

Original papers and short communications on all aspects of the chemistry of organic compounds, including their synthesis, properties, reactions, and applications.

- J. J. Aaron, A. Tine, C. Villiers, C. Párkányi and D. Bouin **Electronic Absorption and Fluorescence Spectra of Indole Derivatives. Quantitative Treatment of the Substituent Effects and a Theoretical Study**

The electronic absorption and fluorescence spectra of nine substituted indoles are discussed. Satisfactory linear Hammett correlations have been established for the electronic absorption and fluorescence maxima. PPP (LCI-SCF-MO) calculations for the indoles have been carried out and the results used to interpret the experimental data.

157—168

- J. Bertrán, J. Anguiano, and A. Oliva

The Electrophile Desolvation Process in Nucleophile-Electrophile Combination Reactions

The protonation and methylation of ethylene in different solvents are studied, by means of the MINDO/3 method, in order to check the hypothesis of partial desolvation of the electrophile at the transition state. It has been found that degree of desolvation depends on both the electrophile and the solvent. The consequences of these results are discussed.

169—174

- N. Bodor, A. J. Harget and E. Phillips

Structure — Activity Relationships in the Antiinflammatory Steroids: A Pattern Recognition Approach

A pattern recognition method was applied to a set of steroids for which experimental results using two different bioassays were known. The resulting structure-activity relationships are discussed. The method predicts a synergistic effect to be in operation between certain pairs of substituents.

175—183

G. Frenking and
H. Schwarz

SCF-Calculations as Tool to Interpret Reaction Pathes — The Cases of CO₂ Loss from Ionized ClCH₂—COOCH₃ and C₂H₄ Loss from Ionized (CH₃)₂N—COCH₂X

The reaction courses of CO₂ loss from ionized ClCH₂—COOCH₃ and C₂H₄ loss from ionized (CH₃)₂N—COCH₂X have been determined by semi-empirical and ab-initio SCF calculations.

185—190

A. Furka

An Empirical Formula Expressing the Mutual Dependence of C—C Bond Distances

An empirical formula is suggested to describe the mutual dependence of the length of the bonds formed by a central atom with other equivalent carbon atoms. Reference and deduced systems: diamond, graphite, cumulene and polyyne chains, and intermediate structures between them. A geometrical representation of the formula is also discussed.

191—197

A. Furka

Comparison of the Heats of Formation of Hydrocarbons

An equation is suggested to convert heats of formation of hydrocarbons into relative enthalpies which numerically demonstrate the structure dependent stabilizing and destabilizing effects.

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P. Ilić and N.
Trinajstić

Structural Normalization of the Topological Resonance Energy

Normalizing method for topological resonance energy values of conjugated compounds is presented. It is based on a graph-theoretical procedure which observes the number, the size, and the organization of circles of the respective conjugated compound.

203—213

A. Jurić and N.
Trinajstić

Topological Resonance Energies of Thiocoumarins

Topological resonance energies of thiocoumarin, thiochromone, and related structures are reported. Comparison of the calculated values gave information about the degree of distortion of benzene rings in these compounds.

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E. C. Kirby

The Use of a Micro-Computer as an Aid to Theoretical Calculations.

Some programs have been written in the computer language BASIC to help in the applications of simple Hückel theory.

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G. Marconi and G. Orlandi

Vibronic Coupling in Carbonyl Compounds

The vibronic coupling integrals for Formaldehyde, Glyoxal, and Biacetyl, calculated using MINDO/3 and CNDO/S programs, are found to provide a satisfactory basis for the explanation of the radiative and radiationless behavior of carbonyl compounds.

225—235

**Ov. Mekenyanyan,
D. Bonchev, and
N. Trinajstić**

Structural Complexity and Molecular Properties of Cyclic Systems with Acyclic Branches

The degree of structural complexity of cyclic molecules with acyclic branches is expressed by means of rules based on the sum of the distances in the corresponding graphs. It is shown that these topological rules are well reflected in the properties of isomeric alkyl-benzenes.

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**M. Mintas,
K. Jakopčić, and
L. Klasinc**

Photoelectron Spectroscopy of Some trans-2,2'-Disubstituted Stilbenes

The He I photoelectron spectra of trans-stilbene, -2,2'-dichlorostilbene, -2-amino-2'-chlorostilbene, -2,2'-diaminostilbene, and -2,2'-dinitrostilbene are recorded. Their electronic structure can well be described on the basis of the composite molecule method and simple Hückel molecular orbital calculations.

263—267

**U. Norinder,
D. Tanner, and
O. Wennerström**

Dianions from Cyclophanes with Unsaturated Bridges; Cyclic Conjugation in 22—38 π -Electron Perimeter Systems

A series of cyclophanes with unsaturated bridges has been synthesized. Several of these cyclophanes containing $4n$ perimeters can be reversibly reduced to their dianions, a process which conveniently can be studied by cyclic voltammetry. A

- 269—280
- I. Novak,
L. Klasinc, G.
Karminski-Zamola
and K. Jakopčić**
- good correlation between calculated energies of the LUMO orbitals and the reversible reduction potentials has been obtained. Recent theories of resonance energies in neutral molecules and dianions with conjugated perimeters have been tested on the unsaturated cyclophanes.
- Photoelectron Spectroscopy of Heterocycles. 1-(2-Furyl)-2-arylethenes**
- HeI photoelectron spectra of 2-ethenylfuran, cis-2-(2-phenylethenyl)-furan, trans-2-(2- α -naphthylethenyl)-furan, cis-2-(2-phenylethenyl)-5-methylfuran and trans-2-(2- α -naphthylethenyl)-5-methylfuran were recorded. Their assignment is attempted.
- 281—287
- P. R. Surján**
- Conformation Analysis in Light of Localization and Delocalization**
- The origin of barriers to internal rotations around single bonds is discussed in terms of localized molecular orbitals and through space and through bond delocalization interactions. Bond-bond pair potentials for giving rise to barrier forces are analysed.
- 289—295
- H. Vogler**
- On the Existence of Biradical-Ionic States of Donor-Acceptor Cyclophanes. A Simple MO-Theoretical Study**
- The ground and lowest excited states of donor-acceptor cyclophanes are investigated by means of MO-theory. Conditions for the occurrence of biradical-ionic ground states are derived.
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