

## Electronic Supplementary Information (ESI)

Ternary coordination compounds of Cu(II) with glycine and 2,2'-bipyridine:  
Synthesis, structural characterization, magnetic and biological properties

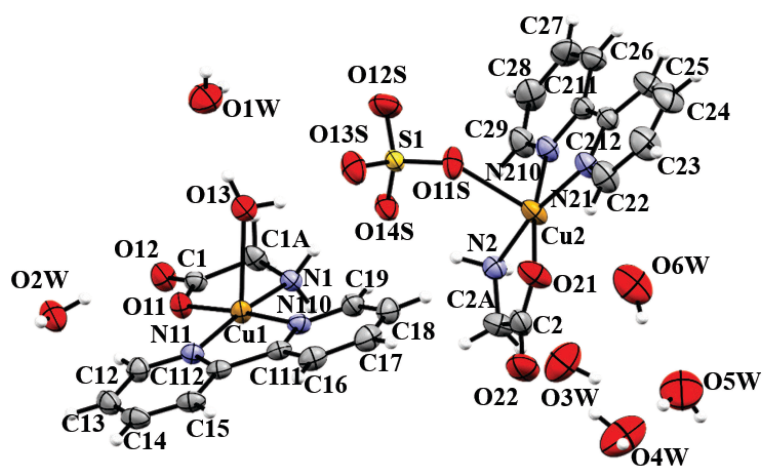
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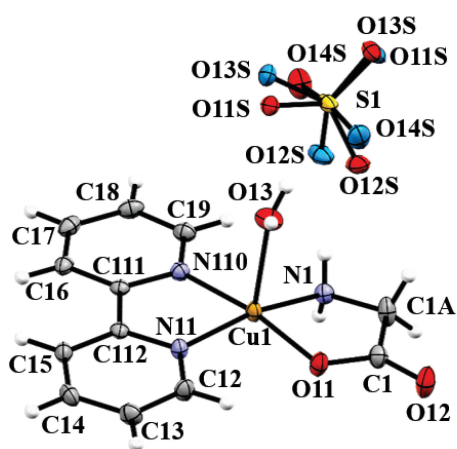
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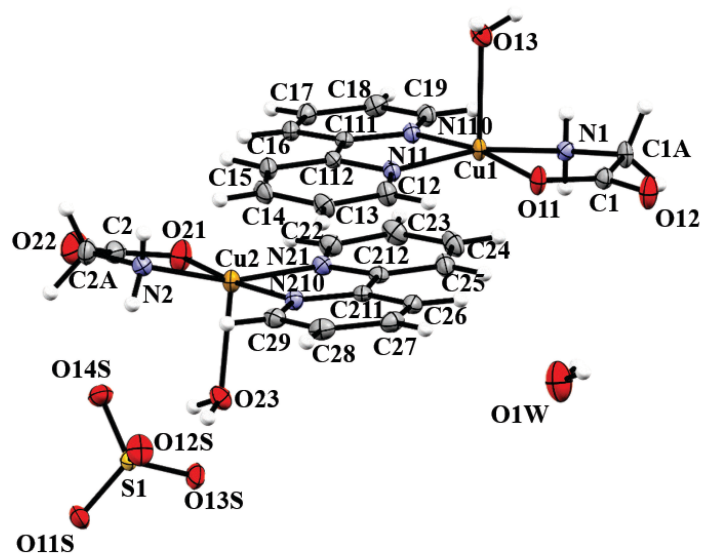
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**1a·6H<sub>2</sub>O**



**1b**



**1b·H<sub>2</sub>O**

Figure S1. Asymmetric unit of **1a·6H<sub>2</sub>O**, **1b** and **1b·H<sub>2</sub>O**. Displacement ellipsoids of non-hydrogen atoms are drawn at the 50% probability level. Oxygen atoms of one disordered part of sulfate ions are shown in blue colour.

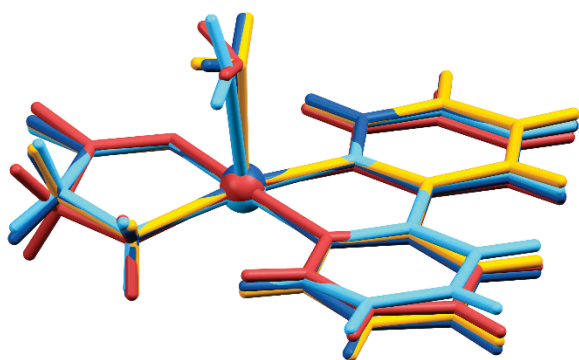


Figure S2. Overlapped complex cations of **1a**·6H<sub>2</sub>O (light blue), **1b** (dark blue) and **1b**·H<sub>2</sub>O (yellow and red).

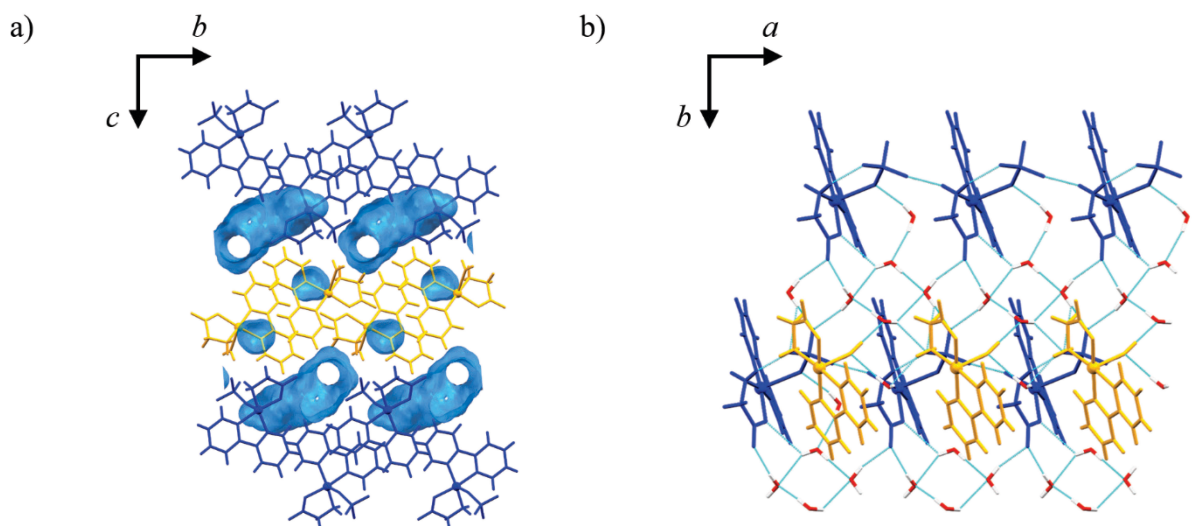


Figure S3. Packing of complex anions (blue) and cations (yellow) in **1a**·6H<sub>2</sub>O a) in *b*-*c* plane and b) *a*-*b* plane. Blue surfaces in a) show contact surface around water molecules. Hydrogen bonds in b) are shown as cyan lines.

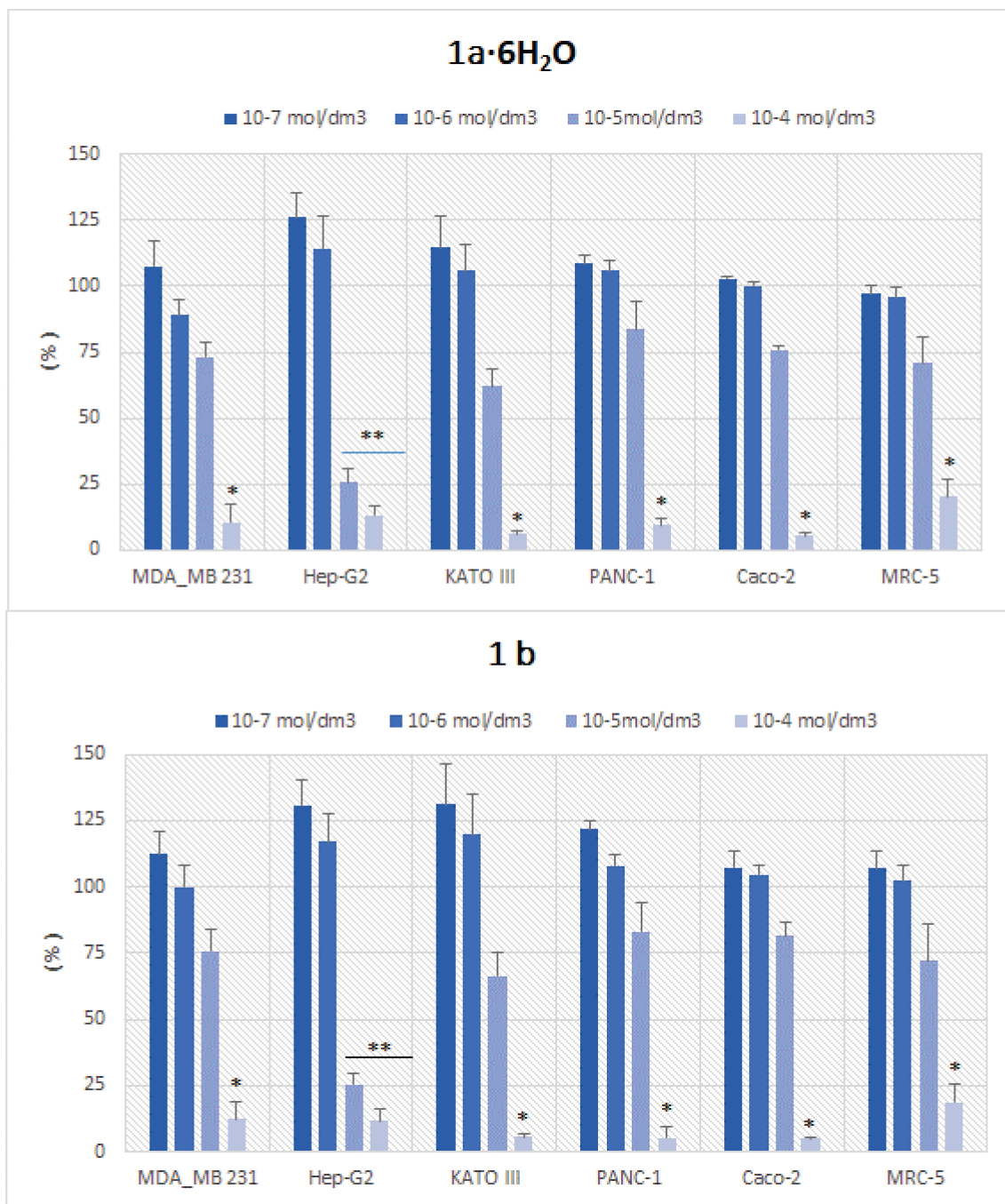


Figure S4. Antiproliferative effect of compounds **1a** and **1b**. Data are presented as mean (SD) of cell viability (%) obtained from three independent experiments done in triplicate. (\*) statistically significant ( $P < 0.05$ ) in respect to control group determined by Mann-Whitney U – test.



**Table S1.** Crystallographic data for coordination compounds **1a·6H<sub>2</sub>O**, **1b** and **1b·H<sub>2</sub>O**

Compound	<b>1a·6H<sub>2</sub>O</b>	<b>1b</b>	<b>1b·H<sub>2</sub>O</b>
Formula	C <sub>24</sub> H <sub>38</sub> Cu <sub>2</sub> N <sub>6</sub> O <sub>15</sub> S	C <sub>24</sub> H <sub>28</sub> Cu <sub>2</sub> N <sub>6</sub> O <sub>10</sub> S	C <sub>24</sub> H <sub>30</sub> Cu <sub>2</sub> N <sub>6</sub> O <sub>11</sub> S
Formula weight	809.74	719.66	737.68
Space group	<i>P</i> $\bar{1}$	<i>C</i> 2/ <i>c</i>	<i>P</i> 2 <sub>1</sub> / <i>c</i>
<i>a</i> /Å	7.3674(3)	21.6670(12)	20.1814(9)
<i>b</i> /Å	10.6440(5)	6.7922(3)	6.7748(3)
<i>c</i> /Å	21.6080(8)	21.0213(11)	20.9225(8)
$\alpha$ /°	85.638(3)	90	90
$\beta$ /°	87.793(3)	121.389(7)	106.977(4)
$\gamma$ /°	71.320(4)	90	90
<i>V</i> /Å <sup>3</sup>	1600.38(12)	2640.9(3)	2736.0(2)
<i>D</i> <sub>calc</sub> /g cm <sup>-3</sup>	1.680	1.810	1.791
$\mu$ /mm <sup>-1</sup>	1.474	1.761	1.705
<i>F</i> (000)	836	1472	1512
$\theta$ range/°	4.3–30.0	4.2–27.0	4.2–27.0
<i>T</i> /K	293	150	150
Radiation wavelength	0.71073	0.71073	0.71073
Range of <i>h, k, l</i>	-9–9, -13–14, -28–28	-27–27, -8–8, -26–26	-25–23, -8–8, -26–26
Reflections collected	30616	21309	31716
Independent reflections	7685	2875	5961
Observed reflections ( <i>I</i> ≥ 2 $\sigma$ )	6622	2489	5172
<i>R</i> <sub>int</sub>	0.041	0.051	0.040
<i>R</i> <sup>a</sup> , <i>wR</i> <sup>b</sup> [ <i>I</i> ≥ 2 $\sigma$ ( <i>I</i> )]	0.0389, 0.0992	0.0280, 0.0669	0.0316, 0.0751
Goodness-of-fit, <i>S</i> <sup>c</sup>	1.05	1.05	1.05
No. of parameters	521	237	453
$\Delta\rho$ <sub>min</sub> , $\Delta\rho$ <sub>max</sub> (e Å <sup>-3</sup> )	-0.89, 0.61	-0.27, 0.40	-0.57, 0.40
CCDC no.	2217106	2217105	2217107

<sup>a</sup>  $R = \sum ||F_o| - |F_d| | / \sum |F_o|$ ; <sup>b</sup>  $wR = [\sum(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$ ; <sup>c</sup>  $S = \sum[w(F_o^2 - F_c^2)^2 / (N_{obs} - N_{param})]^{1/2}$

**Table S2.** Selected bond lengths (Å) and angles (°) for compounds **1a·6H<sub>2</sub>O**, **1b** and **1b·H<sub>2</sub>O**.

	<b>1a·6H<sub>2</sub>O</b>	<b>1b</b>	<b>1b·H<sub>2</sub>O</b>
Cu1–O11	1.9423(15)	1.9469(19)	1.9473(16)
Cu1–O13	2.608(2)	2.2429(15)	2.2535(14)
Cu1–N1	1.981(2)	1.9860(19)	1.989(2)
Cu1–N11	1.9898(17)	1.9949(17)	1.9955(19)
Cu1–N110	1.9982(17)	2.017(2)	2.0140(19)
Cu2–O11S	2.3589(17)	-	-
Cu2–O21	1.9526(18)	-	1.9222(18)
Cu2–O23	-	-	2.2809(14)
Cu2–N2	1.978(2)	-	1.998(2)
Cu2–N21	1.988(2)	-	1.998(2)
Cu2–N210	2.0069(19)	-	2.0073(19)

**Table S3.** The geometry of selected hydrogen bonds for compounds **1a·6H<sub>2</sub>O**, **1b** and **1b·H<sub>2</sub>O**.

	D-H...A	D-H / Å	H...A / Å	D...A / Å	D-H...A / °
<b>1a·6H<sub>2</sub>O</b>	N2-H2A...O14S	0.92(3)	2.07(4)	2.930(3)	155(3)
	N2-H2B...O13S <sup>a</sup>	0.85(3)	2.14(3)	2.925(3)	154(3)
	O1W-H1W2...O14S <sup>b</sup>	0.85(4)	2.40(5)	3.077(3)	138(4)
	O13-H13A...O13S	0.85(3)	1.92(3)	2.736(3)	164(3)
	O6W-H6W1...O11S <sup>a</sup>	0.85(3)	2.04(3)	2.876(3)	167(3)
<b>1b</b>	N1-H1A...O12S	0.84(3)	2.17(3)	3.002(4)	175(3)
	N1-H1A...O12S <sup>c</sup>	0.84(3)	1.97(3)	2.752(4)	154(3)
	N1-H1B...O11S <sup>d</sup>	0.88(3)	2.14(3)	2.956(4)	154(3)
	N1-H1B...O13S <sup>d</sup>	0.88(3)	2.00(3)	2.868(4)	168(3)
	O13-H13A...O11 <sup>e</sup>	0.83(3)	2.05(3)	2.856(3)	163(3)
	O13-H13B...O11S	0.829(18)	2.09(3)	2.807(4)	145(3)
	O13-H13B...O14S <sup>f</sup>	0.829(18)	1.967(16)	2.737(4)	154(3)
<b>1b·H<sub>2</sub>O</b>	N1-H1A...O12S <sup>g</sup>	0.87(3)	2.10(3)	2.958(3)	170(3)
	N1-H1A...O12S <sup>g</sup>	0.89(3)	2.01(3)	2.890(3)	170(2)
	N1-H1A...O12S <sup>h</sup>	0.84(3)	2.14(3)	2.947(3)	159(3)
	N1-H1A...O12S <sup>i</sup>	0.87(3)	2.03(3)	2.899(3)	173(2)
	O1W-H1W1...O14S <sup>g</sup>	0.84(2)	2.05(2)	2.868(3)	164(3)

<sup>a</sup>-1+x,y,z; <sup>b</sup>1+x,y,z; <sup>c</sup>-x,y,1/2-z; <sup>d</sup>x,1+y,z; <sup>e</sup>-x,-y,1-z; <sup>f</sup>-x,y,1/2-z; <sup>g</sup>x,-1+y,-1+z; <sup>h</sup>x,1/2-y,-1/2+z; <sup>i</sup>x,3/2-y,-1/2+z