

# Benzene-1,3-Dihydrogensulfate as a Non-Cyclic Ionophore: Selective Extraction and Transport of Metal Ions

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## SUPPLEMENTARY INFORMATION

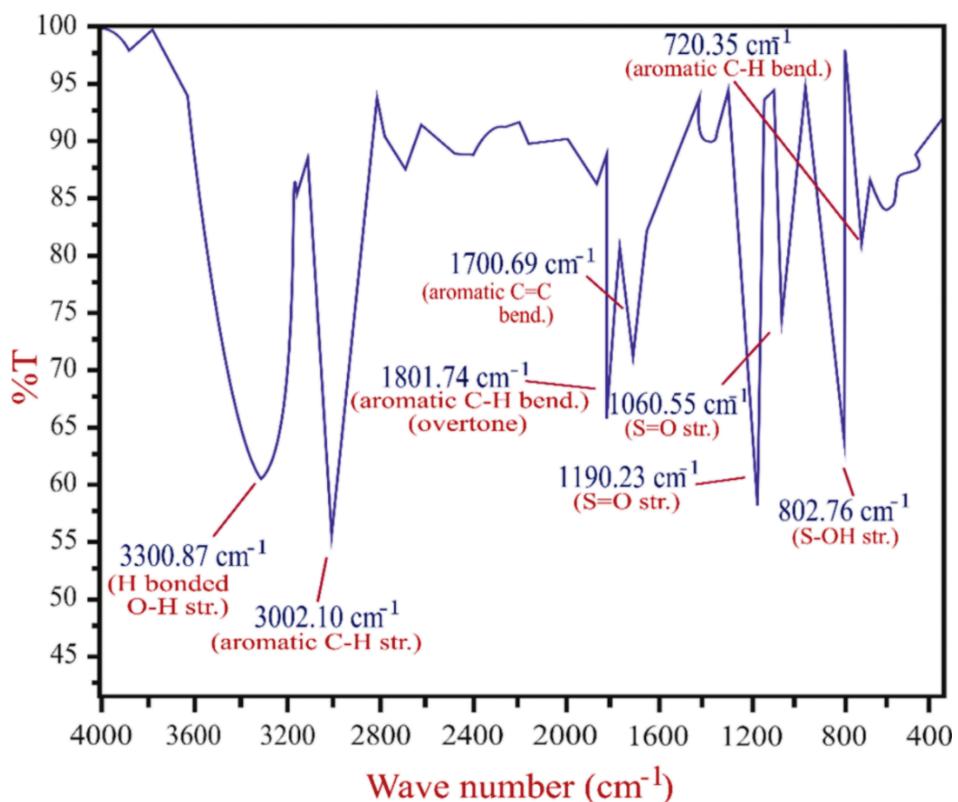
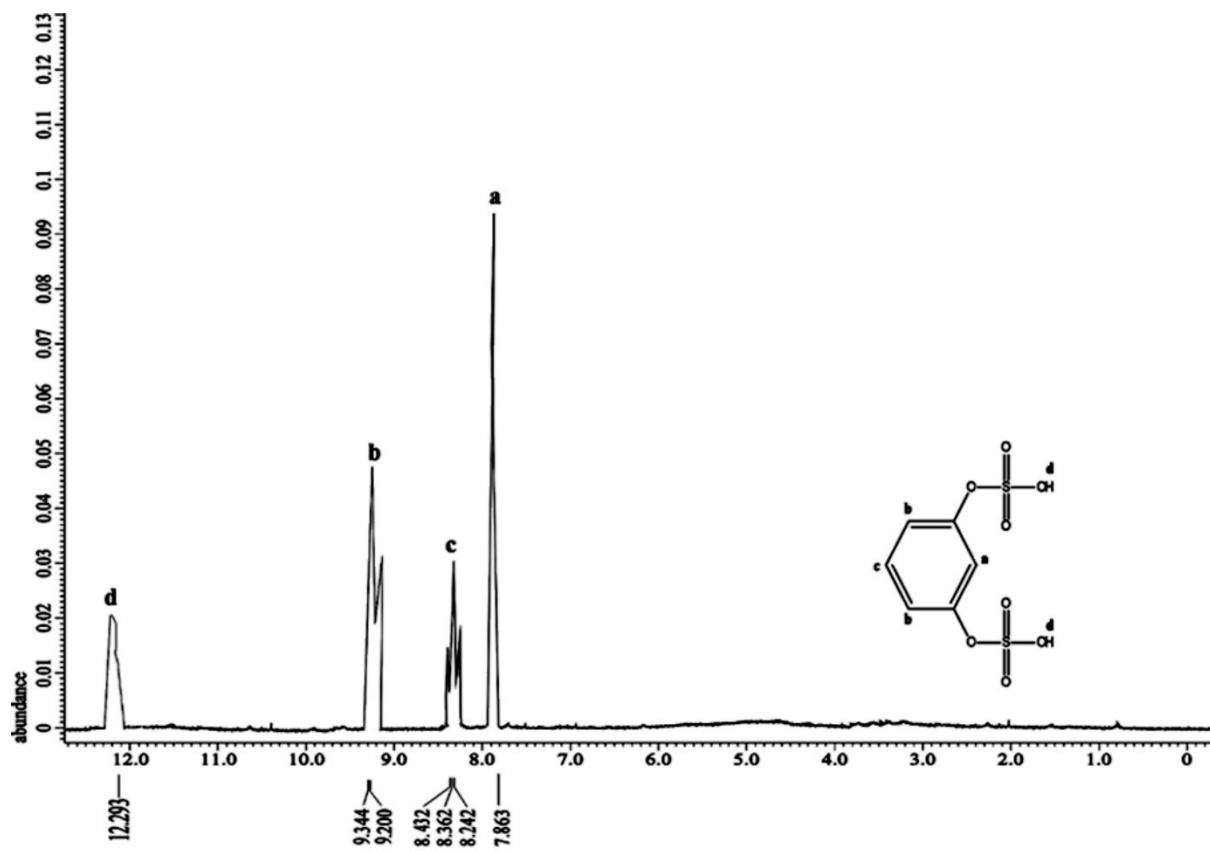
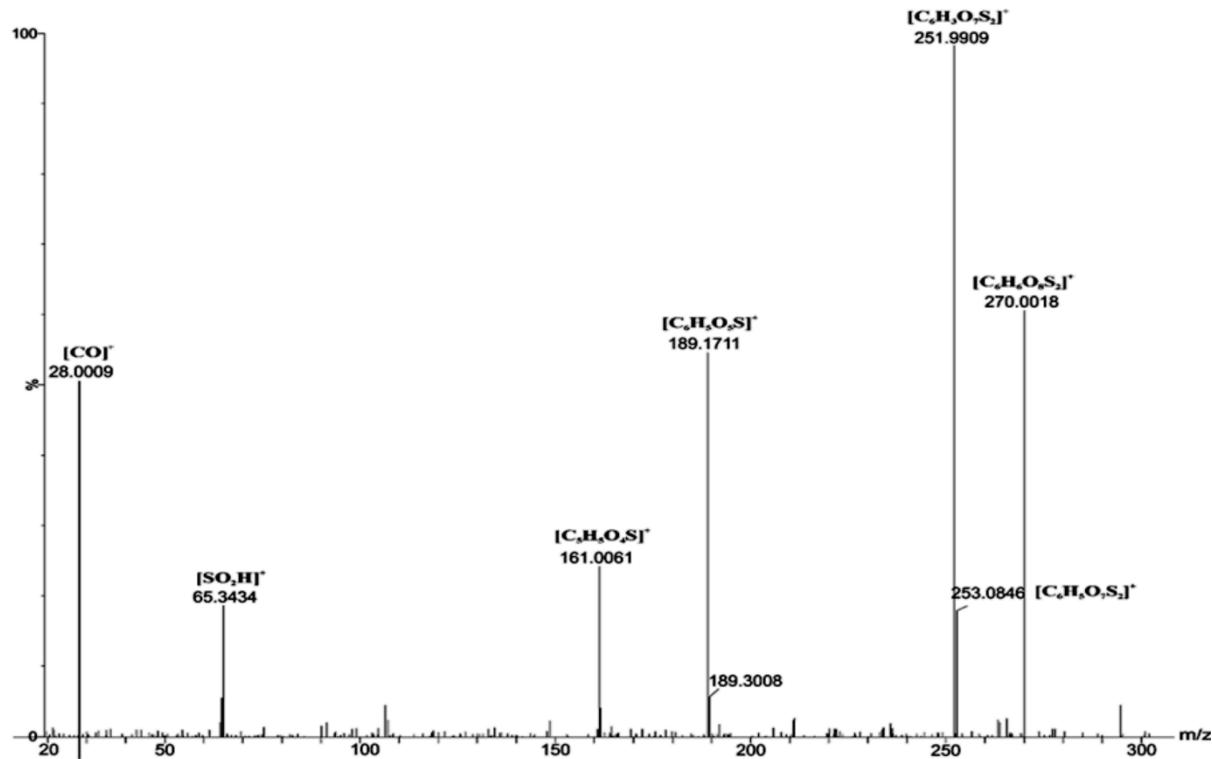


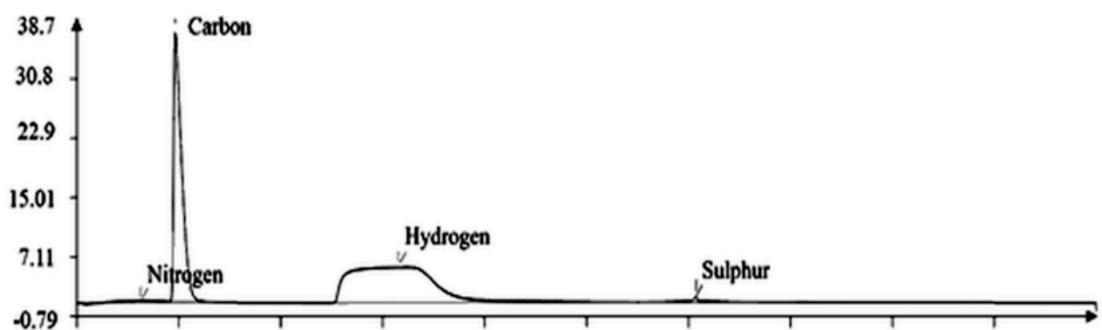
Figure S1. IR spectrum of benzene-1,3-dihydrogensulfate.



**Figure 2.**  $^1\text{H}$  NMR spectrum of benzene-1,3-dihydrogensulfate.



**Figure 3.** Mass spectrum benzene-1,3-dihydrogensulfate.



Peak Number (#)	Retention Time (min)	Area (.1* uV * sec)	Element %	Component Name
1	0.767	97082	0.0005	Nitrogen
2	1.158	2316367	26.5789	Carbon
3	3.892	3373631	2.1908	Hydrogen
4	7.292	50927	23.7989	Sulphur
		0.531	52.5691	

**Figure 4.** Elemental analysis of benzene-1,3-dihydrogensulfate.

**Table 1.** Extraction of metal ions from picrate, dinitrophenolate, and *o*-nitrophenolate salts into chloroform.

Metal Ion	Parameter	picrate	dinitrophenolate	<i>o</i> -nitrophenolate
<b>Zn<sup>2+</sup></b>	R1	0.35	0.32	0.05
	R2	0.32	0.29	0.05
	R3	0.31	0.30	0.04
	<b>Mean(ppm)</b>	<b>0.33</b>	<b>0.30</b>	<b>0.05</b>
<b>Na<sup>+</sup></b>	R1	0.70	0.28	0.07
	R2	0.69	0.29	0.06
	R3	0.65	0.28	0.05
	<b>Mean(ppm)</b>	<b>0.68</b>	<b>0.28</b>	<b>0.06</b>
<b>K<sup>+</sup></b>	R1	0.48	0.19	0.03
	R2	0.47	0.17	0.02
	R3	0.44	0.19	0.03
	<b>Mean(ppm)</b>	<b>0.46</b>	<b>0.18</b>	<b>0.03</b>
<b>Ni<sup>2+</sup></b>	R1	0.30	0.21	0.04
	R2	0.31	0.20	0.04
	R3	0.32	0.21	0.04
	<b>Mean(ppm)</b>	<b>0.31</b>	<b>0.21</b>	<b>0.04</b>

**Table 2.** Extraction of metal ions from picrate, dinitrophenolate, and *o*-nitrophenolate salts into 1,2-dichloroethane membranes.

Metal Ion	Parameter	picrate	dinitrophenolate	<i>o</i> -nitrophenolate
<b>Zn<sup>2+</sup></b>	R1	0.42	0.21	0.15
	R2	0.41	0.19	0.16
	R3	0.42	0.21	0.15
	<b>Mean(ppm)</b>	<b>0.42</b>	<b>0.20</b>	<b>0.15</b>
<b>Na<sup>+</sup></b>	R1	0.68	0.25	0.11
	R2	0.65	0.24	0.09
	R3	0.67	0.25	0.10
	<b>Mean(ppm)</b>	<b>0.67</b>	<b>0.25</b>	<b>0.10</b>
<b>K<sup>+</sup></b>	R1	0.55	0.18	0.08
	R2	0.52	0.19	0.09
	R3	0.54	0.18	0.08
	<b>Mean(ppm)</b>	<b>0.54</b>	<b>0.18</b>	<b>0.08</b>
<b>Ni<sup>2+</sup></b>	R1	0.35	0.2	0.12
	R2	0.34	0.21	0.11
	R3	0.33	0.22	0.11
	<b>Mean(ppm)</b>	<b>0.34</b>	<b>0.21</b>	<b>0.11</b>

**Table 3.** Transport of metal ions from picrate, dinitrophenolate, and *o*-nitrophenolate salts into chloroform membrane.

Metal Ion	Parameter	picrate	dinitrophenolate	<i>o</i> -nitrophenolate
<b>Zn<sup>2+</sup></b>	R1	0.91	0.52	0.02
	R2	0.89	0.53	0.01
	R3	0.88	0.52	0.02
	<b>Mean(ppm)</b>	<b>0.89</b>	<b>0.52</b>	<b>0.02</b>
<b>Na<sup>+</sup></b>	R1	1.39	0.88	0.03
	R2	1.35	0.87	0.01
	R3	1.37	0.89	0.02
	<b>Mean(ppm)</b>	<b>1.37</b>	<b>0.88</b>	<b>0.02</b>
<b>K<sup>+</sup></b>	R1	1.22	0.79	0.05
	R2	1.20	0.79	0.04
	R3	1.22	0.78	0.05
	<b>Mean(ppm)</b>	<b>1.21</b>	<b>0.79</b>	<b>0.05</b>
<b>Ni<sup>2+</sup></b>	R1	0.51	0.32	0.01
	R2	0.48	0.30	0.00
	R3	0.49	0.29	0.01
	<b>Mean(ppm)</b>	<b>0.49</b>	<b>0.30</b>	<b>0.01</b>

**Table 4.** Transport of metal ions from picrate, dinitrophenolate and *o*-nitrophenolate salts into 1,2-dichloroethane membrane.

Metal Ion	Parameter	Picrate	dinitrophenolate	<i>o</i> -nitrophenolate
<b>Zn<sup>2+</sup></b>	R1	0.98	0.87	0.11
	R2	0.96	0.85	0.09
	R3	0.96	0.85	0.10
	<b>Mean(ppm)</b>	<b>0.97</b>	<b>0.86</b>	<b>0.10</b>
<b>Na<sup>+</sup></b>	R1	1.46	1.08	0.04
	R2	1.44	1.06	0.03
	R3	1.45	1.06	0.03
	<b>Mean(ppm)</b>	<b>1.45</b>	<b>1.07</b>	<b>0.03</b>
<b>K<sup>+</sup></b>	R1	1.26	1.11	0.10
	R2	1.20	1.10	0.09
	R3	1.23	1.09	0.09
	<b>Mean(ppm)</b>	<b>1.23</b>	<b>1.10</b>	<b>0.09</b>
<b>Ni<sup>2+</sup></b>	R1	0.62	0.55	0.01
	R2	0.60	0.52	0.00
	R3	0.62	0.53	0.01
	<b>Mean(ppm)</b>	<b>0.61</b>	<b>0.53</b>	<b>0.01</b>

**Table 5.** Spectral and elemental analysis data of ionophore

Analysis Type	Key Findings
IR Spectrum ( $\text{cm}^{-1}$ )	720.35 (aromatic C-H), 802.76 (S-OH stretching), 1190.23 and 1060.55 (S=O stretching), 1700.69 (aromatic C=C bending), 1801.74 (aromatic C-H bending, overtone), 3002.10 (aromatic C-H stretching), 3300.87 (H bonded O-H stretching)
$^1\text{H-NMR}$ ( $\delta$ , ppm)	7.9 (s, 1H, CH), 9.2-9.3 (d, 1H, CH), 8.2-8.4 (t, 1H, CH), 12.0-12.3 (br, 1H, OH)
Mass Spectrum (m/z)	Calculated: 270.2384, Found: 270.0018
Elemental Analysis (%)	Found: C 26.58; H 2.19; S 23.8. Calculated: C 26.67; H 2.22; S 23.7

### Coordinates from wxMacMolPlt

C	6.0	-4.22105789	2.56104136	0.87929606
C	6.0	-3.78582478	3.06105185	2.10459566
C	6.0	-3.82661891	1.32444715	0.38240701
C	6.0	-2.94572186	0.55710131	1.14697719
H	1.0	-4.20384407	0.97606522	-0.57165587
C	6.0	-2.46966314	1.02734649	2.37027550
H	1.0	-2.62313795	-0.41402283	0.78671163
C	6.0	-2.89707589	2.27168226	2.82995582
H	1.0	-4.14578962	4.01444197	2.46638036
O	8.0	-5.17554140	3.31973147	0.16655333
H	1.0	-1.78084779	0.44589975	2.97276664
O	8.0	-2.38414860	2.62434483	4.09527683
S	16.0	-4.66585493	4.29821396	-1.04557037
S	16.0	-2.25384045	4.16454315	4.62257147
O	8.0	-5.89506721	4.78596926	-1.64086330
O	8.0	-3.63000298	3.64115834	-1.81722879
O	8.0	-1.42265844	4.05043650	5.80344057
O	8.0	-3.56048155	4.79236174	4.65689278
O	8.0	-3.91954684	5.48352718	-0.24967983
H	1.0	-4.57415247	6.13096380	0.07627337
O	8.0	-1.46493208	4.88559580	3.42304635
H	1.0	-0.50483137	4.71732950	3.48836708

### Energies in solvent

FREE ENERGY IN SOLVENT =  $\langle \Psi | H(0) + V/2 | \Psi \rangle$  = -1629.7674850920 A.U.  
INTERNAL ENERGY IN SOLVENT =  $\langle \Psi | H(0) | \Psi \rangle$  = -1629.7412170299 A.U.  
DELTA INTERNAL ENERGY =  $\langle D-\Psi | H(0) | D-\Psi \rangle$  = 0.0000000000 A.U.  
ELECTROSTATIC INTERACTION = -0.0262680621 A.U.  
PIEROTTI CAVITATION ENERGY = 0.0000000000 A.U.  
DISPERSION FREE ENERGY = 0.0000000000 A.U.  
REPULSION FREE ENERGY = 0.0000000000 A.U.  
TOTAL INTERACTION (DELTA + ES + CAV + DISP + REP) = -0.0262680621 A.U.  
TOTAL FREE ENERGY IN SOLVENT = -1629.7674850920 A.U.

FREE ENERGY IN SOLVENT = -1022694.65 KCAL/MOL  
INTERNAL ENERGY IN SOLVENT = -1022678.16 KCAL/MOL  
DELTA INTERNAL ENERGY = 0.00 KCAL/MOL  
ELECTROSTATIC INTERACTION = -16.48 KCAL/MOL  
PIEROTTI CAVITATION ENERGY = 0.00 KCAL/MOL  
DISPERSION FREE ENERGY = 0.00 KCAL/MOL  
REPULSION FREE ENERGY = 0.00 KCAL/MOL  
TOTAL INTERACTION = -16.48 KCAL/MOL  
TOTAL FREE ENERGY IN SOLVENT = -1022694.65 KCAL/MOL