

MICROSCOPIC STUDY OF THE MOLECULAR-DIPOLE DEGREE
OF FREEDOM IN NUCLEI

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Recent theoretical^{1,2} and experimental³ studies give strong evidence for molecular-dipole modes to be present in nuclei; this is indicated by the existence of rotational bands of mixed parity with selective intraband E1-transitions, which are enhanced on a molecular scale^{2,3}. In this contribution the idea of molecular-dipole modes is tested on a microscopic level for α -cluster configurations in ^{18}O and $^{46,52}\text{Ti}$.

The many-body wave function of the two-cluster system is taken as the antisymmetrized product of the internal wave functions of the clusters and their relative motion wave function,

$$\Psi_{\ell} = \mathcal{A} \{ \phi_{\alpha} \phi_A \sum_n \frac{1}{\sqrt{\mu_{n\ell}}} \langle u_{n\ell} | g_{\ell} \rangle u_{n\ell}(\vec{r}) \} \quad (1)$$

where $\phi_A = \phi_{1^4\text{C}}, \phi_{4^2\text{Ca}}, \phi_{4^4\text{Ca}}$ resp.

The internal wave functions describe the ground states of the nuclei in the harmonic oscillator shell model with the same b -value for both clusters. The relative motion of the two clusters can be determined from a Schrödinger-like equation⁴

$$\Lambda_{\ell} \left\{ -\frac{\hbar^2}{2\mu} \Delta_{\vec{r}} + V_{\ell}(r) - E \right\} g_{\ell}(\vec{r}) = 0 \quad (2)$$

$u_{n\ell}(\vec{r})$ are the spherical harmonic oscillator wave functions and the normalization constants are

$$\mu_{n\ell} = \langle \phi_{\alpha} \phi_A u_{n\ell} | \mathcal{A} | \phi_{\alpha} \phi_A u_{n\ell} \rangle \quad (3)$$

In eq. (2) the projector Λ_{ℓ} eliminates redundant states of relative motion. The ℓ -dependent potentials $V_{\ell}(r)$ are local approximations to the true non-local potential of the orthogonalized resonating group method⁴. The E1-transitions between molecular-dipole states are governed by matrix elements of the electric dipole operator between states of the form (1). This many-body matrix element reduces to simple matrix elements of the relative motion dipole operator, if conservation of angular momentum and energy is taken into account.

For ^{18}O eq. (2) gives rise to a rotational $0^+, 1^-, 2^+, 3^-, 4^+$ band of molecular dipole type in addition to the $0^+, 2^+, 4^+$ ground state band. Changing the strength parameters of $V_{\ell}(r)$ of ref. 4 by some percent it is possible to obtain a perfect fit to the experimental energies³ of the $0^+, 1^-, 2^+, 3^-$ and 4^+ -states of ^{18}O . The results for the B(E1)-values are listed in table I together with the experimental data³ and the B(E1)-values obtained from

the simple rotator model, normalized to the $1 \rightarrow 0$ transition. The calculated $B(E1)$ -values are enhanced on a molecular scale³ and allow for an interpretation of the respective states in terms of a molecular-dipole rotational band. The deviations from the ideal rotator rule can be understood by examining the mean cluster distance d , which grows considerably with l . Compared to the experimental data³ the calculated $B(E1)$ -values are too large, due to the fact that the cluster wave functions of type (1) turn out more collective than the experimental ones.

Table I $B(E1, l \rightarrow l')$ -values (in standard Weisskopf units) for ^{18}O

$l \rightarrow l'$	experimental	present calc.	ideal rotator
$1 \rightarrow 0$	0.028	0.084	0.084
$2 \rightarrow 0$	0.009	0.107	0.101
$3 \rightarrow 2$	0.022	0.140	0.108
$4 \rightarrow 3$		0.218	0.112

Solving eq.(2) for ^{46}Ti (^{52}Ti) with the potentials $V_l(r)$ of ref.4 we obtain rotational bands starting at $E^* = 11.4$ MeV (8.3 MeV) with a rotational constant of 98 keV (94 keV) and with a parity splitting of ≈ 0.8 MeV. The calculated $B(E1)$ -values (table II) are of the same order of magnitude for ^{52}Ti as for ^{18}O , whereas for ^{46}Ti they are suppressed because the difference of the ratios of charge and mass of the clusters is smaller.

Table II $B(E1, l \rightarrow l')$ -values (in standard Weisskopf units) for $^{46}, ^{52}\text{Ti}$.

$l \rightarrow l'$	^{46}Ti	$l \rightarrow l'$	^{52}Ti
$1 \rightarrow 0$	0.022	$1 \rightarrow 0$	0.19
$1 \rightarrow 2$	0.043	$1 \rightarrow 2$	0.38
$3 \rightarrow 2$	0.028	$3 \rightarrow 2$	0.24
$4 \rightarrow 3$	0.028	$3 \rightarrow 4$	0.30
$5 \rightarrow 4$	0.029	$5 \rightarrow 4$	0.24

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4. D. Wintgen, H. Friedrich, K. Langanke, Nucl.Phys. A408 (1983) 239.