In vitro antioxidant and antityrosinase activities of Manilkara kauki

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³ Molecular Crop Research Unit, Faculty of Science, Chulalongkorn University Bangkok 10330, Thailand Manilkara of family Sapotaceae. This study investigated the total phenolic and flavonoid contents, and antioxidant and antityrosinase activities in different parts of M. kauki (fruits, leaves, seeds, stem barks and woods) and in fractions of stem barks. The total phenolic and flavonoid contents of the methanol and aqueous crude extracts of different parts of M. kauki ranged from 10.87 to 176.56 mg GAE (gallic acid equivalents) per gram of crude extract and 14.33 to 821.67 mg QE (quercetin equivalents) per gram of crude extract, resp. Leaves and stem barks exhibited higher total phenolic and flavonoid contents and antioxidant and antityrosinase activities than fruits, seeds and woods. Stem barks were sequentially extracted with *n*-hexane, ethyl acetate, methanol and water and then the fractionated extracts were subjected to antioxidant and antityrosinase activities testing. The ethyl acetate and methanol extracts of stem barks exhibited higher total phenolic and flavonoid contents and antioxidant and antityrosinase activities than the *n*-hexane and aqueous extracts. Moreover, ethyl acetate extract of M. kauki stem exhibited the highest antityrosinase activity. It may be a potential source of tyrosinase inhibitors for pharmaceutical and cosmetic applications.

Manilkara kauki L. Dubard is a tropical plant in the genus

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In recent years, trends in using medicinal plants, such as phytotherapy, functional foods, herbal teas and especially cosmetics have been growing rapidly. Medicinal plants are commonly used as active ingredients of skin whitening products to treat hyperpigmentation. Melanin is essential for protecting human skin cells from ultraviolet radiation. However, overproduction of melanin leads to skin hyperpigmentation disorders such as blemishes, freckles, melasma, age spots and melanoma (1). Tyrosinase (EC 1.14.18.1) is a key enzyme in melanin biosynthesis. It catalyzes the hydroxylation of L-tyrosine to L-3,4-dihydroxyphenylalanine (L-DOPA) and the oxidation of L-DOPA to (2*S*)-2-amino-3-(3,4-dioxo-

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cyclohexa-1,5-dien-1-yl)propanoic acid (DOPAquinone). DOPAquinone is spontaneously converted to 5,6-dioxo-2,3,5,6-tetrahydro-1*H*-indole-2-carboxylic acid (DOPAchrome). Then, DOPAchrome is transformed into 5,6-dihydroxyindole (DHI) and further oxidized and polymerized to melanin (2). Tyrosinase inhibitors prevent overproduction of melanin in the skin; kojic acid, α -arbutin, hydroquinone and vitamin C have been used as tyrosinase inhibitors in commercial skin whitening cosmetics (3). However, these tyrosinase inhibitors may have side-effects or risks causing skin allergies, inflammation, genotoxicity (4). Furthermore, some countries have banned kojic acid because of a potential connection to the development of cancer (5). Therefore, various phytochemicals have been investigated to be less toxic yet effective tyrosinase inhibitors. Several medicinal plants have been investigated and their *in vitro* and *in vivo* antityrosinase activities have been reported (6, 7).

Manilkara kauki (L.) Dubard is a tropical evergreen tree that belongs to the genus *Manilkara* of family *Sapotaceae*. The fresh ripe fruits are edible with a sweet taste and contain many nutrients such as proteins, fats, calcium, iron, magnesium, phosphorus, potassium, sodium, zinc, copper, vitamin C, thiamin, riboflavin and niacin (8). In traditional medicine in India, barks and roots of this plant are used as an astringent and to treat infantile diarrhea (9). Seeds are used as febrifuge, anthelmintic and antileprotic agents (9). Leaves are used as poultices for tumors (9). However, limited scientific studies of this plant have been found.

The aim of this study was to investigate the total phenolic and flavonoid contents as well as *in vitro* antioxidant and antityrosinase activities of crude extracts from various plant parts (fruits, leaves, seeds, stem barks and woods) of *M. kauki*.

EXPERIMENTAL

Chemicals and equipment

L-tyrosine, kojic acid, α -arbutin, 2,2-diphenyl-1-picrylhydrazyl (DPPH), Folin-Ciocalteu reagent, gallic acid and quercetin were purchased from Merck (Germany). Mushroom tyrosinase, 2,4,6-tri(2-pyridyl)-s-triazine (TPTZ), ferric chloride hexahydrate, aluminum chloride and Trolox were purchased from Sigma-Aldrich (USA).

A vacuum rotary evaporator (EYELA, Japan) and a microplate reader Multiscan GO (Thermo Fisher Scientific, USA) were used.

Plant materials

The fruits, leaves, seeds, stem barks and woods of *M. kauki* were collected from Bangkok, Thailand, in December 2015. The plants were identified by plant taxonomist in the Department of Botany, Faculty of Science, Chulalongkorn University, Thailand. The voucher specimens (BCU No. A015371) were deposited in the Professor Kasin Suvatabhandhu Herbarium, Chulalongkorn University, Thailand.

Preparation of plant parts extracts

Fresh fruits, leaves, seeds, stem barks and woods of *M. kauki* were dried in a hot air oven at 60 °C, then ground into powder. The ground samples (10 g each) were extracted

with methanol (70 mL × 3) for 72 h at room temperature or with water (70 mL × 3) for 20 minutes at 60 °C. Each extract was filtered through Whatman No. 1 filter paper and the filtrate was evaporated under reduced pressure to obtain methanol and aqueous crude extracts of five different parts of *M. kauki*.

In the case of stem barks the ground sample (500 g) was successively soaked in *n*-hexane (3 × 1 L), ethyl acetate (3 × 1 L) and methanol (3 × 1 L) for 72 h at room temperature and water (3 × 1 L) for 20 minutes at 60 °C. Each extract was filtered through Whatman No. 1 filter paper and the filtrate was evaporated under reduced pressure to obtain *n*-hexane, ethyl acetate, methanol and aqueous extracts of stem barks of *M. kauki*.

Determination of total phenolic content

The total phenolic content was determined using Folin-Ciocalteu reagent as previously described (10). The sample was dissolved in methanol (1 mg mL⁻¹). The assay was performed in triplicate for each analysis. Gallic acid (0–190 mg L⁻¹) was used as a standard phenolic compound. The total phenolic content was expressed in milligrams of gallic acid equivalents (GAE) per gram of extract.

Determination of total flavonoid content

The total flavonoid content was measured using the aluminum chloride colorimetric assay (11). The sample was prepared in a concentration of 1 mg mL⁻¹ in methanol. All analyses were carried out in triplicate. Quercetin (0–1000 mg L⁻¹) was used as a standard flavonoid. The total flavonoid content was expressed in milligrams of quercetin equivalents (QE) per gram of extract.

DPPH radical scavenging assay

DPPH assay was performed according to the previously reported method (12). The sample was dissolved in 95 % methanol (1 mg mL⁻¹). All analyses were performed in triplicate. The DPPH radical scavenging activity was expressed in the percentage of inhibition.

Ferric reducing antioxidant power (FRAP) assay

The FRAP assay was performed by a modified spectrophotometric assay as previously reported (13). The sample was dissolved in 95 % methanol (1 mg mL⁻¹). The standard curve was linear between 25 and 1000 μ g mL⁻¹ of Trolox. The results were expressed in milligrams of Trolox equivalents (TE) per gram of extract.

Antityrosinase activity assay

The antityrosinase activity was measured by a slightly modified spectrophotometric method (14). L-tyrosine was used as a substrate and dissolved in 0.1 mol L⁻¹ phosphate buffer (pH 6.5). Kojic acid and α -arbutin were used as positive controls and were also dissolved in 0.1 mol L⁻¹ phosphate buffer (pH 6.5). The sample was prepared in 20 % dimethyl sulfoxide (DMSO) in ethanol. In the assay, 8 µL of the sample solution was mixed with 133 µL of 0.1

mol L⁻¹ phosphate buffer (pH 6.5) and 47 μ L of 2.5 mmol L⁻¹ L-tyrosine. The reaction mixture was incubated for 5 min at room temperature. Then, 12 μ L of mushroom tyrosinase solution (3224 U mL⁻¹) in 0.1 mol L⁻¹ phosphate buffer (pH 6.5) was added to the reaction mixture. After incubation for 20 min at room temperature, the absorbance of the reaction mixture was recorded at 475 nm. The assay for each sample was performed in triplicate. The half-maximal inhibitory concentration (*IC*₅₀) values were calculated by linear regression analysis.

Statistical analysis

Statistical analyses were performed using a one-way analysis of variance (ANOVA) by SPSS program (version 17.0, SPSS Inc., USA). The significant difference between the means was determined by Duncan's multiple range test. p < 0.05 were considered significant.

RESULTS AND DISCUSSION

Extraction and chemical composition of M. kauki parts

The results of the percentage yield of the methanol and aqueous crude extracts of different parts of *M. kauki* are shown in Table I.

	Yield (%)		
Plant part	Methanol crude extract	Aqueous crude extract	
Fruits	11.0	19.1	
Leaves	11.6	9.0	
Seeds	5.3	8.7	
Stem barks	5.9	5.9	
Woods	3.4	4.5	

Table I. Percentage yield of crude extracts of M. kauki plant parts

The total phenolic content in the methanol crude extract of five different parts of M. *kauki* ranged between 14.83 and 176.56 mg GAE g⁻¹ crude extract whereas in aqueous crude extract it ranged 10.87 to 105.98 mg GAE g⁻¹ crude extract (Table II).

Total flavonoid content in the methanol crude extract of different parts of *M. kauki*, ranged from 15.67 to 821.67 mg QE g⁻¹ crude extract but in the aqueous crude extracts it was 14.33 to 800.67 mg QE g⁻¹ crude extract (Table II).

Leaves and stem barks showed higher amounts of total phenolic and flavonoid contents than fruits, seeds and woods.

	Total phenolic content		Total flavonoid content	
	(mg GAE g ⁻¹ crude extract)		(mg QE g ⁻¹ crude extract)	
Plant part	Methanol crude extract	Aqueous crude extract	Methanol crude extract	Aqueous crude extract
Fruits	$17.52 \pm 0.10^{\rm d}$	$14.83\pm0.10^{\rm d}$	$99.00 \pm 0.73^{\circ}$	$105.67 \pm 0.15^{\circ}$
Leaves	$101.75 \pm 1.63^{\rm b}$	$105.98\pm0.10^{\rm a}$	$776.67 \pm 2.08^{\rm b}$	$745.33 \pm 2.52^{\rm b}$
Seeds	$14.83\pm0.10^{\rm e}$	$10.87\pm0.02^{\rm e}$	$38.33\pm0.58^{\rm d}$	$43.33\pm0.42^{\rm d}$
Stem barks	176.56 ± 0.10^{a}	$99.73\pm0.77^{\rm b}$	821.67 ± 2.89^{a}	800.67 ± 2.08^a
Woods	$83.68 \pm 0.87^{\circ}$	$47.71 \pm 0.67^{\circ}$	$15.67 \pm 0.31^{\rm e}$	14.33 ± 0.25^{e}

Table II. Total phenolic and flavonoid contents of M. kauki plant parts

Data are presented as mean ± SD of triplicates.

a-e Values with different superscript letters within the same column are significantly different (p < 0.05).

Antioxidant activity of M. kauki parts

DPPH radical scavenging activity. – The results of DPPH radical scavenging activity testing of different parts of *M. kauki* are presented in Table III. Concerning the methanol crude extracts of five different parts of this plant, stem barks exhibited the highest scavenging DPPH capacity (57.9 ± 0.3 %), followed by leaves (47.0 ± 0.1 %), woods (43.5 ± 0.3 %), fruits (8.8 ± 0.2 %) and seeds (2.2 ± 0.1 %), resp., at a concentration of 1 mg mL⁻¹. For the aqueous crude extracts, the order of DPPH activity was as follows: leaves (55.5 ± 0.1 %), stem barks (48.9 ± 0.1 %), woods (14.8 ± 0.1 %), fruits (1.0 ± 0.1 %) and seeds (0.4 ± 0.03 %), resp., at a concentration of 1 mg mL⁻¹.

FRAP activity. – As shown in Table III, the methanolic crude extracts of leaves (209.74 \pm 1.28 mg TE g⁻¹ crude extract) and stem barks (211.59 \pm 1.40 mg TE g⁻¹ crude extract) exhibited strong FRAP activity but the aqueous crude extract of leaves (219.37 \pm 0.32 mg TE g⁻¹ crude extract) showed the highest FRAP activity. Methanol and aqueous crude extracts of fruits, seeds and woods showed weak FRAP activity.

	DPPH radical scavenging activity (%)		FRAP activity (mg TE g ⁻¹ crude extract)	
Plant part	Methanol crude extract	Aqueous crude extract	Methanol crude extract	Aqueous crude extract
Fruits	$8.80\pm0.16^{\rm d}$	$1.03\pm0.05^{\rm d}$	$11.96\pm0.64^{\rm d}$	$5.48\pm0.32^{\rm d}$
Leaves	$47.01\pm0.14^{\rm b}$	55.49 ± 0.06^{a}	$209.74\pm1.28^{\rm a}$	$219.37\pm0.32^{\rm a}$
Seeds	$2.21\pm0.05^{\rm e}$	$0.41\pm0.03^{\rm e}$	$16.41 \pm 0.85^{\circ}$	$4.19\pm0.32^{\rm d}$
Stem barks	57.87 ± 0.27^{a}	$48.89\pm0.14^{\rm b}$	$211.59 \pm 1.40^{\rm a}$	107.33 ± 0.96^{b}
Woods	$43.51 \pm 0.27^{\circ}$	$14.75 \pm 0.06^{\circ}$	$72.70 \pm 0.85^{\rm b}$	$41.59 \pm 0.85^{\circ}$

Table III. DPPH radical scavenging and FRAP activities of M. kauki plant parts

Data are presented as mean \pm SD of triplicates.

 $^{a-e}$ Values with different superscript letters within the same column are significantly different (p < 0.05).

Antityrosinase activity of M. kauki parts

Among the methanolic crude extracts, stem barks exhibited the strongest antityrosinase activity with an IC_{50} value of 0.25 ± 0.03 mg mL⁻¹, followed by leaves, fruits and seeds, resp. For the aqueous crude extracts, the order of antityrosinase activity was as follows: stem barks, leaves, fruits and seeds, resp., whereas, the methanol and aqueous crude extracts of woods showed no antityrosinase activity (Table IV). Furthermore, the IC_{50} values for mushroom tyrosinase inhibitory activity of α -arbutin and kojic acid were 7.05 \pm 0.22 mmol L⁻¹ and 57.49 \pm 0.35 μ mol L⁻¹, resp.

	<i>IC</i> ₅₀ (mg mL ⁻¹)		
Plant part	Methanol crude extract	Aqueous crude extract	
Fruits	$0.63 \pm 0.01^{\circ}$	$0.86 \pm 0.03^{\circ}$	
Leaves	$0.38\pm0.01^{\rm b}$	$0.50\pm0.01^{\rm b}$	
Seeds	$5.56\pm0.03^{\rm d}$	$6.73\pm0.03^{\rm d}$	
Stem barks	$0.25\pm0.03^{\rm a}$	$0.41\pm0.01^{\rm a}$	
Woods	No activity	No activity	

Table IV. Antityrosinase activity of M. kauki parts

Data are presented as mean ± SD of triplicates.

^{a-d} Values with different superscript letters within the same column are significantly different (p < 0.05).

The *Manilkara* spp. have been reported to possess pharmacological activities such as antioxidant, antimicrobial, anticancer, antidiabetic, antiulcer, analgesic, antityrosinase and immunomodulatory (15–19). The secondary metabolites of *Manilkara* spp. have also been reported to consist of flavonoids, phenolic compounds, tannins, triterpenes and saponins (18–21). Phenolic compounds and flavonoids are secondary metabolites of plants that have been shown to play an important role in antioxidant, antityrosinase, anticancer and other activities (22, 23). The inhibition of tyrosinase activity also depends on the hydroxyl group of the phenolic compounds. The hydroxyl group can develop a hydrogen bond at the active site of tyrosinase, resulting in steric hindrance or conformation changing and then leading to a lower enzymatic activity (24).

In the present study, both the methanol and aqueous crude extracts of leaves and stem barks showed strong DPPH radical scavenging, FRAP and antityrosinase activities, which might be ascribed to the presence of phenolic compounds and flavonoids. Similarly, previous studies of *Manilkara* spp. have reported a correlation between the antioxidant and antityrosinase activities and the amount of phytochemicals, including phenolic compounds and flavonoids (25). Some studies have reported that the plant extracts which contain high total phenolics or flavonoid contents exhibit strong antioxidant activity (15). Expectedly, the plant extracts with high amounts of phenolic compounds also possess high amounts of flavonoids (26, 27). In this study, stem barks of *M. kauki* displayed the highest *in vitro* antityrosinase activity compared with fruits, leaves, seeds and woods. The attention was, therefore, focused on stem barks. Hence, different polarities of solvents (*n*-hexane, ethyl acetate, methanol and water) were used for the extraction of active substances of diverse lipophilicities. These fractions were evaluated for phytochemical composition (phenolic compounds and flavonoids), antioxidant and antityrosinase activities.

Extraction and chemical composition of M. kauki stem barks

The extraction yield of *n*-hexane, ethyl acetate, methanol and aqueous extracts of *M*. *kauki* stem barks were 3.3, 1.5, 16.9 and 9.8 %, respectively.

The total phenolic and flavonoid contents of the extracts of stem barks are shown in Table V. The methanol extract (160.12 \pm 0.93 mg GAE g⁻¹ extract) exhibited the highest total phenolic content, followed by ethyl acetate, aqueous and *n*-hexane extracts, resp. In addition, the ethyl acetate extract (755.00 \pm 1.00 mg QE g⁻¹ extract) showed the highest total flavonoid content, followed by methanol, aqueous and *n*-hexane extracts, resp.

Fraction extract	Total phenolic content	Total flavonoid content
Fraction extract	(mg GAE g ⁻¹ extract)	(mg QE g ⁻¹ extract)
<i>n</i> -Hexane	10.98 ± 0.10^{d}	283.67 ± 1.15^{d}
Ethyl acetate	$111.46\pm0.02^{\rm b}$	755.00 ± 1.00^{a}
Methanol	160.12 ± 0.93^{a}	527.33 ± 0.58^{b}
Aqueous	$66.17 \pm 0.48^{\circ}$	$316.67 \pm 1.15^{\circ}$

Table V. Total phenolic and flavonoid contents of M. kauki stem barks

Data are presented as mean ± SD of triplicates.

^{a-d} Values with different superscript letters within the same column are significantly different (p < 0.05).

Antioxidant activity of M. kauki stem barks

DPPH radical scavenging activity. – The ethyl acetate and methanol extracts exhibited high DPPH radical scavenging activity with 55.7 ± 0.1 % and 54.2 ± 0.2 %, resp., at a concentration of 1 mg mL⁻¹, whereas *n*-hexane and aqueous extracts showed weak DPPH radical scavenging activity (Table VI).

Table VI. Antioxidant and antityrosinase activities of M. kauki stem barks

	Antioxidant	A tit	
Extract	DPPH radical scavenging activity (%)	FRAP activity (mg TE g ⁻¹ extract)	 Antityrosinase activity IC₅₀ (mg mL⁻¹)
<i>n</i> -Hexane	0.66 ± 0.03^{d}	$10.11\pm0.56^{\rm d}$	$9.58\pm0.10^{\rm d}$
Ethyl acetate	55.69 ± 0.13^{a}	$180.83 \pm 0.81^{\rm b}$	$0.24\pm0.04^{\rm a}$
Methanol	$54.22\pm0.17^{\rm b}$	221.59 ± 0.32^{a}	$1.62 \pm 0.01^{\mathrm{b}}$
Aqueous	$7.24 \pm 0.05^{\circ}$	$50.85 \pm 0.32^{\circ}$	$1.91 \pm 0.02^{\circ}$

Data are presented as mean ± SD of triplicates.

 $^{a-d}$ Values with different superscript letters within the same column are significantly different (p < 0.05).

FRAP activity. – The ethyl acetate and methanol extracts also exhibited potent FRAP activity (180.83 \pm 0.81 and 221.59 \pm 0.32 mg TE g⁻¹ extract, resp.), while, the *n*-hexane and aqueous extracts exhibited weak FRAP activity (Table VI).

Antityrosinase activity of M. kauki stem barks

The ethyl acetate extract ($IC_{50} = 0.24 \pm 0.04 \text{ mg mL}^{-1}$) exhibited the strongest antityrosinase activity, followed by methanol ($IC_{50} = 1.62 \pm 0.01 \text{ mg mL}^{-1}$), aqueous ($IC_{50} = 1.91 \pm 0.02 \text{ mg mL}^{-1}$) and *n*-hexane ($IC_{50} = 9.58 \pm 0.10 \text{ mg mL}^{-1}$) extract.

The partial purification of *M. kauki* stem barks indicated that high amounts of phenolic compounds and flavonoids were found in the ethyl acetate and methanol extracts as a medium to high polarity media. These extracts also demonstrated high antioxidant and antityrosinase activities. According to the reports on some other plants such as *Zingiber officinale, Armoracia rusticana, Bucida buceras* and *Phoradendron californicum,* the amounts of phenolic compounds and flavonoids, antioxidant and antityrosinase capacities also differed depending on the polarity of solvents for extraction (28–30).

CONCLUSIONS

This study was the first report on comparing phytochemical composition, antioxidant and antityrosinase activities of different parts of *M. kauki*. Its leaves and stem barks exhibited higher total phenolic and flavonoid contents, DPPH radical scavenging, FRAP and antityrosinase activities than fruits, seeds and woods. Furthermore, extraction with solvents of different polarities indicated that the ethyl acetate and methanol extracts of stem barks demonstrated high total phenolic and flavonoid contents, and antioxidant activity. However, only the ethyl acetate extract exhibited potent antityrosinase activity and may serve as a source of tyrosinase inhibitor for pharmaceutical and cosmetic applications.

We are inclined to believe that a high level of phenolic compounds/flavonoids led to good antioxidant and antityrosinase activities. This comes along with the fact that ethyl acetate extract of stem barks with a significantly higher amount of flavonoids exhibited also significantly higher antityrosinase activity, pointing to the fact that flavonoids play an important role in tyrosinase inhibition.

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