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MODELLING, INVESTIGATION OF PROCESS RESPONSES, SURFACE ASSESSMENT AND PARAMETRIC OPTIMIZATION IN POWDER MIXED ELECTRICAL DISCHARGE DIAMOND GRINDING OF TI6AL4V UTILIZING GREY-BASED TAGUCHI APPROACH

Summary

In this study, the powder mixed electrical discharge diamond grinding (PMEDDG) process was designed and a set-up for machining hard electrically conductive material surfaces built. The paper presents empirical models, investigation, an optimal setting of factors and the distinct surface production in the PMEDDG of Ti6Al4V with aluminium (Al) and silicon carbide (SiC) powder mixed dielectric fluid. The response surface methodology was applied to the modelling. One set of 32 experiments with Al powder and another set of 32 experiments with SiC powder mixed dielectric fluid were performed on a PMEDDG set-up. Current, pulse on time, wheel speed, duty cycle, and volumetric proportion of the powder were taken as input machining variables. Material removal rate and surface roughness were computed as outputs. The behaviour of the input factors against the responses was studied and compared with the SiC and Al powder mixed dielectric fluid used in the PMEDDG process. Further, scanning electron microscopy (SEM) investigations were carried out to determine the impact of different factors on the PMEDDG-produced surfaces as well as the effect of powder presence in the dielectric fluid on white recast layer thickness of the produced surfaces. The grey-based Taguchi approach was used to determine an optimal set of process variables when aluminium powder is used and a confirmation test was conducted on the optimal set to estimate the effectiveness of this approach.

Key words: powder mixed electrical discharge diamond grinding, analysis of variance (ANOVA), grey-based Taguchi approach, hybrid process, RSM, SEM, Ti6Al4V

1. Introduction

Titanium and its alloys are generally utilized in aerospace, automotive and biomedical industries as well as in various assembly companies because of their good strength/weight ratio, high resistance to corrosion, etc. Jilani et al. [1] carried out experiments with electrical discharge machining (EDM) of cemented carbide and proposed a model for single release. This study showed the impact of various factors and diverse carbide grades on the material removal rate (MRR), relative electrode wear (REW) and cavity size. The formulated models were utilized to determine the cavity size, swept volume/release and thickness of the resolidified layer. Mahdivian et al. [2] performed experiments with EDM of mild steel. They

formulated hypothetical models for MRR and surface quality. They also determined the impact of various factors on the responses. In their experimental work, Singh et al. [3] proposed the grey-based Taguchi approach for multiple response optimizations of EDM factors for the machining of aluminium-silicon carbide composites (10% wt). Tsai et al. [4] investigated EDM and suggested the use of the response surface methodology (RSM) for the investigation and modelling of the re-solidified layer of spheroidal graphite cast iron. Gangopadhyay et al. [5] performed experiments with EDM of M2 die steel and utilized the RSM approach for the formulation of surface roughness (SR), white layer thickness (WLT), and surface crack density models. They also revealed that current (I) and pulse on time (T_{on}) influence the integrity of the produced surface. In their experiments, Jeswani et al. [6] observed that blending of 4 gm/litre graphite powder in lamp oil improves the MRR and tool wear rate (TWR) in the EDM of mild steel. Modi et al. [7] performed the experiments with surface electrical diamond discharge grinding (SEDDG) of Ti6Al4V. They found that the application of the grey-based Taguchi methodology improves the productivity of machining. Lim et al. [8] reported that pulse duration, property of powder, electrodes polarity, current, and workpiece material influence the generation of mirror-like surfaces in powder mixed EDM. They also found that the spark gap in the powder mixing state is double when contrasted with EDM. Oczos et al. [9] clarified the philosophy and the uses of the hybrid machining process (HMP). They also described the issue related to the HMP and the necessity of additional progression in research endeavours to upgrade its uses. Mote et al. [10] conducted their experimental work with electrical diamond discharge grinding of high speed steel and observed that the inclusion of EDM and the grinding procedure improves the EDM effectiveness with respect to specific energy. They also formulated numerical models of the input factors and output responses to examine the relationship between input factors and the output responses. Jain, V K [27] reported on the performance of the HMP. Chakraborty et al. [11] proposed a modified principal component analysis method for multiple response optimization of process factors. This technique was utilized to acquire an optimal setting in two data sets in the wire EDM procedure. Lin et al. [12] performed experiments with computerised numerically controlled turning of SKD-11 steel and utilized the grey-based Taguchi approach in the multiple response optimization of process variables. Ravindra et al. [13] carried out experiments with the electrical discharge face grinding of erosion-abrasion based compound Al/SiC/B4C wheel. They studied the influence of the process factors on MRR and R_a. Umacharan et al. [14] performed tests with diamond electrical discharge drilling of Nimonic alloy. They described in detail the influence of process factors on R_a and the circularity of the drilled hole. Yadav et al. [15] conducted experiments with the cut-off grinding of Al/SiC/B₄C in a hybrid process. They reported the impact of process variables on MRR and R_a. Mali et al. [16] conducted experiments with the cut-off diamond electrical discharge grinding of Inconel 718 and optimized the process variables through the combined approach of orthogonal array and grey relational analysis. Modi et al. [17] did tests with the powder mixed EDM of Nimonic 80A with Al and Cr powder mixed dielectric fluid. They described the adequacy of EDM and concluded that the properties of the powders included in the dielectric fluid have impact on the EDM. Rao et al. [18] performed experiments with EDDG of reaction-bonded silicon carbide (RBSiC) ceramics. They reported on the impacts of grit sizes and polarities of diamond grinding wheel on the quality of a machined surface. Singh et al. [19] did experiments with the abrasive mixed diamond electrical discharge grinding of Nimonic alloy. They found that the MRR is affected by speed, current and concentration of powder. Nirala et al. [20] conducted tests with the abrasive mixed diamond electrical discharge grinding of Monel K-500. They developed an ANN model with the input process factors and responses. They also optimised the process factors by applying the ANN-NSGA-II approach. Barzani et al. [21] performed experiments with the abrasive mixed diamond electrical discharge grinding of Nimonic 80A. They utilized the fuzzy logic

approach to estimate the performance of this hybrid machining process. Unune et al. [22, 23] established numerical models of responses and analysed the effectiveness of artificial neural network and RSM models in forecasting the MRR and surface roughness in the machining of Inconel 718 with the SEDDG process. They also reported that the blending of powders in dielectric fluid enhances the MRR and improves the surface quality. Tripathy et al. [24] did tests with powder mixed EDM of H-11 tool steel. They studied the influence of the input process factors on the responses and utilised the TOPSIS method for the estimation of an optimal setting of process factors. Modi et al. [25] used the dimensional analysis method for establishing MRR and surface roughness models for drilling with the powder mixed electrical discharge machining of EN 31 steel. They also explored the effect of procedure factors on the procedure results.

From the above literature survey, it can be seen that the machining of Ti6Al4V by applying the PMEDDG procedure with silicon carbide (SiC) and aluminium (Al) powder mixed dielectric fluid is another area under development. The goals of this study are as follows: (a) to investigate the impact of process factors on MRR and average roughness of the surface (R_a) in the machining of Ti6Al4V by applying the PMEDDG process with SiC and Al powder mixed dielectric fluid; (b) to formulate numerical models of MRR and R_a by applying the RSM method; (c) to examine the effect of process factors on the machined surfaces and on the white recast layer thickness (WRLT) through the analysis of scanning electron microscopy (SEM) micrographs; (d) to compute an optimal set of factors in the machining of Ti6Al4V by applying the PMEDDG process with Al powder mixed dielectric fluid in multiple output optimizations by utilizing the grey-based Taguchi technique.

To fulfill these objectives, one set of 32 experiments with Al powder and another set of 32 experiments with SiC powder mixed dielectric fluid were performed on a PMEDDG setup. Mathematical models of responses were developed with Design-Expert software. The grey-based Taguchi approach was utilized to discover an optimal set of process variables in the PMEDDG of Ti6Al4V with aluminium powder mixed dielectric fluid in the multiple response optimization. For this purpose, one set of 18 experiments with Al powder mixed dielectric fluid was conducted in accordance with the L18 Taguchi-orthogonal array using the PMEDDG set-up. To verify the effectiveness of the approach, the optimum combination suggested by this approach and the obtained results were compared with initial values.

2. Experimentation on the PMEDDG set-up

Two sets of 32 experiments i.e. one set of 32 experiments with aluminium (Al) powder and the second set of 32 experiments with silicon carbide (SiC) powder mixed dielectric fluid were performed on a PMEDDG set-up for the processing of Ti6Al4V. All details of this set-up are available in Modi et al. [26]. The schematic photographic diagram of the PMEDDG set-up is shown in Fig. 1.



Fig.1 Schematic diagram of electrical discharge diamond grinding set-up

0.02%

3.7 %

0.13%

The details about the bronze diamond wheel are given in Table 1. The material of the workpiece is Ti6Al4V. Its composition is shown in Table 2. The workpiece is rectangular and flat in shape. Its dimensions are 65 mm x 20 mm x 20 mm.

			88	-			
Abrasive	Diameter	Thickness	Bond material	Concentration	Bore	Depth of abrasive	Grit size
Diamond	100 mm	10 mm	Bronze	75 %	32 mm	5 mm	80/100
Table 2 Co	mposition o	f Ti6Al4V (C	Grade-5)				
(Carbon		Ti	Al		V	Fe

Table 1 Particulars of bronze diamond grinding wheel

90.1 %

Eqs 1 and 2 are utilized to compute the MRR (mg/minute) and MRR (mm³/minute), respectively, after every experiment.

$$MRR(mg/minute) = \left[\frac{(WWBM) - (WWAM)}{t}\right] \cdot 1000$$
(1)

6.05%

MRR (mm³/minute) =
$$\left[\frac{(WWBM) - (WWAM)}{t \cdot \rho}\right] \cdot 1000$$
 (2)

The machining time (t) in this procedure is equivalent to 45 minutes and the density of the material is $\rho = 4.37 \text{ gm/cm}^3$. The terms WWBM and WWAM, used in Eqs 1 and 2, refer to the weight of the workpiece before and after the machining. The MRR was evaluated by using a Wensar electronic balance. R_a was assessed by a Surtronic-25 SR analyser at the c-off value of 2.5 mm. An advanced tachometer was used for the estimation of the grinding wheel speed.

3. RS modelling

The RSM is a merger of arithmetic and statistical methodology. This method is utilized to determine the relationship between input and output machining factors. All experiments were performed as per central composite rotary design (CCRD) plan.

The response surface is described by Eq. 3. It is the adequate method to access the second order model.

$$Z = b_0 + \sum_{p=1}^m b_p y_p + \sum_{p=1}^m b_{pp} y_p^2 + \sum_{q>p} \sum_{p \neq p} b_{pq} y_p y_q.$$
(3)

Where,

Ζ	predicted response	b_{pp}	quadratic term coefficient
b_0	free term coefficient	b_{pq}	cross product term coefficient
b_p	linear term coefficient	p	1,2,,m
ģ	2,3,, m+1	m	No. of variables
v	coded variables		

4. Experimental data with aluminium powder

In total, 32 experiments were conducted on the PMEDDG set-up with aluminium powder mixed dielectric fluid. The level and value of the factors associated with these experiments are shown in Table 3.

Table 3	Level and v	value of	f factors	in alum	inium/s	ilicon o	carbide	powder	mixed	dielectric	fluid in	PMEI	DDG
1	process												

Parameters	Unit	-2	-1	0	+1	+2
Current (I)	Ampere	1	3	5	7	9
Pulse on time (T_{on})	μ-s	40	70	100	130	160
Wheel speed (S)	RPM	350	450	550	650	750
Duty cycle (DC)	%	0.59	0.63	0.67	0.71	0.75
Al/SiC powder concentration (CP)	gm/litre	1.5	2	2.5	3	3.5

The practical CCRD plan and its responses in the aluminium powder mixed dielectric fluid in the PMEDDG process are shown in Table 4.

 Table 4
 Practical CCRD plan and its responses in aluminium powder mixed dielectric fluid in PMEDDG process

Eve	Current	Dulso on time	Wheel speed	Duty avala	Powder	Experime	ental
No.	<i>I</i> / Ampere	T_{on}/μ -s	S / RPM	Duty Cycle DC / %	concentration	MRR /	R / µm
	1	0 1			<i>CP</i> / gm/litre	mm ³ /minute	$R_a / \mu m$
1	-2.00	0.00	0.00	0.00	0.00	0.272	2.69
2	-1.00	1.00	1.00	1.00	-1.00	0.44	3.80
3	0.00	0.00	0.00	0.00	0.00	0.48	3.84
4	-1.00	-1.00	-1.00	1.00	-1.00	0.206	3.93
5	0.00	0.00	0.00	0.00	0.00	0.49	3.82
6	1.00	1.00	-1.00	1.00	-1.00	0.614	5.46
7	-1.00	1.00	-1.00	1.00	1.00	0.33	4.41
8	0.00	0.00	0.00	0.00	0.00	0.55	3.76
9	0.00	0.00	-2.00	0.00	0.00	0.336	4.45
10	0.00	0.00	0.00	0.00	2.00	0.65	4.73
11	0.00	0.00	0.00	0.00	0.00	0.51	3.96
12	0.00	0.00	0.00	0.00	-2.00	0.46	3.58
13	1.00	-1.00	-1.00	1.00	1.00	0.482	5.15
14	1.00	1.00	-1.00	-1.00	1.00	0.84	4.27
15	-1.00	-1.00	1.00	-1.00	-1.00	0.398	2.54
16	-1.00	1.00	-1.00	-1.00	-1.00	0.384	3.02
17	0.00	2.00	0.00	0.00	0.00	0.63	3.98
18	-1.00	1.00	1.00	-1.00	1.00	0.54	2.87
19	-1.00	-1.00	-1.00	-1.00	1.00	0.294	3.06
20	1.00	1.00	1.00	1.00	1.00	0.76	5.02
21	0.00	0.00	0.00	0.00	0.00	0.53	3.77
22	-1.00	-1.00	1.00	1.00	1.00	0.36	3.50
23	0.00	0.00	0.00	0.00	0.00	0.48	3.95
24	0.00	0.00	2.00	0.00	0.00	0.66	3.53
25	1.00	-1.00	-1.00	-1.00	-1.00	0.58	3.69
26	1.00	1.00	1.00	-1.00	-1.00	0.88	3.65
27	2.00	0.00	0.00	0.00	0.00	0.921	4.80
28	1.00	-1.00	1.00	1.00	-1.00	0.63	4.28
29	0.00	-2.00	0.00	0.00	0.00	0.374	3.13
30	0.00	0.00	0.00	2.00	0.00	0.405	4.88
31	0.00	0.00	0.00	-2.00	0.00	0.552	3.01
32	1.00	-1.00	1.00	-1.00	1.00	0.678	3.23

4.1 Numerical modelling of practical test responses for aluminium powder mixed dielectric fluid

The numerical models for MRR and R_a are acquired by the use of Design Expert software [31]. These models are delineated by Eqs 4 and 5 as follows:

$$MRR_{AI} = 0.51 + 0.16I + 0.070T_{on}^{2} + 0.067S - 0.044DC + 0.022CP + 0.022I^{2}$$
(4)

$$R_{a4l} = 3.88 + 0.49 I + 0.20 T_{on}^2 + 0.25 S + 0.54 DC + 0.14 CP - 0.078 T_{on}^2 + 0.072 CP^2$$
(5)

From Figs 2 and 3 it can be seen that the predicted values from the empirical model are in a good agreement with the actual values of MRR and R_a . Different graphs have been created between the input factors and the responses by using Eqs 4 and 5. These graphs are shown in Figs 6 to 13.



for MRR with Al powder

Fig. 3 Plot of actual and predicted output for R_a with Al powder

4.2 ANOVA investigation for aluminium powder mixed dielectric fluid

The results of the analysis of variance (ANOVA) are shown in Table 5. It is observed that the measure of p is under 0.05 in the models. Hence, it is presumed that the above developed numerical models of the process outcomes are satisfactory to explain the relationship between the process factors and the responses.

			MRR m	odel				R _a mo	odel	
Source	DF	SS	MS	F	Р	DF	SS	MS	F	Р
Model	6	0.90	0.15	95.7	< 0.0001	7	16.1	2.31	84.8	< 0.0001
Residual error	25	0.04	0.0015			24	0.65	0.027		
Total	31	0.94				31	16.7			

Table 5 ANOVA of MRR and Ra models with aluminium powder mixed dielectric fluid

5. Experimental data with silicon carbide powder

Another set of 32 experiments were conducted on the PMEDDG set-up with silicon carbide powder mixed dielectric fluid. The level and value of factors associated with these experiments are shown in Table 3.

The practical CCRD plan and its responses in the silicon carbide powder mixed dielectric fluid in the PMEDDG process are shown in Table 6.

Eve	Cumont	Dulas on time	Wheel aread	Duty avala	Powder	Experime	ental
Exp.	Current		wheel speed		concentration	MRR /	_ /
No.	I / Ampere	T_{on}/μ -s	S/RPM	DC / %	CP / gm/litre	mm ³ /minute	$R_a / \mu m$
1	1.00	-1.00	1.00	1.00	-1.00	0.567	4.32
2	-1.00	1.00	1.00	1.00	-1.00	0.406	3.89
3	-1.00	-1.00	-1.00	1.00	-1.00	0.189	4.02
4	0.00	0.00	0.00	0.00	0.00	0.442	3.93
5	0.00	0.00	2.00	0.00	0.00	0.595	3.57
6	1.00	1.00	1.00	1.00	1.00	0.687	5.1
7	0.00	0.00	0.00	0.00	0.00	0.503	3.83
8	-1.00	-1.00	1.00	-1.00	-1.00	0.362	2.57
9	0.00	0.00	0.00	0.00	0.00	0.449	3.9
10	-1.00	-1.00	1.00	1.00	1.00	0.325	3.55
11	0.00	0.00	-2.00	0.00	0.00	0.307	4.53
12	1.00	1.00	1.00	-1.00	-1.00	0.792	3.69
13	-1.00	1.00	1.00	-1.00	1.00	0.49	2.92
14	0.00	0.00	0.00	0.00	-2.00	0.419	3.63
15	-1.00	1.00	-1.00	1.00	1.00	0.302	4.5
16	0.00	0.00	0.00	0.00	0.00	0.465	4.02
17	-1.00	1.00	-1.00	-1.00	-1.00	0.349	3.05
18	1.00	-1.00	1.00	-1.00	1.00	0.6102	3.26
19	-2.00	0.00	0.00	0.00	0.00	0.251	2.76
20	0.00	0.00	0.00	2.00	0.00	0.3645	4.93
21	1.00	1.00	-1.00	-1.00	1.00	0.765	4.32
22	1.00	-1.00	-1.00	-1.00	-1.00	0.522	3.73
23	0.00	0.00	0.00	0.00	2.00	0.593	4.81
24	0.00	2.00	0.00	0.00	0.00	0.572	4.02
25	-1.00	-1.00	-1.00	-1.00	1.00	0.266	3.11
26	0.00	0.00	0.00	0.00	0.00	0.479	3.83
27	0.00	-2.00	0.00	0.00	0.00	0.3366	3.16
28	1.00	1.00	-1.00	1.00	-1.00	0.562	5.57
29	0.00	0.00	0.00	-2.00	0.00	0.4968	3.04
30	0.00	0.00	0.00	0.00	0.00	0.433	4
31	2.00	0.00	0.00	0.00	0.00	0.8289	4.85
32	1.00	-1.00	-1.00	1.00	1.00	0.439	5.22

Table 6	Practical plan for CCRD	and outputs for P	M EDDG process	with silicon ca	arbide mixed	dielectric fluid
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5.1 Numerical modelling of practical test responses for silicon carbide powder mixed dielectric fluid

The empirical models for MRR and R_a are obtained by the utilization of Design-Expert software [31]. These developed models are outlined by Eqs 6 and 7.

$$MRR_{SiC} = 0.46 + 0.14I + 0.064T_{on} + 0.059 - 0.039DC + 0.020CP + 0.019I^{2}$$
(6)

$$R_{aSiC} = 3.93 + 0.49I + 0.21T_{on} - 0.26S + 0.55DC + 0.15CP - 0.085T_{on}^{2} + 0.073CP^{2}$$
(7)

The predicted values obtained through the developed model are found to be in a good agreement with the actual experimental values of MRR and R_a as depicted in Figs 4 and 5, respectively.



for R_a with SiC powder

5.2 ANOVA investigation for silicon carbide powder mixed dielectric fluid

The obtained ANOVA results are shown in Table 7. It can be seen that the proportion of p is under 0.05 for the models. Thus, it is assumed that the developed empirical models of the process results are essential to clarify the relationship between the process factors and the responses.

Table 7 ANOVA of MRR and R_a models with silicon carbide powder mixed dielectric fluid

			MRR M	odel				$R_a \operatorname{Mo}$	odel	
Source	DF	SS	MS	F	Р	DF	SS	MS	F	Р
Model	6	0.73	0.12	90.4	< 0.0001	7	16.67	2.38	81.77	< 0.0001
Residual error	25	0.033	0.0013			24	0.70	0.029		
Total	31	0.76				31	17.36			

Various graphs were created for input factors and responses by using Eqs 6 and 7. These are shown in Figs 6 to 13.



Fig. 6 Effect of I on MRR with distinct DC (S = 550 RPM, T_{on} = 100µ-s, CP = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)



Fig. 7 Effect of I on MRR with distinct T_{on} (S = 550 RPM, DC = 0.67, CP = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)



Fig. 8 Effect of *I* on MRR with distinct $S(T_{on} = 100\mu$ -s, DC = 0.67, CP = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)



Fig. 9 Effect of S on MRR for with distinct $I(T_{on}=100\mu\text{-s}, DC=0.67, CP=2.5 \text{ gm/litre}, \text{Al-P}=\text{presence of aluminium powder in dielectric fluid}, SiC-P=\text{presence of silicon carbide powder in dielectric fluid})$



Fig. 10 Effect of *I* on R_a with distinct *DC* (*S* = 550 RPM, $T_{on} = 100\mu$ -s, *CP* = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)



Fig. 11 Effect of *I* on R_a with distinct T_{on} (S = 550 RPM, DC = 0.67, CP = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)



Fig. 12 Effect of *I* on R_a with distinct *S* ($T_{on} = 100\mu$ -s, DC = 0.67, CP = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)



Fig. 13 Effect of *S* on R_a with distinct $I(T_{on} = 100\mu$ -s, DC = 0.67, CP = 2.5 gm/litre, Al-P = presence of aluminium powder in dielectric fluid, SiC-P = presence of silicon carbide powder in dielectric fluid)

6. Analysis and interpretations of results

The analysis of the results is summarised in Table 8 which includes also the explanation of Figs 6 to 13.

Fig. number	Two dashed/ Three dotted line	Dielectric fluid (with Al and SiC powder)	Pattern 1 in Fig.	Cause for curve trend 1 in Fig.	Pattern 2 in Fig.	Cause for curve trend 2 in Fig.
6	Two dashed line	With Al powder	MRR increase with a rise	Extra energy is created by spark	MRR increase with a falling	T _{on} is rising up consistently to get the decreasing tendency in DC. This is why
	Three dotted line	With SiC powder	in ampere current.	pattern in I.	pattern in DC.	ionization and flushing of DF.

Table 8 Analysis and explanation of results from Figs 6 to 13

7	dashed line Three	With Al powder With SiC	<u> </u>	-	MRR increase with a decrease in T _{on} .	T _{on} is rising up consistently. This is why enough time is available for the conduction.		
	dotted line	powder			011			
8	dashed line	With Al powder	-	-	MRR increase with	Extra energy created by spark is scattered due to an increase in S. It		
	Three dotted line	With SiC powder			in S.	the crushing zone increases.		
9	Two dashed line	With Al powder	MRR increase with a rise	Flushing rises and gap-width falls because of	MRR increase with	Extra energy is created by spark		
	Three dotted line	With SiC powder	in wheel speed.	an increase in wheel speed.	in I.	occause of fising pattern in I.		
10	Two dashed line	With Al powder	R _a increase with an	Extra energy is created by spark	R _a increase with an	T _{on} is diminishing consistently to get the rising pattern in DC. This is why less time is qualitable for deinsisting and		
	Three dotted line	With SiC powder	increase in I.	pattern in I.	DC.	flushing of DF.		
11	Two dashed line	With Al powder		_	R _a increases with a rise in	T _{on} is rising constantly with the goal that enough time is available for warmth conduction of workpiece. MRR rises		
	Three dotted line	With SiC powder			pulse on time.	because of conditioning of workpiece and subsequently Ra increases because of generation of larger craters.		
12	Two dashed line	With Al powder	-	-	R_a falls with	Every single evacuated molecule from IEG is removed by productive flushing. This is acquired by an increase in S.		
_	Three dotted line	With SiC powder	-		a rise in S.	The adherence of evacuated molecules from workpiece is reduced.		
13	Two dashed line	With Al powder	R_a falls with	Flushing increases with an	R _a increase with an	Extra energy is created by spark		
	Three dotted line	$\begin{tabular}{ccc} \hline Three & With SiC \\ dotted line & powder \end{tabular} a rise in S. increase in S and \\ reduces R_{a.} \end{tabular}$		increase in I.	because of rising pattern in I.			

*Variation in MRR and R_a in PMEDDG procedure with SiC and Al powder mixed dielectric fluid (comment on Table 8)

Figs 6 to 9 show that the mechanism of the removal of metal in the PMEDDG process with the powder mixed dielectric fluid is such because of the occurrence of the (spark + series) discharge and the grinding between the wheel and the workpiece. The molecules of the SiC/Al powder in the interelectrode gap (IEG) get empowered and quickened by the connected electric field and then they move along criss-cross paths. The powder molecules help the breakdown in the IEG and increase the spark gap between the electrodes. The series discharge happens in the IEG because of the occurrence of short circuiting between the electrodes. This happens due to a reduction in the insulating strength of the dielectric fluid because of the bridging effect of powder molecules. The thermal conductivity of an aluminium powder molecule is higher than that of the silicon carbide powder, which subsequently results in a higher MRR. Figs 10 to 13 show that the spark gap between the electrodes is larger with aluminium powder mixed dielectric fluid. In addition, more silicon carbide powder molecules are available in the IEG because silicon carbide powder possesses higher density resulting in the worst surface generation as compared to the case with aluminium powder in the IEG. This investigation suggested that the best results are acquired in the PMEDDG process with the aluminium powder mixed dielectric fluid as compared to the SiC powder mixed dielectric fluid.

7. Investigation of PMEDDG-produced surface

SEM examinations were conducted to study the impact of different procedure factors on surfaces created using the PMEDDG process with SiC and Al powder mixed dielectric fluid. The SEM investigation has also been conducted to see the effect of the presence of SiC and Al powder mixed dielectric fluid on the white recast layer thickness.

SEM was used at 500X magnification. The scale used in all SEM images was 20 μ m. A flat machined workpiece was prepared for the SEM investigation in the following steps: a) 4/0 sand paper was utilized to granulate the generated surfaces, b) the generated surfaces were rinsed and brightened by polishing followed by etching with an etching solution (accumulation of H₂O = 88 %, HNO₃ = 10 %, and HF = 2 %) for one minute.

The influence of current (I), and T_{on} in the case of the surfaces produced by the PMEDDG of Ti6Al4V (with aluminium and silicon carbide powder mixed dielectric fluid) is shown in Figs 14 (a to d), and 15 (a to d), respectively. The following has been deduced by analysing the SEM pictures. The craters on the surface are broader and larger with an increase in I or T_{on} . This happens because of the creation of extra energy due to the rising pattern in I or T_{on} . It is also observed from the SEM pictures that the craters on the surface are smaller and cramped with aluminium in comparison to the silicon carbide powder mixed dielectric fluid in the case of the EDDG of Ti6Al4V.



Fig. 14 Effect of current on PMEDDG-produced surface; a) with Al powder, I = 2 ampere; b) with SiC powder, I = 2 ampere; c) with Al powder, I = 10 ampere; d) with SiC powder, I = 10 ampere; $(S = 550 \text{ RPM}, DC = 0.80, T_{on} = 300 \text{ }\mu\text{-s}, CP = 5 \text{ gm/litre})$



Fig. 15 Effect of T_{on} on PMEDDG-produced surfaces; a) with Al powder, $T_{on} = 100 \ \mu$ -s; b) with SiC powder, $T_{on} = 100 \ \mu$ -s; c) with Al powder, $T_{on} = 200 \ \mu$ -s; d) with SiC powder, $T_{on} = 200 \ \mu$ -s; (I = 6 ampere, S = 550 RPM, DC = 0.80, CP = 5 gm/litre)

The influence of DC in the case of the surfaces produced by the PMEDDG of Ti6Al4V (with aluminium and silicon carbide powder mixed dielectric fluid) is shown in Figs 16a to d. The following has been deduced by analysing the SEM pictures. The craters on the surface are broader and larger with an increase in DC, so that R_a rises due to the occurrence of a continuous spark. It is also observed from the SEM pictures that the craters on the surface are smaller and cramped with aluminium in comparison to silicon carbide powder mixed dielectric fluid in the case of the EDDG of Ti6Al4V.



Fig. 16 Effect of *DC* on PMEDDG-produced surface; a) with Al powder, DC = 0.70; b) with SiC powder, DC = 0.70; c) with Al powder, DC = 0.75; d) with SiC powder, DC = 0.75; (*I* = 6 ampere, *S* = 550 RPM, $T_{on} = 100 \mu$ -s, CP = 5 gm/litre)

The influence of wheel speed (S) in the case of the surfaces produced by the PMEDDG of Ti6Al4V (with aluminium and silicon carbide powder mixed dielectric fluid) is shown in Figs 17a to d. The following has been deduced by analysing the SEM pictures. The craters on the surface are thinner and smaller with an increase in S, so that R_a falls because of productive flushing in the IEG and, furthermore, the chances of arcing in the IEG decrease. Subsequently, the increase in S is the main reason for the production of thinner and smaller craters on the machined surface.



Fig. 17 Effect of S on PMEDDG-produced surface; a) with Al powder, S = 650 RPM; b) with SiC powder, S = 650 RPM; c) with Al powder, S = 750 RPM; d) with SiC powder, S = 750 RPM; $(I = 6 \text{ ampere}, DC = 0.80, T_{on} = 100 \text{ }\mu\text{-s}, CP = 5 \text{ gm/litre})$

It can be seen in all SEM pictures that craters on the PMEDDG-generated surfaces are narrower and smoother when aluminium powder was used as compared to the silicon carbide powder mixed dielectric fluid. It was observed that the gap between the anode and the cathode when aluminium powder was used is marginally larger when contrasted with the silicon carbide powder mixed dielectric liquid, which is attributable to its lower electrical resistivity. This smallest spark gap would be accountable for extreme gas blast pressure when silicon carbide powder is used. Furthermore, the silicon carbide powder density is higher than the aluminium powder density, resulting in arcing rather than sparking. Consequently, R_a is higher in the case of silicon carbide when contrasted with aluminium powder. Al powder delivered superior surface finish as compared to SiC powder.

7.1 Investigation of PMEDDG-produced WRLT

By investigating the SEM pictures shown in Figs 18a and b, it is seen that WRLT is lower in the case of the Al powder mixed dielectric fluid when contrasted with the SiC powder mixed dielectric fluid in the PMEDDG of Ti6Al4V. This happens because the presence of SiC powder in the IEG is responsible for the occurrence of arcing instead of sparking.



Fig. 18 Effect of powders on PMEDDG-produced WRLT; a) with Al powder, $t = 30 \,\mu\text{m}$, (b) with SiC powder, $t = 44 \,\mu\text{m}$, (I = 6 ampere, $T_{on} = 200 \,\mu\text{s}$, DC = 0.80, S = 750 RPM, CP = 5 gm/litre)

It is noted from the investigation and interpretation of the results as well as from the study of the SEM images that the best result is obtained in the PMEDDG process of Ti6Al4V in the presence of aluminium powder mixed dielectric fluid. The reason for this is that one set of 18 tests was conducted in accordance with L18 Taguchi orthogonal array on the set-up in the presence of aluminium powder mixed dielectric fluid.

8. Multiple response optimization by applying grey-based Taguchi method

The procedure for the application of the grey-based Taguchi method is shown in Fig. 19.

There are three sorts of signal-to-noise (SN) proportions: the lower-is-the-better, the higher-is-the-better, and the nominal-is-the-better, and they are communicated as [29]. The SN ratio is determined by using Eqs 8, 9, and 10.

(i) Lower-is-the-better

$$\frac{S}{N} = -10\log\left(\frac{1}{n}\sum_{i=1}^{n}y_{i}^{2}\right)$$
(8)

(ii) Higher-is-the-better

$$\frac{S}{N} = -10\log\left(\frac{1}{n}\sum_{i=1}^{n}\frac{1}{y_{i}^{2}}\right)$$
(9)

(iii) Normal-is-the-better

$$\frac{S}{N} = -10\log\left(\frac{1}{ns}\sum_{i=1}^{n}y_{i}^{2}\right)$$
(10)

where y_i is the *i*th estimation of the estimated output, *n* is the number of runs and *s* is the standard deviation.



Fig.19 Procedure for grey-based Taguchi method

Here,			
$x_{i}^{\left(0 ight) }\left(t ight)$	original sequence	$x_{0}^{\left(st ight) }\left(t ight)$	reference sequence
$\min x_i^{(0)}(t)$	minimum value of $x_i^{(0)}(t)$	$x_i^*(t)$	comparability sequence
$\max x_i^{(0)}(t)$	maximum value of $x_i^{(0)}(t)$	ζ	distinguishing coefficient $\zeta \in [0,1]; \zeta = 0.5$
$\Delta_{0i}(t)$	deviation sequence	Q	total observation data
Δ_{\min}	smallest value of Δ_{0i}	$\Delta_{ m max}$	largest value of Δ_{0i}

The level and value of factors associated with these experimental tests are shown in Table 9.

Parameters	Unit	Level 1	Level 2	Level 3
Al powder concentration (CP)	gm/litre	1.5	3	-
Current (I)	Ampere	3	7	11
Pulse on time (T_{on})	μ-s	100	150	200
Wheel speed (S)	RPM	450	650	850
Duty cycle (DC)	%	0.61	0.69	0.77

Table 9 Level and value of factors in aluminium powder mixed dielectric fluid in PMEDDG process

The L18 orthogonal array, responses and their signal-to-noise ratios (SNR), grey relational normalization (GRN), grey relational coefficient (GRC), grey relational grade (GRG), and signal-to noise-ratio value for the overall GR grade are given in Table 10.

 Table 10
 Grey relational normalization (GRN), grey relational coefficient (GRC), grey relational grade (GRG), and signal-to-noise ratio (SNR) value for overall GR grade

Ex.	Control factor Responses		SNR value Normalized SNR value		Grey relational co-efficient (GRC)		GR grade	SN ratio value							
No	СР	I	Ton	S	DC	MRR / mm ³ /minute	<i>Ra /</i> μm	MRR	Ra	MRR	Ra	MRR	Ra	(GRG)	GR grade
1	1	1	1	1	1	0.521	2.21	-5.66	-6.89	0.000	0.00	0.333	0.333	0.33	-9.56
2	1	1	2	2	2	0.540	2.60	-5.35	-8.30	0.052	0.15	0.345	0.370	0.36	-8.94
3	1	1	3	3	3	0.660	3.20	-3.61	-10.10	0.345	0.34	0.433	0.431	0.43	-7.31
4	1	2	1	1	2	0.790	3.40	-2.05	-10.63	0.606	0.40	0.560	0.453	0.51	-5.93
5	1	2	2	2	3	0.880	4.01	-1.11	-12.06	0.764	0.55	0.679	0.526	0.60	-4.42
6	1	2	3	3	1	0.659	4.65	-3.62	-13.35	0.342	0.69	0.432	0.614	0.52	-5.65
7	1	3	1	2	1	0.956	4.15	-0.39	-12.36	0.884	0.58	0.812	0.544	0.68	-3.39
8	1	3	2	3	2	0.980	4.63	-0.18	-13.31	0.920	0.68	0.863	0.611	0.74	-2.67
9	1	3	3	1	3	0.835	5.80	-1.57	-15.27	0.687	0.89	0.615	0.819	0.72	-2.91
10	2	1	1	3	3	0.780	3.60	-2.16	-11.82	0.588	0.45	0.548	0.476	0.51	-5.83
11	2	1	2	1	1	0.625	4.70	-4.08	-14.15	0.265	0.70	0.405	0.621	0.51	-5.81
12	2	1	3	2	2	0.726	4.98	-2.78	-13.94	0.483	0.75	0.492	0.666	0.58	-4.77
13	2	2	1	2	3	1.025	4.35	0.21	-12.77	0.986	0.62	0.972	0.571	0.77	-2.27
14	2	2	2	3	1	0.880	5.80	-1.11	-15.27	0.764	0.89	0.679	0.819	0.75	-2.53
15	2	2	3	1	2	0.760	5.40	-2.38	-14.65	0.550	0.82	0.526	0.739	0.63	-3.99
16	2	3	1	3	2	0.915	4.70	-0.77	-13.44	0.820	0.70	0.736	0.621	0.68	-3.38
17	2	3	2	1	3	1.035	5.90	0.30	-16.31	1.000	1.00	1.000	1.000	1.00	-0.02
18	2	3	3	2	1	0.840	6.54	-1.51	-15.42	0.696	0.91	0.622	0.840	0.73	-2.74

The response and the ANOVA results for GRG are given in Tables 11 and 12, respectively.

Table 11 Response table for grey relational grade

Symbol	Response table for GRG					
Symbol	Level 1	Level 2	Level 3	Max-Min		
СР	0.5419	0.6839		0.1420		
Ι	0.4536	0.6296	0.7554	0.3018		
T_{on}	0.5789	0.6586	0.6012	0.0797		
S	0.6159	0.6187	0.6041	0.0147		
DC	0.5867	0.5807	0.6712	0.0905		

 Table 12
 ANOVA table for grey relational grade

Sympol	ANOVA table for GRG						
Symbol	DF	SS	MS	F	Р	C (%)	
СР	1	0.0907	0.0907	3.96	0.064	15	
Ι	2	0.2757	0.1379	11.3	0.001	61	
Error	8	0.0394	0.0049			11	
Total	17	0.4575				100	

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For conducting an ANOVA investigation, *Mini-Tab* software is utilized. From Table 12 it can be seen that I and CP are important factors. This investigation was undertaken for a degree of significance of 5%, that is, for a degree of certainty of 95%. The F examination [30] is utilized to find out which process factors significantly affect the output. As a rule, the change of process factors significantly affects the output when the F value is higher. The percentages of distinct factors in the grey-based Taguchi method are CP (15%), T_{on} (5%), S (1%), DC (7%), and I (61%). The response graph for the average grey relational grade (AGRG) is delineated in Fig. 20.



Fig. 20 Response chart of average grey relational grade

An optimal set of various factors is $PC_2I_3.T_{on2}.S_2.DC_3$. A confirmation test is performed at the optimal setting and its outcomes are given in Table 13. An improvement in the MRR by 100%, an increase in R_a by 135% and an improvement in GRG of 0.29 can be seen.

Factor level	Initial process parameters	Optimal process parameters (grey-based Taguchi approach)			
	$PC_1 I_1 T_{on_1} S_1 DC_1$	$P C_2 I_3 T_{on_2} S_2 DC_3$			
MRR / mm ³ /minute	0.521	1.045			
R_a / $\mu { m m}$	2.21	5.21			
GR grade	0.33	0.62			
Improvement in grey relational grade is 0.29					

Table 13 Outcome of the tests conducted at optimal setting

9. Conclusions

The conclusions could be drawn from the experimental results, interpretation of Figs 6 to 13, the application of the multiple response optimization technique and the examination of the SEM images.

- 1. In comparison to silicon carbide powder mixed dielectric fluid, aluminium powder mixed dielectric fluid results in a higher MRR and superior surface finish.
- 2. The maximal MRR is accomplished when *I*, T_{on} and *S* are at the maximal level and *DC* is at the lowest level. Similar trends in the MRR with few parameters have been obtained by [20, 22].
- 3. The maximal R_a is accomplished when *I*, T_{on} and *DC* are at the maximal level and *S* is at the lowest level. Similar trends in R_a with few parameters have been obtained by [20, 22].
- 4. The craters on the machined surfaces are broader and larger with an increase in *I*, *T*_{on}, and *DC*. Similarly, the craters on machined surfaces are thinner and smaller with an increase in *S*.

- 5. In comparison to aluminium powder mixed dielectric fluid, silicon carbide powder mixed dielectric fluid results in higher WRLT.
- 6. An optimal set of various factors is $PC_2I_3.T_{on2}.S_2.DC_3$, computed by the grey-based Taguchi method. In the ANOVA investigation, it is seen that *I* and *CP* are important factors. The percentages of distinct factors in the grey-based Taguchi method are *CP* (15 %), T_{on} (5 %), S (1 %), DC (7 %), and I (61 %).
- 7. An increase in the MRR by 100 %, in R_a by 135 % and the overall improvement in GRG of 0.29 has been established.

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