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MEASUREMENT OF DISSOCIATION IN A HYDROGEN GLOW DISCHARGE*

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Abstract: Refractivity index measurements of the gas combined with manometric data and information on the energy dissipation in the positive column are utilized for the determination of the degree of dissociation in hydrog *discharge. The abscnce of pcrccptiblc distortion of intcrfcrcnce frigges confirmes the basic assumption of the proposed mcthod, that the radial density gradient is small and consequently the transfer of heat to the walls rathcr fast. Thc accuracy of the method is estimated to be better than ± 10%.*

1. lntroduction

Knowledgc of thc conccntration of hydrogen atoms appeared indispcnsable in our investigation of the stationary striated positive column in hydrogen *discharges. Thus far several mcthods for determining thc degree of dissociation of molecular gases have been proposed'·* 2, 3), *but only the Wrede's diffusion gauge***⁴** *> and thc manometric method***⁵** *> havc found more application. Both methods, however, require assumptions concerning the temperature of the molecular gas.*

Here we describc a mcthod for determining the degree of dissociation of hydrogen based on intcrfcrometric and manometnc measurements combined with data on the powcr input in the positivc column.

2. Theory

The gas under consideration, at an initial pressure p_1 and ambient temperature T_1 occupies the discharge tube of volume V_1 and an auxiliary recipient

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of volume V² , communicating with the discharge tube. When the discharge is switched on, a fraction x of molecules is dissociatcd in the discharge and the gas temperature raised to T_2 . Owing to both effects, a fraction f of mo*leculcs is pressed from* V_1 *into the side enclosure and the pressure in the* system increases to p_2 . If $V_2 > V_1$ the gas temperature in V_2 will little change; *if, furthermore, the gas displaced from* V_1 *into* V_2 *consists only of molecules, or what is equivalent, if the atoms in* V_2 *quickly recombine one can write*

$$
(N_2 V_2 + f N_1 V_1) / V_2 = \frac{p_2}{k T_1},
$$
 (1)

where f is related to the interferometric measurements of density in the discharge tube by

$$
f = K \cdot \Delta s. \tag{2}
$$

*A s denotes the number of displaced interfercncc fringes, suitably e:xpressed by the anglc of rotation of the Jamin compensation plate; K is a constant of proportionality, depending on the length of thc gas cdls and the refractivity of the gas at a given wavelength; N***¹ ,** *N2 and k havc thcir usual mcaning.*

Because of disturbing thcrmal cffccts, thc measurment of f can bc pcrformed only shortly after switching on thc dischargc and thc establishment of a stationary value of x. It is thus necessary to determine the value of T *² attained at an instant t after the onsct of thc discharge.*

The material balance equation for thc systcm is given by

$$
N_1 V_1 + N_1 V_2 = N_1 V_1 (1 - f) (1 - x) + 2 x N_1 V_1 (1 - f) + N_2 V_2,
$$
 (3)

where

$$
N_1(1-f)(1+x) = \frac{p_2}{k T_2}.
$$

Taking into account that practically ali cnergy dissipatcd in thc positivc column of a glow discharge is transferred to thc walls and that there is no perceptiblc radial temperature gradicnt in thc dischargc tube (sce bclow), one obtains T_2 by calculating the temperature of the inner surface of the *discharge tube, T,.*

According to the well-known relation for transicnt heat flow through a semi-infinite medium, thc encrgy W transfcrred to the wall is givcn by

$$
W = \frac{\chi A \cdot (T_s - T_1)}{\sqrt{\pi \alpha t}}, \tag{5}
$$

where x is the coefficient of heat conductivity, A is the inner surface area, α is given by $\chi/(c_p \rho)$ with c_p the specific heat and ρ the density of the wall *material.*

Since $T_2 \sim T$, and $W = i \cdot X$, where X is the field and *i* the strength of the *current flowing through the discharge tube, one obtains from Equs. (4) and (5) the degree of dissociation in the form*

$$
x = \frac{A \cdot p_2 \cdot V \cdot \rho c_p}{k \cdot N_1 (1 - f) \left[A T_1 V \cdot \rho c_p + i X (\pi t)^{N_2} \right]} - 1.
$$
 (6)

This equation will be valid for other molecular gascs too, providcd thc rate of heat transfer to the walls is sufficiently high.

3. Apparatus

A discharge and a reference tube, both of 100 cm length and $\varnothing_{\text{in}} = 9.5$ mm, *were placed in a Rayleigh-Jeans interferometer and connected with a suitable vacuum and gas filling system (Fig. 1).*

Fig. 1. Experimental arrangement: $s^{i\alpha}$ — the pointolit source, $s^{i\alpha}$ — condensing *lens, »p« - inlet slit of the interferometer, »j« - Jamin compcusating plate, l>O« - auxiliary eyepiece, »t,« - discharge tube, »t2« - reference tube, »r« - side enclosure, »m«* — differential manometer, *»d₁«* and *»d₂«* — glass valves allowing the isolation of *»r*« and *»t₁«* respectively *»t₁«*.

The discharge tube is fcd from a 20 kV rectificr through a set of rcsistanccs. The reference tube is filled to the initial pressure $\mathbf{v}p_i \cdot \mathbf{v}$ and afterwards sepa*rated from the discharge tube by means of the valve »d2«. The pressurc in the discharge tube and reference tube is measured by a calibrated LKB-gaugc and the pressure increase in »11« and »r«, after separation from »/***²** *«, by the differential manometer m.*

The entrance slit of the interferometer was illuminated by a 100 W *pointolit lamp, and in that way the brightness of fringes largely exceeded the luminosity of the discharge. The shift of the interference fringes was measured with respect to the cross hair of an auxiliary eyepiece »o«, as reference. In order to exclude any influence of the dilatation of the discharge tube upon the fringe shift, individual measurements were accomplished within time intervals* ≤ 1 *s* after the onset of the discharge.

4. Results and discussion

Taking into account the limit of accuracy of the Zeiss interferometer and the absence of any perceptible distortion of interference fringes in the observed range of pressure and current strength, it can be concluded that radial

Fig. 2. A plot of degree of dissociation <i>x vs reduced field strength *X/p*, with current strength as parameter. The dashed line denotes the limit above which *standing striations are present in the positive column.*

temperature gradient in the discharge tube was cerlainly smaller than 4 °K/cm at 4.5 torr or 12 °K/cm at 1.5 torr. Furthermore, the thermal boundary layer at the walls was found to be thinner than 0.33 cm, i.e., of the order of the mean free paths. Thus thc maximum error introduced by the assumption $T_2 = T_s$ was $\leq 50/6$

Fig. 2 shows thc degree of dissociation as a function of the reduced ficld strength X/p. Each point rcpresents an averagc of several measurcments, and the accuracy is estimated to be bctter than 100/o, including the crror which might originate from the departure of $T₂$, from $T₃$. As expected, the *plot x versus X/p is similar to the behaviour of the first Townscnd cocfficient, at first increasing almost lincarly with the reduced field strength.*

In the preceding consideration it was implicitly assumed that a Maxwellian energy distribution of atoms and molecules was established in the discharge. However, this does not strictly apply in the prcscnt case. Spectroscopic mcasurements of the Balmer line profiles showed that a small fraction of *H-atoms was out of thermal equilibrium. This effcct is most probably due to the formation of the anlibound l* **³***l:***u** *H***2** *molecules which dissociate into fast H-atoms of 2.2 cV kinetic energy. The contribution of the recently discussed mechanism***⁶** *> of acceleration ol' protons in the space charge zone, followed by neutralization at thc surface, is of minor importance because of low ion densities and current strength in thc positive column.*

*Although the interaction of atomic hydrogcn with glass is rather complex and not yet sufficiently understood***⁷** *>, thc only slight departure from thermal equilibrium may be considercd as a proof that the energy transfer of H-atoms to the walls procccds at a rather high rate.*

Furthermore, H-atoms w1th energies cxceeding 0.3 eV enter with largc probabilities in the metathetic reaction $H + H_2 \rightarrow H_2 + H_3$, thus transferring *efficiently momentum to the H***2** *molecules. Afterwards hydrogen molecules lose quickly their excess energy in collisions with the glass surface, for the accommodation coefficient of H***2** *on glass is about 0.3***⁸** *1. In the present geometry the time interval between two consecutive collisions with the walls,* $\tau \approx 4 r^2/D$, is estimated to be about $3 \cdot 10^{-4}$ s, and therefore the relaxation *time is about 10-2 s.*

lt follows from the above discussion that the error arising from thc basic assumptions used is small, and one may thus obtain reliable information on the degree of dissociation of hydrogcn by Equ. (6) and the simple manometric method.

A fuller interpretation of the results shown in the Fig. 2 is not at prescnl possible, sincc it requires data on the population of excited species and on the electron distribution function. Further work on the striated column is in progress.

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MJERENJE STUPNJA DISOCJJACIJE U TINJAVOM IZBIJANJU U VODIKU

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S a d r žaj

Stupanj disocijacije vodika u tinjavom izbijanju je određen na osnovi mjerenja indeksa loma plina, tlaka i disipacije energije u pozitivnom stupu. Odsustvo zamjetljive distorzije interferentnih pruga potvrdilo je osnovnu pretpostavku metode da je radijalni gradijent gustoće slobodnih i vezanih **H-atoma u izboju malen i da je prema tome prijenos topline na stjenke brz proces. Procjene pokazuju da je točnost metode bolja od** ± **100/o.**