# Preparation of γ-di-tertiary Glycols From the Esters of γ-keto Carboxylic Acids, II

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Some  $\gamma$ -di-tertiary glycols have been prepared by the Grignard's method from the Et esters of  $\gamma$ -keto carboxylic acids.

In a previous communication we have described the synthesis of some  $\gamma$ -ditertiary glycols from the esters of  $\gamma$ -ketocarboxylic acids, prepared by the method of Peters Jr. and R. Fischer III-a was prepared earlier by Blaise and the same way 1.1.4-triphenylheptane-1,4-diol [IV-b] and 5.8-dibutyldodecane-5.8-diol [IV-a]. The compound IV-a has been prepared earlier by Petrov and Sanin from ethyl succinate and n-butyl-magnesium bromide.

As starting material for the preparation of 5,8-dibutyldodecane-5.8-diol we used ethyl 3-ketoheptane-1-carboxylate [III-a], which was synthesized by the method of Chichibabin<sup>2</sup> from  $\alpha$ -furyl-propylcarbinol [II-a], prepared by method of Peters J. and R. Fischer<sup>4</sup>. III-a was prepared earlier by Blaise and Koehler<sup>5</sup> from carboetoxymethanecarboxyl chloride and n-amyl-zinc-chloride.

Intending to prepare some ditertiary glycols from the esters of 3-keto-dodecane-1-carboxylic acid [III-c], we synthesized for that purpose by the Grignard's method  $\alpha$ -furyloctylcarbinol [II-c]. The attempt to prepare III-c by clevage of II-c according to the mentioned method² proved to be till now unsuccessful.

The synthesized compounds IV-a and IV-b gave positive Denigès-reaction<sup>6</sup> on tertiary alcohols.

#### EXPERIMENTAL

## x-Furyl-propylcarbinol [II-a]

11.0 g. (0.45 g-atom) of magnesium, 51.0 g. (0.41 mole) of n-propyl bromide and 20.0 g. (0.21 mole) of furfural [I] in 250 ml. of dry ether yielded following a previosly described method 18.0 g. (62%) of  $\alpha$ -furyl-propylcarbinol (b. p. 90.5—01%)15 mm in an atmosphere of carbon dioxide). Peters and Fischer recorded 55% yield b. p. 91—92% 2 mm.

# Ethyl 3-ketoheptane-1-carboxylate [III-a]

To 16.0 g. (0.11 mole) of  $\alpha$ -furyl-propylcarbinol [II-a] in 53 ml. of abs. ethanol was added 1.7 ml. of a 26% ethanolic solution of hydrochloric acid. The reaction mixture was refluxed for 30 minutes, and the ethanol distilled off on a water bath in an atmosphere of carbon aloxide. The residue was cooled, added to a cool 1% solution of sodium hydroxide, and the mixture extracted with ether (3×50 ml.). The ethereal solution was dried over sodium sulfate. The solvent evaporated on a water bath in an atmosphere of carbon dioxide. After fractionation in vacuo in atmosphere of carbon dioxide, 8.5 g. (40%) of colourless liquid of an agreable »aromatic« odour was obtained, b. p. 122—124% mm. For the analysis a sample was redistilled and the fraction boiling at 115—116% mm was used. Blaise and Koehler5 recorded b. p. 125% mm.

 $d_4^{20} : 0.971_0 \quad n_D^{20} : 1.438_2; \text{ calc'd} : MR^7 \quad 50.2, \text{ found} : MR \quad 50.4.$ 

Anal. 29.530 mg. subst.: 69.820 mg. CO<sub>2</sub>, 25.230 mg.  $\rm H_2O$   $\rm C_{10}H_{18}O_3$  (186.24) calc'd: C 64.49; H 9.74 $^0$ / $^0$  found: C 64.39; H 9.54 $^0$ / $^0$ .

## 3-Ketoheptane-1-carboxylic acid

The ethyl ester II-a was saponified by refluxing it with  $5^{0/0}$  aqueous hydrochloric acid for two hours. Thin and colourless glistening plate-like crystals were obtained, m. p.  $53^{0/8}$ .

### 5.8-Dibutuldodecane-5.8-diol [IV-a]

To a solution of butyl-magnesium bromide prepared from 4.6 g. (0.19 g-atom) of magnesium and 24.0 g. (0.18 mole) of n-butyl bromide in 200 ml. of dry ether, 8.0 g. (0.04 mole) of ethyl 3-ketoheptane-1-carboxylate was added. Following a procedurel reported earlier an orange-red semicrystalline solid (15.0 g.) was obtained. Removal of the oily by-product by suction yielded crystalline IV-a (10.5 g., 80%). For the analysis the compound was washed with petroleum ether, and recrystallized from dilute acetic acid. The pure compound consisted of very thin, shiny, transparent leaflets, m. p. 103% (recorded m. p. 103%), which were soluble in acetone, ether, chloroform and benzene, slightly in petroleum ether, and sparingly soluble in dilute ethanol (1:1).

Anal. 29.980 mg. subst.: 83,645 mg. CO<sub>2</sub>, 35.975 mg. H<sub>2</sub>O  $C_{20}H_{42}O_2$  (314.54) calc'd: C 76.36; H 13.46 $^0$ /o found: C 76.14; H 13,42 $^0$ /o.

The oily by-product decomposed when distilled in vacuo.

# 1,1,4-Triphenylheptane-1,4-diol [IV-b]

To a solution of phenyl magnesium bromide prepared from 5.4 g. (0.22 g-atom) of magnesium and 32.0 g. (0.20 mole) of bromobenzene in 100 ml of dry ether, 5.0 g. (0.03 mole) of ethyl 3-ketohexane-1-carboxylate<sup>1</sup> in 50 ml. of dry ether was added. Following a procedure reported earlier<sup>1</sup>, a semicrystalline solid was obtained. Removal of the oily by-product by trituration with a cooled ether-acetone-petroleum ether-mixture (1:1:1) yielded crystalline IV-b (5,2 g., 50%), which after recristallization from petroleum ether gave small white silky needles, m. p. 113.5—114%.

The compound was easily soluble in acctone, ether, chloroform and benzene, less readily in petroleum ether and sparingly soluble in dilute ethanol (1:1).

Anal. 29.570 mg. subst.: 90.230 mg. CO<sub>2</sub>, 20.680 mg. H<sub>2</sub>O  $C_{25}H_{28}O_2$  (360.47) calc'd: C 83.29; H 7.83 $^0$ /o found: C 83.27; H 7.83 $^0$ /o.

## $\alpha$ -Furyl-octylcarbinol [II-c]

To a solution of octyl magnesium bromide prepared from 3.15 g. (0.13 g-atom) of magnesium and 24.0 g. (0.13 mole) of n-octyl bromide in 100 ml. of dry ether, a solution of 10.0 g. (0.10 mole) of freshly distilled furfural [I] in 25 ml. of dry ether was added. The reaction was accomplished in the same way as described previously¹. Tre crude  $\alpha$ -furyloctylcarbinol was distilled twice in vacuo. 9.8 g (41%) of the pure product was obtained, b. p. 137—138% mm. The compound was a clear yellowish liquid, with an aromatic, fruitlike odour. By cooling the liquid hardens at 2%, giving a white, waxy, amorphous mass which melts at 9%.

 $d_{\,\,4}^{\,20}$  : 0.9287  $\,\,n_{\rm D}^{\,20}$  : 1.4678; calc'd: MR 62.72, found: MR 62.9.

Anal. 30.525 mg, subst.; 82.892 mg, CO<sub>2</sub>, 28.044 mg, H<sub>2</sub>O  $C_{13}H_{22}O_2$  (210.31) calc'd; C 74.24; H 10.54% found; C 74.10; H 10.28%.

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#### IZVOD

#### Priprava y-ditercijarnih dialkohola iz estera y-ketokarbonskih kiselina. H

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Prema modificiranoj metodi¹ Čičibabina² pripravljen je iz  $\alpha$ -furil-propilkarbinola [II-a] etilni ester 3-ketoheptankarbonske kiseline-1 [III-a] ( $40^{0}/_{0}$ ) T. k.4115— $6^{0}$ . Iz [III-a] i butilmagnezijevog bromida dobiven je 5,8-dibutildodekandiol-5,8 [IV-a] ( $80^{0}/_{0}$  sirovog produkta). T. t.  $103^{0}$  (iz razr. octene kiseline). Etilni ester 3-ketoheksankarbonske kiseline-1¹ i fenilmagnezijev bromid daje 1,1,4-trifenilheptandiol-1,4 ( $50^{0}/_{0}$  sirovog produkta). T. t. 113,5— $114^{0}$  (iz petroletera).

Iz furfurala [I] i oktilmagnezijevog bromida sintetiziran je  $\alpha$ -furil-oktilkarbinol [II-c] (41%) T. k.3137—8%; T. t. 9%, koji za sada nije uspjelo metodom Čičibabina² prevesti u ester odnosne  $\gamma$ -ketokarbonske kiseline [III-c].

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