

M. Mintas and
W. G. Filby

Surface Charging of Zinc Oxide During
XPS Examination

The magnitude of the charge is 0.4 to 0.6 eV at room temperature and decreases at higher temperatures to almost zero

403—406

C. J. Porter,
C. J. Proctor,
T. Ast, and
J. H. Beynon

Charge Stripping Reactions in Mass Spectrometry: A Study of Diatomic and Triatomic Inorganic and Organic Ions

Charge stripping reactions have been studied for a variety of diatomic and triatomic inorganic and organic ions. The method is described as fast and straightforward. Ionisation energy and the relative cross-sections for the charge stripping processes have been determined. Cases of possible interferences are described and discussed

407—419

K. S. Siddiqi,
N. S. Neelam,
F. R. Zaidi, and
S. A. A. Zaidi

Complexes of Some Group(IV) Metal Halides with 5-Aminoindazole

The synthesis and characterization of Sn(IV) halides, Ge(IV), Ti(IV), and Zr (IV) chloride complexes of the type $\text{MX}_4 : \text{L}_{1-2}$ with 5-aminoindazole is described

421—425

A. Mehlhorn
and J. Fabian

Aromatic Electronic Delocalization of Benzo Derivatives of Five-membered Heterocycles

The aromaticity of some bicyclic heterocycles is discussed with the aid of the configuration analysis of the expansion coefficient matrices and the similarity analysis of the density matrices

427—434

I. Butula,
B. Zorc und
V. Vela

Reaktionen mit 1-Benzotriazolcarbonsäurechlorid. VII. Die Umsetzung mit Aminosäuren

Das 1-Benzotriazolcarbonsäurechlorid reagiert mit Aminosäuren unter Bildung von *N*-(1-Benzotriazolylcarbonyl)-amino-säuren am besten in Dioxan . . .

435—440

B. Zorc und
I. Butula

Reaktionen mit *N*-(1-Benzotriazolyl-carbonyl)-amino-säuren. I. Synthese von Hydantoinen und Hydantoinsäure-amiden

Die mit der BTCO-Gruppe geschützten Aminosäure-chloride werden mit verschiedenen Aminen indirekt oder direkt in die entsprechend substituierten Hydantoinsäure-amide umgesetzt . . .

441—449

K. Tabaković
and
I. Tabaković

Electrochemical Synthesis of Heterocyclic Compounds. XI. Annulation of Coumarin Ring via Cathodic Reduction of 3-Nitrocoumarin Derivatives

The reaction of the carbonyl compound, as an electrophile, with an electrochemically generated amino group as an nucleophile has been studied by employing the annelation of the coumarin ring via cathodic reduction of 3-nitrocoumarin derivatives by controlled potential. The reduction reactions of several 3-nitrocoumarine derivatives are described . . .

451—458

V. C. Koshy
and
S. G. Tandon

Extractive Spectrophotometric Determination of Vanadium(V) with *N*-*p*-Chlorophenyl-2-naphthohydroxamic Acid and Investigation of Its Solid Complex

The extractive spectrophotometric determination of vanadium(V) by using *N*-*p*-chlorophenyl-2-naphthohydroxamic acid is described as a simple, selective and sensitive quantitative method. The solid complex, $\text{VOCl}(\text{C}_{17}\text{H}_{11}\text{NO}_2\text{Cl})_2$, was prepared and characterised by melting point, elemental analysis, visible and infrared spectra . . .

459—464

R. Kant,
R. Srivastava,
and
O. Prakash

Palladium Ternary Complex with Chromazurol S and Cetyltrimethylammonium Bromide and Cetylpyridinium Bormide

Formation of green coloured ternary complexes between palladium, chromazurol S and cetyltrimethylammonium bromide or cetylpyridinium bromide has been studied and proposed as a spectrophotometric method for micro-determination of palladium. The method is described as sensitive, precise and selective

465—472

V. Simeon,
Z. Radić, and
E. Reiner

Inhibition of Cholinesterases by the Oximes P2AM and Toxogonin

The reversible inhibition of electric eel acetylcholinesterase by P2AM (2-(hydroxyamino)methyl-1-methyl-pyridinium chloride) and Toxogonin (1,1'-[oxybis(methylene)]bis (4-(hydroxyimino)-methyl-pyridinium dichloride) was studied using acetylthiocholine as substrate. The correlation between the degree of inhibition and acetylcholine and oxime concentrations fits a theoretical model which postulates that the substrate and the inhibitor bind to two sites on the enzyme: the catalytic site and an allosteric, substrate-inhibition site

473—480

S. Lapanje,
M. Simić, and
A. Pavlič

Interactions of α -Chymotrypsinogen A with Some Alkylureas

The interactions of α -chymotrypsinogen A with urea, methyl-, N,N' -dimethyl-, ethyl-, N,N' -diethyl-, and propylurea were studied by means of calorimetry and circular dichroism. The transfer of the protein from water to aqueous urea and methylurea solutions is accompanied by release of heat, whereas the transfer of the same protein to solutions of other alkylureas is characterized by consumption of heat. The alkylureas are found to be less efficient denaturants than urea

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