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Kinetics and Mechanism of Interactions Between Iron(III) and Desferrioxamine B. The Formation and Hydrolysis of Ferrioxamine B in Acidic Aqueous Solution

The kinetics of the formation and hydrolysis of ferrioxamine B complexes have been studied in acidic aqueous solution (0.001—1.0 M HCl) at 25.0 °C, $\mu = 1.0$ M (maintained by NaCl). Two stage kinetics have been observed in both the formation and in the hydrolysis reactions. The proposed reaction model involves formation/ /hydrolysis of bidentate, tetradentate, and hexadentate bonded desferrioxamine B to iron(III). . . . . 61—77

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DOCUMENTA CHEMICA YUGOSLAVICA

CROATICA CHEMICA ACTA

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Zagreb, 15. travanj 1983

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Zagreb, April 15, 1983

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