

## ERRATA

**Investigation of Selectivity of a Sulfide Ion-selective Electrode [Croat. Chem. Acta]**

**59 (2) (1986) 451—462 by Dinko Tuhtar.**

In page 460 the following Table should be added:

TABLE I  
Published Values of  $k_{S, X}^{\text{pot}}$ <sup>a</sup>

Anion, $X^{n-}$	$k_{S, X}^{\text{pot}}$
Chloride	$10^{-27} \leq 10^{-3} \quad 10^{-8} - 10^{-10} \quad 6.3 \times 10^{-15} \quad 3 \times 10^{-31} \quad 7 \times 10^{-30}$ $10^{-15}$
Bromide	$5 \times 10^{-23} \leq 10^{-3} \quad 10^{-8} - 10^{-10} \quad 2.5 \times 10^{-12} \quad 5 \times 10^{-26} \quad 2 \times 10^{-26}$ $10^{-13}$
Iodide	$1.3 \times 10^{-15} \quad 10^{-3} \quad 10^{-8} - 10^{-10} \quad 1.6 \times 10^{-8} \quad 2 \times 10^{-18} \quad 10^{-9}$
Cyanide	$\leq 10^{-3} \quad 10^{-8} - 10^{-10} \quad 10^{-2} \quad 6.3 \times 10^{-4}$
Hydroxide	$4 \times 10^{-32} \quad 6.3 \times 10^{-17} \quad 8 \times 10^{-27} \sim 10^{-35} \quad 10^{-16}$
Thiosulfate	$\leq 10^{-3} \quad 10^{-8} - 10^{-10} \quad 2 \times 10^{-11}$
Sulfate	$\leq 10^{-3} < 10^{-21}$
Carbonate	$< 10^{-3}$
Hydrocarbonate	$\leq 10^{-3}$
Thiocyanate	$8 \times 10^{-13} \quad 10^{-3}$
Fluoride	$\leq 10^{-3} \quad 10^{-16}$
Phosphate	$8.0 \times 10^{-17}$

<sup>a</sup> Taken from ref. (16) and references therein.