# **Original articles**

# Changes of salivary biomarkers under different storage conditions: effects of temperature and length of storage

Tomás Barranco¹, Camila P Rubio¹, Asta Tvarijonaviciute¹, Mónica Rubio², Elena Damia², Elsa Lamy³, Ramón Cugat⁴, José J Cerón¹, Fernando Tecles\*¹. Damián Escribano⁵

#### Abstract

**Introduction**: In this report, we aimed to examine the stability of various analytes in saliva under different storage conditions.

**Materials and methods**: Alpha-amylase (AMY), cholinesterase (CHE), lipase (Lip), total esterase (TEA), creatine kinase (CK), aspartate aminotransferase (AST), lactate dehydrogenase (LD), lactate (Lact), adenosine deaminase (ADA), Trolox equivalent antioxidant capacity (TEAC), ferric reducing ability (FRAS), cupric reducing antioxidant capacity (CUPRAC), uric acid (UA), catalase (CAT), advanced oxidation protein products (AOPP) and hydrogen peroxide ( $H_2O_2$ ) were colorimetrically measured in saliva obtained by passive drool from 12 healthy voluntary donors at baseline and after 3, 6, 24, 72 hours, 7 and 14 days at room temperature (RT) and 4 °C, and after 14 days, 1, 3 and 6 months at -20 °C and -80 °C.

**Results**: At RT, changes appeared at 6 hours for TEA and  $H_2O_2$ ; 24 hours for Lip, CK, ADA and CUPRAC; and 72 hours for LD, Lact, FRAS, UA and AOPP. At 4 °C changes were observed after 6 hours for TEA and  $H_2O_2$ ; 24 hours for Lip and CUPRAC; 72 hours for CK; and 7 days for LD, FRAS and UA. At -20 °C changes appeared after 14 days for AST, Lip, CK and LD; and 3 months for TEA and  $H_2O_2$ . At -80 °C observed changes were after 3 months for TEA and  $H_2O_3$ .

**Conclusions**: In short-term storage, the analytes were more stable at  $4 \, ^{\circ}$ C than at room temperature, whereas in long-term storage they were more stable at  $-80 \, ^{\circ}$ C than at  $-20 \, ^{\circ}$ C.

**Keywords**: biomarker; enzyme stability; oxidative stress; saliva; temperature

Received: September 07, 2018

Accepted: December 21, 2018

# Introduction

Interest in saliva assays for clinical purposes has increased during recent years because this fluid has important advantages: its collection is easy, does not produce evident stress or pain and does not require expensive material or medical personnel. The main areas in which saliva can be used for testing are psychology and stress research, endocrinology, occupational and sports medicine, drug monitoring, metabolism and oxidative status evaluation, immunology and inflammation (1).

The use of salivary biomarkers for diagnostic purposes would be of benefit if standardised procedures for saliva collection were used, as well as the validation/verification of the methods performed in saliva. In addition, knowledge of how the storage conditions can affect the stability of measured analytes is of major importance, especially as saliva is usually less stable than plasma (due to bacterial multiplication, viscosity or extensive proteolytic cleavage by endogenous and exogenous proteas-

<sup>&</sup>lt;sup>1</sup>Interdisciplinary Laboratory of Clinical Analysis (Interlab-UMU), Campus of Excellence Mare Nostrum, University of Murcia, Murcia, Spain

<sup>&</sup>lt;sup>2</sup>Department of Animal Medicine and Surgery, University CEU Cardenal Herrera, Valencia, Spain

<sup>&</sup>lt;sup>3</sup>Institute of Mediterranean Agricultural and Environmental Sciences (ICAAM), University of Évora, Évora, Portugal

<sup>&</sup>lt;sup>4</sup>Arthroscopy and Traumatology Unit of the Quirón Hospital, Barcelona, Spain

<sup>&</sup>lt;sup>5</sup>Department of Animal and Food Science, Faculty of Veterinary Medicine, Autonomous University of Barcelona, Barcelona, Spain The first two authors contributed equally to this work.

<sup>\*</sup>Corresponding author: ftecles@um.es

es) (2). Stability would be of particular importance when retrospective studies or studies involving multiple experimental sampling time-points are designed, since they usually involve the storage of samples, and an inappropriate temperature can affect enzymatic activities in saliva samples during sampling and storage (3).

Salivary α-amylase (AMY) increases in situations of acute stress and activation of the sympathetic nervous system (SNS), due to different psychological causes or physical efforts (4). Changes in salivary cholinesterase (CHE) activity have been described in Alzheimer's disease and they have been related to situations of stress, although it is traditionally measured as a biomarker of anti-CHE insecticides exposure (5,6). Salivary lipase (Lip) secretion also seems to be promoted by the activation of the SNS (7). The total esterase activity (TEA) of saliva comprises several enzymatic activities, and it is increased in situations of physical stress (8). Creatine kinase (CK), aspartate transaminase (AST) and lactate dehydrogenase (LD) can increase in human saliva in conditions such as intensive exercise (9). Lactate (Lact) is considered a marker of muscle function and its quantification in saliva is important in internal and sport medicine to monitor the maximum performance level of athletes (10). Adenosine deaminase (ADA) is involved in various processes related with the immune system, it is considered a biomarker of inflammation and it has been found to change in saliva in oral diseases (11).

In situations of oxidative stress, reactive oxygen species (ROS) are produced in high amounts that cannot be removed by antioxidants. Total antioxidant capacity (TAC) represents the antioxidant status of a sample and comprises the sum of the concentrations of almost all non-enzymatic antioxidants (12). It can be evaluated in saliva by different assays, such as ferric reducing ability of saliva (FRAS), cupric reducing antioxidant capacity (CUPRAC), and the 6-hydroxy-2,5,7,8-tetramethylchroman-2-carboxylic acid (Trolox) equivalent antioxidant capacity (TEAC) (13). The Trolox equivalent antioxidant capacity, FRAS and CUPRAC are assays based on the ability of antioxidants present in a sample to reduce or inhibit oxidized products gen-

erated in the assay (14). The Trolox equivalent antioxidant capacity measures the capacity of the sample to reduce the 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulfonic acid) (ABTS) radical, whereas FRAS and CUPRAC measure the ability of a sample to reduce Fe<sup>3+</sup> and Cu<sup>2+</sup> to Fe<sup>2+</sup> and Cu<sup>1+</sup>, respectively. They have been evaluated in saliva in patients with diseases such as oral lichen planus (13). In addition, individual components of the antioxidant system such as uric acid (UA) and catalase (CAT) can be measured. Uric acid is the final metabolite of purines and represents approximately 70% of salivary TAC (15). It can increase in hypoxia due to the appearance of oxidant metabolites and there is evidence that salivary UA is altered in diseases such as oral lichen planus or diabetes (13,16). Catalase is an enzyme capable of removing ROS from saliva and its activity is altered in patients with different diseases such as human immunodeficiency virus (17). Components of the oxidant system can also be measured in saliva, namely the advanced oxidation protein products (AOPP) and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>). Advanced oxidation protein products represent a sensitive biomarker of oxidative-modified proteins and it has been measured in human saliva before and after acute resistance exercise (18). Hydrogen peroxide is a reactive species produced during normal metabolism, with increased concentrations found in situations of inflammation and tissue damage (19).

Studies on the stability for some of these analytes already have been published; such as stability of AMY at room temperature (RT), 4 °C and after freeze-thaw cycle; stability of LD and AST at RT, 4 °C and – 20 °C for 3 months and at – 80 °C for 28 days, or the stability of AOPP for 4 weeks at – 20 °C and – 80 °C (2,3,20-25). However, long-term studies longer than 3 months in which different storage conditions were compared have not been published. In addition, to the authors' knowledge, there have not been studies about the stability of CHE, Lip, TEA, CK, Lact, ADA, TEAC, UA, CAT and  $H_2O_2$  in human saliva under different conditions of storage.

Our hypothesis was that the analytes in saliva can have different changes depending of the sample storage conditions and used times. In this report, we aimed to examine the stability of various analytes (AMY, CHE, Lip, TEA, CK, AST, LD, Lact, ADA, TEAC, FRAS, CUPRAC, UA, CAT, AOPP and  $\rm H_2O_2$ ) in saliva under different storage conditions.

#### Material and methods

#### **Subjects**

This experimental study was conducted at the Interdisciplinary Laboratory of Clinical Analyses of the University of Murcia, Spain, from June to December 2017. The experimental protocol was approved by the Investigation Ethics Committee of the University of Murcia (Spain) and followed the Declaration of Helsinki of the World Medical Association for research with humans.

Twelve subjects, six men (29-58 years) and six women (28-56 years) participated in this study. They were healthy volunteers from the personnel of Murcia University. All participants filled a questionnaire in which they were asked about habits, the presence and description of any acute or chronic disease, any type of symptom in the days before the experiment took place, or whether they were receiving or they had recently received any medical treatment. The inclusion criteria for the participants were that they should be adults without any acute or chronic disease, not under physician's care for any disease for the last 6 months, not receiving any medication (including hormones, steroids or non-steroidal anti-inflammatories), food supplements (vitamins or sport supplements) and not being smokers or alcohol consumers. In addition, subjects should not have oral diseases, such as gingivitis or periodontitis, after complete examination of the oral cavity performed by an oral medicine professional.

#### **Methods**

The participants themselves performed sample collection. All participants received detailed information by oral communication and written guidelines about the aims and experimental protocol, the saliva collection procedure, and signed written consent. They were informed about the need to

avoid coughing or clearing the throat into the collection tube and were to abstain from brushing teeth or using mouthwash, ingesting any food or chewing gum for 1 hour before saliva collection.

The participants rinsed their mouth with water five minutes before saliva collection. Then, unstimulated saliva was collected by passive drool in the absence of chewing movements into 10 mL plain tubes (Proquilab, Murcia, Spain). Collection started in all subjects at 9:30 am and lasted between 2 and 5 minutes. The volunteers sat in a relaxed position throughout the sampling procedure to avoid any stress. Between 3 to 5 mL of saliva was obtained from each participant, all samples were checked for blood contamination by visual inspection and no reddish samples indicating blood contamination were included in the study. Immediately after collection, the saliva samples were centrifuged (Universal 320R, Hettich, Tuttlingen, Germany) at 5000xg and 4 °C for 5 minutes, then the supernatant was collected and divided into aliquots, discarding the sediment. To evaluate the effects of different storage conditions, 19 aliquots of each sample were prepared: (a) 1 aliquot was freshly analysed (baseline) and used as the reference value for all analytes; (b) 4 aliquots of each sample were stored at room temperature (RT); (c) 6 aliquots were refrigerated at 4 °C, (d) 4 were frozen at – 20 °C and (e) 4 were frozen at – 80 °C. The aliquots stored at RT were analysed at 3 (T1), 6 (T2), 24 (T3) and 72 (T4) hours after sampling; the refrigerated aliquots were analysed at T1, T2, T3 and T4, and in addition 7 (T5) and 14 days (T6) from collection. Finally, the aliquots stored at - 20 °C and at - 80 °C were analysed 14 days (T6), 1 (T7), 3 (T8) and 6 months (T9) from collection.

The analytical methods used, which were adapted in the authors' laboratory for saliva samples, as well as their fundamentals, details of the reagents and analytical performance, appear in Table 1. All assays were performed on an automated biochemistry analyser (Olympus AU400, Olympus Diagnostica GmbH, Ennis, Ireland) at 37 °C. Manufacturers' control solutions of two different values were used for the quality control analysis of AMY, Lip, CK, AST, LD, Lact and UA (Beckman Coulter, lot 0037 and 0038) and one control solution was used

TABLE 1. Analytical methods used and results of their analytical performance in human saliva

Paramylase   Caamylase   Caamylase caamylase   Caamylase caa	Method	Reference in which the method was validated for saliva or basis of the method (in those assays in which validation was performed specifically for this research)	Manufacturer or reagents (in home-made assays)	Intra-assay CV (%)	Inter-assay CV (%)	Linearity (R²)	TLOD
Hydrolysis of ATCI to thiccholine in presence in MDDNB, in 0.1M  DTNB; non-enzatic hydrolysis of ATCI must be phosphate buffer sham ATCI and the subtracted of an experiment by the phosphate buffer by phosphate buffer actions calcidable and father and subtracted elipses glycerol thosphate exceptions calcidable and peroxidase peroxidase and peroxidase peroxidase (25)  Deamination of adenosine to inosine, which is phosphorylase (PNI). Hypoxanthine is then converted to una codiace (25) by xanthine accidant dividuogen peroxida (10) by xanthine accidant dividuogen peroxida (10) becomerted of una codiace (25) by a parthine accidant dividuogen peroxida (10) becomerted of una codiace (25) by a parthine accida with N-Ethyl-N-(21-hydroxy-3-sufforoxy)-3-methylaniline (EHSPT) and 4-aminoantityrine (4-AA) in the presence of peroxidase (POID) to generate quinnone dye  (10) Beckman Coulter*  (25) 8 6.1°							
Hydrolysis of ATCI to thiocholine in presence ImM DTNB. in 0.1M  DTNB, non-enzymatic Ardroloysis of ATCI must be phosphate buffer 55mM ATCI to thiocholine in presence subtracted to 2-monoglyceride monoglyceride is hydrolyzed to 2-monoglyceride monoglyceride is the measured by coupled enzyme reactions catalyzed by enonoglyceride in pass, glycerol phosphate oxidase and peroxidase  (8) 0.1M Tris-HCI buffer A-nitrophenyl 2.46 5.18 > 0.999  (25) Beckman Coulter* < 6.0 < 11.0 > 0.99  Deamination of adenosine to inosine, which is converted to unic acid and hydrogen peroxide phosphorylase (PIQ). Hypoxanthine is then converted to unic acid and hydrogen peroxide phosphorylase (PIQ). Hypoxanthine is then converted to unic acid and hydrogen peroxide phosphorylase (PIQ). Hypoxanthine is then converted to unic acid and hydrogen peroxide phosphorylase (PIQ). Hypoxanthine is then converted to unic acid and hydrogen peroxide yautine (EHSPT) and 4-aminoantipyrine (EHSPT) and 4-aminoantipyrine (EHSPT) and 4-aminoantipyrine (EAAA) in the presence of peroxidase (POD) to generate quinone dye	α-amylase	(24)	Beckman Coulter <sup>a</sup>	< 3.0	< 3.0	> 0.99	11.7 U/L
1,2-Diglyceride is hydrolyzed to 2-monoglyceride and faty acid. The 2-monoglyceride inpase, glycerol kinase, glycerol phosphate oxidase and peroxidase  (8)  (15)  (15)  (25)  (25)  (25)  (25)  (25)  (25)  (25)  (25)  (25)  (25)  (26)  (27)  (27)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (29)  (20)	holinesterase	Hydrolysis of ATCI to thiocholine in presence DTNB; non-enzymatic hydrolysis of ATCI must be subtracted		4.5 <sup>b</sup>	6.2 <sup>b</sup>	0.90°	1.4 <sup>d</sup> µM/ mL/min
(25)  Beckman Coultera  Converted to hypoxanthine by purine nucleoside phosphorylase (PNP). Hypoxanthine by turther reacted with N-Ethyl-N-(2-hydroxy-3-s-uflopropyl)-3-methylaniline (EHSPT) and 4-minoantipyrine (4-AA) in the presence of peroxidase (POD) to generate quinone dyse  (10)  Beckman Coultera  4-nitrophenyl  2.46  5.18  5.0.99  5.0.99  6.10  7.3b  6.1b  7.3b  6.1b  7.3c  6.1b  7.3c  6.1b  7.3c  6.1b  7.3c  6.1b  7.3c  6.1b  7.3c  6.1d  7.3c  6.1d  7.3c  6.1d  7.3c  6.1d  7.3c  6.1d  7.3c  6.1d  7.3c  7	Lipase	1,2-Diglyceride is hydrolyzed to 2-monoglyceride and fatty acid. The 2-monoglyceride is then measured by coupled enzyme reactions catalyzed by monoglyceride lipase, glycerol kinase, glycerol phosphate oxidase and peroxidase	Beckman Coulter <sup>a</sup>	4.3 <sup>b</sup>	5.7 <sup>b</sup>	> 0.99°	1.0 <sup>d</sup> U/L
(25) Beckman Coulter <sup>a</sup> < 6.0 < 11.0 > 0.99  Deamination of adenosine to inosine, which is converted to hypoxanthine by purine nucleoside phosphorylase (PNP). Hypoxanthine is then converted to hypoxanthine is then converted to the hypoxanthine is then converted to the hypoxanthine is then converted to the hypoxanthine is then hypoxanthine is the hypoxan	otal esterase activity	(8)		2.46	5.18	> 0.99	0.8 U/L
Deamination of adenosine to inosine, which is converted to hypoxanthine by purine nucleoside phosphorylase (PNP). Hypoxanthine is then converted to uric acid and hydrogen peroxide (H <sub>2</sub> O <sub>2</sub> ) by xanthine oxidase (XOD). H <sub>2</sub> O <sub>2</sub> is further reacted with N-Ethyl-N-(2-hydroxy-3-sulfopropyl)-3-methylaniline (EHSPT) and 4-aminoantipyrine (4-AA) in the presence of peroxidase (POD) to generate quinone dye (10) Beckman Coulter³ 2.86 5.23 0.99	reatin kinase	(25)	Beckman Coulter <sup>a</sup>	> 8.0	< 9.0	> 0.99	2.0 U/L
Deamination of adenosine to inosine, which is converted to hypoxanthine by purine nucleoside phosphorylase (PNP). Hypoxanthine is then converted to uric acid and hydrogen peroxide (H <sub>2</sub> O <sub>2</sub> ) by xanthine oxidase (XOD). H <sub>2</sub> O <sub>2</sub> is further reacted with N-Ethyl-N-(2-hydroxy-3-sulfopropyl)-3-methylaniline (EHSPT) and 4-aminoantipyrine (4-AA) in the presence of peroxidase (POD) to generate quinone dye (10) Beckman Coultera 2.86 5.23 0.99	Aspartate ransaminase	(25)	Beckman Coulter <sup>a</sup>	< 6.0	< 11.0	> 0.99	3.0 U/L
Deamination of adenosine to inosine, which is converted to hypoxanthine by purine nucleoside phosphorylase (PNP). Hypoxanthine is then converted to uric acid and hydrogen peroxide (H <sub>2</sub> O <sub>2</sub> ) by xanthine oxidase (XOD). H <sub>2</sub> O <sub>2</sub> is further reacted with N-Ethyl-N-(2-hydroxy-3-sulfopropyl)-3-methylaniline (EHSPT) and 4-aminoantipyrine (4-AA) in the presence of peroxidase (POD) to generate quinone dye (10) Beckman Coulter <sup>a</sup> 2.86 5.23 0.99	Lactate ehydrogenase	(25)	BioSystems <sup>e</sup>	< 1.0	< 3.0	> 0.99	10.0 U/L
(10) Beckman Coulter <sup>a</sup> 2.86 5.23 0.99	Adenosine deaminase	Deamination of adenosine to inosine, which is converted to hypoxanthine by purine nucleoside phosphorylase (PNP). Hypoxanthine is then converted to uric acid and hydrogen peroxide (H <sub>2</sub> O <sub>2</sub> ) by xanthine oxidase (XOD). H <sub>2</sub> O <sub>2</sub> is further reacted with N-Ethyl-N-(2-hydroxy-3-sulfopropyl)-3-methylaniline (EHSPT) and 4-aminoantipyrine (4-AA) in the presence of peroxidase (POD) to generate quinone dye	Diazyme <sup>f</sup>	7.3b	6.1 <sup>b</sup>	> 0.99°	0.07 <sup>d</sup> U/L
	Lactate	(10)	Beckman Coulter <sup>a</sup>	2.86	5.23	0.99	0.01 mM

Method	Reference in which the method was validated for saliva or basis of the method (in those assays in which validation was performed specifically for this research)	Manufacturer or reagents (in home-made assays)	r reagents (in e assays)	Intra-assay CV (%)	Inter-assay CV (%)	Linearity (R²)	ПСОБ
		R1	R2				
FRAS	Reduction of Fe <sup>3+</sup> to Fe <sup>2+</sup> by the antioxidants present in the sample	0.7mM of TPTZ, and 1.5mM of FeCl <sub>3</sub> ·6H <sub>2</sub> O in acetate buffer	None	0.89 <sup>b</sup>	2.25 b	0.996	0.031 mM
TEAC	Reduction of ABTS radical to ABTS by the antioxidants present in the sample	2mM of ABTS, 0.25µM of HRP, and 40µM of H <sub>2</sub> O <sub>2</sub> in 50mM of phosphate buffer	None	1.91 <sup>b</sup>	4.51 <sup>b</sup>	0.99°	0.09 mM
CUPRAC	Reduction of Cu <sup>2+</sup> to Cu <sup>1+</sup> by the antioxidants present in the sample	0.25mM of BCS in 10mM of phosphate buffer	0.5mM of CuSO <sub>4</sub> in ultrapure water	0.60 <sup>b</sup>	1.25 <sup>b</sup>	0.99€	0.003 mM
Uricacid	(10)	Beckman Coulter <sup>a</sup>	Coulter <sup>a</sup>	0.57	6.41	>0.99	0.22 µmol/L
Catalase	Inhibition of colour development in a $\rm H_2O_2$ -producing urate assay	2mM of DHBS, 0.5mM of AP, 50mM of Fe(CN) <sub>6</sub> ·3H <sub>2</sub> O, 28 U/L of uricase, and 200 U/L of HRP in 50mM of phosphate buffer	500mM of uric acid	2.09 <sup>b</sup>	13.03 <sup>b</sup>	0.91c	0.03 units/ mL
AOPP	Producing solutions which in the presence of potassium iodide absorb at 340nm	0.059 M of KI in ultrapure water	Acetic acid, 50% (v/v)	1.48 <sup>b</sup>	6.25 <sup>b</sup>	0.98°	3.67 µM
H <sub>2</sub> O <sub>2</sub>	Production of diamine (yellow-coloured oxidation product) which absorbs at 450 nm	0.7mM of TMB and 25mU of HRP in 100mM of phosphate buffer	500mM of sulphuric acid	3.43 <sup>b</sup>	16.58 <sup>b</sup>	0.95°	0.01 µМ

2-hydroxybenzenesulphonate. AP - 4-aminophenazone. K<sub>a</sub>Fe(CN)<sub>A</sub>:3H<sub>2</sub>O - potassium hexacyanoferrate II. KI - potassium iodide. TMB, - 3,5,3050-tetramethylbenzidine. Beckman Coulter Inc, Fullerton, CA, USA. bMean of CVs for 6 repeated measurements of three pooled saliva samples of different concentrations as follows: standard deviation from the 6 measurements divided by mean and multiplied by 100. "Mean of R2 measured after linearity under dilution of two saliva samples. "Calculated as ylbenzthiazoline-6-sulfonic acid. HRP - horseradish peroxidase. BCS - bathocuproinedisulfonic acid disodium salt. CuSO4 – copper (II) sulphate. DHBS - 3,5-dichloro-CV - coefficient of variation. R² - coefficient of determination. LLOD - lower limit of detection. R1 - reagent 1. R2 - reagent 2. ATCl - acetylthiocholine iodide. DTNB - 5,5′dithiobis-2-nitrobenzoic acid. TPTZ – tripyridyltriazine. FeCl<sub>3</sub>·6H<sub>3</sub>O - ferric chloride hexahydrate. FeSO<sub>4</sub>·7H<sub>3</sub>O - ferrous sulfate heptahydrate. ABTS - 2,2′-azino-bis-3-ethmean + 2 standard deviations of 20 replicates of the zero standard. "BioSystems, Barcelona, Spain. 「Diazyme Laboratories, Poway, CA, USA. for ADA (Diazyme, DZ177A-Con). Daily in-house controls of two different values were used for analysis of FRAS, TEAC, CUPRAC, CAT, AOPP and  $\rm H_2O_2$ . The control for FRAS was a ferric chloride hexahydrate solution, for TEAC and CUPRAC a Trolox solution, an enzyme solution for CAT, Chloramine-T solution for AOPP, and a  $\rm H_2O_2$  solution for  $\rm H_2O_2$  assay. A single measurement was made in all cases since all analytical methods showed an intra-assay imprecision lower than 15%, which indicated adequate assay repeatability.

## Statistical analysis

Descriptive statistical procedures and spreadsheets (Excel 2000, Microsoft Corporation, Redmond, USA) and software (Graph Pad Prism, Version 5 for Windows, Graph Pad Software Inc, San Diego, USA, and IBM SPSS statistic for Windows, version 24.0, IBM Corp., Armonk, USA) were used. Since only 12 data were included, normality was not assumed. Therefore, the values of the analytes at different times and conditions after storage were compared with aliquots analysed immediately using the Friedman test, followed by Dunn's multiple comparison test. The results were considered significant when P < 0.05.

#### Results

The results for the stability of the studied analytes, under different storage conditions, are shown in Table 2.

At RT, AMY, CHE, AST, TEAC and CAT were stable during the whole experimental period (72 hours). The analytes that showed significant decreases were: TEA at 6 hours; Lip, CK and CUPRAC at 24 hours; and LD, Lact, FRAS, UA and AOPP at 72 hours. Significant increases were detected for  $H_2O_2$  at 6 hours and for ADA at 24 hours.

At 4 °C, AMY, CHE, AST, Lact, ADA, TEAC, CAT and AOPP were stable after 14 days of storage. Significant decreases were observed after 6 hours for TEA; after 24 hours for Lip and CUPRAC; after 72 hours for CK; and after 7 days for LD, FRAS and UA. Significant increases were recorded for H<sub>2</sub>O<sub>2</sub> after 6 hours.

When samples were stored at - 20 °C, AMY, CHE, Lact, ADA, TEAC, FRAS, CUPRAC, UA, CAT and AOPP were stable for 6 months. Significant decreases were recorded after 14 days of storage for AST, Lip, CK and LD; and after 3 months for TEA. A significant increase was observed for  $H_2O_2$  after 3 months of storage.

At - 80 °C, AMY, CHE, AST, Lip, CK, LD, Lact, ADA, TEAC, FRAS, CUPRAC, UA, CAT and AOPP were stable for 6 months. Significant decreases were observed after 3 months for TEA. A significant increase was detected for  $H_2O_2$  at 3 months.

#### **Discussion**

This study found that although there was a variability in the results depending on the studied analyte, in general, in the short-term storage conditions tested, the storage at 4° C provided longer stability for analytes in saliva than at RT. On the other hand, in the long-term storage conditions tested, - 80° C provided longer stability than - 20° C. In the short-term storage conditions, we also included storage for 72 hours at RT and 7 days at 4° C. We are aware that samples are not usually stored in these conditions; however, other researchers in their stability studies have used similar time points and they were included in our study in order to obtain information regarding stability in those conditions (26).

Regarding individual analytes, AMY, CHE and ADA were the enzymes that showed fewer changes in the different storage conditions. The high stability of AMY is in accordance with the results of other studies where, for example, AMY was stable for 5 days at RT or for 10 days at RT or 4 °C (20,21). Cholinesterase was also stable in all conditions, so it seems that its activity is not affected after storage, although there is a lack of previous reports to compare with. Adenosine deaminase was also stable in most of the conditions with the exception of RT, where it showed a significant increase. Stability of ADA has been studied in porcine saliva and was considered as stable for up to 1 year at - 80 °C (27). The reason why the activity of this enzyme in saliva increases at RT needs to be further elucidated.

**TABLE 2.** Stability results for 12 different saliva analytes obtained after measuring samples at different processing time and under different storage conditions

Analyte (unit)	ОТ	Temperature		T	12	T3	T4	<b>T</b> 5	16	11	18	19
				58,570 (47,055 – 101,200)	71,340 (48,460 – 94,915)	74,920 (44,770 – 92,445)	79,850 (48,175 -100,690)	ı	ı	ı	,	1
		RT	Variation (%)	1.9	24.1	30.4	38.9	ı	ı	ı	ı	ı
			۵	> 0.999	> 0.999	> 0.999	> 0.999	,	1	1	,	1
				78,810 (48,060 – 102,305)	68,670 (41,470 – 90,655)	76,480 (39,935 – 95,885)	79,970 (47,300 – 101,295)	76,540 (47,395 – 99,160)	81,000 (48,325 – 100,705)	ı	,	1
		4 °C	Variation (%)	37.1	19.5	33.1	39.2	33.2	40.9	ı	ı	ı
			۵	0.494	> 0.999	> 0.999	0.769	> 0.999	> 0.999	1	,	1
AMY (U/L)	57,470 (39,615 – 101,310)			,	1	1	1	ı	79,870 (47,165 – 100,730)	68,840 (40,750 – 89,215)	74,160 (42,280 – 97,095)	77,610 (47,965 – 99,245)
		- 20 °C	Variation (%)	ı	,	ı	ı	ı	39.0	19.8	29.0	35.0
			۵	,	,	,	,		> 0.999	0.054	> 0.999	> 0.999
				1	1	ı	ı	ı	79,300 (47,730 – 100,625)	68,140 (39,715 – 87,200)	71,250 (42,175 – 89,315)	77,460 (49,580 – 104,120)
		J∘ 08 -	Variation (%)	ı	,	ı	ı	,	38.0	18.6	24.0	34.8
			Ь	,	'	'	'	'	> 0.999	0.156	0.248	> 0.999
				7.4 (3.7 -10.3)	5.5 (0.1 – 7.7	5.3 (2.6 - 11.1)	5.3 (3.6 – 11.9)	,	ı	,	,	,
		RT	Variation (%)	2.8	23.6	26.4	26.4	1	ı	ı	1	ı
			Ь	> 0.999	> 0.999	> 0.999	> 0.999	-	-	-	-	-
				7.2 (3.6 -10.3)	6.6 (3.2 – 8.8)	11.2 (7.8 – 14.5)	7.1 (4.0 – 10.7)	11.4 (9.3 – 14.7)	7.4 (6-6 – 10.3)	,	1	ı
		4 °C	Variation (%)	0.0	8.3	55.6	4.1	58.3	2.8	ı	,	ı
			Ь	> 0.999	> 0.999	0.223	> 0.999	0.097	> 0.999		-	
CHE (μΜ/mL/ min)	7.2 (2.6 - 9.5)			1	1	,	1	,	4.9 (3.1 – 8.3)	6.7 (4.2 – 8.0)	5.9 (4.2 – 7.2)	4.0 (0.6 – 4.8)
		- 20 °C	Variation (%)	,	,	'	,		31.9	6.9	18.1	44.4

Analyte (unit)	10	Temperature		1	12	T3	<b>4</b> T	T5	16	12	18 18	L 61
			۵					,	> 0.999	> 0.999	> 0.999	> 0.999
				,	1	1	,	ı	7.2 (6.1 – 11.3)	8.1 (5.8 – 11.0)	6.9 (4.9 –	4.5 (3.8 – 7.3)
		-80 °C	Variation (%)		,	•			0:0	12.5	4.2	37.5
			Ь		1	1		ı	> 0.999	> 0.999	> 0.999	> 0.999
				10.6 (4.5 – 25.5)	11.6 (4.3 – 15.9)	4.4 (3.6 – 8.2)	4.1 (3.6 -4.7)	ı	ı	ı	ı	ı
		RT	Variation (%)	20.9	13.4	67.2	69.4	ı				
			۵	> 0.999	> 0.999	< 0.001	< 0.001	,		,		
				14.3 (5.1 -23.3)	15.7 (3.7 - 22.1)	6.3 (2.4 – 16)	2.1 (0.4 - 3.3)	1.7 (1.4 - 4.1)	0.6 (0.5 - 3.9)		'	
		4 °C	Variation (%)	2.9	17.2	53.0	84.3	87.3	95.5		,	,
			۵	> 0.999	> 0.999	< 0.001	< 0.001	<0.001	< 0.001	1		
Lip (U/L)	13.4 (7.3 - 33.2)			,			,	1	3.8 (0.9 – 8.0)	2.0 (1.4 – 14.0)	2.0 (1.2 - 3.5)	3.5 (2.4 -
		- 20 °C	Variation (%)	,	,	ı	,	ı	71.6	85.1	85.1	73.9
			۵		•	•		,	0.018	0.017	< 0.001	0.031
				,	1	1	,	ı	12.5 (3.5 - 32.4)	9.9 (2.7 - 36.4)	11.6 (4.1 - 34.7)	12.9 (4.5 - 54.4)
		J₀ 08 -	Variation (%)	,	1	ı		ı	6.7	26.1	13.4	3.7
			۵		•	,		,	> 0.999	> 0.999	> 0.999	> 0.999
				20.1 (14.4 -48.8)	15.7 (12.4 - 39.1)	27.7 (23.8 - 32.4)	18.1 (16.4 - 25.4)	1	ı	,	1	1
		RT	Variation (%)	6.6	29.3	24.5	18.6	1	ı	1	1	1
			Ь	> 0.999	0.002	> 0.999	0.035	ı	1			
				19.7 (15.7 - 51.3)	16.4 (13.3 - 42.4)	28.8 (25.4 - 57.4)	14.8 (12.2 - 27.2)	26.5 (20.4 -42.6)	23.2 (18.4 - 34.2)	,	ı	
		4 °C	Variation (%)	11.5	26.5	29.2	33.7	18.9	4.3		1	
			۵	> 0.999	0.041	> 0.999	< 0.001	> 0.999	> 0.999	•	-	

TEA (U/L) (15.6-59.4)  -20 °C (%a)  -20 °C (%b)  -20 °C (	Analyte (unit)	10	Temperature		F	T2	T3	4	TS	16	11	<b>8</b> 2	T9
-20°C (%)  P -80°C (%)  -80°C (%)		22.3 (15.6 - 59.4)			1	1	1	1	1	19.9 (15.6 - 49.7)	24.4 (17 - 56.9)	16.8 (9.9 -35.4)	21.5 (13.9 -43.4)
-80°C Variation -80°C Variatio			- 20 °C	Variation (%)	1	1	ı	1	1	10.3	6.7	24.5	3.4
-80°C Variation -80°C Variation -80°C (%) -81(36 - 10.0(3.8 - 5.5(1.5 - 6.0(2.4 - 21.6) 20.4) 9.8) 11.2) -81 (%) -80°C Variation -9				۵	1	1	ı	1	•	> 0.999	> 0.999	< 0.001	0.315
Po 1/3.6 10.0 (3.8 - 5.5 (1.5 - 6.0 (2.4 - 1.6))  RT Variation 45.5 40.1 67.2 6.0 (2.4 - 1.6)  P					ı	ı	ı	ı	ı	20.5 (15.4 - 50.9)	24.3 (17.8 - 56.7)	13.6 (9.0 -34.2)	21.0 (15.6 -53)
HT Variation (%)  16.6 (7.000- 33.2)  -80 C Variation (%)  -20 C Variation (%)  -80 C Variation (%)  -80 C (%			J∘ 08-	Variation (%)	1	1	ı	ı	1	8.1	9.0	38.9	5.6
RT   Variation   45.5   10.0 (3.8 - 5.5 (1.5 - 6.0 (2.4 - 1.6))   11.2)   11.2				۵		•	ı	ı	1	> 0.999	> 0.999	< 0.001	> 0.999
RT   Variation   45.5   40.1   67.2   63.9     P   > 0.999   0.248   0.002   0.003     1.7 (5.8 - 11.5 (6.6 - 10.9 (47 - 5.7 (2.7 - 4.7))     A \cdot C   Variation   29.8   30.7   34.6   66.0     16.6 (7.000 - 33.2)   -20 \cdot C   Variation   -80 \cdot C   (%)					9.1 (3.6 - 21.6)	10.0 (3.8 - 20.4)	5.5 (1.5 - 9.8)	6.0 (2.4 -	1	ı	ı	1	ı
Horizon P > 0.999 0.248 0.002 0.003 1.1.7 (5.8 - 11.5 (6.6 - 10.9 (4.7 - 5.7 (2.7 - 1.4) 16.5) 18.0) 10.1)			RT	Variation (%)	45.5	40.1	67.2	63.9	1	ı	ı	1	ı
4°C Variation 29.8 30.7 34.6 66.0 (%) 20.099 > 0.999 0.006 10.1) 11.7 (5.8 - 11.5 (6.6 - 10.9 (4.7 - 5.7 (2.7 - 1.4) 16.5) 18.0) 10.1) 10.1) 21.4) 29.8 30.7 34.6 66.0 66.0 66.0   16.6 (7.00 - 1.20 °C Variation				۵	> 0.999	0.248	0.002	0.003	ı	1	,	1	ı
4 °C   Variation   29.8   30.7   34.6   66.0     16.6		•			11.7 (5.8 - 21.4)	11.5 (6.6 - 16.5)	10.9 (4.7 - 18.0)	5.7 (2.7 - 10.1)	4.4 (0.9 - 7.8)	2.0 ((- 1.0)	ı	1	ı
16.6 (7.00- 33.2)  -20°C			4 °C	Variation (%)	29.8	30.7	34.6	0.99	70.8	88.0	ı	ı	ı
16.6 (7.00- 33.2)  -20 °C (%)  P  P  -  RT Variation  -  11.2 (9.0- 11.5 (10.2- 15.3 (12.4- 25.5) 24.0) 24.7)  RT Variation  (7.00- 11.2 (9.0- 11.5 (10.2- 15.3 (12.4- 25.5) 24.0) 24.7)		'		۵	> 0.999	> 0.999	> 0.999	90000	<0.001	< 0.001			
-20 °C   Variation     -   -	כא (ח/ר)	16.6 (7.00 - 33.2)			1	ı	ı	ı	1	7.0 (3.4 - 10.8)	5.9 (4.7	7.2 (5.7 - 11.5)	6.9 (4.0 - 8.5)
Variation (%)  Variation (%)  P			-20 °C	Variation (%)	1	ı	ı	1	1	57.8	64.8	9.99	58.4
Variation (%)  P				۵	•	1	,	ı	1	0.003	0.043	0.045	0.020
Variation       -					1	ı	ı	ı	ı	12.9 (6.7 - 22.4)	13.6 (8.7 - 20.7)	15.9 (8.1 - 20.5)	13.6 (10.1 - 20.1)
P			J∘ 08 -	Variation (%)		1	ı	1		22.6	18.1	4.2	18.4
11.2 (9.0 - 11.5 (10.2 - 15.3 (12.4 - 25.5) 24.0) 24.7)  Variation 7.1 4.6 27.0 (%)				۵	1	ı	ı	1	ı	> 0.999	> 0.999	> 0.999	> 0.999
<b>Variation</b> 7.1 4.6 27.0 (%)					11.2 (9.0 - 25.5)	11.5 (10.2 - 24.0)		22.2 (15.3 - 32.5)	ı	ı	ı	ı	ı
			RT	Variation (%)	7.1	4.6	27.0	83.8		1	1	,	1

Analyte (unit)	01	Temperature		1	12	T3	T4	T5	16	11	M2	19
			۵	> 0.999	> 0.999	> 0.999	> 0.999	1	1	1	1	1
				11.7 (9.5 - 25.2)	12.7 (10.5 - 24.4)	12.8 (10.3 - 23.6)	10.3 (8.3 - 24.1)	10.7 (8.4 - 29.3)	10.3 (8.2 - 24.7)		1	,
		4 °C	Variation (%)	3.3	5.0	6.2	14.5	11.2	14.9	,	1	1
			۵	> 0.999	> 0.999	> 0.999	0.760	0.508	0.222	1	ı	
AST (U/L)	12.1 (10.1 - 26.7)			1	ı	ı	1	1	9.7 (7.7 - 17.2)	9.1 (5.9 - 11.3)	7.9 (6.1 - 11.5)	5.2 (2.8 - 6.0)
		- 20 °C	Variation (%)		ı		ı	1	19.5	24.9	34.4	57.3
			Ь	1	ı	1	ı	1	0.028	< 0.001	< 0.001	< 0.001
				1	ı	1	ı	1	12.8 (11.0 - 25.1)	13.1 (11.1 - 26.1)	13.5 (11.4 - 26.1)	13.9 (9.7 - 25.0)
		J∘ 08 -	Variation (%)	1	ı	1	1	1	6.2	8.3	12.0	15.4
			۵	1	ı		ı	ı	> 0.999	> 0.999	> 0.999	999.0
				338 (214 -409)	339 (212 – 399)	276 (189 – 344)	204 (139 – 295)	1	1	1	ı	1
		RT	Variation (%)	0.3	9.0	18.1	39.3	1	,	1	ı	1
			۵	> 0.999	> 0.999	0.365	900.0	ı	,	ı	ı	ı
				333 (224- 402)	336 (193 – 404)	320 (211 – 383)	220 (172 – 338)	196 (156 – 302)	131 (77 – 261)	ı	ı	1
		4 °C	Variation (%)	1.0	0.1	5.0	34.8	41.7	61.1	1	ı	ı
			Ь	> 0.999	> 0.999	> 0.999	0.222	0.006	< 0.001	1	-	1
(U/L)	337 (221-403)			1	ı	ı	ı	ı	69 (27 – 168)	54 (18 – 97)	27 (5 – 59)	31 (15 – 59)
		- 20 °C	Variation (%)	1	ı	1	ı	1	79.5	84.0	92.1	6.06
			۵	1	ı		ı	ı	< 0.001	< 0.001	< 0.001	< 0.001
				1	1	1	1	1	323 (214 – 391)	321 (223 – 399)	335 (234 – 414)	370 (279 – 445)

Analyte (unit)	Т0	Temperature		T1	12	Т3	Т4	Т5	Т6	77	Т8	Т9
		J° 08 -	Variation (%)	ı	ı	ı	ı	ı	4.0	1.8	9.0	10.0
			۵		,	•	,	,	> 0.999	> 0.999	> 0.999	> 0.999
				0.5 (0.4 -	0.5 (0.4 -	0.3 (0.1 -	0.01((- 0.003) -0.04)	ı	1	ı	1	1
		RT	Variation (%)	1.0	1.0	58.2	98.1	ı	1	ı	1	1
			Ь	> 0.999	> 0.999	0.634	0.003	ı	1	ı	ı	ı
				0.5 (0.4 -	0.5 (0.4 - 1.2)	0.4 (0.3 -1.3)	0.4 (0.3 -	0.5 (0.4 -1.5)	0.5 (0.1 - 0.7)	ı	1	ı
		4 °C	Variation (%)	1.9	1.0	15.5	17.5	8.7	6.7	ı	1	
			۵	> 0.999	> 0.999	> 0.999	> 0.999	>0.999	> 0.999	1	1	,
Lact (mmol/L)	0.5 (0.4 - 1.2)			-	1	1	1	1	0.5 (0.4 -	0.5 (0.4 -	0.5 (0.4 - 1.2)	0.5 (0.4 -
		-20 °C	Variation (%)		ı	ı	1	ı	1.0	10.7	5.8	3.9
			۵	,	,	1	,	1	> 0.999	> 0.999	> 0.999	> 0.999
				1	ı	1	1	ı	0.5 (0.4 -	0.5 (0.4 -	0.5 (0.4 -	0.5 (0.4 -1.2)
		-80 °C	Variation (%)	•	ı	ı	ı	ı	1.0	6.8	1.9	3.9
			Ь	1	1	1	1	1	> 0.999	> 0.999	> 0.999	> 0.999
				1.2 (0.8 - 1.7)	1.3 (1.0 - 1.9)	2.0 (1.2 -2.2)	1.1 (0.9 - 1.8)	ı	ı	ı	ı	ı
		RT	Variation (%)	18.2	35.4	101	6.1	ı	ı	ı	ı	ı
			Ь	> 0.999	0.059	0.007	> 0.999	,	'	'	1	1
				1.1 (0.9 - 1.6)	1.2 (0.8 - 1.8)	1.0 (0.8 - 1.9)	0.9 (0.6 - 1.7)	0.8 (0.6 -1.2)	0.6 (0.5 - 1.2)	ı	ı	ı
		4 °C	Variation (%)	12.1	21.2	4.0	6.1	20.2	37.4	ı	ı	ı
			۵	> 0.999	0.734	> 0.999	> 0.999	>0.999	> 0.999	1	ı	1

Analyte (unit)	10	Temperature		7	12	T3	<b>T</b> 4	TS	T6	14	<b>18</b>	19
ADA (U/L)	1.0 (0.7 - 1.6)			1	1	1	1	1	1.0 (0.5 -	0.8 (0.3 -	0.5 (0.2 - 1.1)	0.6 (0.2 - 0.8)
		- 20 °C	Variation (%)	1	1	1	1	1	2.0	15.2	45.5	41.4
			۵	1	ı	I	1	ı	> 0.999	> 0.999	0.189	0.064
				,	1	ı	1	ı	1.2 (0.8 - 1.8)	1.2 (0.8 - 1.8)	1.0 (0.7 - 1.5)	1.2 (1.0 - 2.0)
		J∘ 08 -	Variation (%)	1	ı	1	ı	ı	17.2	16.2	2.0	18.2
			۵	,	1	1	1	1	> 0.999	> 0.999	> 0.999	0.291
				0.2 (0.2 - 0.3)	0.2 (0.2 - 0.3)	0.1 (0.1 - 0.2)	0.2 (0.1 - 0.2)	ı	1	1	-	-
		RT	Variation (%)	31.6	26.2	13.9	9.1	ı	,	ı	,	,
			۵	0.701	0.529	0.582	0.582	1	,	,	•	
	•			0.2 (0.2 - 0.3)	0.2 (0.2 - 0.3)	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.2 -0.3)	0.2 (0.1 - 0.2)	1	1	1
		4 °C	Variation (%)	29.5	24.1	4.6	7.4	17.1	10.5	1	•	1
			۵	> 0.999	> 0.999	> 0.999	> 0.999	0.064	> 0.999	1	1	1
TEAC (mM)	0.2 (0.1 – 0.2)			1			1	1	,	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.3)
		-20 °C	Variation (%)	1			1	ı		0.9	8.9	33.8
			۵	ı	ı	ı	ı	ı	ı	> 0.999	> 0.999	0.457
	-			ı	1	1	ı	ı	1	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.3)	0.2 (0.2 - 0.3)
		J∘ 08 -	Variation (%)	ı	1	ı	ı	ı	1	8.3	11.0	32.0
			Ь	ı	•	1	ı	ı	1	> 0.999	0.291	0.841
				0.4 (0.3 - 0.5)	0.4 (0.3 - 0.5)	0.4 (0.3 - 0.5)	0.3 (0.2 - 0.3)	ı	1	1	ı	ı
		RT	Variation (%)	3.3	9.1	9.5	13.3	1	1	1		1

Analyte (unit)	10	Temperature		Ħ	12	Т3	4	T2	<b>T6</b>	11	<b>8</b> 2	<b>6</b> 1
			۵	> 0.999	0.669	> 0.999	< 0.001		1			
				0.4 (0.2 - 0.5)	0.3 (0.2 - 0.5)	0.4 (0.3 - 0.5)	0.3 (0.3 - 0.4)	0.3 (0.2 -0.4)	0.4 (0.3 - 0.4)		,	,
		4 °C	Variation (%)	3.0	1.3	2.0	4.5	9.4	4.3	1		
			۵	> 0.999	> 0.999	> 0.999	0.555	0.018	0.047			
FRAS (mM)	0.3 (0.3 - 0.5)			1	ı	ı	ı	ı	0.3 (0.3 - 0.5)	0.4 (0.3 - 0.5)	0.3 (0.3 - 0.5)	0.3 (0.3 - 0.5)
		- 20 °C	Variation (%)	,	ı	ı		,	0.1	10.4	4.8	0.3
			۵		,	,	•	•	> 0.999	> 0.999	> 0.999	> 0.999
				1	ı	ı		1	0.3 (0.2 - 0.5)	0.3 (0.2 - 0.5)	0.3 (0.2 - 0.5)	0.3 (0.2 - 0.5)
		- 80 °C	Variation (%)	1	ı	ı	1	1	0.1	9.0	1.3	2.1
			۵	•	ı	1	•	•	> 0.999	> 0.999	> 0.999	0.394
				0.2 (0.2 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.2)	,	ı		,	,
		RT	Variation (%)	Ξ:	1.5	0.1	27.2	,	ı	1	1	1
			۵	> 0.999	> 0.999	0.010	< 0.001	•	ı	ı	ı	ı
				0.2 (0.2 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.1 - 0.2)	0.2 (0.1	0.2 (0.1 - 0.2)	,	1	1
		4 °C	Variation (%)	8.0	0.1	5.9	12.7	14.3	10.1			
			۵	> 0.999	0.794	0.002	< 0.001	<0.001	< 0.001	,	,	,
CUPRAC (mM)	0.2 (0.2 - 0.3)			1	ı	ı	1	1	0.2 (0.2 - 0.2)	0.2 (0.2 - 0.2)	0.2 (0.2 - 0.2)	0.2 (0.2 - 0.3)
		-20 °C	Variation (%)	1	ı	ı	1	1	1.2	0.3	3.3	2.1
			۵	1	1	1	1	1	> 0.999	> 0.999	0.162	> 0.999
				1	ı	ı	-	1	0.2 (0.2 - 0.2)	0.2 (0.2 - 0.3)	0.2 (0.2 - 0.2)	0.2 (0.2 - 0.3)

Analyte (unit)	10	Temperature		1	12	Т3	47	T5	<b>T6</b>	17	18 18	19
		-80 °C	Variation (%)	1		1	1	1	0.1	3.7	0.2	2.4
			۵		1	1	1	1	> 0.999	> 0.999	0.131	> 0.999
				13.1 (9.5 - 16.7)	13.7 (9.5 - 16.7)	11.9 (7.7 -	9.5 (5.9 -	ı	1	1	1	1
		RT	Variation (%)	2.6	1.3	12.2	28.6	ı	ı	1	1	,
			۵	> 0.999	> 0.999	0.084	< 0.001	,	ı	,	•	,
				13.7 (9.5 - 17.3)	13.1 (9.5 - 17.3)	13.1 (8.9 -	12.5 (8.3 -	11.9 7.7 -	10.7 (7.1 - 13.7)	1	1	1
		4 °C	Variation (%)	1.7	1.5	4.6	10.2	12.1	20.6	1	1	1
			۵	> 0.999	> 0.999	> 0.999	0.383	0.014	0.002	•	•	•
Uric acid (µmol/L)	13.7 (9.5 - 16.1)				1	-		1	13.7 (9.5 - 18.4)	13.7 (9.5 - 16.7)	14.3 (9.5 - 17.3)	14.3 (10.1 - 17.8)
		- 20 °C	Variation (%)		1	1	,	1	1.5	2.4	2.8	5.6
			۵	1	ı	1	,	ı	> 0.999	> 0.999	> 0.999	0.365
				1	ı	ı	1	ı	13.7 (9.5 - 16.7)	13.1 (9.5 - 16.7)	13.7 (9.5 - 17.3)	14.3 (9.5 - 16.7)
		J. 08 -	Variation (%)	1	ı	1	,	ı	0.2	2.8	0.0	2.8
			Ь	1	1	-	-	ı	> 0.999	> 0.999	> 0.999	999:0
				0.2 (0.2 - 0.3)	0.2 (0.2 - 0.3)	0.2 (0.2 - 0.4)	0.4 (0.2 - 0.5)	ı	ı	1	1	1
		RT	Variation (%)	2.2	13.3	12.5	50.0	ı	ı	1	1	1
			Ь	> 0.999	0.960	> 0.999	> 0.999	-	'	-	'	
				0.2 (0.2 - 0.3)	0.2 (0.2 - 0.3)	0.3 (0.2 - 0.3)	0.2 (0.2 - 0.3)	0.3 (0.2 - 0.4)	0.2 (0.2 - 0.4)	1		1
		4 °C	Variation (%)	8.9	1.2	2.0	6.5	33.5	0.4	ı	1	1
			۵	> 0.999	> 0.999	> 0.999	> 0.999	>0.999	> 0.999	1	-	1

Analyte (unit)	10	Temperature		Ε	12	<b>E</b>	<b>4</b> 1	TS	16	12	<b>T8</b>	T9
CAT (units/ mL)	0.2 (0.2 – 0.3)			1	1	ı	1	1	0.2 (0.2 - 0.3)	0.3 (0.2 - 0.4)	0.3 (0.2 - 0.4)	0.3 (0.2 - 0.4)
		- 20 °C	Variation (%)	1	•		•	ı	8.1	14.1	19.4	26.2
			۵	1	1	1	1	1	> 0.999	> 0.999	> 0.999	> 0.999
				1	'	,	,	1	0.3 (0.2 - 0.3)	0.3 (0.2 - 0.4)	0.2 (0.2 - 0.4)	0.2 (0.2 - 0.4)
		J∘ 08 -	Variation (%)	1				ı	4.0	11.3	2.4	4.8
			۵	ı	•	•	,		> 0.999	> 0.999	> 0.999	> 0.999
				103 (80 – 169)	106 (83 – 168)	121 (79 – 147)	59 (41 – 69)	1	1	-	1	1
		RT	Variation (%)	6.7	3.7	8.6	46.9	ı	ı	1	ı	ı
			۵	> 0.999	> 0.999	> 0.999	0.001	1	,	,	,	,
				116 (81 – 169)	119 (89 – 165)	115 (67 – 142)	102 (60 – 126)	109 (56 – 173)	95 (69 – 123)	,	1	1
		4 °C	Variation (%)	4.6	8.1	4.	8.1	1.2	14.2	•	1	1
			۵	> 0.999	> 0.999	0.914	0.331	>0.999	> 0.999	1	1	1
AOPP (µM)	111 (76 -180)			1	,	,	1	1	109 (97 – 184)	115 (100 – 186)	102 (81 – 155)	96 (82 – 146)
		- 20 °C	Variation (%)	1				1	1.9	4.1	7.3	13.3
			۵	1	1		,	1	> 0.999	> 0.999	> 0.999	> 0.999
				ı	1	1	1	ı	114 (90 – 185)	116 (98 – 182)	110 (84 – 171)	124 (83 – 207)
		J∘ 08 -	Variation (%)	ı	1	,	1	ı	2.9	4.7	0.5	11.8
			۵	1	•	•	1	1	> 0.999	> 0.999	> 0.999	> 0.999
				5.4 (2.1 - 6.6)	6.8 (3.2 -10.6)	5.6 (4.3 -10.8)	4.4 (2.2 - 7.2)	1	1	1	1	1
		RT	Variation (%)	120.4	178.8	127.3	80.8	1	1	1	1	1

Analyte (unit)	Т0	Temperature		Ţ	12	T3	T4	T5	16	17	T8	Т9
			۵	> 0.999	0.011	0.001	> 0.999	,	,		1	
				5.3 (2.4 - 6.4)	7.3 (3.1 -10.2)	8.6 (5.7 -12.2)	6.1 (3.2 - 7.6)	4.2 (3.5 -7.2)	4.5 (2.5 - 8.0)	1	1	1
		4 °C	Variation (%)	117.1	198	252.2	146.9	72.2	84.5	1	ı	1
			۵	> 0.999	0.012	< 0.001	096.0	>0.999	> 0.999	ı	ı	ı
H <sub>2</sub> O <sub>2</sub> (µМ)	2.5 (1.7 - 8.9)			1	1	ı	ı	'	6.8 (2.1 - 9.4)	7.5 (3.1 - 9.0)	9.2 (4.1 -11.8)	7.7 (3.8 -10.8)
		- 20 °C	Variation (%)	ı	,	ı	ı	•	176.3	206.9	274.7	212.2
			۵	,	•	ı	,	•	> 0.999	0.248	< 0.001	0.002
				ı	1	ı	ı	1	6.2 (2.5 -10.8)	6.6 (2.8 - 9.0)	7.2 (3.3 -12.4)	9.6 (3.0 -12.2)
		J∘ 08 -	Variation (%)	ı	•	ı	ı	•	153.5	169.8	193.9	291.0
			۵	1	1	1	1	1	0.804	> 0.999	< 0.001	0.002

reducing ability. CUPRAC - cupric reducing antioxidant capacity. CAT - catalase. AOPP - advanced oxidation protein products.  $H_2O_2$  - hydrogen peroxide. P < 0.051 month. T8 – 3 months . T9 – 6months. RT - room temperature. AMY - α-amylase. CHE - cholinesterase. Lip - lipase. TEA - total esterase. CK - creatine kinase. AST - aspartate aminotransferase. LD - lactate dehydrogenase. Lact - lactate. ADA - adenosine deaminase. TEAC - Trolox equivalent antioxidant capacity. FRAS - ferric Quantitative data is presented as median and interquartile range. T0 - baseline. T1 - 3 hours. T2 - 6 hours. T3 - 24 hours. T4 - 72 hours T5 - 7 days. T6 - 14 days. T7 represents statistical significant difference.

Lipase and CK showed a high stability at - 80 °C; however, in the rest of the storage conditions they showed significant changes. To our knowledge, the stability of Lip and CK in saliva under different storage conditions has not been accessed before, and the reason for lipase and CK enzymatic activity changes in saliva is unknown and should be further studied. Aspartate aminotransferase showed also the highest stability at - 80 °C, being stable for 6 months. The lack of stability of AST in other storage conditions, such as - 20 °C, agrees with previous reports (2,3).

Total esterase and LD were the most labile enzymes in our stability study. It is difficult to know the reason for the changes of TEA, it is likely to have been influenced by the instability of any of the various enzymes that integrate in this total activity (8). Lactate dehydrogenase results in our study were similar to other reports that showed a significant decrease of this enzyme in saliva after only 30 min, 3 days and 2 weeks of storage at - 20 °C (2,3,22). These results could be due to the lability of the LD-4 and LD-5 isoenzymes at - 20 °C (28).

With the exception of  $H_2O_2$ , the remaining antioxidants biomarkers showed a high stability under freezing conditions. Uric acid has been observed to remain relatively stable during storage, in accordance with previous studies describing stability at - 20 °C, - 70 °C and at - 196 °C in human serum when stored for 1 year (29). This can also help to explain the stability observed for the TEAC and FRAS in saliva, both at - 20 °C and at - 80 °C, as UA is one of the main contributors to TEAC and FRAS. When oxidant biomarkers were studied, AOPP showed a high level of stability in all conditions, except at RT. Only  $H_2O_2$  was very unstable and

could only be measured after 3 hours at 4 °C or 1 month under freezing conditions. Further studies should be made to elucidate the reason for the production of  $\rm H_2O_2$  in the saliva samples when they are stored.

This study has some limitations that should be taken into account. Firstly, the study has been made in healthy subjects. It would be of interest in the future to perform further studies also involving subjects with different diseases in order to evaluate the possible differences of stability between samples of healthy and sick subjects. Although in line with previous reports, the number of subjects included in this study can be considered low and ideally a higher number of cases should be included (21,25,30). Therefore, this report should be taken as a pilot study and additional studies would be needed prior to making recommendations about the storage conditions. In addition, in the case of the enzymes, it would have been interesting to study the stability of the different isoenzymes.

It can be concluded that in short-term storage the analytes were more stable at 4 °C than at room temperature, whereas in long-term storage they were more stable at - 80 °C than at - 20 °C.

### **Acknowledgments**

This research was supported by the Garcia Cugat Foundation and the Seneca Foundation of Murcia Region. The authors also had financial support from the Portuguese Foundation for Science and Technology (FCT) in the form of Elsa Lamy's FCT investigator contract IF/01778/2013.

#### **Potential conflict of interest**

None declared.

#### References

- Gröschl M. Current status of salivary hormone analysis. Clin Chem. 2008;54:1759-69. https://doi.org/10.1373/ clinchem.2008.108910
- 2. Dos Santos DR, Souza RO, Dias LB, Ribas TB, de Oliveira LCF, Sumida DH, et al. The effects of storage time and temperature on the stability of salivary phosphatases, transaminases and dehydrogenase. Arch Oral Biol. 2018;85:160–5. https://doi.org/10.1016/j.archoralbio.2017.10.016
- 3. Kasim NA, Ariffin SHZ, Shahidan MA, Abidin IZZ, Senafi S, Jemain AA, et al. Stability of lactate dehydrogenase, aspartate aminotransferase, alkaline phosphatase and tartrate resistant acid phosphatase in human saliva and gingival crevicular fluid in the presence of protease inhibitor. Arch Biol Sci 2013;65:1131–40. https://doi.org/10.2298/ABS1303131K
- 4. Nater UM, Rohleder N. Salivary alpha-amylase as a non-invasive biomarker for the sympathetic nervous system: current state of research. Psychoneuroendocrino-

- logy. 2009;34:486–96. https://doi.org/10.1016/j.psyneu-en.2009.01.014
- Boston PF, Gopalkaje K, Manning L, Middleton L, Loxley M. Developing a simple laboratory test for Alzheimer's disease: measuring acetylcholinesterase in saliva - a pilot study. Int J Geriatr Psychiatry. 2008;23:439-40. https://doi.org/10.1002/ gps.1882
- Bulgaroni V, Rovedatti MG, Sabino G, Magnarelli G. Organophosphate pesticide environmental exposure: analysis of salivary cholinesterase and carboxilesterase activities in preschool children and their mothers. Environ Monit Assess. 2012;184:3307-14. https://doi.org/10.1007/s10661-011-2190-8
- Lee PC, Purcell ES, Borysewicz R, Klein RM, Werlin SL. Developmental delay of lingual lipase expression after guanethidine-induced sympathectomy. Proc Soc Exp Biol Med. 1992;199:192-8. https://doi.org/10.3181/00379727-199-43346
- 8. Tecles F, Tvarijonaviciute A, De Torre C, Carrillo JM, Rubio M, García M, et al. Total esterase activity in human saliva: Validation of an automated assay, characterization and behaviour after physical stress. Scand J Clin Lab Invest. 2016;76:324–30. https://doi.org/10.3109/00365513.2016.1163417
- Brancaccio P, Maffulli N, Buonauro R, Limongelli FM. Serum enzyme monitoring in sports medicine. Clin Sports Med. 2008;27:1–18. https://doi.org/10.1016/j.csm.2007.09.005
- 10. Franco-Martínez L, Tvarijonaviciute A, Martínez-Subiela S, Márquez G, Martínez Díaz N, Cugat R, et al. Changes in lactate, ferritin, and uric acid in saliva after repeated explosive effort sequences. J Sports Med Phys Fitness. 2018. [Epub ahead of print].
- 11. Rai B, Kaur J, Jacobs R, Anand SC. Adenosine deaminase in saliva as a diagnostic marker of squamous cell carcinoma of tongue. Clin Oral Investig. 2011;15:347-9. https://doi.org/10.1007/s00784-010-0404-z
- 12. Bartosz G. Non-enzymatic antioxidant capacity assays: limitations of use in biomedicine. Free Radic Res. 2010;44:711-20. https://doi.org/10.3109/10715761003758114
- 13. Tvarijonaviciute A, Aznar-Cayuela C, Rubio CP, Tecles F, Ceron JJ, López-Jornet P. Salivary Antioxidant Status in Patients with Oral Lichen Planus: Correlation with Clinical Signs and Evolution during Treatment with Chamaemelum nobile. Biomed Res Int. 2018;2018:5187549. https://doi.org/10.1155/2018/5187549
- 14. Rubio CP, Hernández-Ruiz J, Martinez-Subiela S, Tvarijonaviciute A, Ceron JJ. Spectrophotometric assays for total antioxidant capacity (TAC) in dog serum: an update. BMC Vet Res. 2016;12:166. https://doi.org/10.1186/s12917-016-0792-7
- 15. Battino M, Ferreiro MS, Gallardo I, Newman HN, Bullon P. The antioxidant capacity of saliva. J Clin Periodontol. 2002;29:189–94. https://doi.org/10.1034/j.1600-051X.2002.290301x.x
- 16. Al-Rawi NH. Oxidative stress, antioxidant status and lipid profile in the saliva of type 2 diabetics. Diab Vasc Dis Res. 2011;8:22–8. https://doi.org/10.1177/1479164110390243
- 17. Ahmadi-Motamayel F, Vaziri-Amjad S, Goodarzi MT, Poorolajal J. Evaluation of salivary vitamin C and catalase in HIV

- positive and healthy HIV negative control group. Infect Disord Drug Targets. 2017;17:101–5. https://doi.org/10.2174/1871526517666170116142547
- 18. Deminice R, Sicchieri T, Payão PO, Jordão AA. Blood and salivary oxidative stress biomarkers following an acute session of resistance exercise in humans. Int J Sports Med. 2010;31:599-603. https://doi.org/10.1055/s-0030-1255107
- Leto TL, Geiszt M. Role of Nox family NADPH oxidases in host defense. Antioxid Redox Signal. 2006;8:1549–61. https://doi. org/10.1089/ars.2006.8.1549
- 20. O'Donnell K, Kammerer M, O'Reilly R, Taylor A, Glover V. Salivary alpha-amylase stability, diurnal profile and lack of response to the cold hand test in young women. Stress. 2009;12:549–54. https://doi.org/10.3109/10253890902822664
- 21. Park JR, Kim MH, Woo J, Lee SJ, Song KE. Measurement of amylase in saliva collected by salivette. Korean J Lab Med 2008;28:438–43. (in Korean) https://doi.org/10.3343/kilm.2008.28.6.438
- 22. Alonso de la Peña V, Diz Dios P, Lojo Rocamonde S, Tojo Sierra R, Rodriguez-Segade S. A standardised protocol for the quantification of lactate dehydrogenase activity in saliva. Arch Oral Biol. 2004;49:23–7. https://doi.org/10.1016/S0003-9969(03)00196-1
- 23. Qing Z, Ling-Ling E, Dong-Sheng W, Hong-Chen L. Relationship of advanced oxidative protein products in human saliva and plasma: age- and gender-related changes and stability during storage. Free Radic Res. 2012;46:1201–6. https://doi.org/10.3109/10715762.2012.700113
- 24. Tecles F, Fuentes-Rubio M, Tvarijonaviciute A, Martínez-Subiela S, Fatjó J, Cerón JJ. Assessment of stress associated with an oral public speech in veterinary students by salivary biomarkers. J Vet Med Educ. 2014;41:37-43. https://doi.org/10.3138/jvme.0513-073R1
- 25. Barranco T, Tvarijonaviciute A, Tecles F, Carrillo JM, Sánchez-Resalt C, Jimenez-Reyes P, et al. Changes in CK, LDH and AST in saliva samples after an intense exercise: a pilot study. J Sports Med Phys Fitness. 2018;58:910-6.
- Jensen EA, Stahl M, Brandslund I, Grinsted P. Stability of heparin blood samples during transport based on defined preanalytical quality goals. Clin Chem Lab Med. 2008;46:225-34. https://doi.org/10.1515/CCLM.2008.053
- 27. Escribano D, Contreras-Aguilar MD, Tvarijonaviciute A, Martínez-Miró S, Martínez-Subiela S, Cerón JJ, et al. Stability of selected enzymes in saliva of pigs under different storage conditions: a pilot study. J Vet Med Sci. 2018;80:1657-61. https://doi: 10.1292/jvms.18-0346
- 28. Nagler RM, Lischinsky S, Diamond E, Klein I, Reznick AZ. New insights into salivary lactate dehydrogenase of human subjects. J Lab Clin Med. 2001;137:363–9. https://doi.org/10.1067/mlc.2001.114710
- 29. Jansen EHJM, Beekhof PK, Viezeliene D, Muzakova V, Skalicky J. Long-term stability of oxidative stress biomarkers in human serum. Free Radic Res. 2017;51:970–7. https://doi.org/10.1080/10715762.2017.1398403
- 30. Nalla AA, Thomsen G, Knudsen GM, Frokjaer VG. The effect of storage conditions on salivary cortisol concentrations using an enzyme immunoassay. Scand J Clin Lab Invest. 2015;75:92-5. https://doi.org/10.3109/00365513.2014.985252