90 : 10) eluted (Z)-3 $\beta$ -hydroxy-5,10-seco-1(10)-cholesten-5-one (52 mg, 5.7<sup>0</sup>%), *m. p.* 118 °C (from MeOH) (lit.<sup>4</sup> *m. p.* 116–118 °C), which was acetylated (with  $\text{Ac}_2\text{O}$  in pyridine) to (Z)-3 $\beta$ -acetoxy-5,10-seco-1(10)-cholesten-5-one (2), *m. p.* 138 °C (from acetone) (lit.<sup>4</sup> *m. p.* 138 °C).

### Catalytic Hydrogenation of the Anthrasteroidal Enone 3

A solution of the anthrasteroidal enone 3 (100 mg) in ethanol (25 ml) was hydrogenated in the presence of Adams catalyst (10 mg) at room temp. for 1 hour. The mixture was filtered through a Celite mat and the filtrate evaporated under reduced pressure, to give the saturated anthrasteroidal ketone 6 (100 mg, 100<sup>0</sup>%), which after recrystallization from acetone (91 mg, 90.5<sup>0</sup>%) had *m. p.* 102–104 °C;  $[\alpha]_D^{20} = +82.5$  ( $c = 0.40\%$ ,  $\text{CHCl}_3$ ); IR (KBr): 1705; <sup>1</sup>H-NMR: 0.66 (s,  $\text{CH}_3$ -18), 0.86 (d,  $\text{CH}_3$ -26 and  $\text{CH}_3$ -27), 0.91 (d,  $\text{CH}_3$ -21), 0.95 (d,  $\text{CH}_3$ -19); <sup>13</sup>C-NMR: 215.3 (s, C-5), 56.3 (d, C-17), 56.0 (d, C-14), 48.5 (d, C-6), 45.0 (d, C-1), 43.3 (s, C-13), 40.9 (d, C-9), 40.1 (t, C-12), 39.5 (t, C-24), 37.9 (t, C-4), 37.3 (d, C-8), 36.2 (t, C-22), 35.7 (d, C-20), 34.0 (d, C-10), 30.9 (t, C-7), 28.2 (t, C-16), 28.0 (d, C-25), 26.4 (t, C-3), 26.1 (t, C-11), 25.7 (t, C-2), 23.9 (t, C-15 and C-23), 22.8 (q, C-27), 22.6 (q, C-26), 18.7 (q, C-21), 14.7 (q, C-19), 12.0 (q, C-18).

*Anal.* Calcd. for  $\text{C}_{27}\text{H}_{46}\text{O}$  ( $M_r = 386.67$ ): C 83.87, H 11.99<sup>0</sup>%; found: C 83.69, H 11.86<sup>0</sup>%.

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**IZVOD**

**Transanularna fotociklizacija kod (E)-3 $\beta$ -acetoksi-5,10-seko-1(10)-holesten-5-ona**

*Ljubinka Lorenc, Vladimir Pavlović, Mihailo Lj. Mihailović, Bernard Tinant,*

*Jean-Paul Declercq i Maurice Van Meerssche*

UV ozračivanjem (E)-5,10-seko-steroidnog ketona 1 u acetonskom rastvoru vrši se, pored ranije opisanih transformacija — tj. E/Z-izomerizacije i intramolekulske Paterno-Büchi-jeve reakcije (koja daje 1 $\alpha$ ,5 $\alpha$ -oksetan 4 u prinosu od 42% i 1 $\beta$ ,5 $\beta$ -oksetan 5 u prinosu od 2—3%) — još i transanularna ciklizacija (praćena eliminacijom sirćetne kiseline), pri čemu se gradi antrasteroidni enon 3 (u prinosu od oko 7%). Struktura proizvoda 3 izvedena je na osnovu spektralnih podataka a potvrđena je rendgenskom strukturnom analizom.