

## Supporting Information

### Transition Metal-Mediated Hydrolysis of the Imine Bond in 2-Azomethine Benzothiazoles

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## 1. X-ray crystal structure determination

Table S1. Experimental data for the X-ray diffraction studies.

Compound	<b>3</b>	<b>1a</b>	<b>2a</b>	<b>2b</b>	<b>2c</b>
Formula	C <sub>14</sub> H <sub>9</sub> BrN <sub>2</sub> OS	C <sub>22</sub> H <sub>17</sub> N <sub>5</sub> O <sub>5</sub> S <sub>2</sub> Zn	C <sub>21</sub> H <sub>14</sub> CuFN <sub>5</sub> O <sub>4</sub> S <sub>2</sub>	C <sub>14</sub> H <sub>8</sub> CuF <sub>2</sub> O <sub>4</sub>	C <sub>14</sub> H <sub>10</sub> CuF <sub>2</sub> O <sub>5</sub>
F <sub>w</sub> (g mol <sup>-1</sup> )	333.20	560.89	547.03	341.74	359.76
Crystal system	Monoclinic	Triclinic	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P</i> 2 <sub>1</sub> / <i>c</i>	<i>P</i> $\bar{1}$	<i>P</i> 2 <sub>1</sub> / <i>c</i>	<i>P</i> 2 <sub>1</sub> / <i>c</i>	<i>P</i> 2 <sub>1</sub> / <i>c</i>
<i>a</i> (Å)	17.9824(3)	8.0897(7)	7.9378(2)	12.5165(5)	12.9621(1)
<i>b</i> (Å)	6.3691(1)	11.7385(10)	13.4701(3)	3.8555(2)	6.9497(1)
<i>c</i> (Å)	11.8792(2)	12.9843(10)	20.3168(6)	12.6338(5)	14.6724(1)
$\alpha$ (°)	90	98.819(7)	90	90	90
$\beta$ (°)	105.743(2)	101.722(7)	99.617(3)	94.421(4)	92.061(1)
$\gamma$ (°)	90	100.623(7)	90	90	90
<i>V</i> (Å <sup>3</sup> )	1309.51(4)	1163.05(17)	2141.80(10)	607.86(5)	1320.87(2)
<i>Z</i>	4	2	4	2	4
<i>Z'</i>	1	1	1	0.5	1
<i>D</i> <sub>calc</sub> (g cm <sup>-3</sup> )	1.690	1.602	1.696	1.867	1.809
<i>F</i> (000)	664	572	1108	342	724
Reflections collected	5768	12137	7761	4215	6296
Independent reflections	2681	4783	4127	1252	2731
<i>R</i> <sub>init</sub>	0.0236	0.0637	0.0390	0.0338	0.0192
Reflections observed	2415	3395	3448	1194	2498
Parameters	176	325	313	97	207
<i>R</i> <sub>1</sub> [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )] <sup>[a]</sup>	0.0542	0.0619	0.0589	0.0743	0.0348
<i>wR</i> <sub>2</sub> (all data) <sup>[b]</sup>	0.1543	0.1843	0.1661	0.2288	0.1152
Goof, <i>S</i> <sup>[c]</sup>	1.068	1.056	1.041	1.147	1.096
Maximum/minimum electron density (e Å <sup>-3</sup> )	1.516/-0.614	0.542/-0.939	0.544/-0.715	1.265/-0.612	0.476/-0.280

<sup>[a]</sup>  $R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ . <sup>[b]</sup>  $wR_2 = \{\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2]\}^{1/2}$ . <sup>[c]</sup>  $S = \{\sum [w(F_o^2 - F_c^2)^2] / (n - p)\}^{1/2}$  where *n* is number of reflections and *p* is the total number of parameters refined.

## 2. Crystal packing

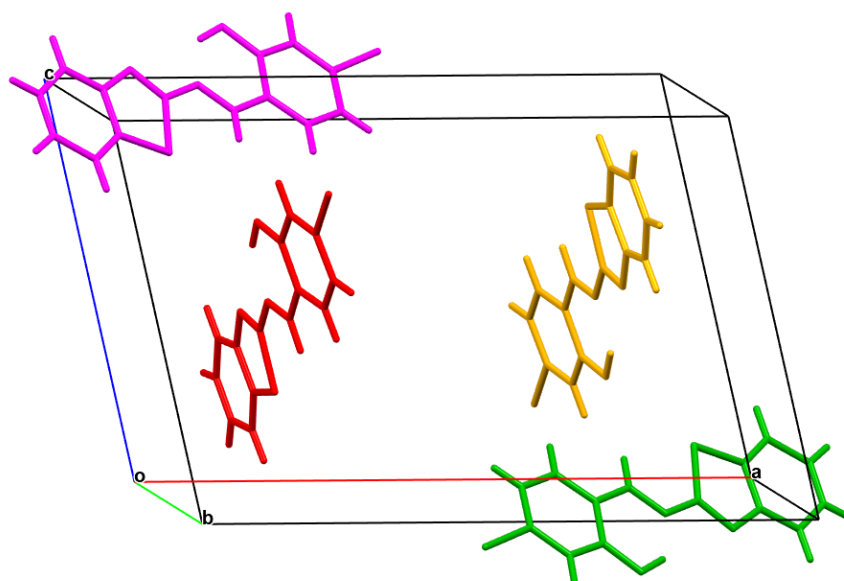


Figure S1. Crystal packing of molecules in **3**. One molecule in the asymmetric unit is given in red color.

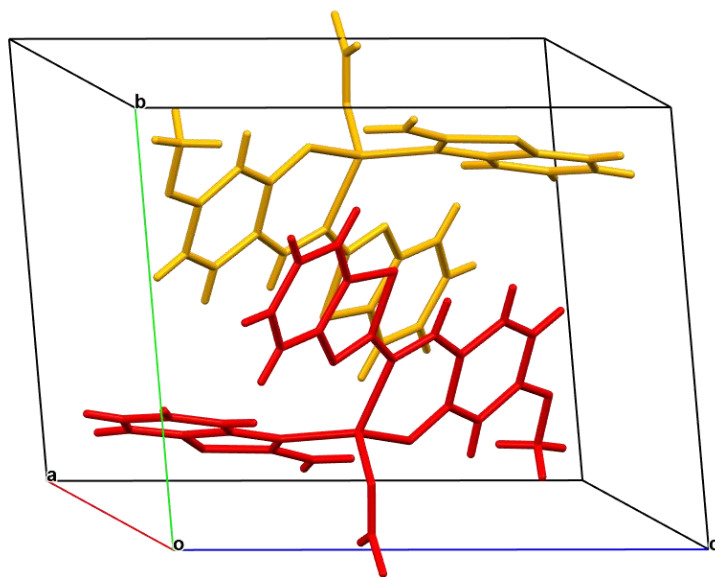


Figure S2. Crystal packing of complexes in **1a**. One complex in the asymmetric unit is given in red color.

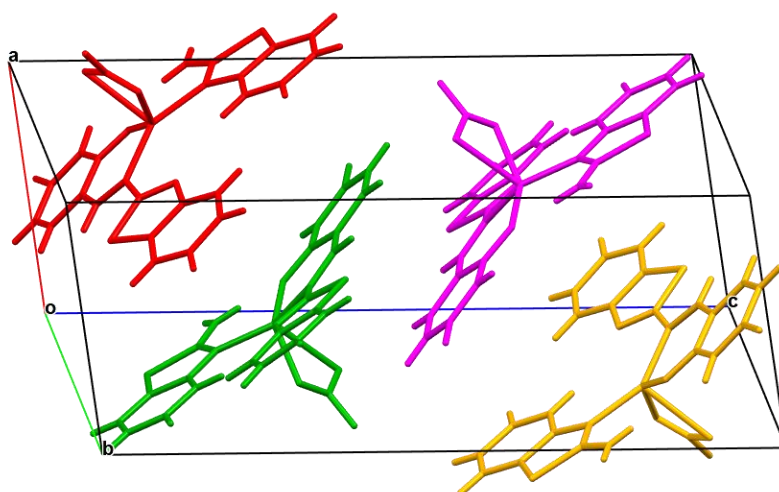


Figure S3. Crystal packing of complexes in **2a**. One complex in the asymmetric unit is given in red color.

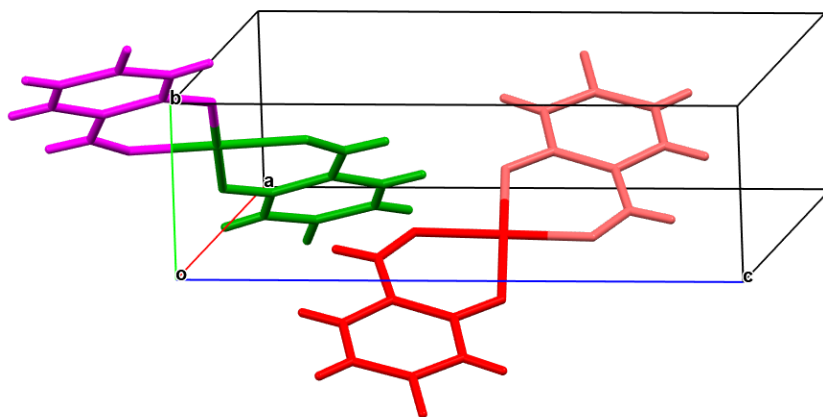


Figure S4. Crystal packing of complexes in **2b**. One half of complex in the asymmetric unit is given in red color.

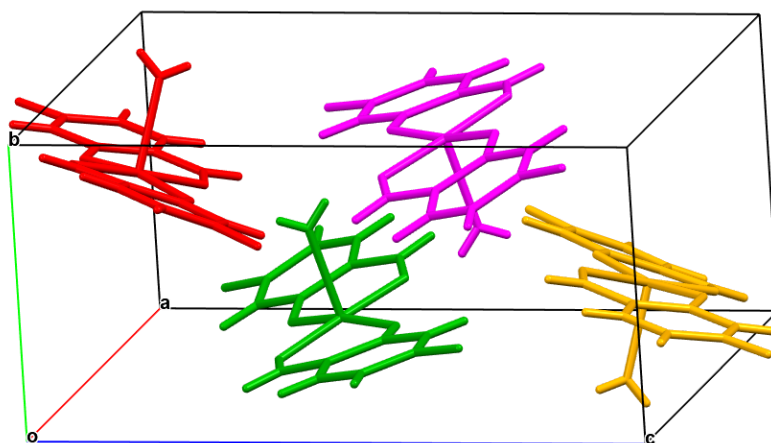
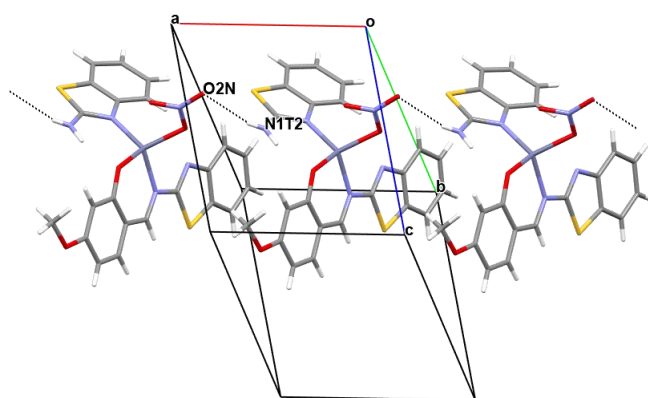
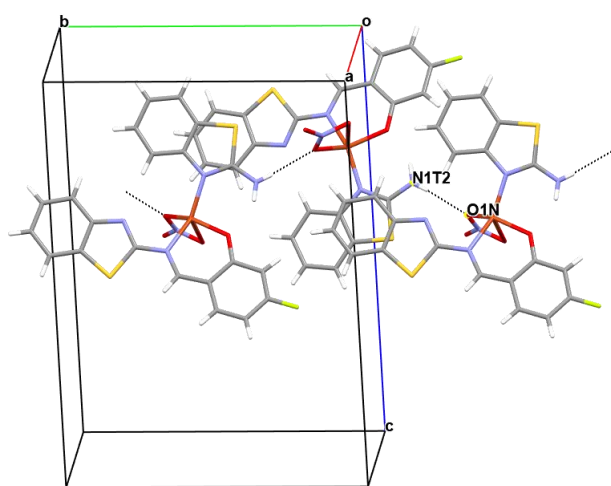


Figure S5. Crystal packing of molecules in **2c**. One complex in the asymmetric unit is given in red color.



a)



b)

Figure S6. One-dimensional hydrogen bonding motifs in **1a** (a) and **2a** (b), formed by intermolecular hydrogen bonds N1T2-HBT2...O2N<sup>i</sup> for **1a** [symmetry code for **1a**: (i)  $x+1, y, z$ ] and N1T2-HBT2...O1N<sup>i</sup> for **2a** [symmetry code for **2a**: (i)  $-x+2, y-1/2, -z+1/2$ ].

### 3. $^1\text{H}$ and $^{13}\text{C}$ NMR spectra of ligands 1–3

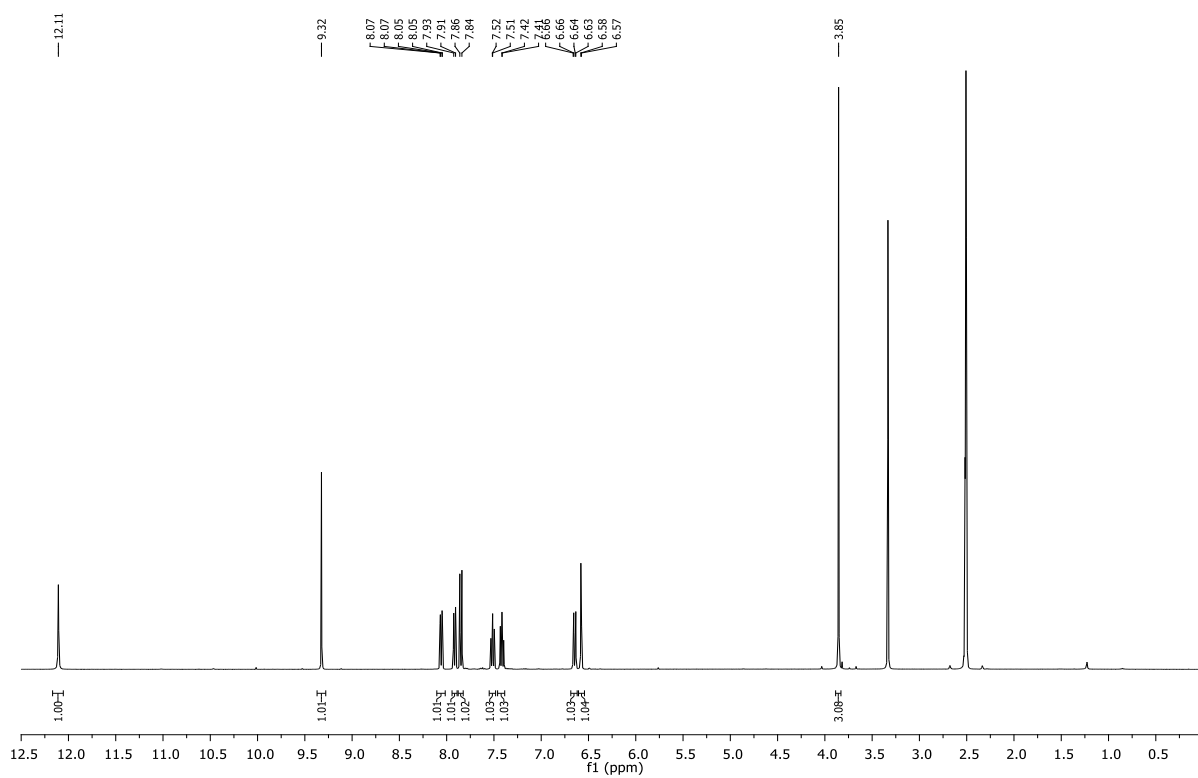


Figure S7.  $^1\text{H}$  NMR spectrum of ligand 1

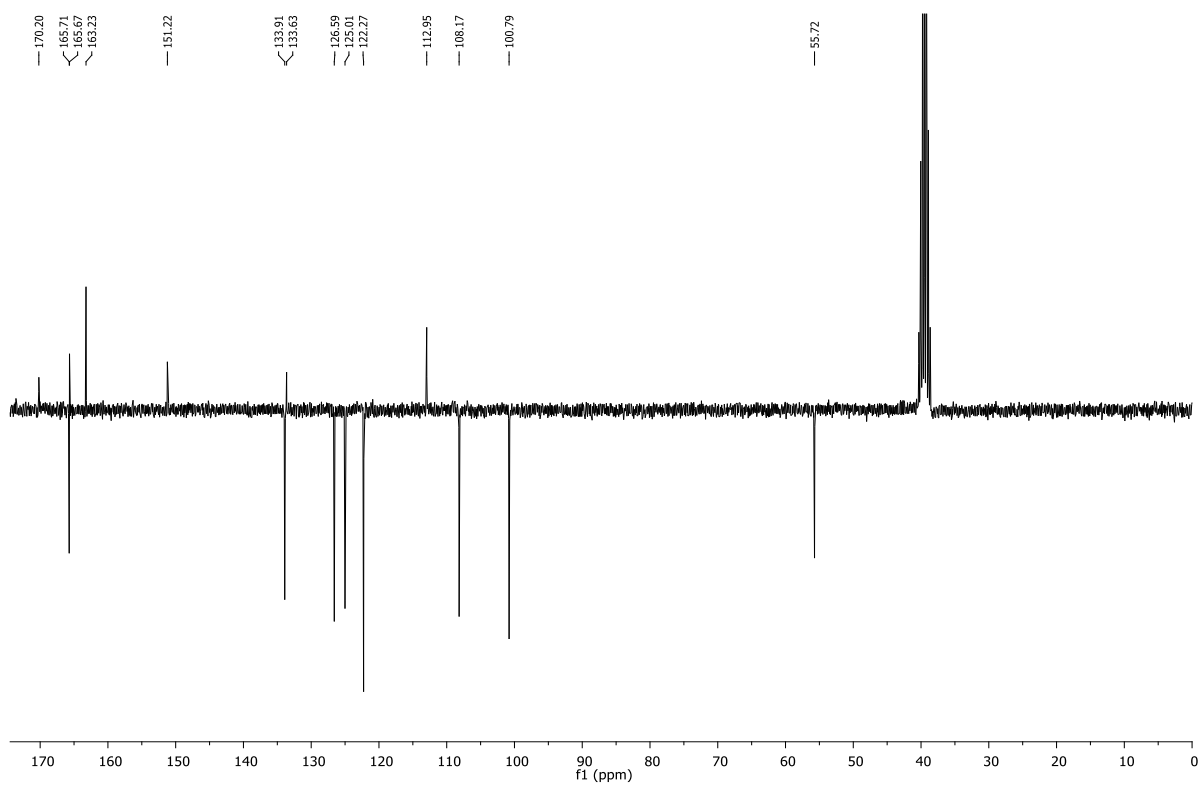


Figure S8.  $^{13}\text{C}$  NMR spectrum of ligand 1

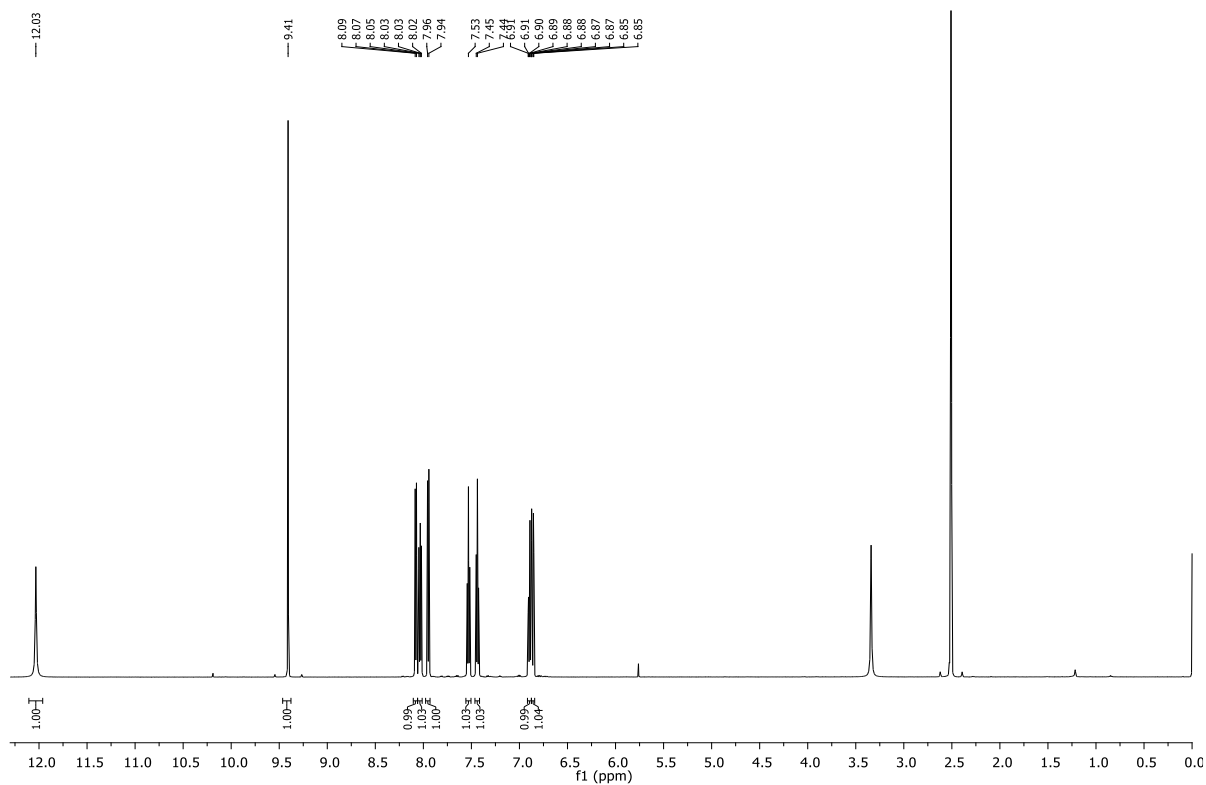


Figure S9.  $^1\text{H}$  NMR spectrum of ligand **2**

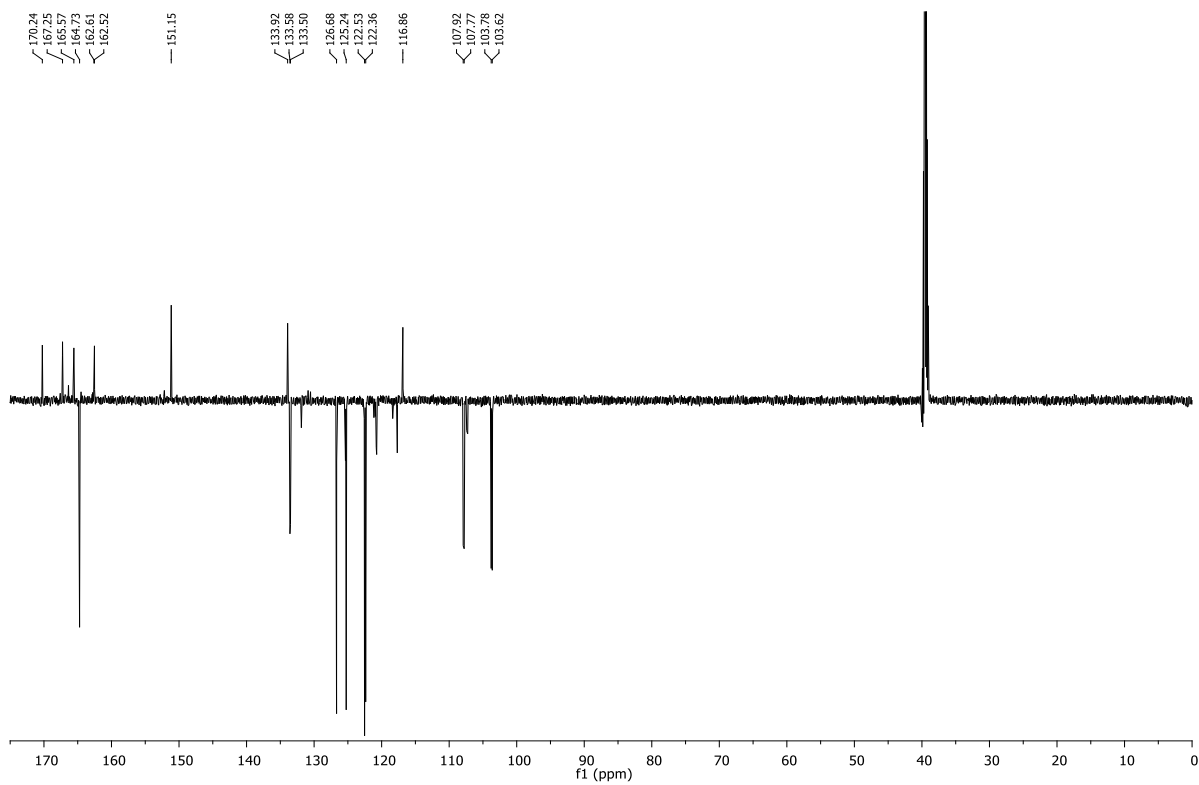


Figure S10.  $^{13}\text{C}$  NMR spectrum of ligand **2**

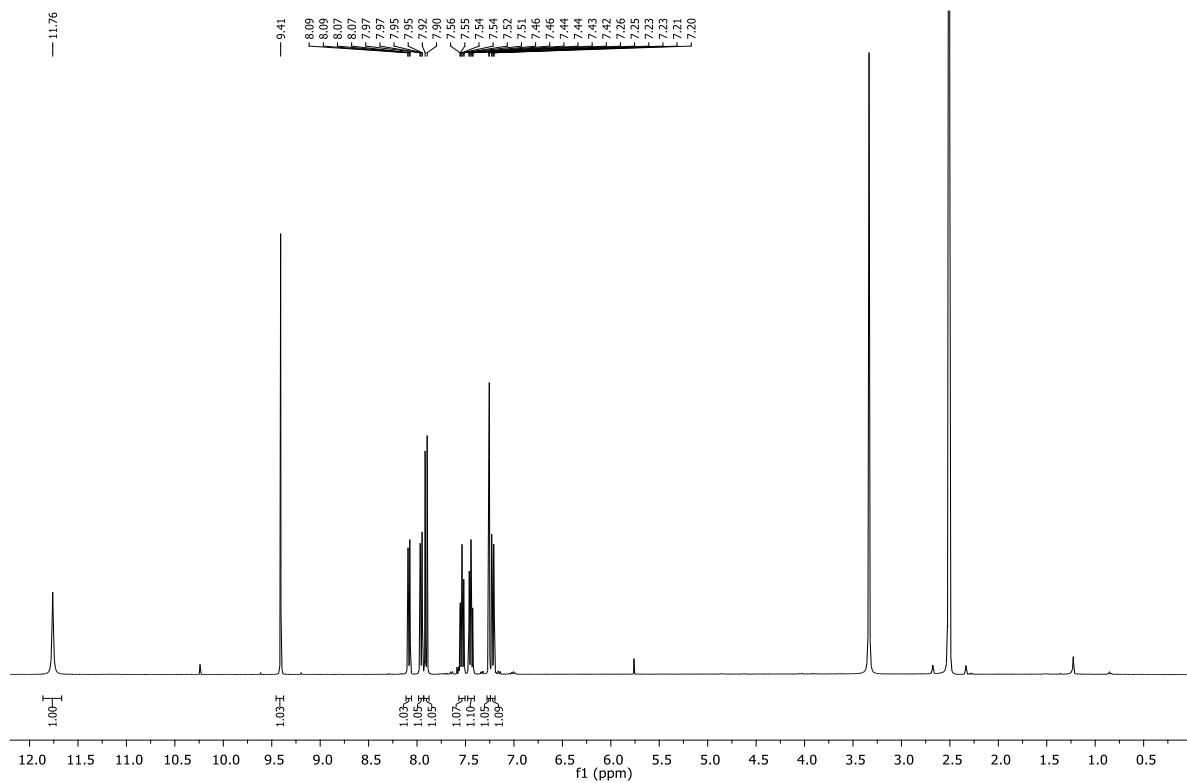


Figure S11.  $^1\text{H}$  NMR spectrum of ligand **3**

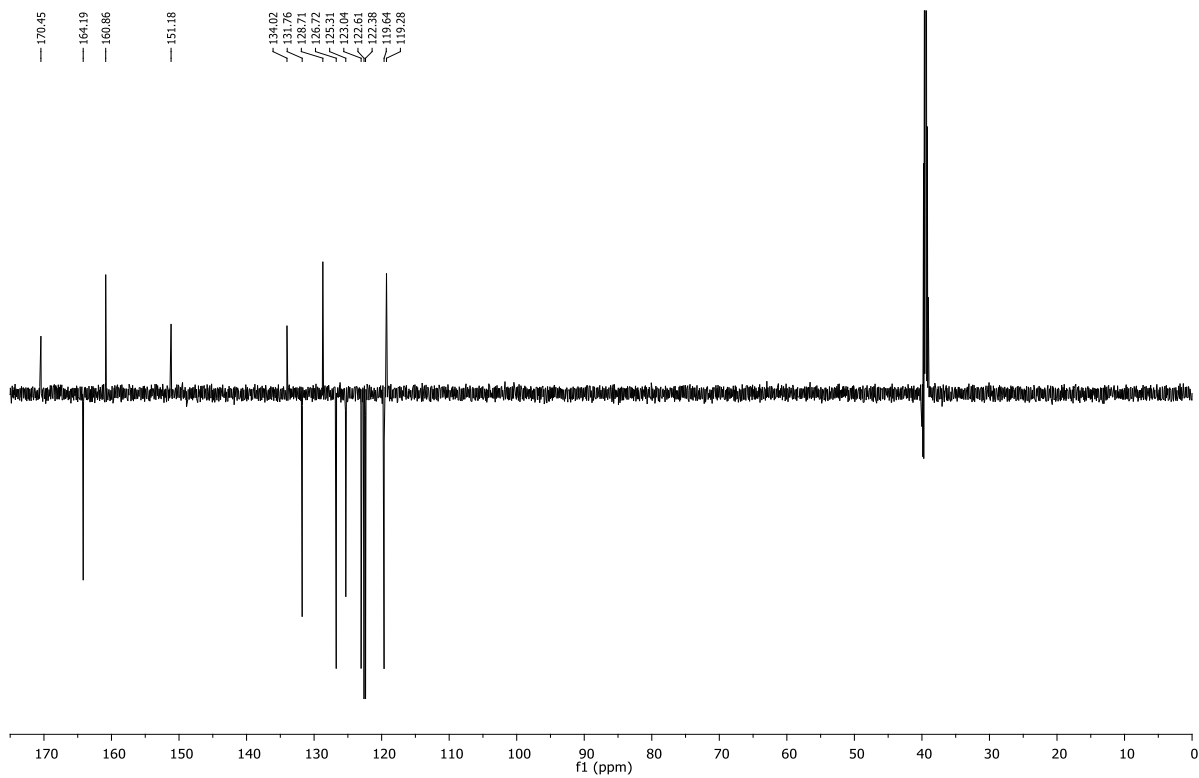


Figure S12.  $^{13}\text{C}$  NMR spectrum of ligand **3**

#### 4. IR(ATR) spectra of ligands 1–3

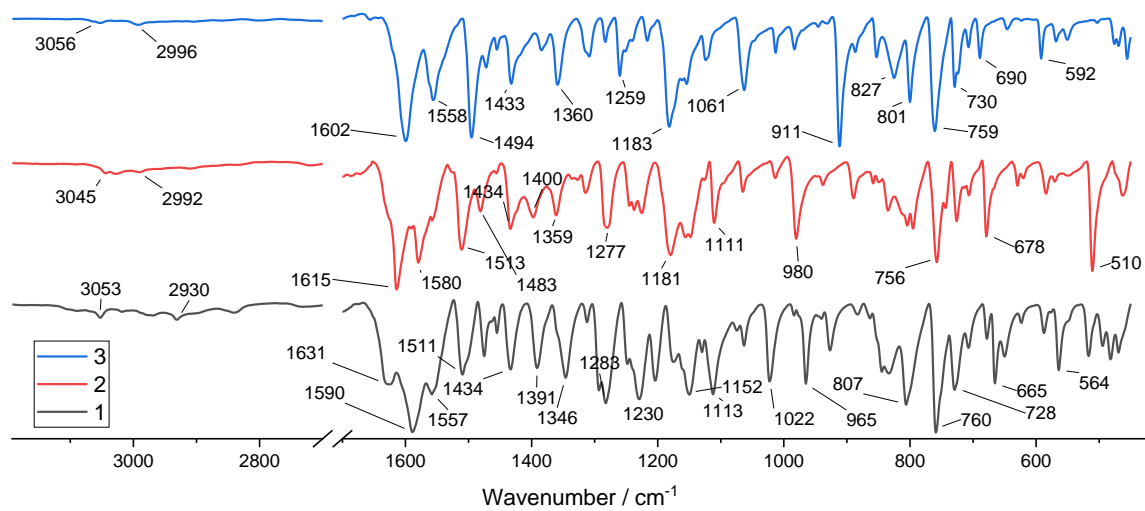


Figure S13. IR(ATR) spectra of ligands 1–3