

Supplementary information

Synthesis of Biphenyl Iodonium Salts as (Radio)labelling Precursors for Fluoroarenes

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Overview of synthesized diaryliodonium salts and their color

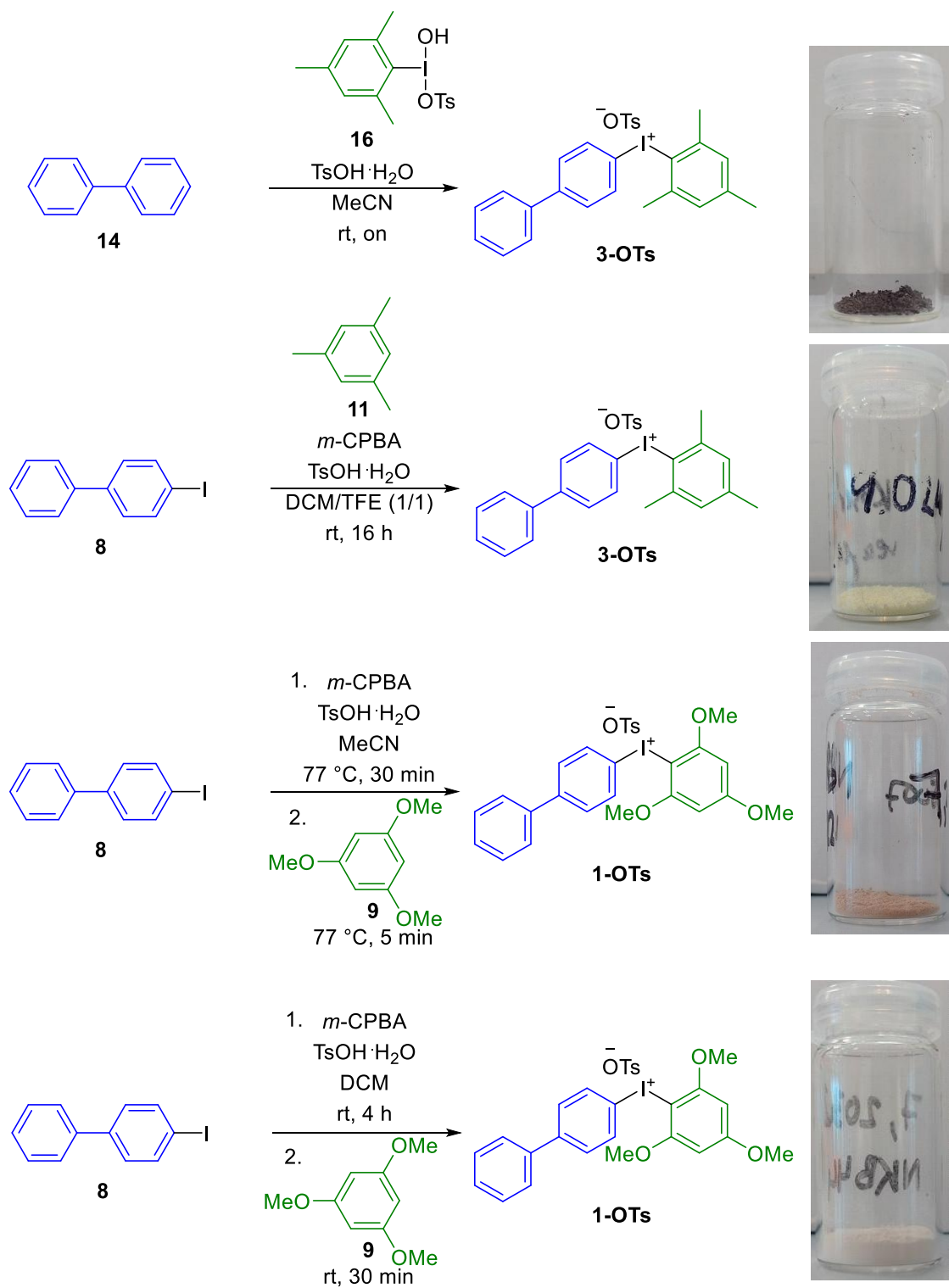


Figure S1: Overview of synthesized diaryliodonium salts and their color.

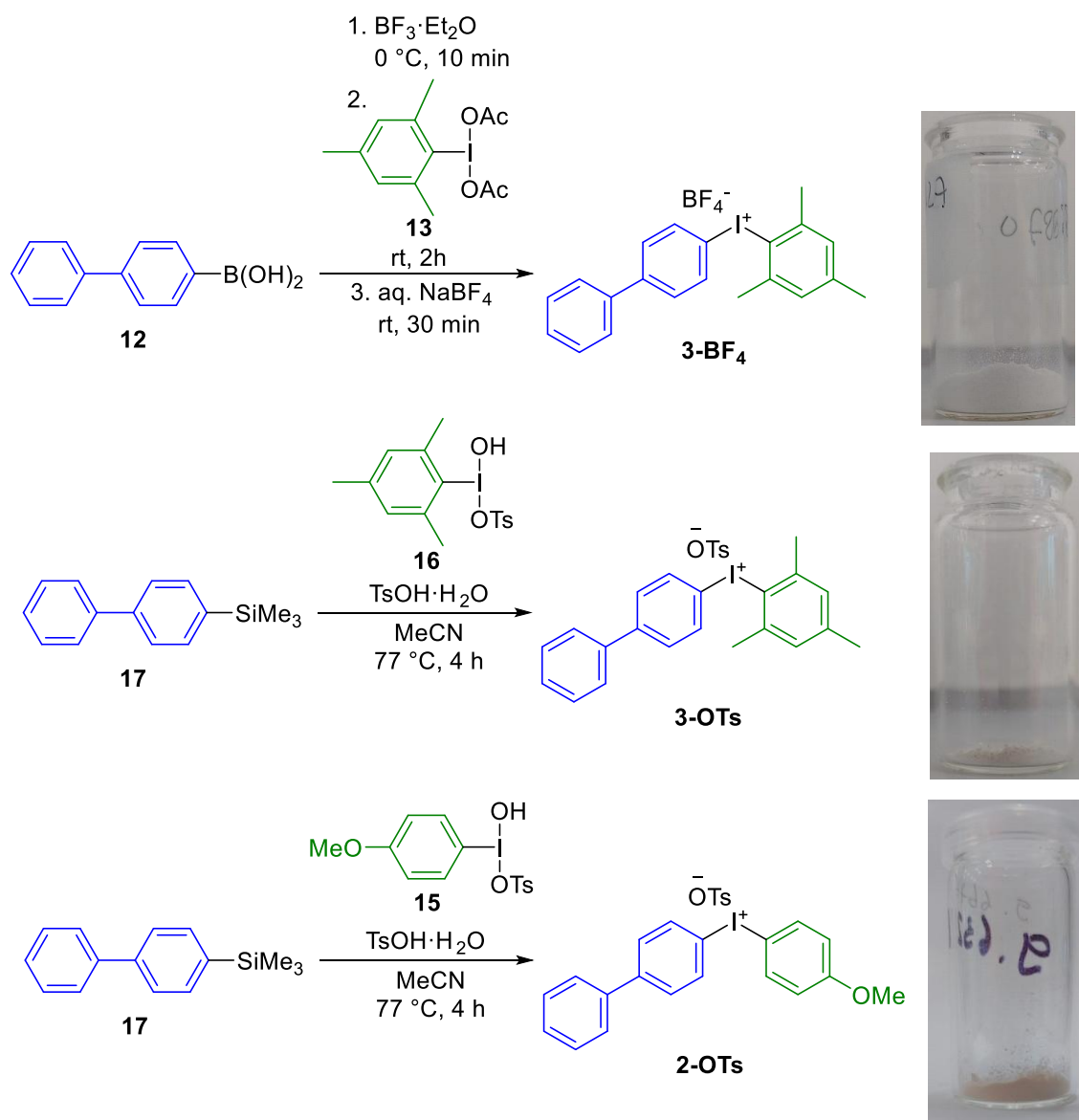


Figure S1 (continued): Overview of synthesized diaryliodonium salts and their color.

Analytical spectra of synthesized compounds

Analytical spectra of compound 1-OTs

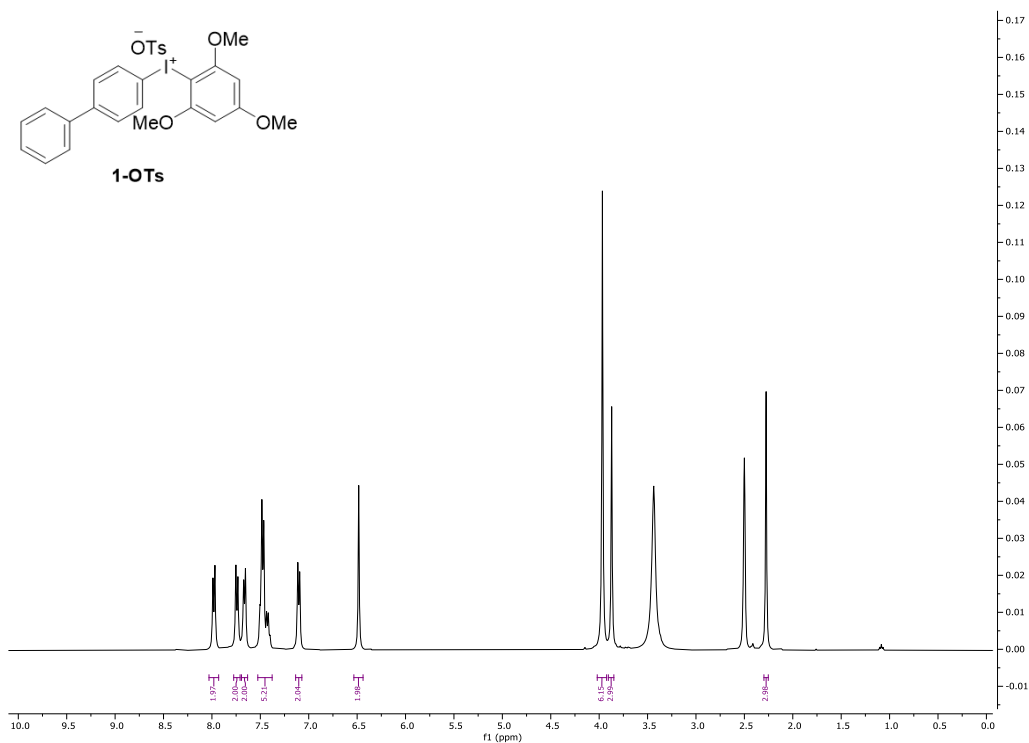


Figure S2. ^1H NMR spectrum (DMSO- d_6 , 400 MHz) of compound 1-OTs (Method A).

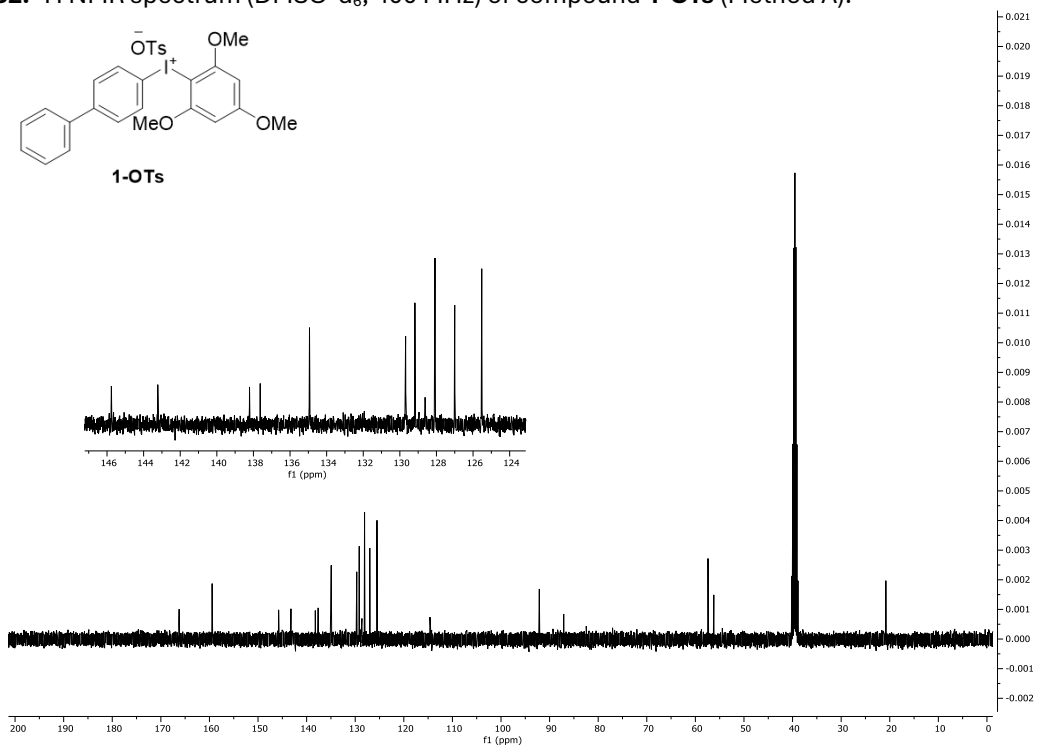


Figure S3. ^{13}C NMR (DMSO- d_6 , 101 MHz) of compound 1-OTs (Method A).

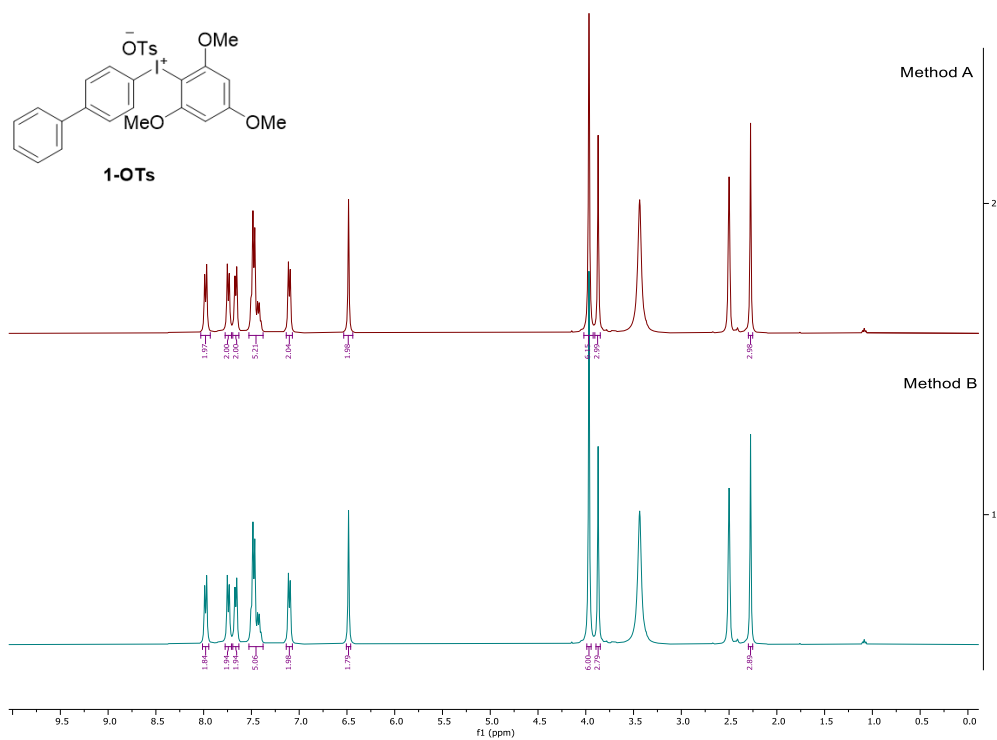


Figure S6. Overlay of ^1H NMR spectra (DMSO- d_6 , 400 MHz) of compound **1-OTs** prepared *via* different methods (Method A: red, top, red; Method B: blue, bottom).

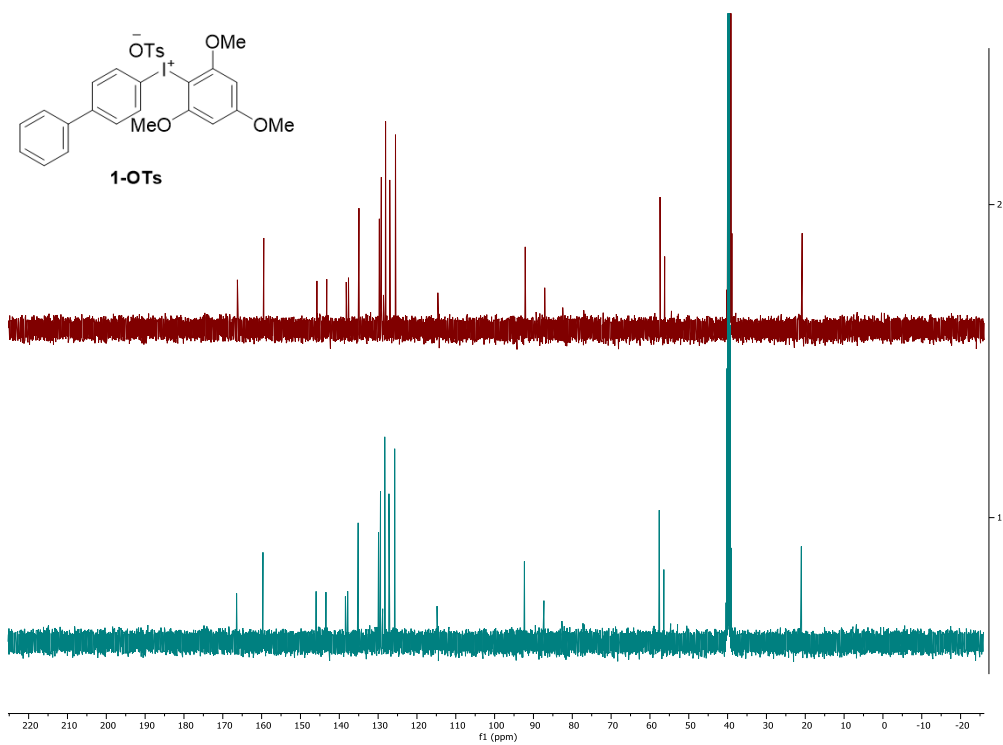


Figure S7. Overlay of ^{13}C NMR spectra (DMSO- d_6 , 101 MHz) of compound **1-OTs** prepared *via* different methods (Method A: red, top, red; Method B: blue, bottom).

Analytical spectra of compound **2-OTs**

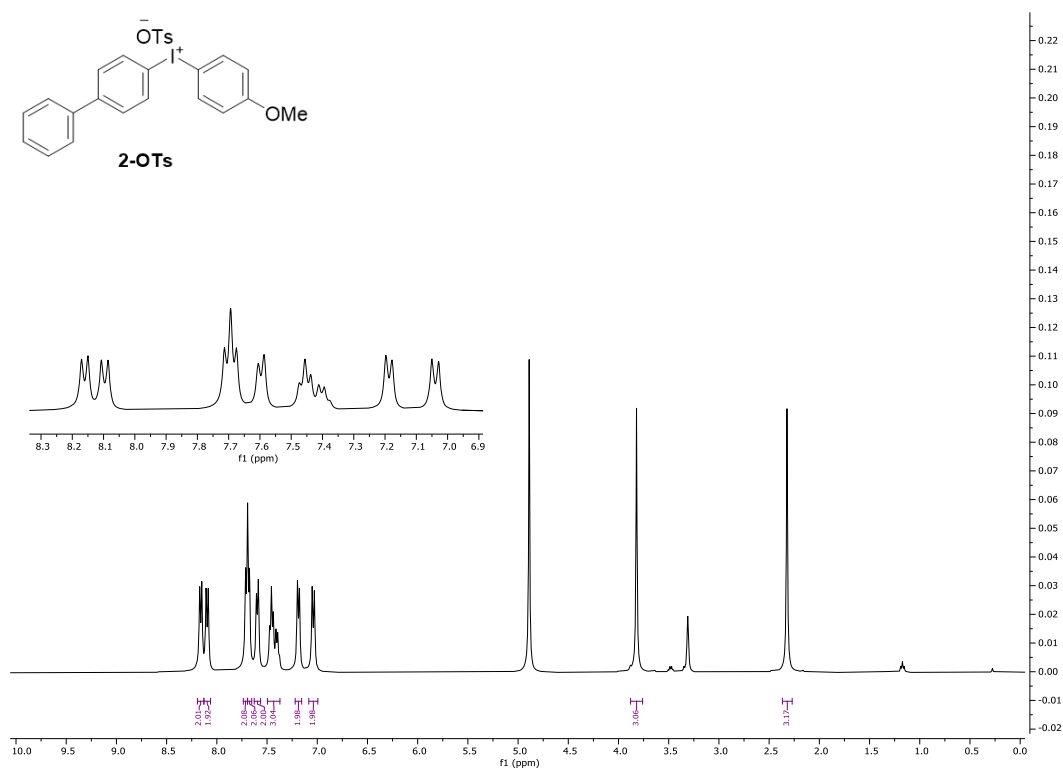


Figure S8. ^1H NMR spectrum (CD_3OD , 400 MHz) of compound **2-OTs**.

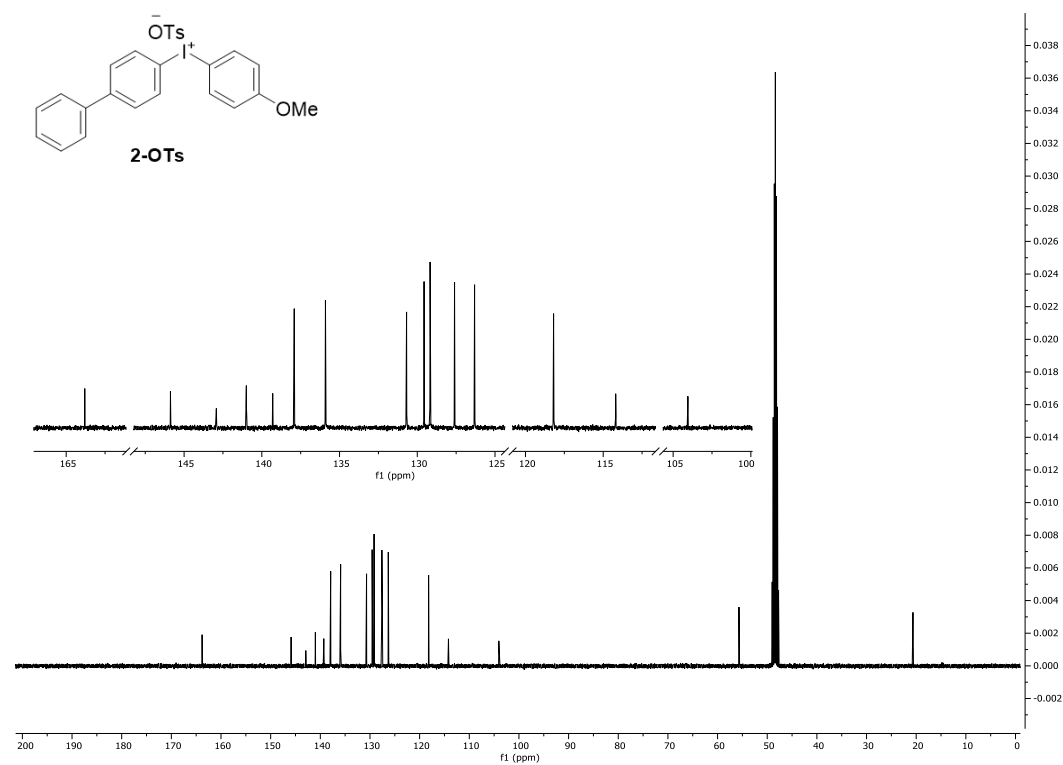


Figure S9. ^{13}C NMR (DMSO-d_6 , 101 MHz) of compound **2-OTs**.

Analytical spectra of compound **3-OTs**

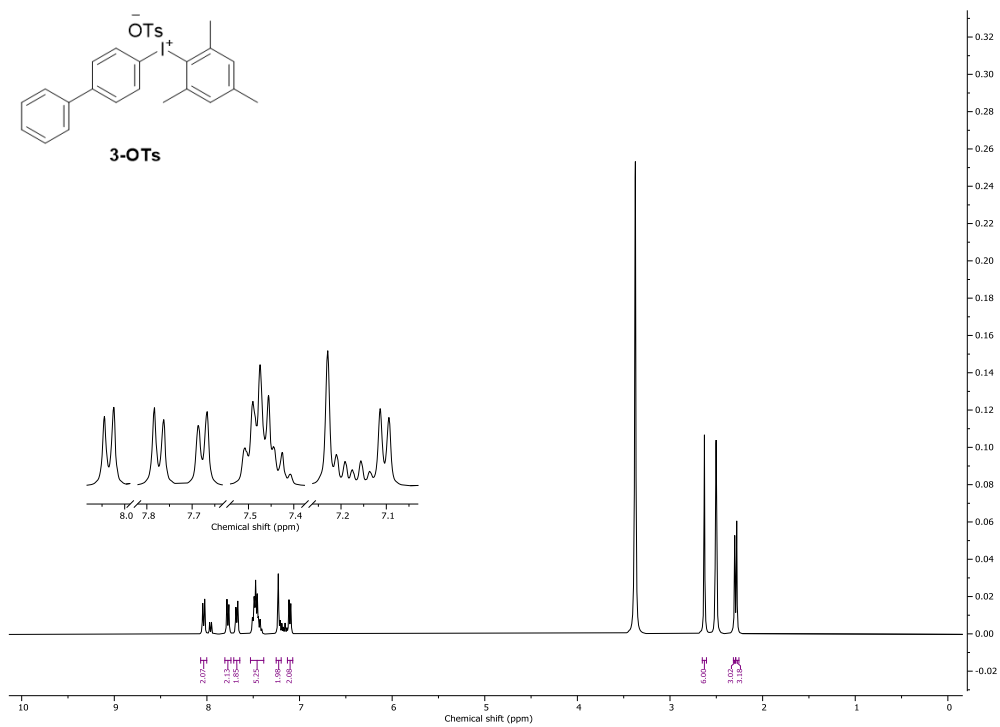


Figure S10. ^1H NMR spectrum (DMSO- d_6 , 400 MHz) of compound **3-OTs** (Method C).

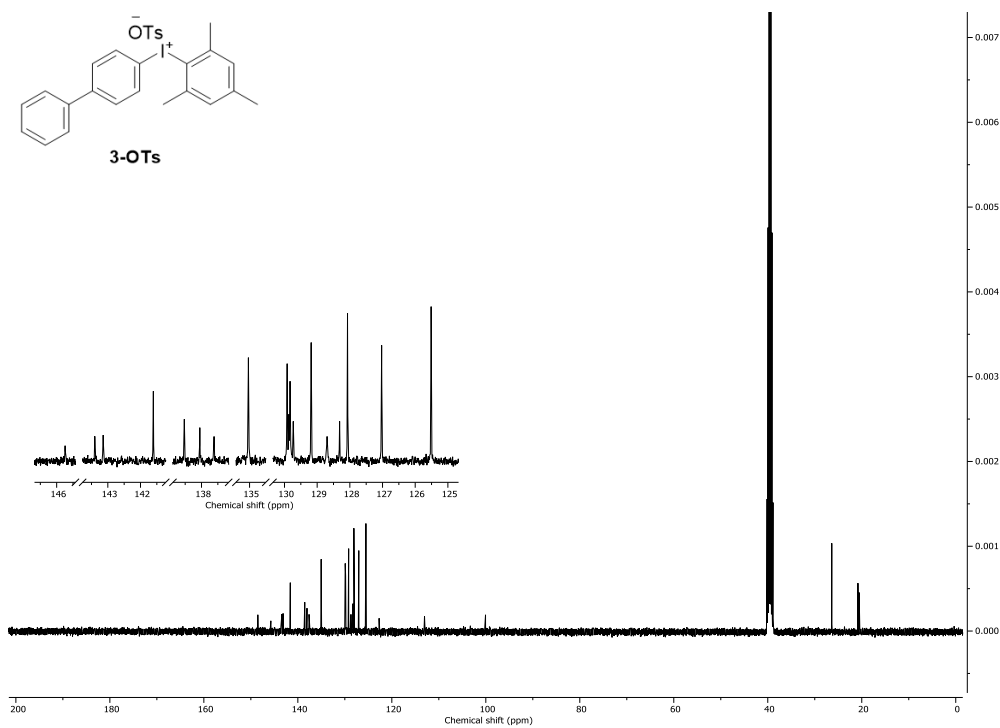


Figure S11. ^{13}C NMR spectrum (DMSO- d_6 , 101 MHz) of compound **3-OTs** (Method C).

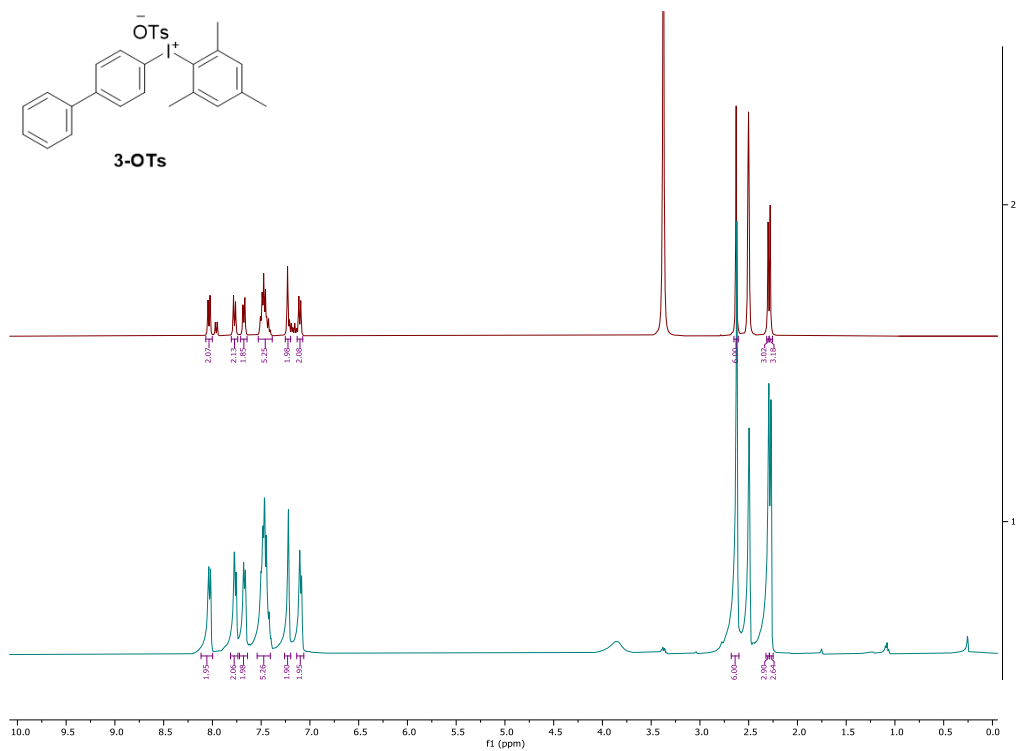


Figure S14. Overlay of ^1H NMR spectra (DMSO- d_6 , 400 MHz) of compound **3-OTs** prepared *via* different methods (Method C: red, top, red; Method D: blue, bottom).

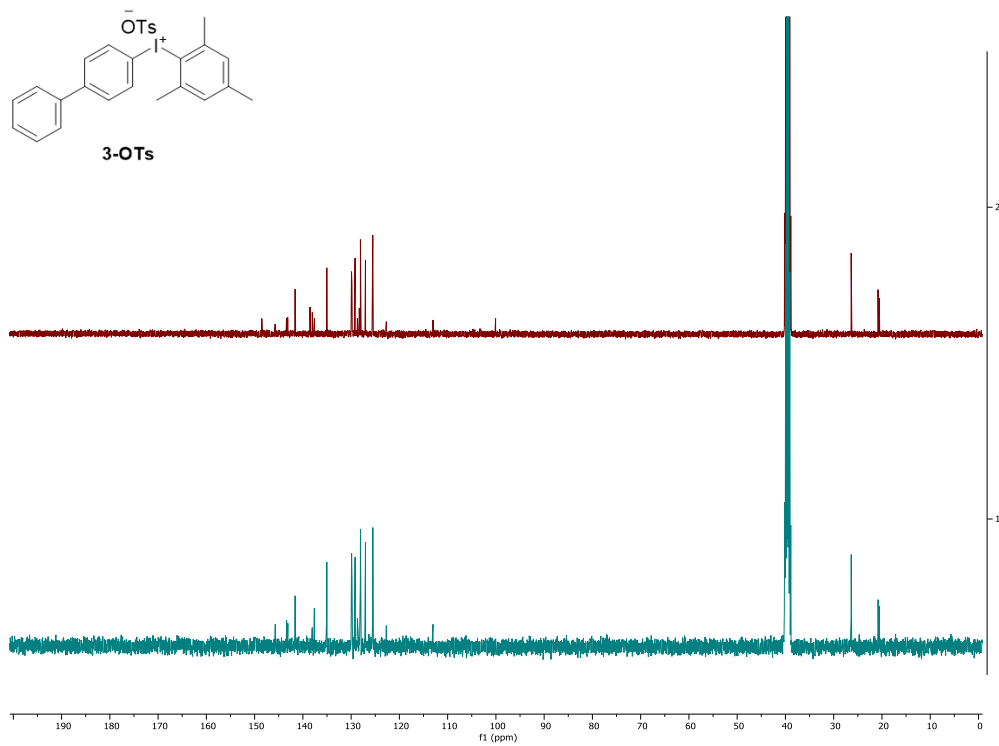


Figure S15. Overlay of ^{13}C NMR spectra (DMSO- d_6 , 101 MHz) of compound **3-OTs** prepared *via* different methods (Method C: red, top, red; Method D: blue, bottom).



Figure S18. ¹⁹F NMR spectrum (DMSO-d₆, 376 MHz) of compound **3-BF₄**. The observed splitting in ¹⁹F NMR is due to the ¹¹B/¹⁰B isotope effects.

Analytical spectra of compound **13**

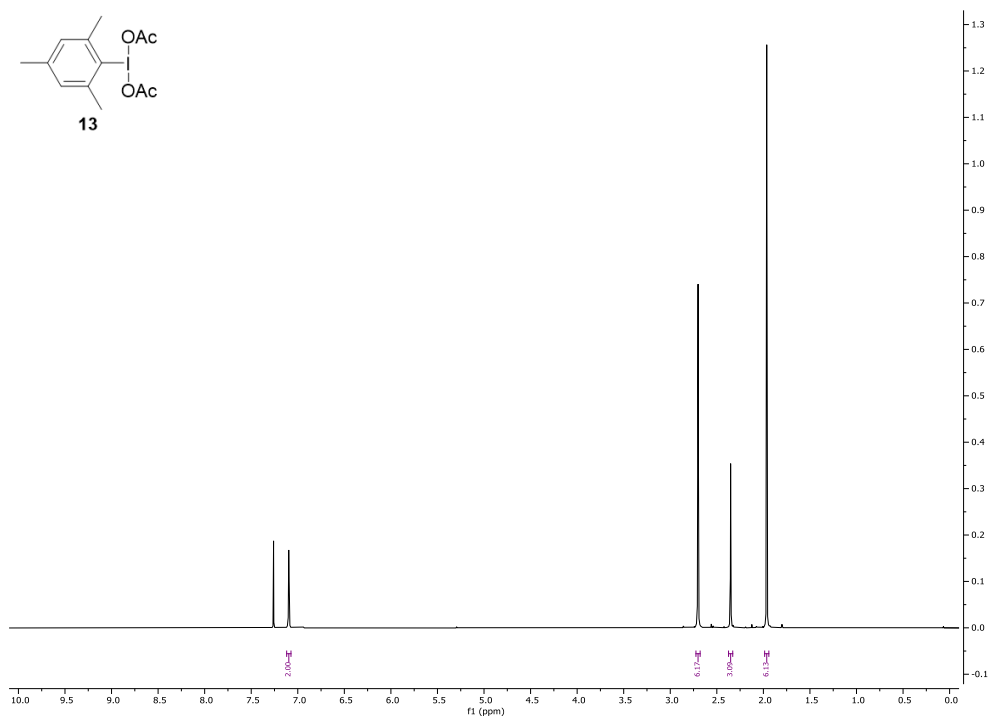


Figure S19: ¹H NMR spectrum (CDCl₃, 400 MHz) of compound **13**.

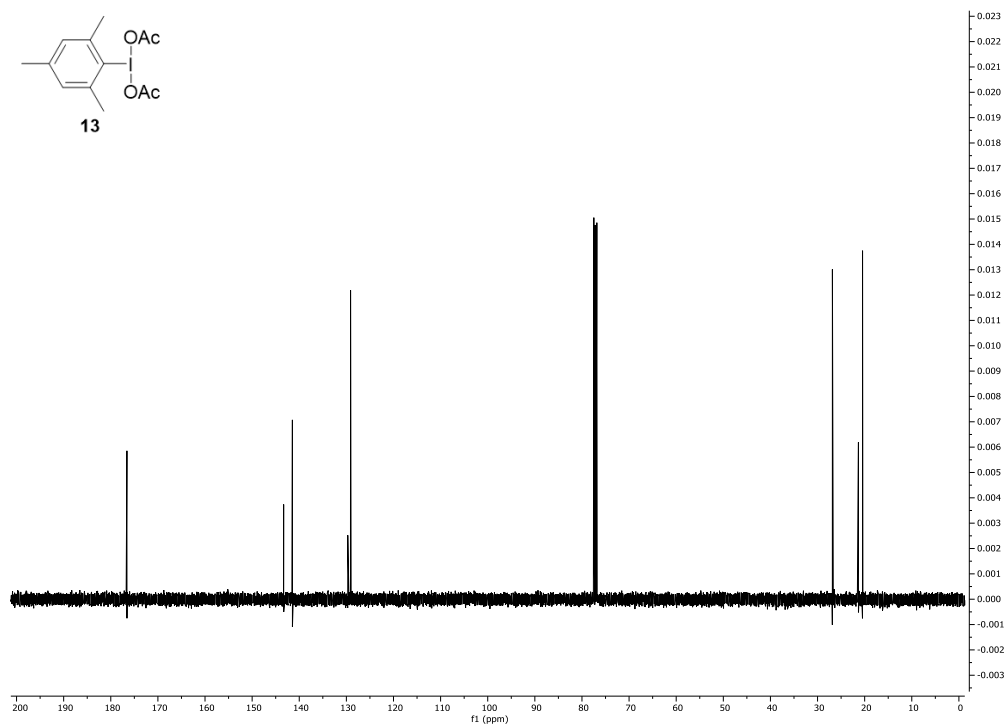


Figure S20: ¹³C NMR spectrum (CDCl₃, 101 MHz) of compound **13**.

Analytical spectra of compound **15**

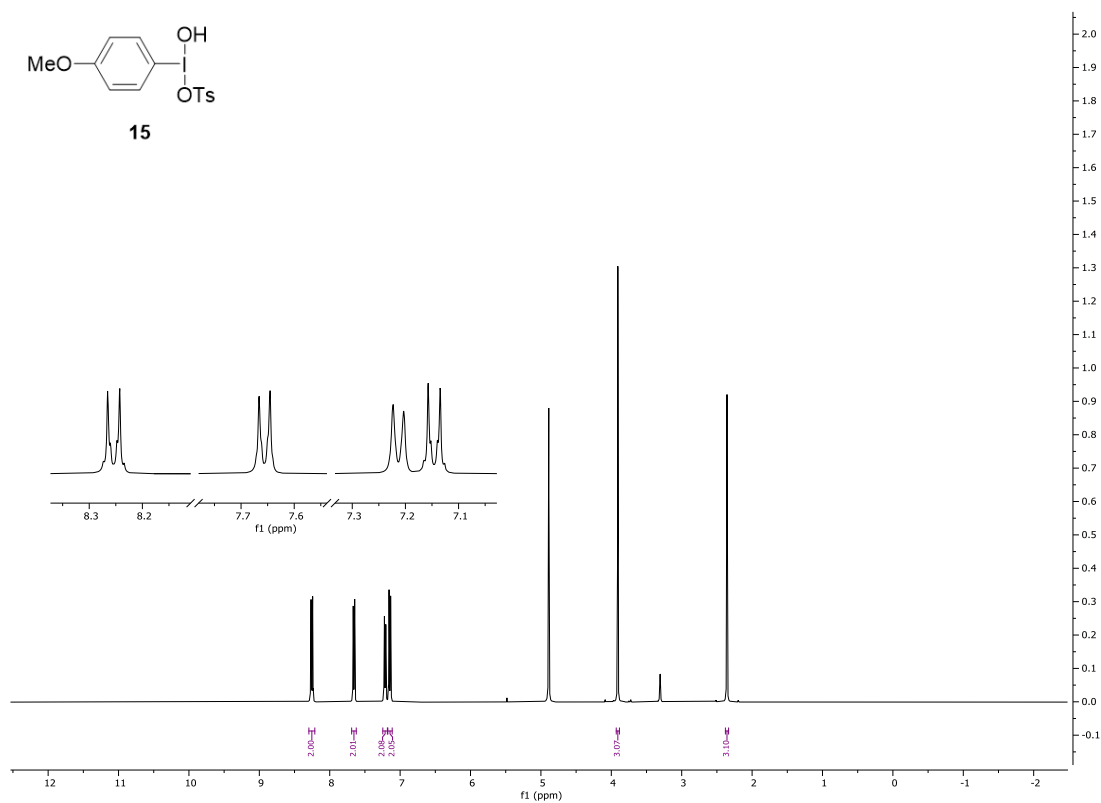
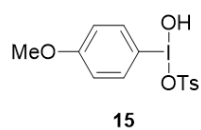


Figure S21: ^1H NMR spectrum (CD_3OD , 400 MHz) of compound **15**.

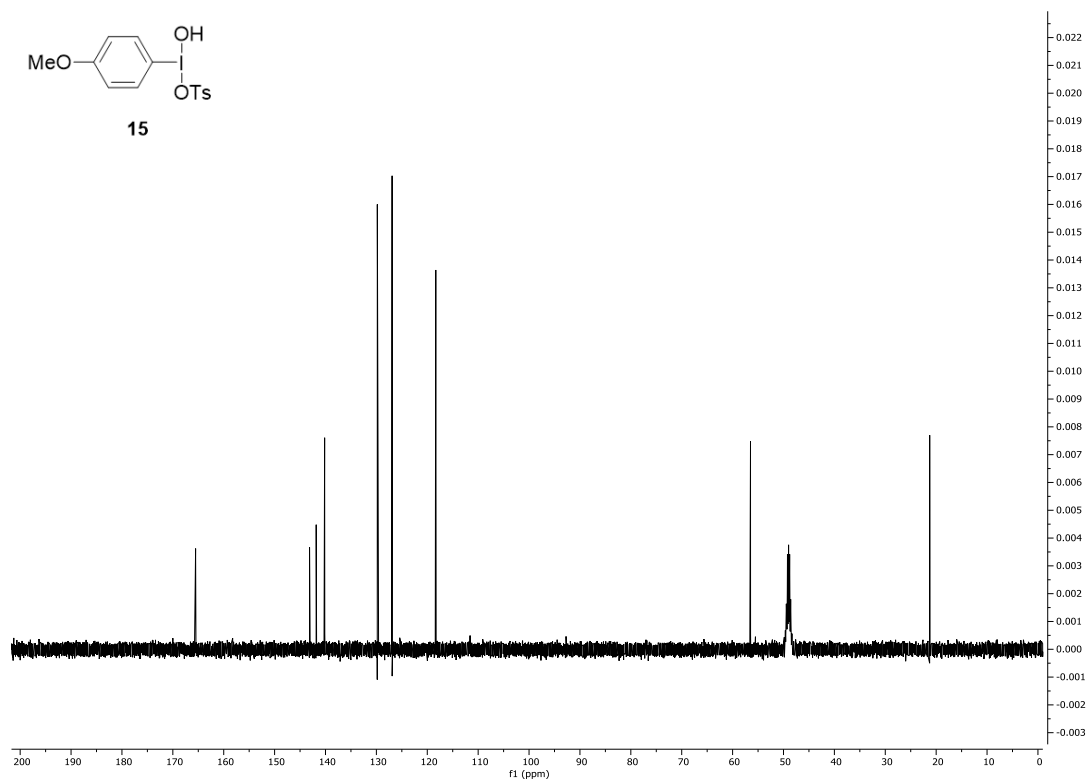
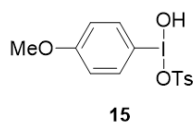


Figure S22: ^{13}C NMR spectrum (CD_3OD , 101 MHz) of compound **15**.

Analytical spectra of compound **16**

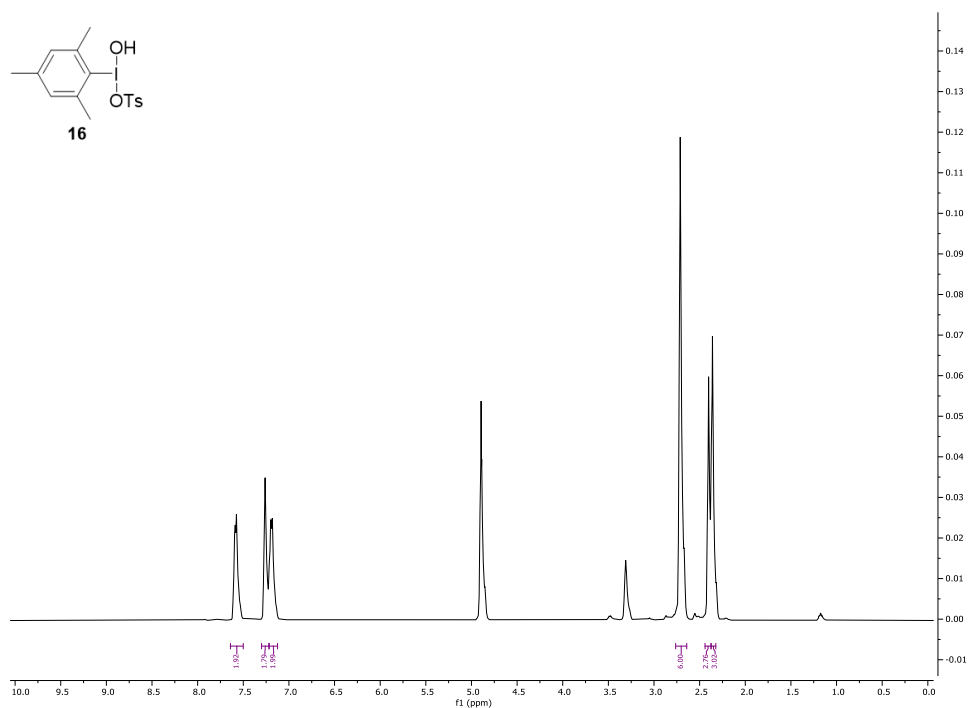


Figure S23: ¹H NMR spectrum (CD₃OD, 400 MHz) of compound **16**.

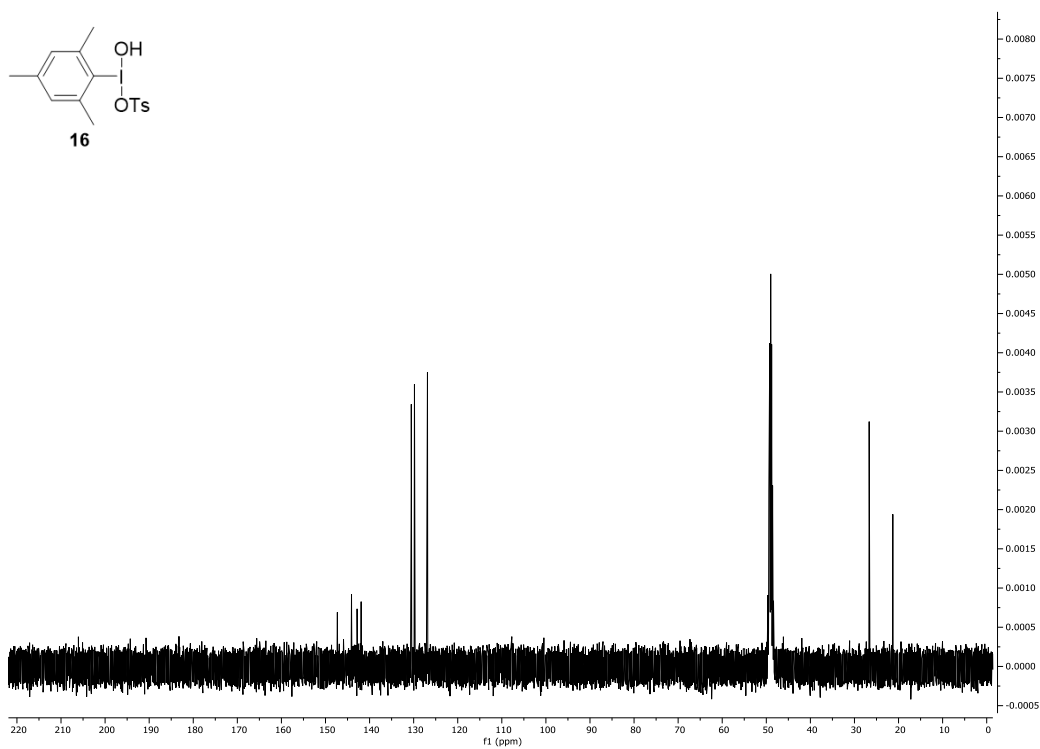


Figure S24: ¹³C NMR spectrum (CD₃OD, 101 MHz) of compound **16**.

Analytical spectra of compound **17**

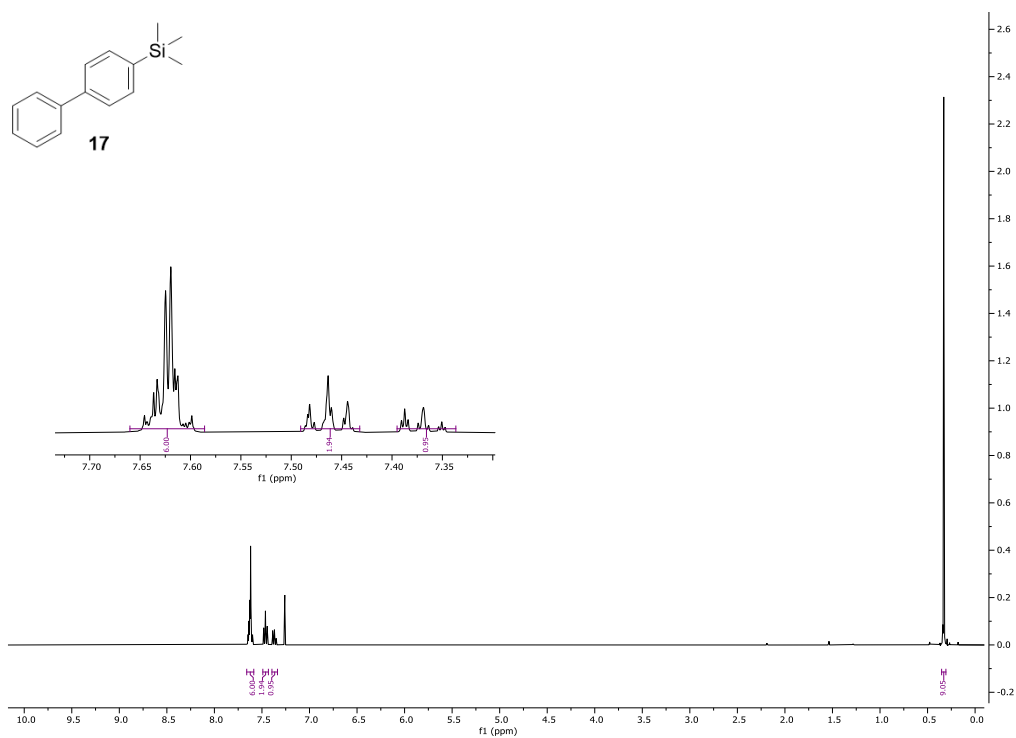


Figure S25: ¹H NMR spectrum (CDCl₃, 400 MHz) of compound **17**.

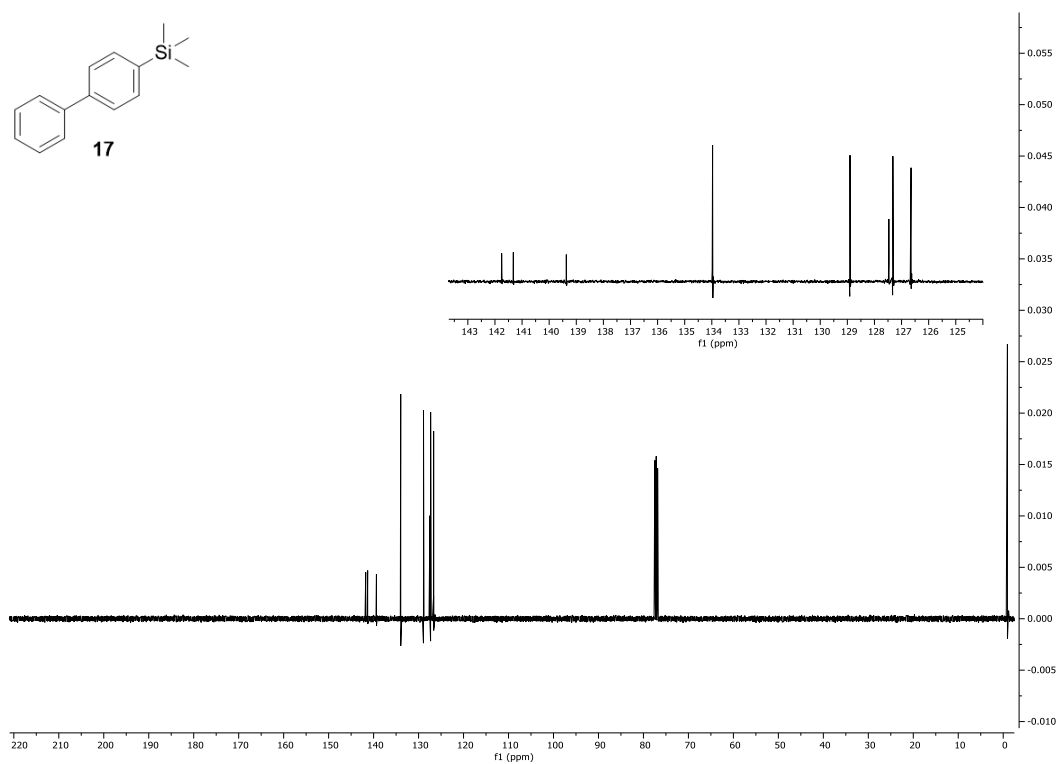


Figure S26: ¹³C NMR spectrum (CDCl₃, 101 MHz) of compound **17**.

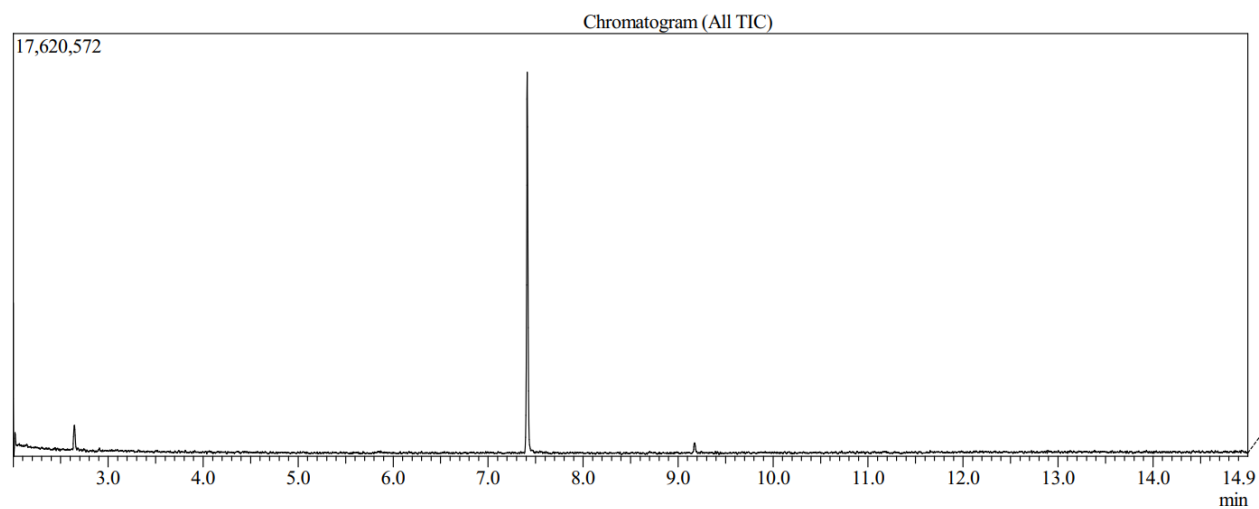


Figure S27: GC-MS chromatogram of compound **17**.

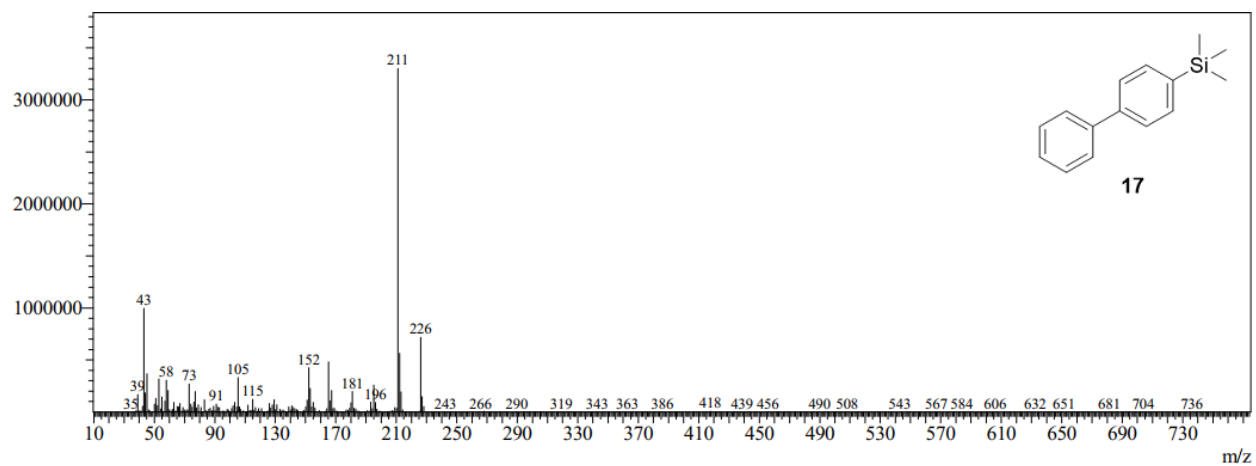


Figure S28: MS spectrum of compound **17** (7.4 min). Calculated mass for $C_{15}H_{18}Si^+ = 226$.

Degradation studies of Koser intermediate 15

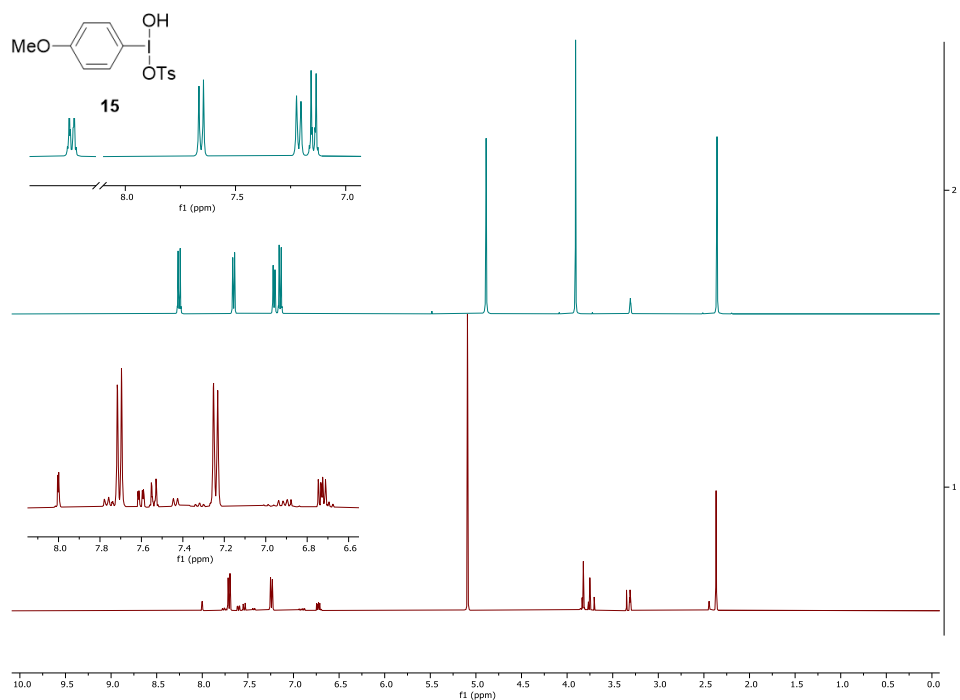


Figure S29: ^1H NMR spectrum (CD_3OD , 400 MHz) of compound **15** directly after isolation (blue, top) and degraded after 20 minutes upon isolation as a solid (red, bottom).

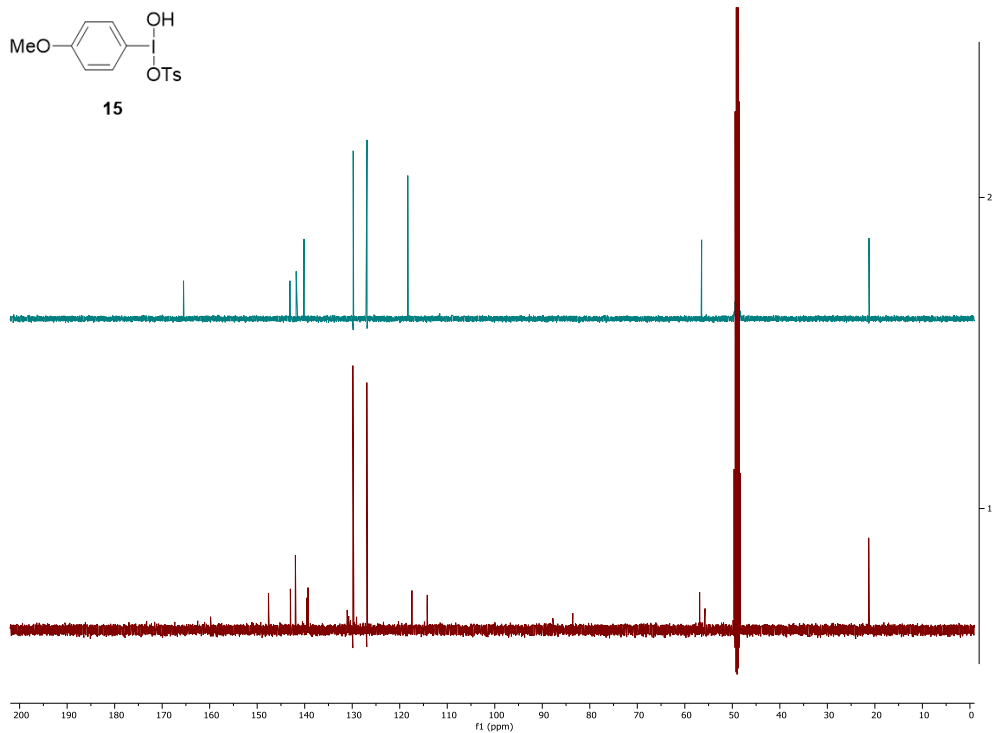


Figure S30: ^{13}C NMR spectrum (CD_3OD , 101 MHz) of compound **15** directly after isolation (blue, top) and degraded after 20 minutes upon isolation as a solid (red, bottom).

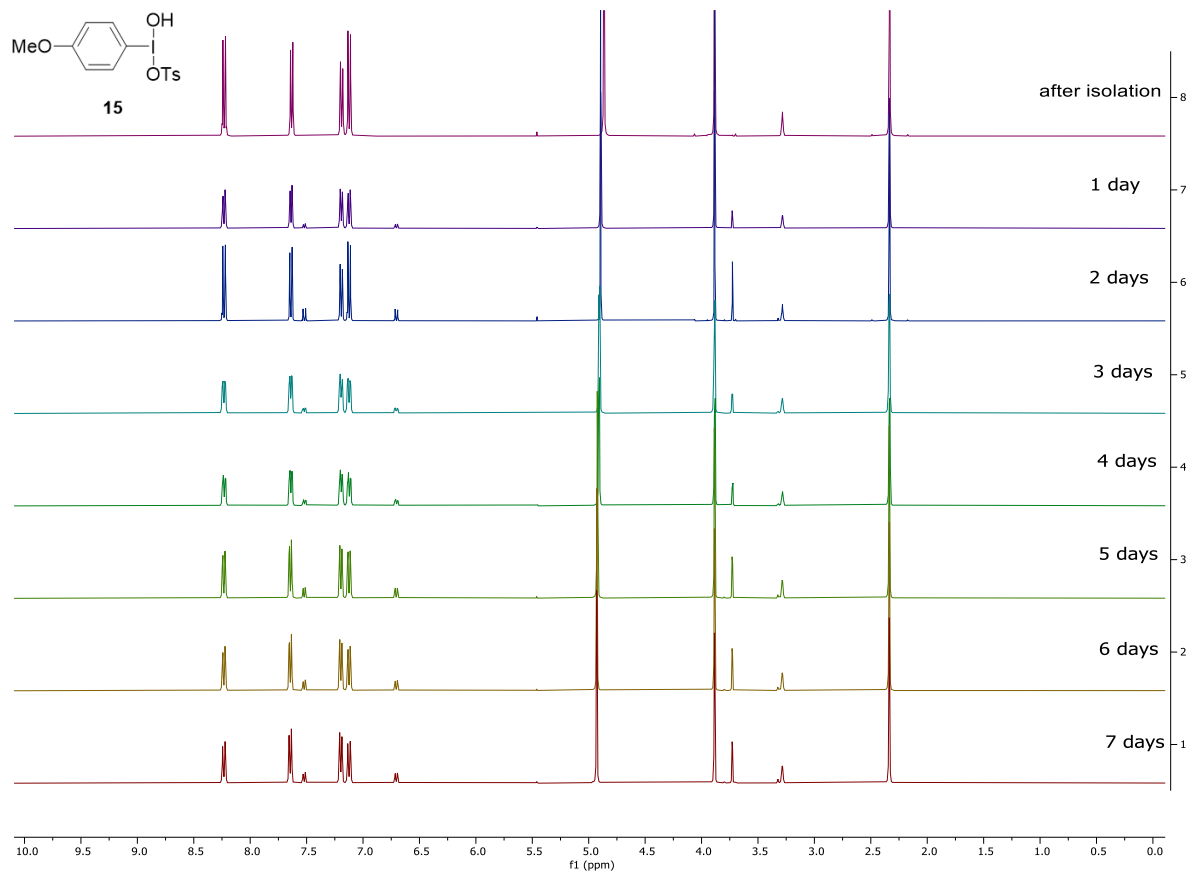


Figure S31: ¹H NMR spectrum (CD₃OD, 400 MHz) of compound **15** directly after isolation (purple, top) and stored in CD₃OD at room temperature over several days (1 day – 7 days, top to bottom). In solution degradation is slowed down with respect to storage as isolated solid (Figure S25).

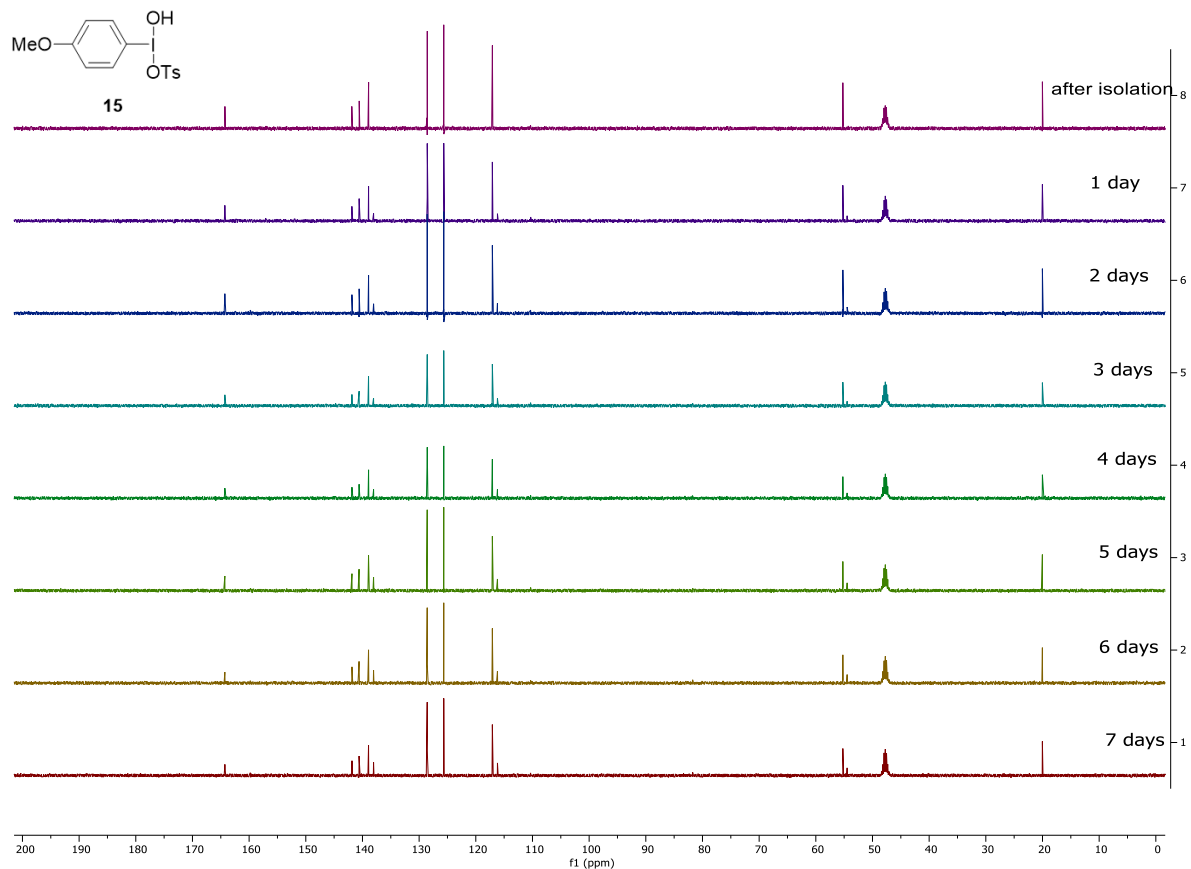


Figure S32: ¹³C NMR spectrum (CD₃OD, 101 MHz) of compound **15** directly after isolation (purple, top) and stored in CD₃OD at room temperature over several days (1 day – 7 days, top to bottom). In solution degradation is slowed down with respect to storage as isolated solid (Figure S26).

Fluorination reactions using synthesized diaryliodonium salts 2-OTs and 3-OTs

Method A (overnight fluorination of 3-OTs)

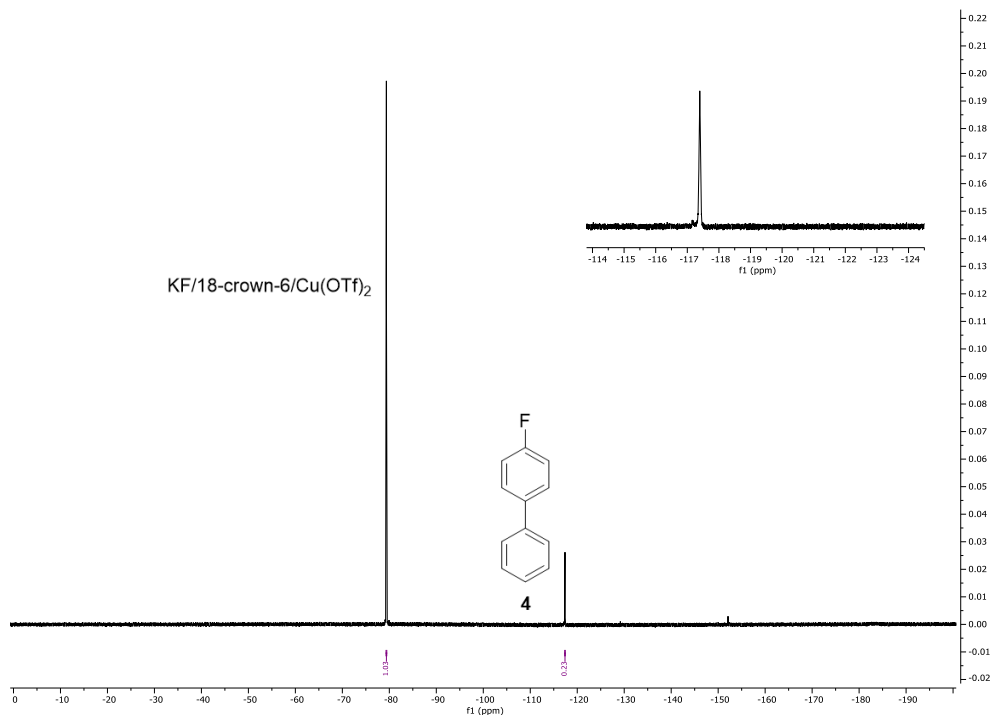


Figure S33: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of the crude overnight fluorination reaction using diaryliodonium salt **3-OTs** (Method A).

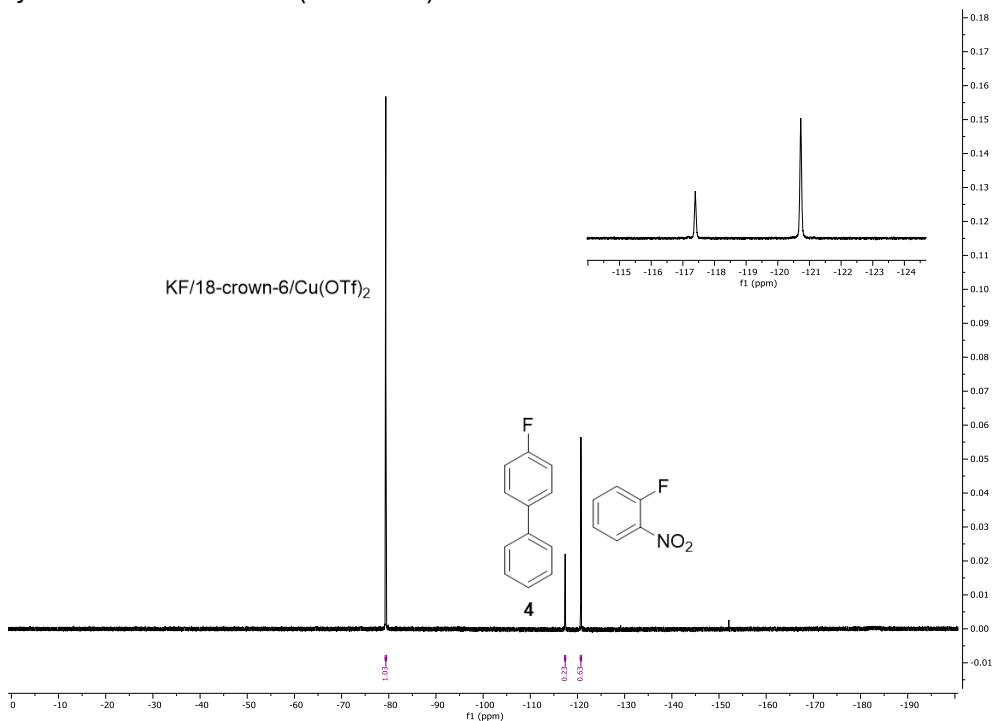


Figure S34: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of the crude overnight fluorination reaction using diaryliodonium salt **3-OTs** (Method A) with added 1-fluoro-2-nitrobenzene (2.6 μL , 25 μmol , 0.5 equiv.) as internal standard.

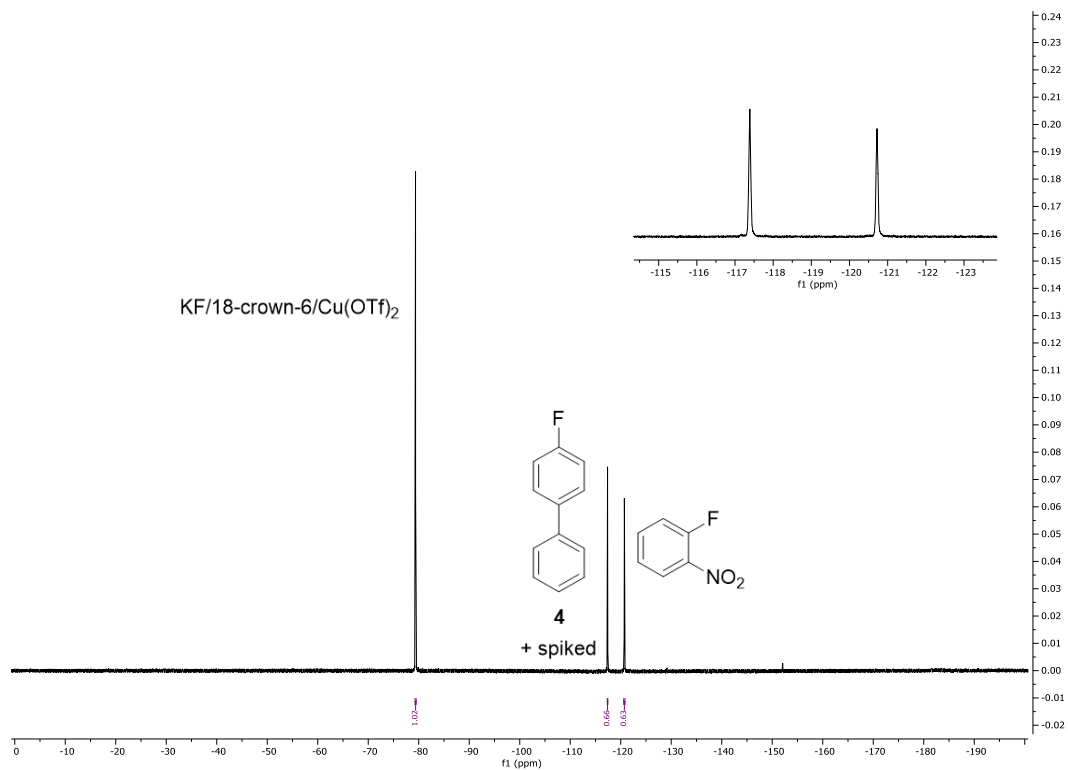


Figure S35: ¹⁹F NMR spectrum (DMF/CDCl₃ 4:1, 376 MHz) of the crude overnight fluorination reaction using diaryliodonium salt **3-OTs** (Method A) with added 1-fluoro-2-nitrobenzene (2.6 μL, 25 μmol, 0.5 equiv.) as internal standard and spiked with commercial 4-fluorobiphenyl (approx. 3.0 mg).

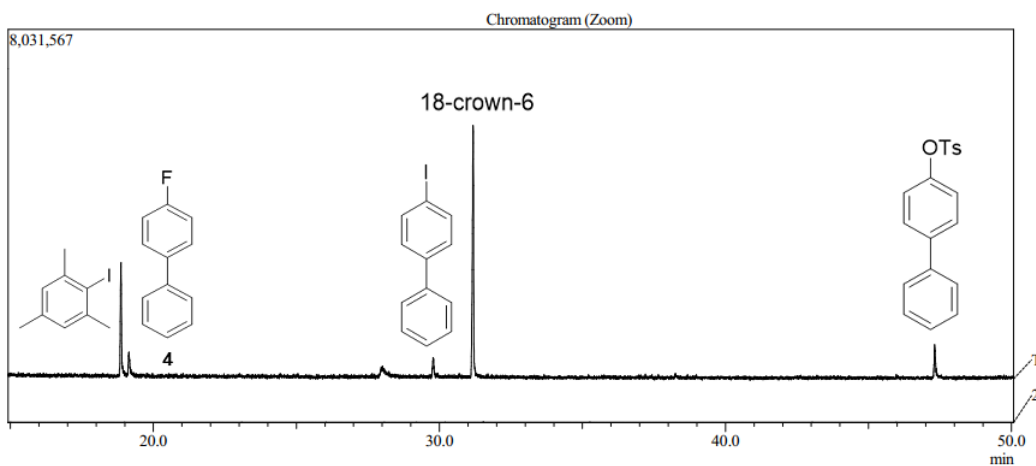
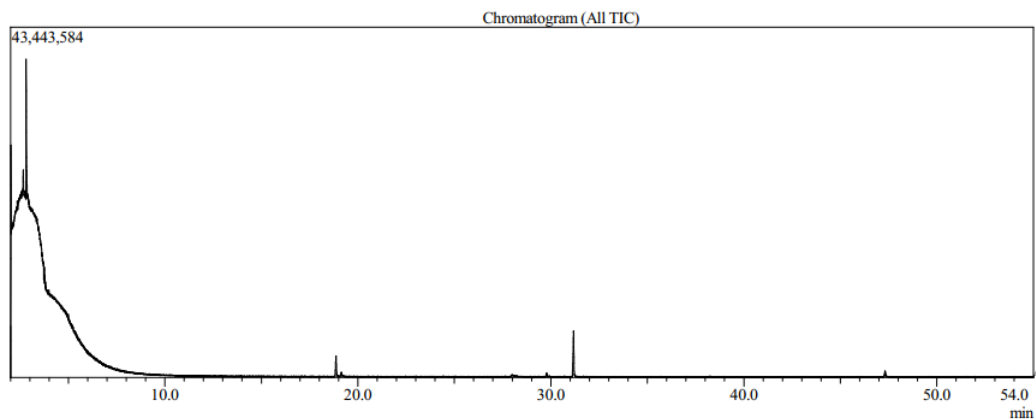


Figure S36: GC-MS chromatogram of the crude overnight fluorination reaction using diaryliodonium salt **3-OTs** (Method A).

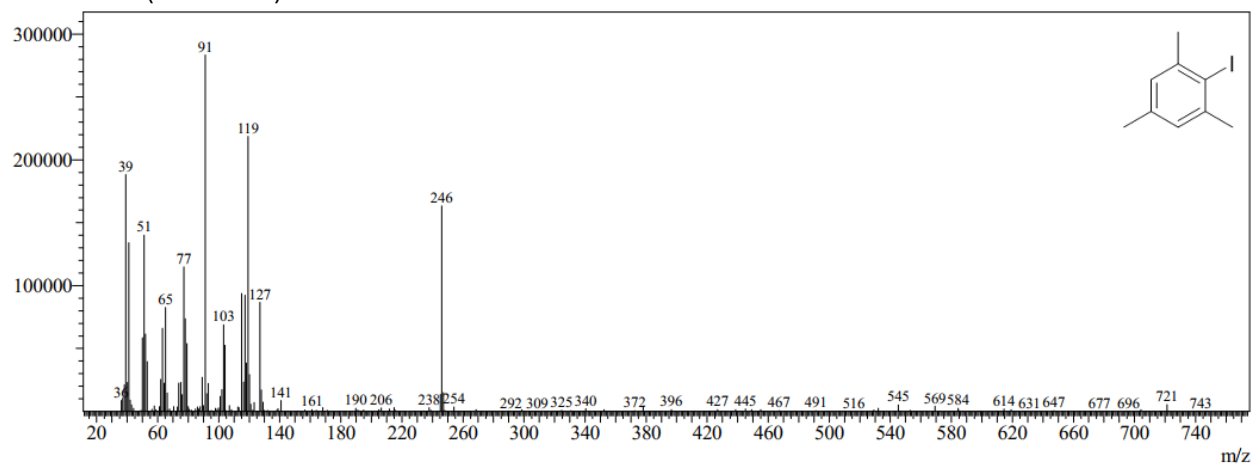
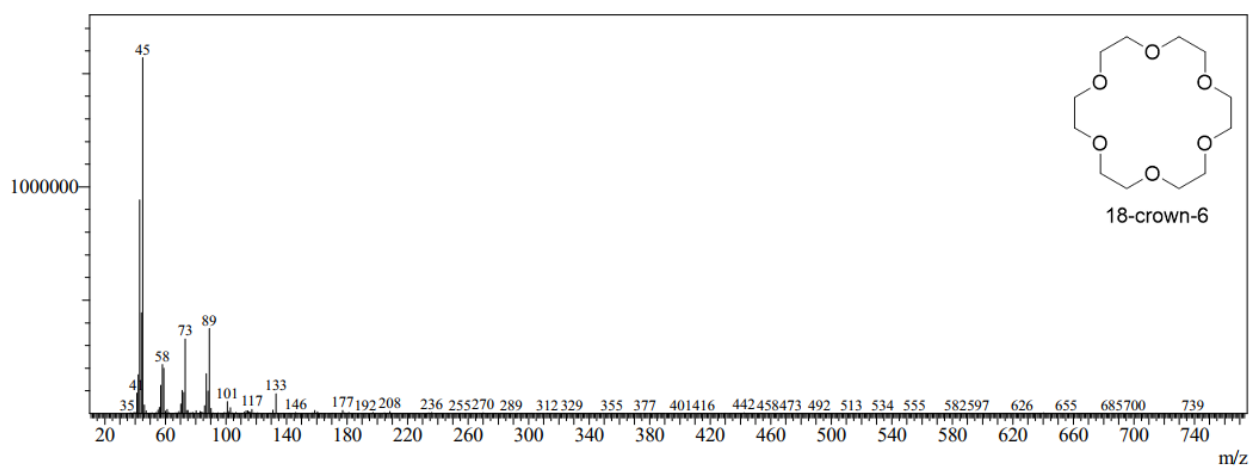
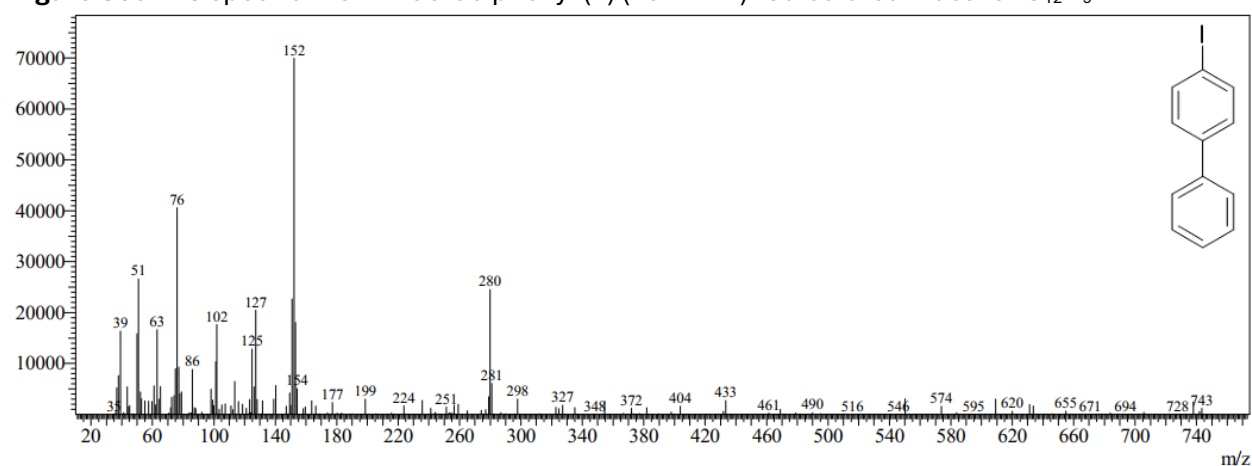
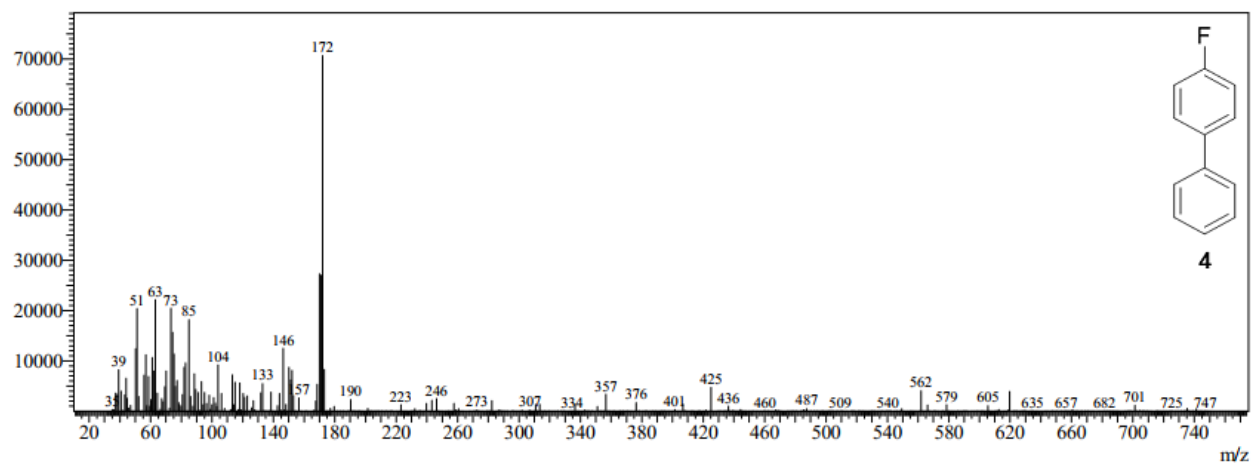


Figure S37: MS spectrum of 2-iodomesitylene (18.9 min). Calculated mass for $C_9H_{11}I^+$ = 246.



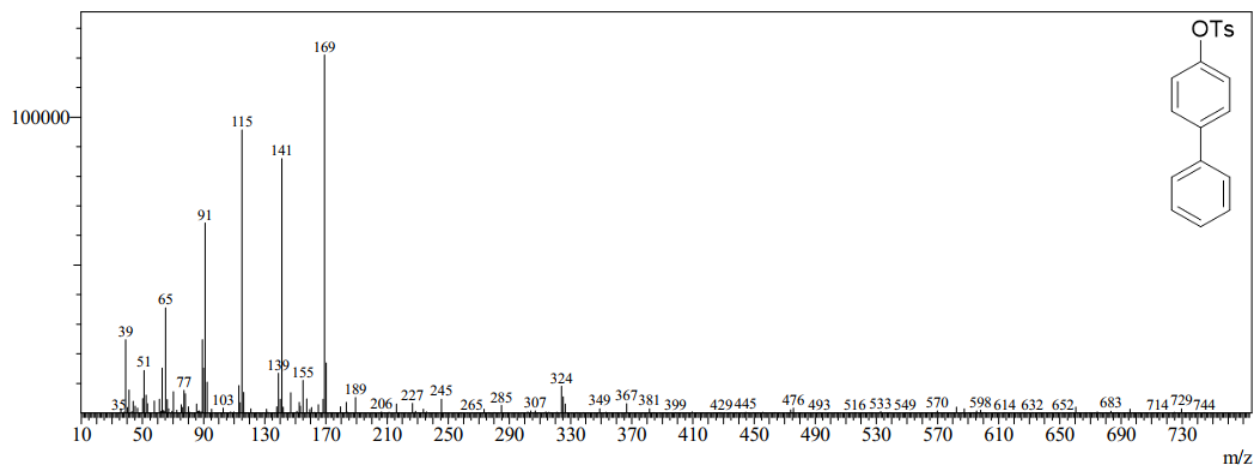


Figure S41: MS spectrum of biphenyl-4-yl tosylate (47.3 min). Calculated mass for $C_{19}H_{16}O_3S^+$ = 324.

Method B (20-minute fluorination of **3-OTs**)

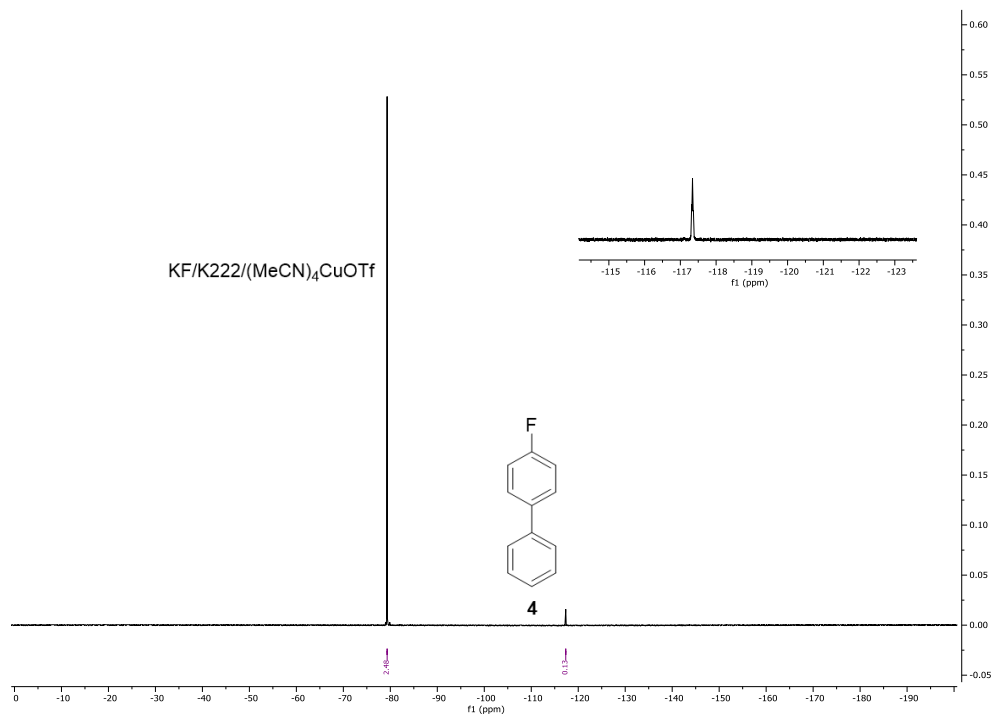


Figure S42: ¹⁹F NMR spectrum (DMF/CDCl₃ 4:1, 376 MHz) of the crude 20-minute fluorination reaction using diaryliodonium salt **3-OTs** (Method B).

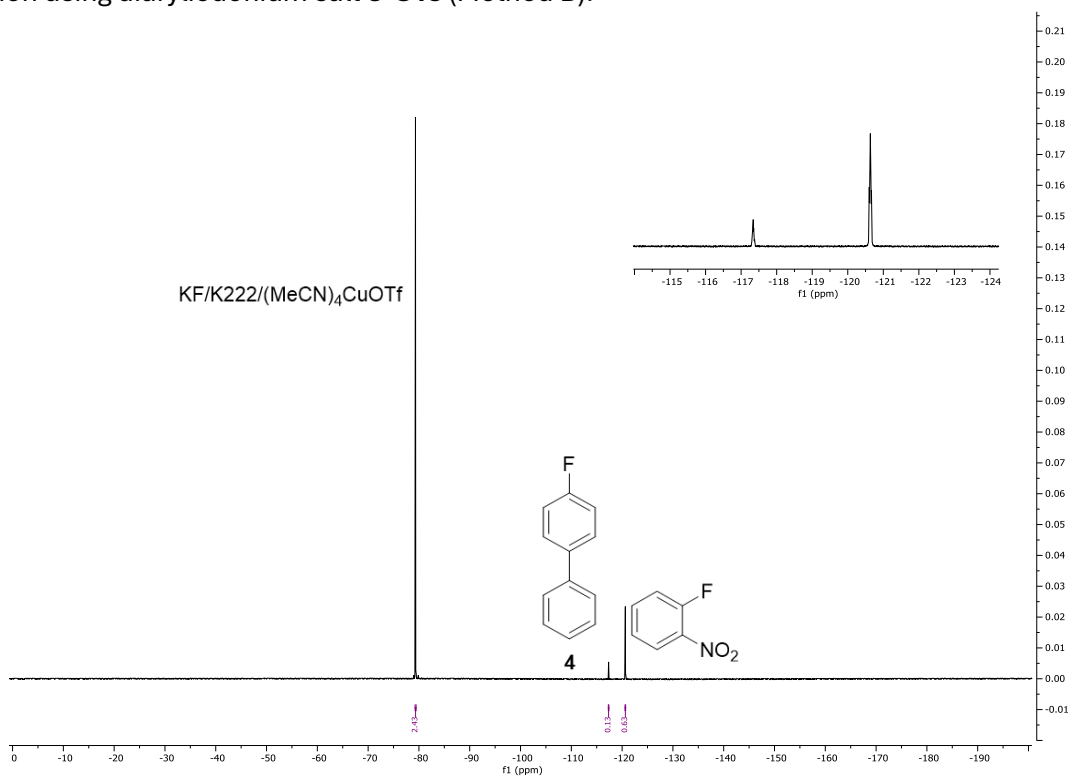


Figure S43: ¹⁹F NMR spectrum (DMF/CDCl₃ 4:1, 376 MHz) of the crude 20-minute fluorination reaction using diaryliodonium salt **3-OTs** (Method B) with added 1-fluoro-2-nitrobenzene (1.6 μ L, 15 μ mol, 0.5 equiv.) as internal standard.

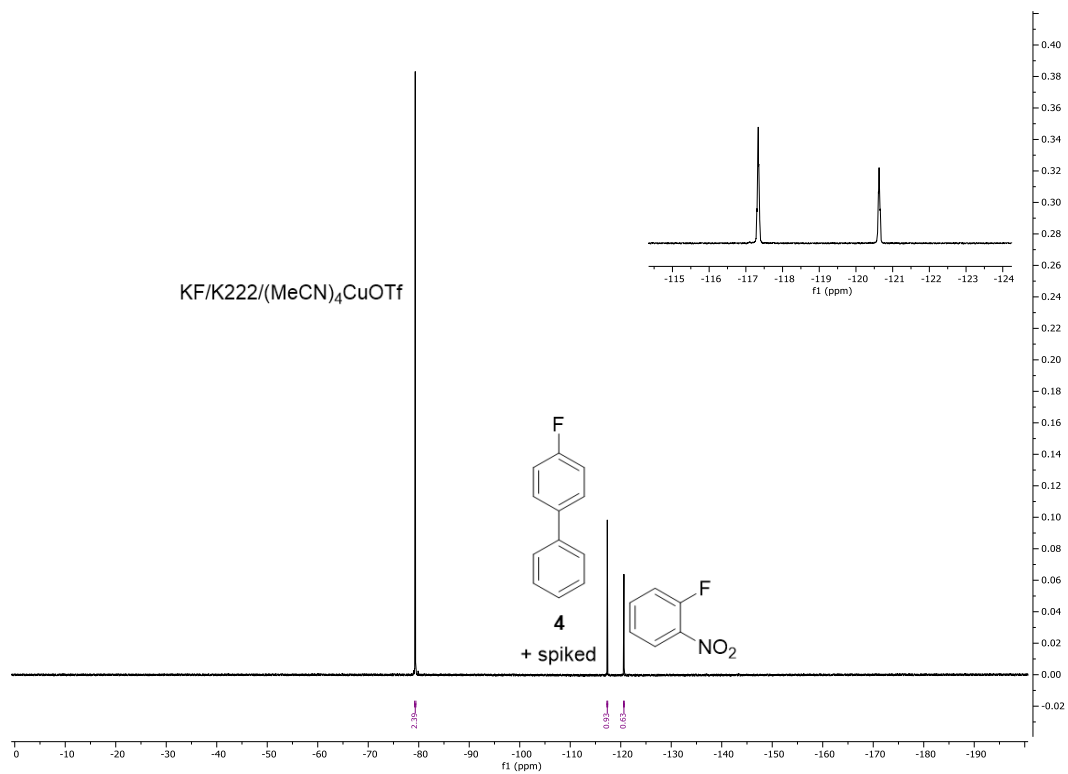


Figure S44: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of the crude 20-minute fluorination reaction using diaryliodonium salt **3-OTs** (Method B) with added 1-fluoro-2-nitrobenzene (1.6 μL , 15 μmol , 0.5 equiv.) as internal standard and spiked with commercial 4-fluorobiphenyl (approx. 3.0 mg).

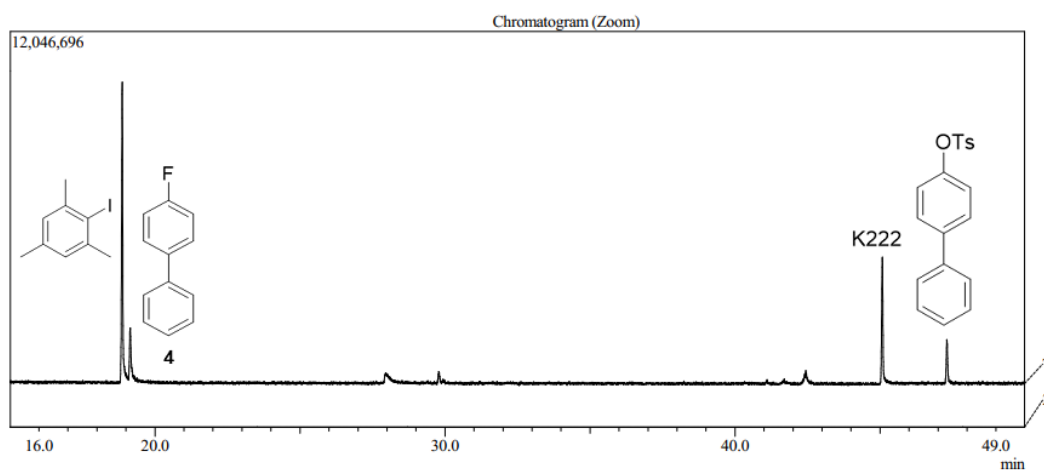
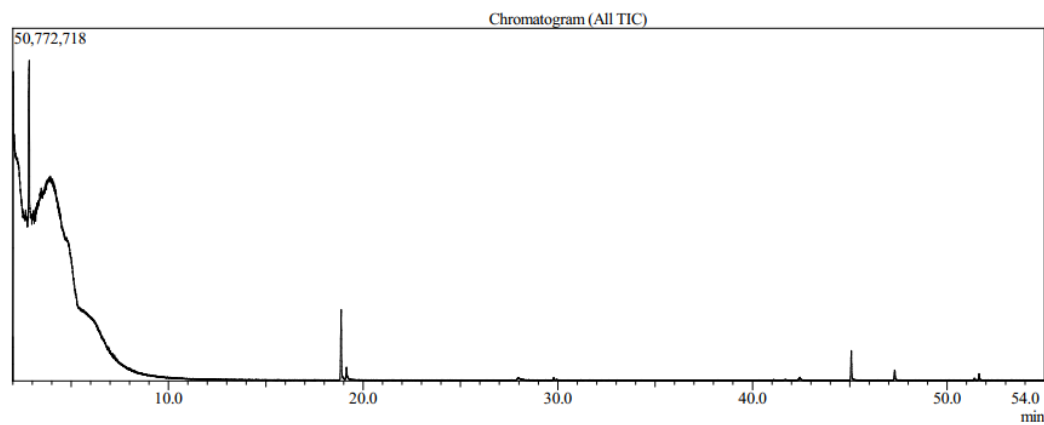


Figure S45: GC-MS chromatogram of the crude 20-minute fluorination reaction using diaryliodonium salt **3-OTs** (Method B).

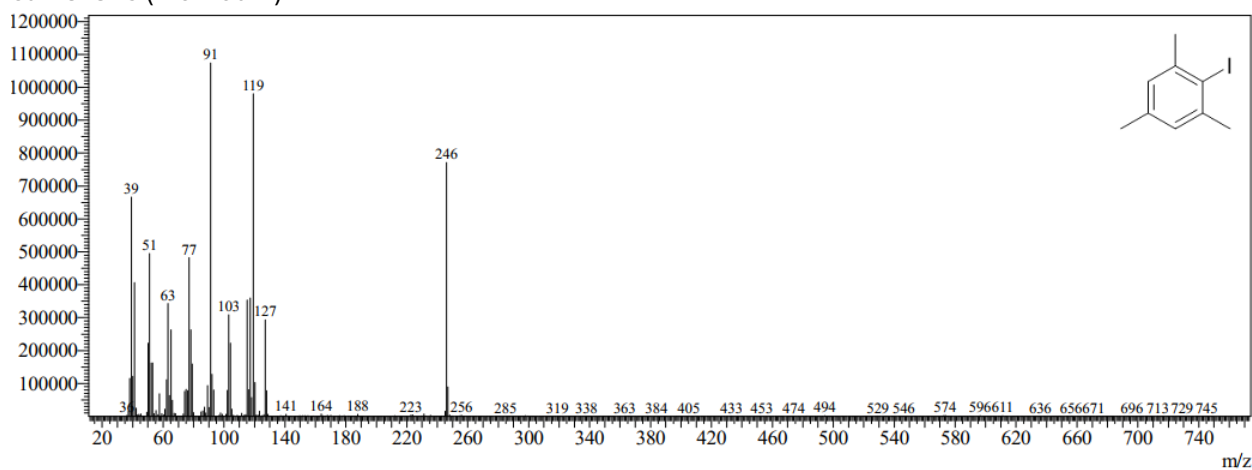


Figure S46: MS spectrum of 2-iodomesitylene (18.9 min). Calculated mass for $C_9H_{11}^+ = 246$.

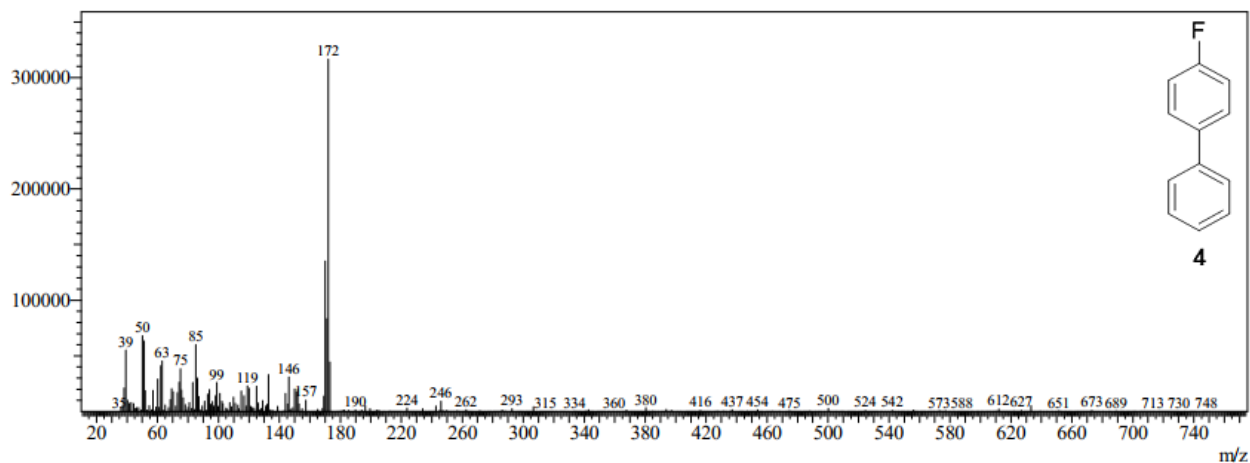


Figure S47: MS spectrum of 4-fluorobiphenyl (**4**) (19.1 min). Calculated mass for $C_{12}H_9F^+$ = 172.

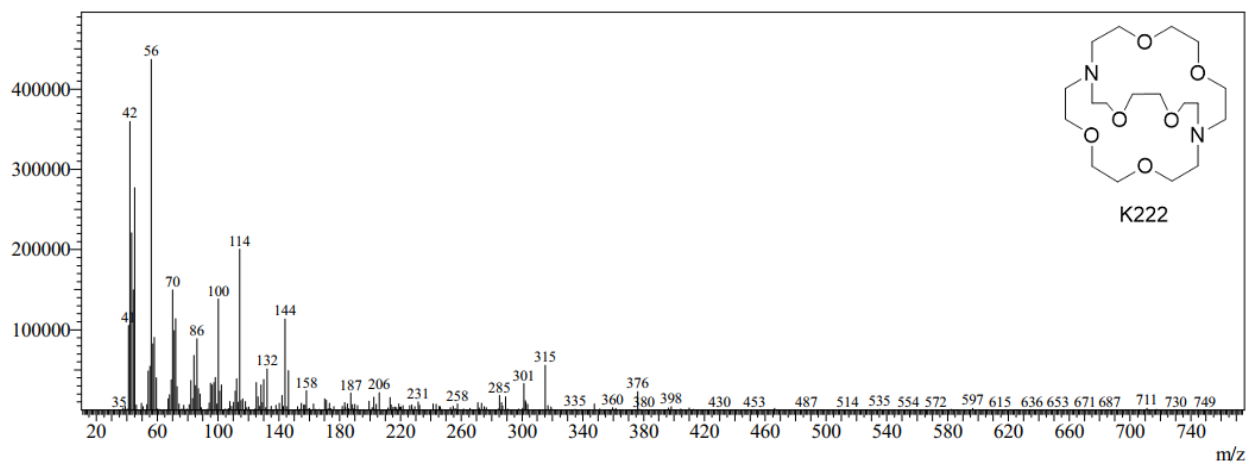


Figure S48: MS spectrum of K222 (45.1 min). Calculated mass for $C_{18}H_{36}N_2O_6^+$ = 376.

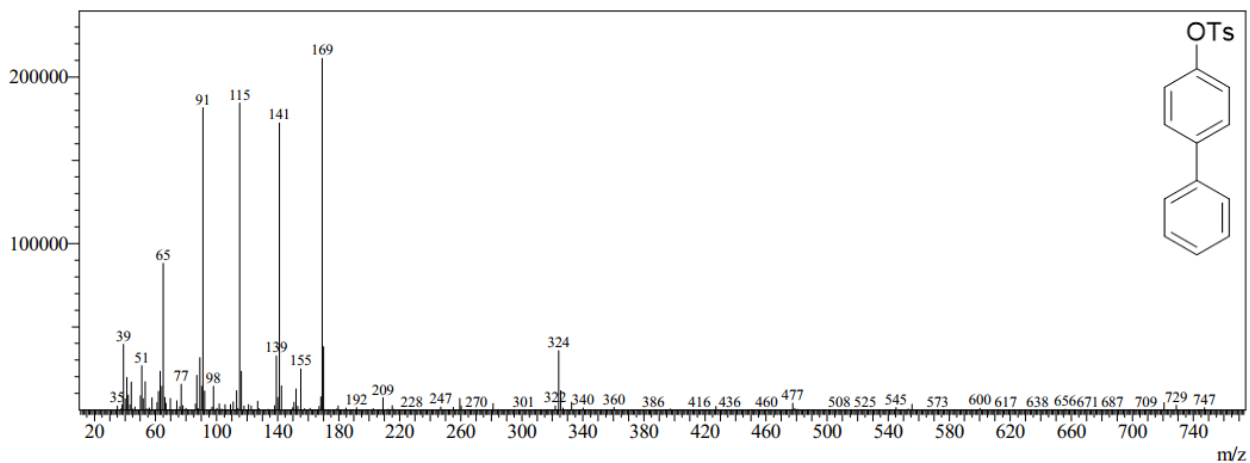


Figure S49: MS spectrum of biphenyl-4-yl tosylate (47.3 min). Calculated mass for $C_{19}H_{16}O_3S^+$ = 324.

Method C (20-minute fluorination of **2-OTs**)

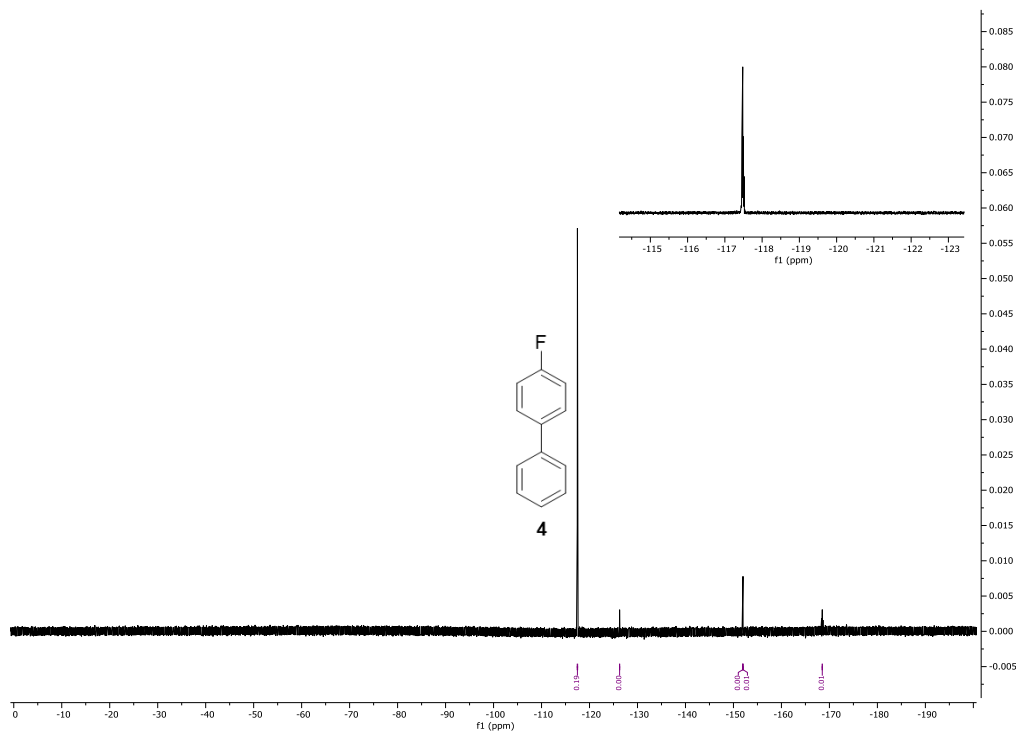


Figure S50: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of the crude 20-minute fluorination reaction using diaryliodonium salt **2-OTs** (Method C).

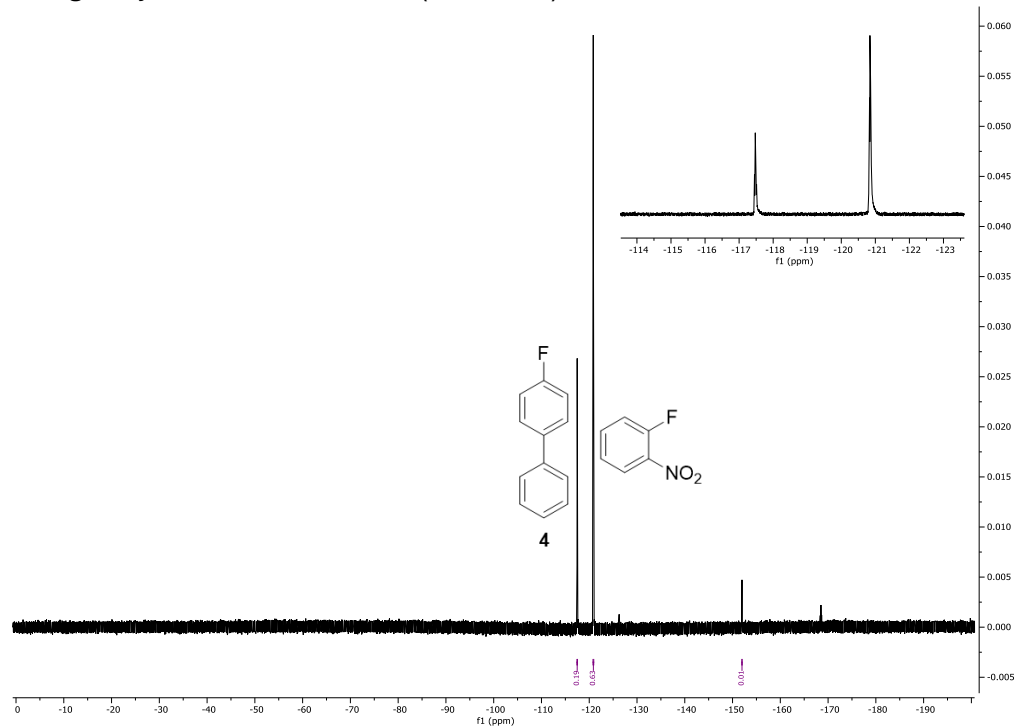


Figure S51: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of the crude 20-minute fluorination reaction using diaryliodonium salt **2-OTs** (Method C) with added 1-fluoro-2-nitrobenzene (1.6 μL , 15 μmol , 0.5 equiv.) as internal standard.

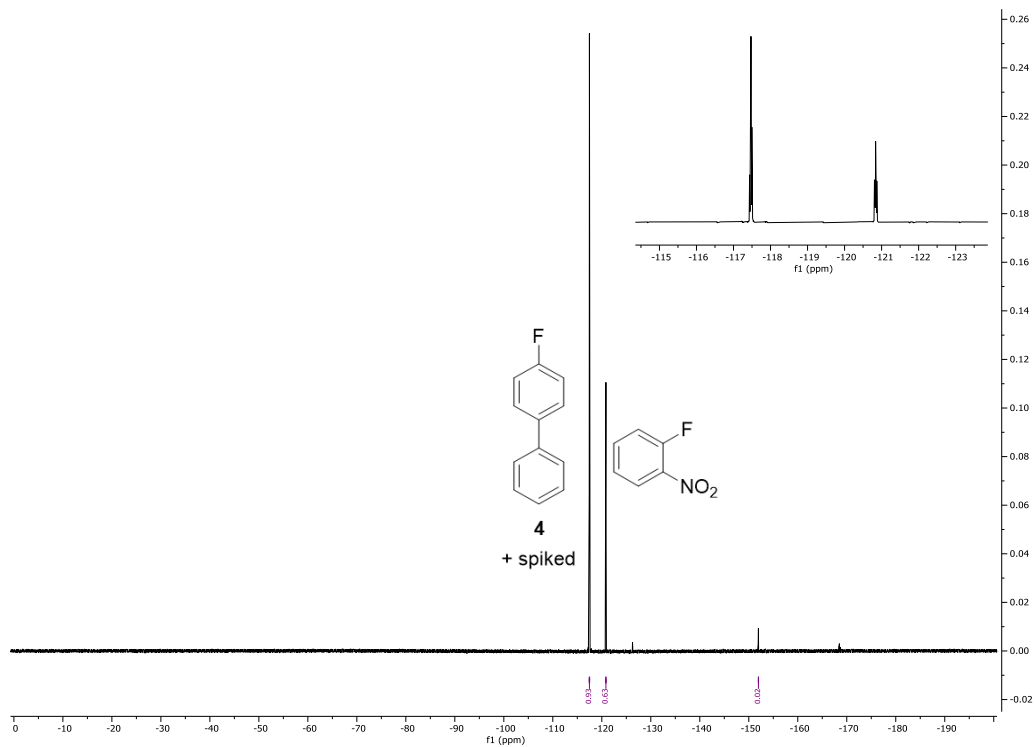


Figure S52: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of the crude 20-minute fluorination reaction using diaryliodonium salt **2-OTs** (Method C) with added 1-fluoro-2-nitrobenzene (1.6 μL , 15 μmol , 0.5 equiv.) as internal standard and spiked with commercial 4-fluorobiphenyl (approx. 3.0 mg).

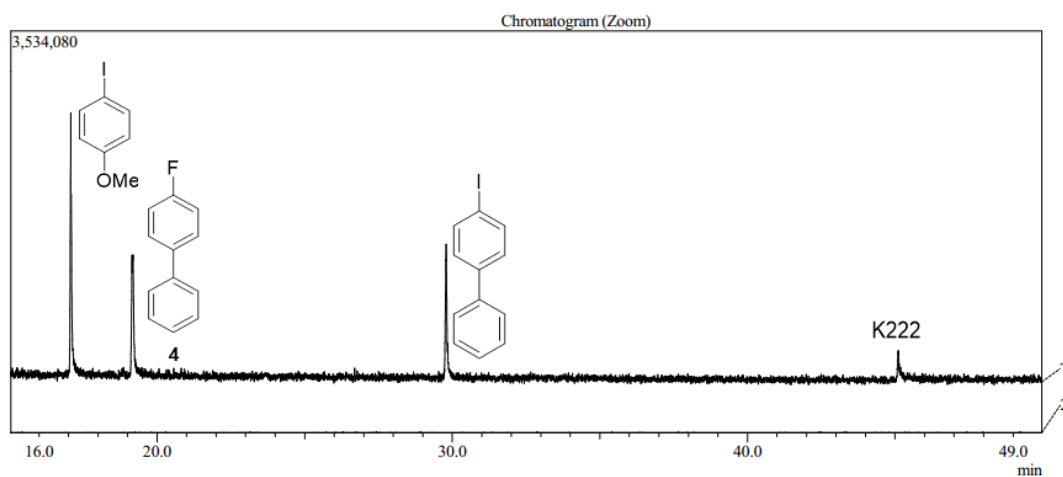
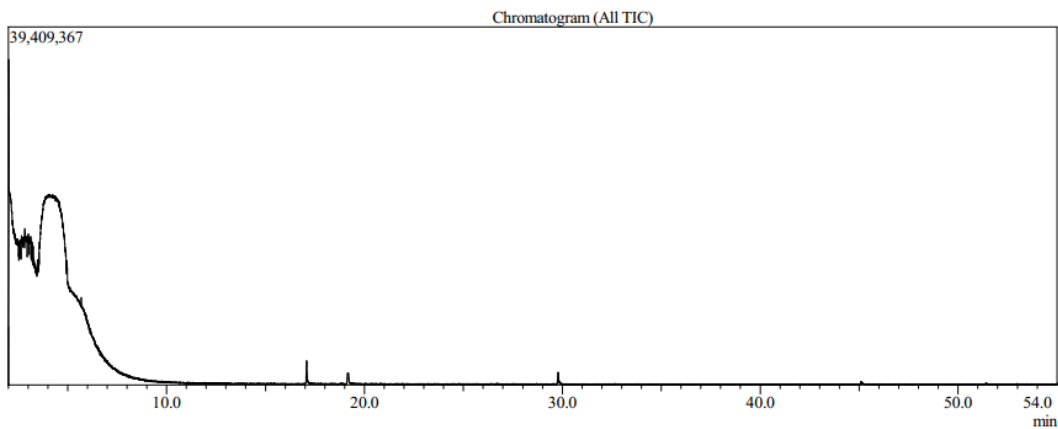


Figure S53 GC-MS chromatogram of the crude 20-minute fluorination reaction using diaryliodonium salt **2-OTs** (Method C).

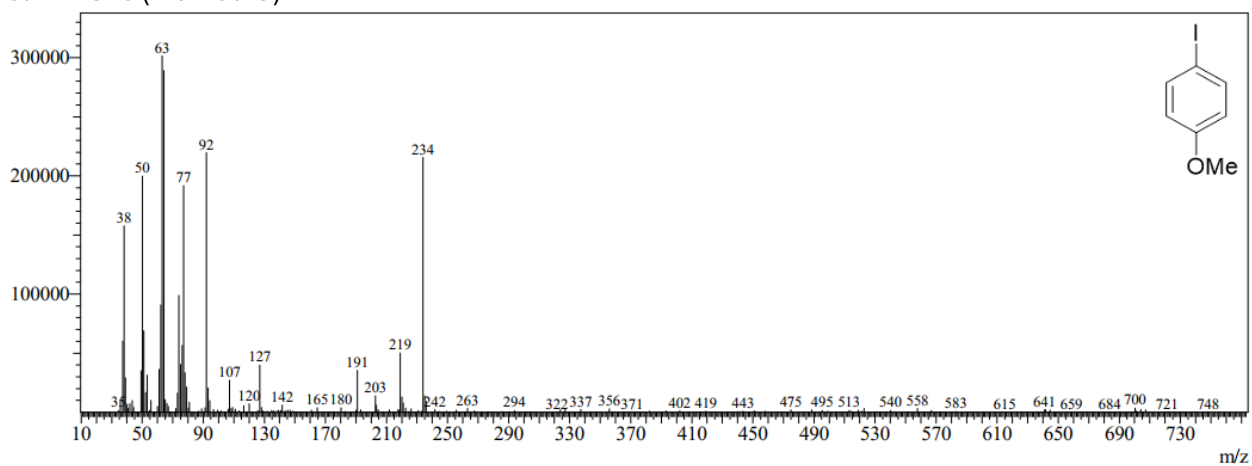
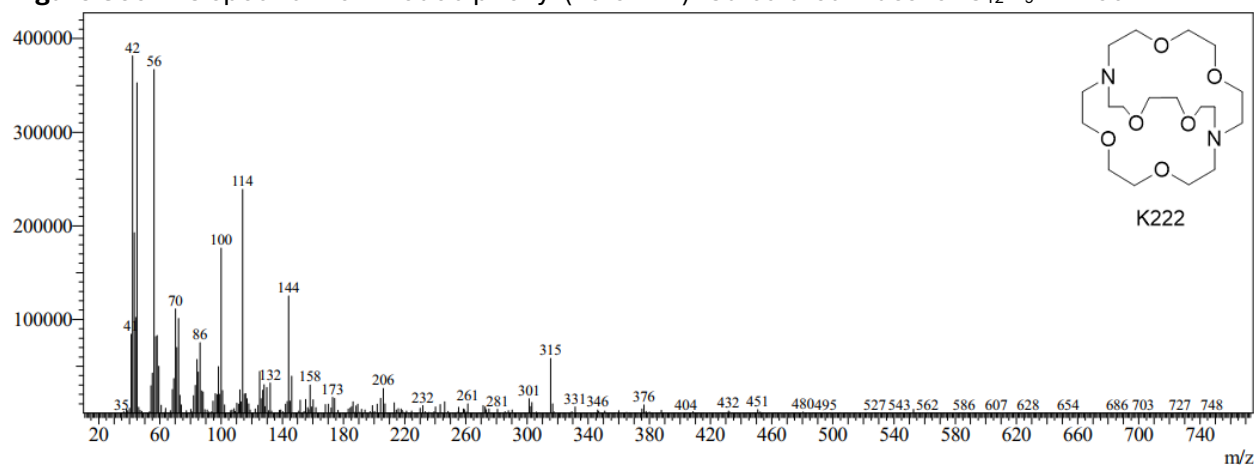
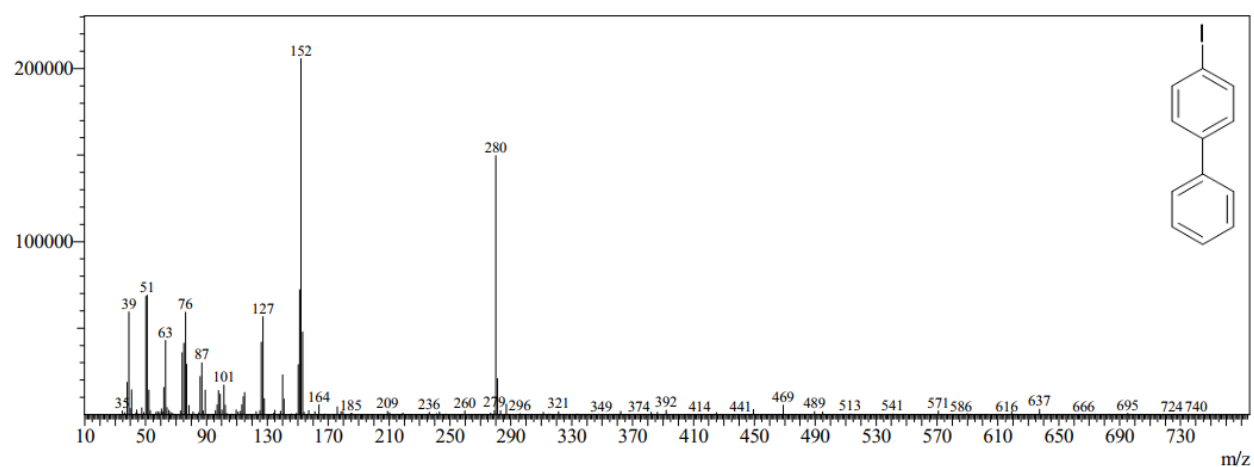
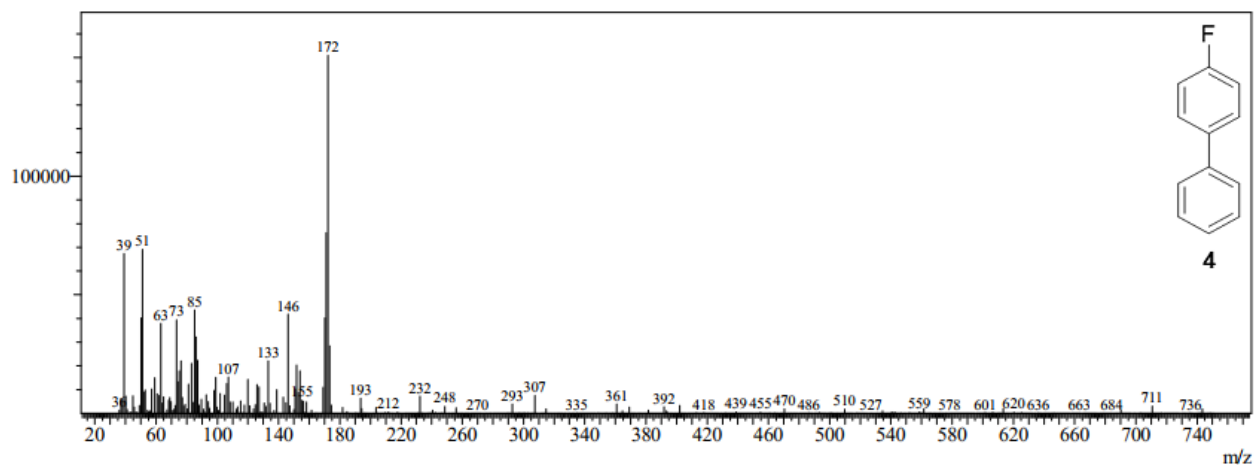


Figure S54: MS spectrum of 4-iodoanisole (17.1 min). Calculated mass for $C_7H_7IO^+$ = 234.



Reference spectra for fluorination reactions

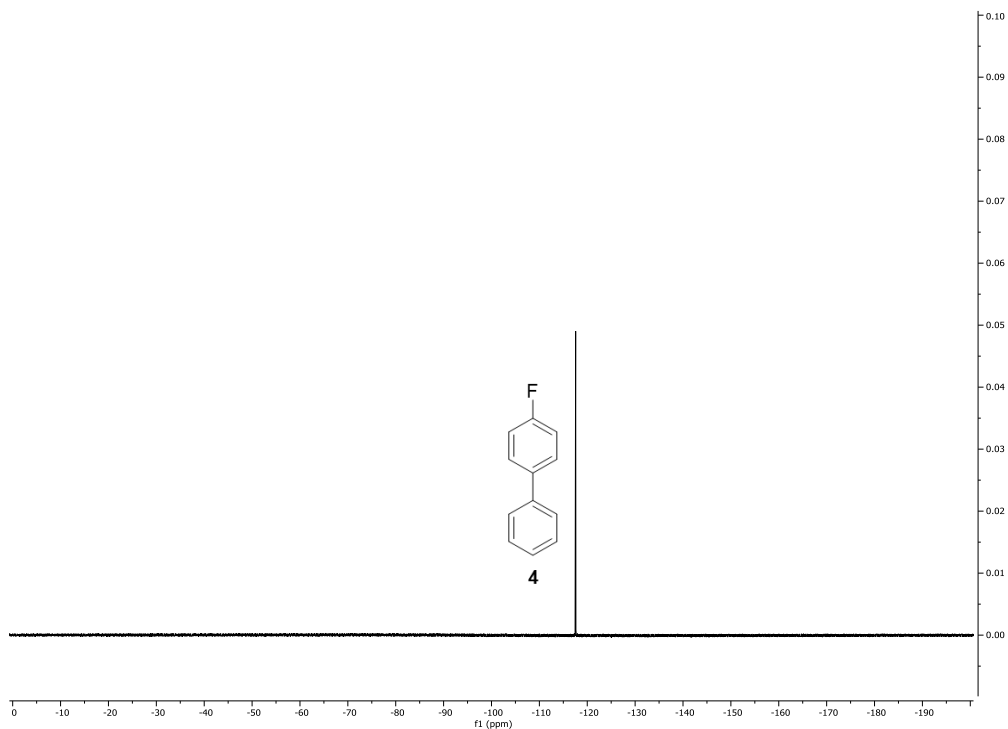


Figure S58: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of commercial 4-fluorobiphenyl (**4**).

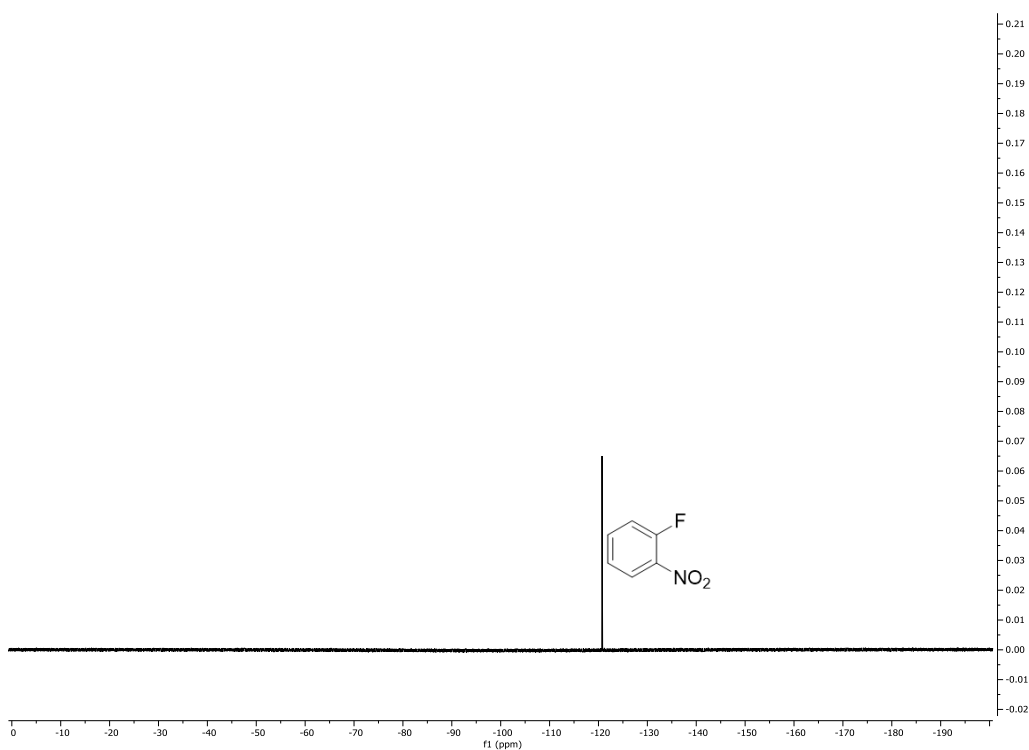


Figure S59: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of 1-fluoro-2-nitrobenzene.

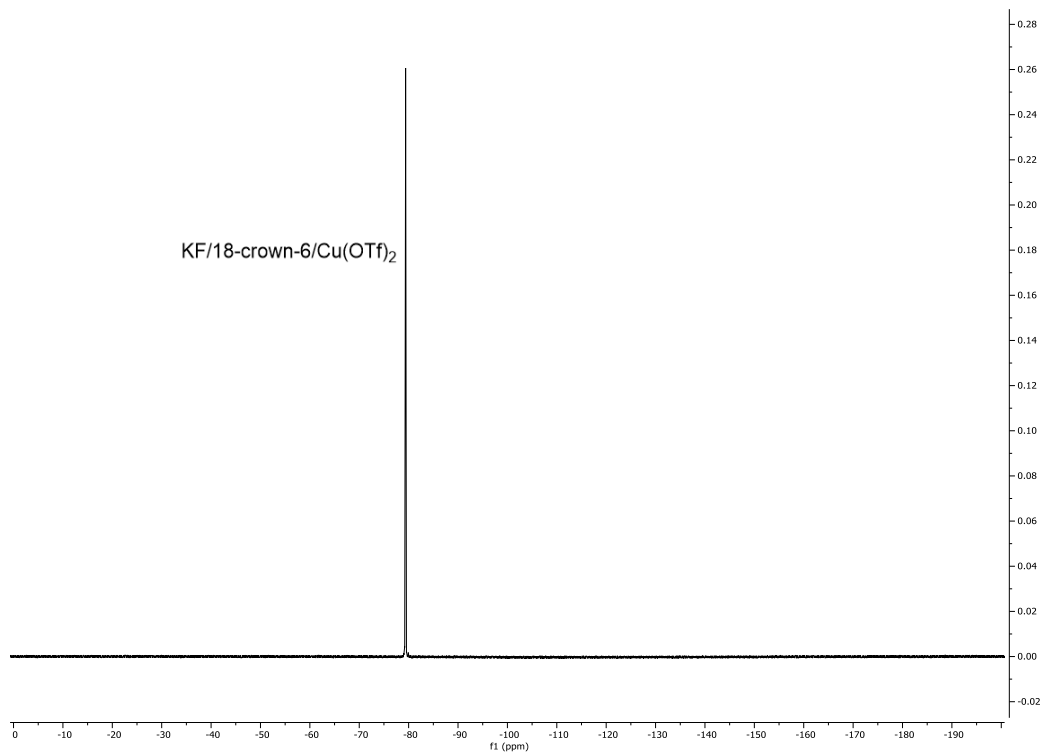


Figure S60: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of a mixture of KF (3.2 mg, 55 μmol), 18-crown-6 (5.3 mg, 20 μmol), and $\text{Cu}(\text{OTf})_2$ (3.6 mg, 10 μmol).

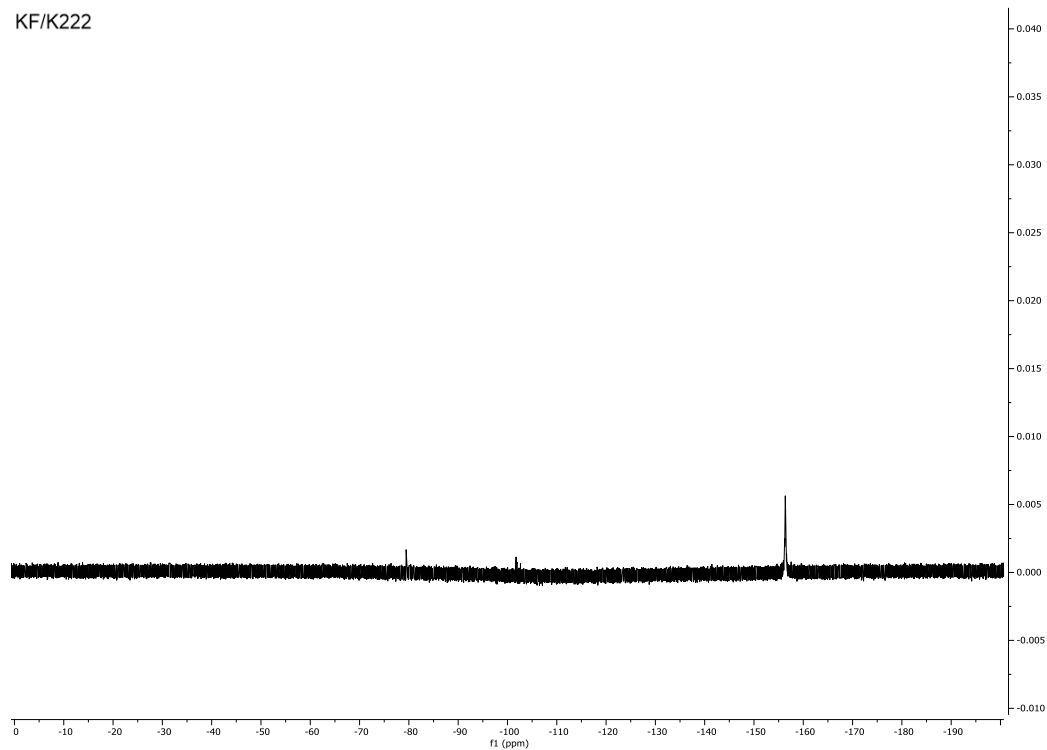


Figure S61: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of a mixture of KF (1.9 mg, 33 μmol) and K222 (11 mg, 30 μmol).

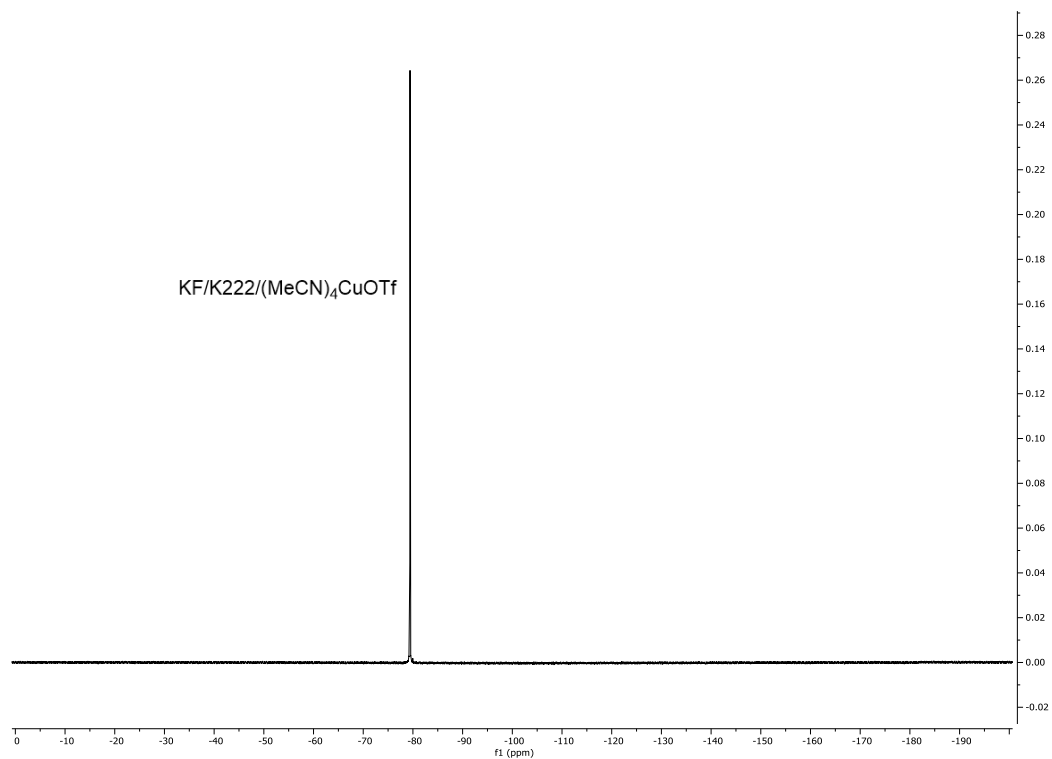


Figure S62: ^{19}F NMR spectrum (DMF/ CDCl_3 4:1, 376 MHz) of a mixture of KF (1.9 mg, 33 μmol), K222 (11 mg, 30 μmol), and $(\text{MeCN})_4\text{CuOTf}$ (11 mg, 30 μmol).

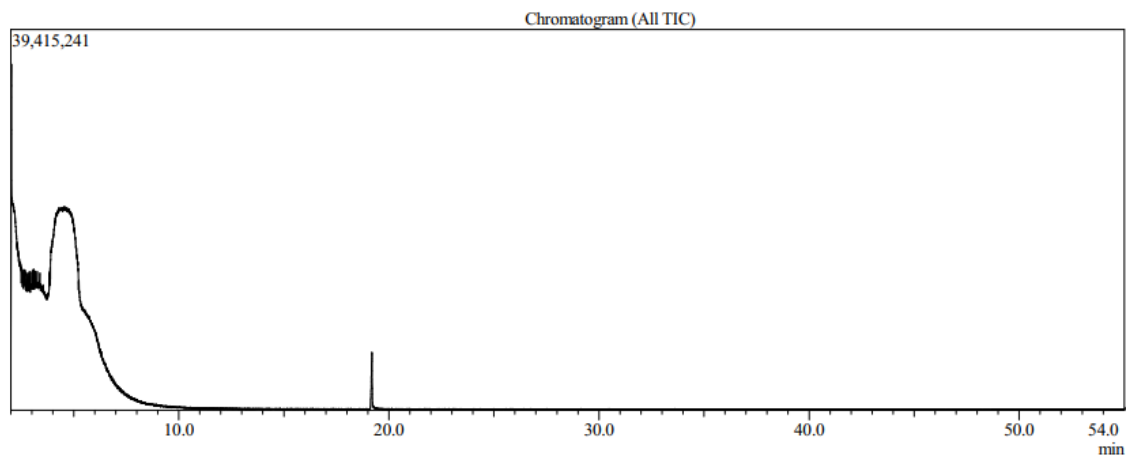


Figure S63: GC-MS chromatogram of commercial 4-fluorobiphenyl (**4**).

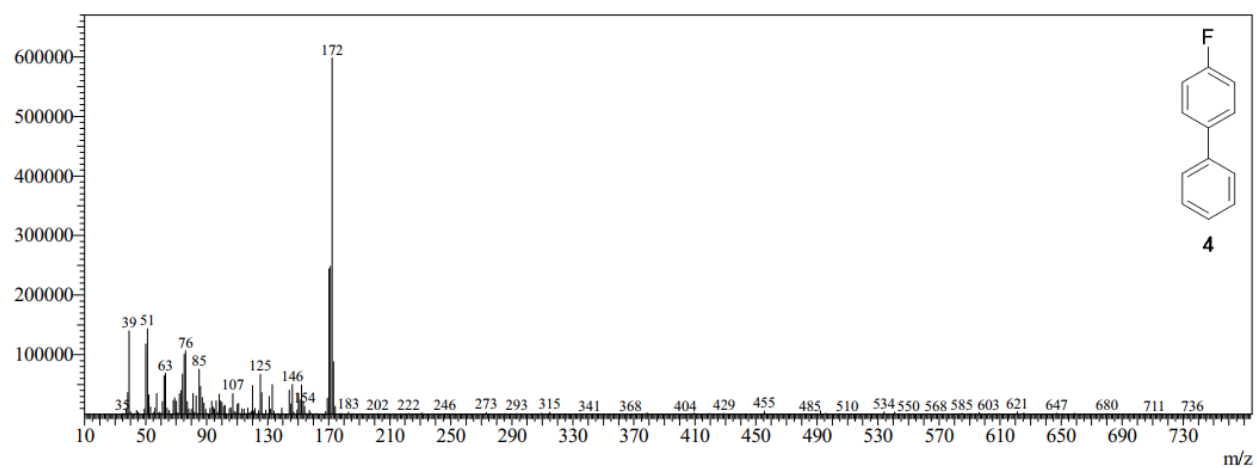


Figure S64: MS spectrum of commercial 4-fluorobiphenyl (**4**) (19.2 min). Calculated mass for $C_{12}H_9F^+$ = 172.